prises dissolving a color former fast to diffusion, in water containing a water-soluble alkali, in the presence of a compound selected from the class consisting of bile acids and salts thereof, mixing the solution with a silver halide emulsion and 5 vasting the resulting mixture.

11. The process as defined in claim 10, wherein the water-soluble alkali is caustic soda.

12. The process as defined in claim 9, wherein the color former is fast to diffusion by virtue of the presence therein of an aliphatic chain of more than 5 carbon atoms.

13. The composition as defined in claim 4, wherein said compound is a salt of cholic acid.

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PHOTOGRAPHIC EMULSION

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13 Claims. (Cl. 95-6)

Our present invention relates to a photographic aulsion and more particularly to a silver halide

iulsion for color photography.

In the production of color photographic images image dyestuff is often directly or indirectly med by means of a silver image from dyestuff imponents one or several of which have already n incorporated into the silver halide emulsion fore exposure. In multi-layer materials it is asary in order to obtain an unobjectionable 10 or separation that the dyestuff components fast to diffusion with respect to the layerming colloid, for instance, gelatin, i. e. that or do not diffuse from layer to layer or from layers into the photographic baths during production of the layers and the treatment reof in the baths even if they are added to emulsion in a soluble form. This fastness diffusion of the dyestuff formers may, for innce, be effected by their special chemical conbution or by the introduction into the molecule certain radicles. Dyestuff formers made fast diffusion by substituting the same with such pups which essentially increase the molecular ight thereof often are sparingly water-soluble. ch sparingly soluble dyestuff formers have the advantage that it is difficult to keep them in dissolved condition when added to the phoraphic emulsion and when drying them toher with the gelatin to give a clear transent layer. It has, therefore, already been posed to disperse these water-insoluble or gringly water-soluble dyestuff formers with table wetting or dispersing agents such as sulized high molecular weight hydroxy fatty acids 35 lalcohols, high molecular weight aliphatic irines or resin soaps. The effect of these addiagents preferably is a physical one for they duce a strong decrease of the surface tension ween the substance to be dispersed and the ter as the solvent so that the substance is re intensely wetted whereby its emulsification attained and its recrystallization or reseparai from the emulsion is prevented or at least de difficult. It is evident that the difficultly ible dyestuff formers which are to be emulsiin this manner have a great tendency to irate from the solution so that it is necessary add a considerable amount of dispersing agents the photographic emulsion in order to render 50 desible a separation and agglomeration of dyestuff components. The photographic r, however, is very sensitive to foreign subices. Moreover traces of crystals and flakes the gelatin layer even if so small that they 55 335,144 filed May 14, 1940, Ser. No. 341,180 filed

can be scarcely distinguished with the naked eye cause much trouble in projection. It is, therefore, desirable to dissolve the water-insoluble dyestuff formers directly in the photographic emulsion.

Our present invention is based on the observation that water-insoluble or sparingly water-soluble dyestuff components can be dissolved in aqueous liquids without influencing their fastness to diffusion with respect to the binding agent (gelatin) of the emulsion by dissolving the components in water in the presence of bile acids or salts thereof, if necessary with an addition of organic solvents. It is sometimes helpful to warm the substances while dissolving them. The solution obtained can be added to a photographic emulsion or a gelatin solution without effecting a precipitation or a turbidity of these aqueous liquids. On drying the gelatin solution a layer clear as glass is obtained. By this invention it is accordingly possible to keep in solution dyestuff formers which have heretofore been thought useless since they could not be added in a dissolved condition to an emulsion due to their low solubility or their tendency to crystallize on drying the emulsion to give a layer. It is furthermore possible to form layers clear as glass with such dyestuff formers.

The dyestuff formers are conveniently dissolved together with the bile acid or a watersoluble salt thereof, if desired, with an addition of organic solvents or alkalies whereupon the solution thus obtained is mixed with the emulsion at any time prior to casting. For removing undesired substances the emulsion may be subjected to a washing with water after the solution of the dyestuff former is added. The solution of the dyestuff former may also be mixed with a gelatin solution whereupon, if necessary, after washing with water, the resulting solution is worked up into a silver halide emulsion.

The dyestuff formers used in the practice of this invention are, for instance, of the general types described in the following patents and applications:

U. S. Patents 2,186,852, 2,186,719, 2,186,734, 2,186,851, 2,186,733, 2,186,732, 2,186,847 dated January 9, 1940, U. S. Patents 2,179,238, 2,179,244, 2,178,612 dated November 7, 1939, French Patents 344,648, 344,649, 844,637, U. S. Patents 2,280,722, 2,292,575, 2,303,928 and 2,307,399 and U. S. Patent applications Ser. No. 284,258 filed July 13, 1939, Ser. No. 333,814 filed May 7, 1940, Ser. No.