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(54) Title: INJECTION GROUTING

(57) Abstract: The invention relates to the use of a composition for injection grouting obtainable by mixing an alkali metal silicate or an organic silicate, colloidal silica particles, and at least one gelling agent, wherein the weight ratio of colloidal silica to silicate is from about 2:1 to about 100:1. The invention also relates to a method of sealing a leaking part or cavity, and method of cutting off a liquid flow in a leaking part or cavity. The invention further relates to a composition for injection grouting and a method for preparing such composition.

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Injection grouting

The present invention relates to the use of a composition for injection grouting, a method of sealing by injecting a grouting composition, a method of cutting off a liquid flow, a grouting composition and a method of providing such composition. The grouting
5 composition comprises colloidal silica particles, an alkali metal silicate or an organic silicate, and at least one gelling agent.

Background of the invention

The requirements and performance of compositions used for sealing rock and
10 soil have increased over time. The requirements involve both improved environmental and technical aspects. Previously, various plastics and polymers involving toxic substances have been employed to seal water leaks in e.g. concrete walls, tunnels, dumps etc. Such sealing chemicals have in a number of cases caused contamination of groundwater and health problems. However, attempts have been made to replace
15 hazardous products with more environmentally adapted ones. New products have also been sought for to meet the recently imposed leakage restrictions. Water leakage levels as low as 1 litre/(min*100 meters) have occasionally been the upper threshold on constructions sites of tunnels. US 5,396,749 describes a method for cutting off a water flow by grouting whereby water leakage and collapse of ground is prevented. The cut-off
20 agent is prepared by mixing e.g. colloidal silica, inorganic salt, and a water-soluble urethane polymer. However, the strength of these agents has shown to be insufficient in several applications, particularly where the agent during injection and gelling is exposed to a high water pressure. Especially unstable agents have during the ageing phase resulted in a fairly low long term strength.

25 Sealing agents have also been used for sealing soil to prevent leakage of contaminants in subterranean areas like buried sources, dumps etc. US 5,836,390 discloses a method of forming subsurface barriers where a viscous liquid comprising polybutenes, polysiloxanes, and colloidal silica is injected. However, such barriers are not always sufficiently strong to prevent aqueous and organic liquids from penetrating.
30 Further to this, stability problems or pre-gelling of the sealing composition have many times rendered the sealing procedure inefficient.

It would be desirable to provide a stable and environmentally adapted liquid cut-off agent having a high early strength as well as long term strength, especially in the field of sealing rock, where the cut-off agents may be subjected to high water pressure from
35 groundwater. It would also be also desirable to provide durable products which resist washout forces and have a low permeability or even impermeable to water and other liquids.

It is therefore an object of the present invention to provide products solving the drawbacks of the prior art.

The invention

5 The present invention relates to the use of a composition for injection grouting obtainable by mixing an alkali metal silicate or an organic silicate, colloidal silica particles, and at least one gelling agent, wherein the weight ratio of colloidal silica to silicate is from about 2:1 to about 100:1.

10 The invention also relates to a method of sealing a leaking part or cavity comprising injecting into the leaking part or cavity a composition obtainable by mixing an alkali metal silicate or organic silicate, colloidal silica particles, and at least one gelling agent, wherein the weight ratio of colloidal silica to silicate is from about 2:1 to about 100:1.

15 The invention also relates to a method of cutting off a liquid flow in a leaking part or cavity by injecting into the leaking part or cavity a composition obtainable by mixing an alkali metal silicate or an organic silicate, colloidal silica particles, and at least one gelling agent, wherein the weight ratio of colloidal silica to silicate is from about 2:1 to about 100:1.

20 The invention also relates to a method of preparing a composition for injection grouting and a grouting composition obtainable from said method. The invention also relates to a composition for injection grouting comprising colloidal silica particles, an alkali metal silicate or an organic silicate, and at least one gelling agent, wherein the weight ratio of colloidal silica to alkali metal silicate is from about 2:1 to about 100:1.

25 Generally, in all inventions defined, the weight ratio of colloidal silica to silicate preferably is from about 3:1 to about 70:1, more preferably from about 6:1 to about 50:1, even more preferably from about 6:1 to about 20:1, and most preferably from about 6:1 to about 10:1. However, according to one embodiment, the weight ratio of colloidal silica to silicate is from about 8:1 to about 100:1, preferably from about 8:1 to about 20:1.

30 According to one embodiment, the composition is dispersed in an aqueous solution, most preferably an aqueous silica sol. According to one embodiment, except for the possible presence of organic silicate and/or organic gelling agent as defined herein, no further organic compounds are present in the composition.

35 The term "injection grouting" as used in this context particularly means sealing by injecting a grouting composition, particularly sealing of cavities or leaking parts such as sealing fissures, cracks, particularly in rock and soil as well as in constructions such as roads, tunnels, bridges, buildings, e.g. floors and walls of parking garages; concrete pipes; well cementing, as well as other subterranean constructions such as marine

constructions e.g. quays, piers, and jetties. "Injection grouting" as used herein also comprises injection in cementitious grouting. The most preferred application is injection in rock and/or soil.

Colloidal silica particles and silica sols can be derived from e.g. precipitated
5 silica, micro silica (silica fume), pyrogenic (fumed silica), silanes, siloxanes, or silica gels with sufficient purity.

The terms "colloidal silica particles" and "silica sol" used herein also comprise e.g. aluminium-modified and boron-modified silica particles and sols. Boron-modified silica sols are further described in e.g. US 2,630,410. The aluminium modified silica
10 particles suitably have an Al_2O_3 content of from about 0.05 to about 3 wt%, preferably from about 0.1 to about 2 wt%. The procedure of preparing an aluminium modified silica sol is further described e.g. in "The Chemistry of Silica", by Iler, K. Ralph, pages 407-409, John Wiley & Sons (1979) and in US 5 368 833.

The colloidal silica particles suitably have an average particle diameter ranging
15 from about 3 to about 150 nm, preferably from about 7 to about 50 nm, and most preferably from about 10 to about 40 nm. Suitably, the colloidal silica particles have a specific surface area from about 20 to about 1000, preferably from about 40 to about 400, and most preferably from about 60 to about 300 m^2/g .

The colloidal silica particles suitably have a narrow particle size distribution, i.e.
20 with a low relative standard deviation of the particle size. The relative standard deviation of the particle size distribution is the ratio between the mean particle size by numbers and the standard deviation of the particle size distribution. The relative standard deviation of the particle size distribution preferably is lower than about 60 % by numbers, more preferably lower than about 30 % by numbers, and most preferably lower than about 15
25 % by numbers.

The colloidal silica particles suitably are dispersed in water in presence of stabilising cations such as K^+ , Na^+ , Li^+ , NH_4^+ , organic cations, quaternary, tertiary, secondary, and primary amines, or mixtures thereof so as to form an aqueous silica sol. However, also other dispersions such as organic solvents, e.g. lower alcohols, acetone or
30 mixtures thereof may be used to prepare organic silica sols. However, use of aqueous silica sols is preferred in the compositions according to the invention, and the colloidal silica particles added to the composition is dispersed in a solvent in a concentration from about 1 to about 70, preferably from about 5 to about 60, and most preferably from about 10 to about 50 wt% counted as dry weight silica. The pH of the dispersion suitably is from
35 about 1 to about 12, preferably from about 7 to about 11. A high silica content is preferred in the composition as long as the colloidal silica particles remain stable without immediate substantial aggregation and/or gelation before injecting the composition after addition of

gel agent. This is beneficial considering the superior technical performance such composition has but also the reduced transportation cost thereof.

The colloidal silica suitably has an S-value from about 20 to about 100, preferably from about 30 to about 90, and most preferably from about 60 to about 90.

5 It has been found that compositions for injection grouting with an S-value within the ranges defined herein can impart improved long-term gel strength.

The S-value characterises the extent of aggregation of colloidal silica particles, i.e. the degree of aggregate or microgel formation. The S-value has been measured and calculated according to the formulas given in Iler, R.K. & Dalton, R.L. in J. Phys. Chem.
10 60(1956), 955-957.

The S-value is dependent on the silica content, the viscosity, and the density of the colloidal silica. A high S-value indicates a low microgel content. The S-value represents the amount of SiO₂ in percent by weight present in the disperse phase of e.g. a silica sol. The degree of microgel can be controlled during the production process as
15 further described in e.g. US 5368833.

The alkali metal silicate used may contain any of the alkali metals potassium, sodium, and lithium, preferably potassium. The organic silicate may contain any suitable organic substance, e.g. amines. The molar ratio of SiO₂/M₂O, wherein M is sodium or potassium preferably is from about 1 to about 4. The molar ratio of SiO₂/M₂O, wherein M
20 is lithium or an organic component, preferably is from about 1 to about 20.

The term "gel strength" as used herein is a measure of the unconfined compressive strength of the gelled composition that develops with time. It has also been found that the composition has low permeability which is crucial in sealing applications.

The gelling agent, which is capable of gelling colloidal silica, suitably is an
25 inorganic salt or acid, e.g. an alkali metal salt, an organic salt or acid such as sodium acetate, acetic acid, preferably an inorganic salt. Examples thereof include aluminium chloride, aluminium nitrate, aluminium sulphate, potassium chloride, calcium chloride or other calcium donor such as cement, sodium chloride, magnesium chloride, magnesium sulphate, potassium iodide, sodium hydrogen phosphate, magnesium nitrate, sodium
30 nitrate, potassium nitrate, calcium nitrate, and sodium silicate or mixtures thereof, preferably calcium chloride, sodium chloride or potassium chloride, and most preferably sodium chloride and/or sodium aluminate, or mixtures thereof.

Preferably, the content of gelling agent in the composition, preferably as an aqueous composition, suitably is in an amount from about 1 to about 30 wt%, preferably
35 from about 2 to about 15 wt% of the total dry weight of silicate and silica particles.

In order to prepare the composition for injection grouting, colloidal silica particles and alkali metal silicate or organic silicate are preferably added as a premix to the gelling agent. According to one preferred embodiment, the premix and the gelling agent are added separately to the point to be sealed. This may be performed e.g. by means of so called jet grouting, wherein the ingredients are mixed in situ, e.g. in the soil.

The injection grouting composition is suitably used in smaller cracks, preferably with a diameter ranging from about 2 μm to about 20mm, preferably from about 2 μm to about 5 mm, and most preferably from about 20 μm to about 1mm.

The composition for injection grouting may further comprise a hydraulic binder or other mineral binders such as lime, gypsum etc, and/or concrete. The hydraulic binder may be e.g. a cement such as Ordinary Portland Cement (OPC), micro cement (highly ground cements) or blended cements as further described in e.g. US 6,008,275. Concrete comprises aggregates such as stone, gravel and sand, crushed concrete, lime, stone filler, slag cement, fly ash, silica flower, glass filler and other preferred inorganic material, suitably with an average particle diameter range from about 0.01 to about 100 mm, preferably from about 0.125 to about 100 mm. Aggregates are suitably comprised in an amount of from about 100 to about 1000 wt% based on the weight of the hydraulic binder. The composition comprising the hydraulic binder can be used for sealing e.g. larger cracks, preferably with a diameter ranging from about 0.02 mm to about 20 mm, preferably from about 0.05 to about 10 mm, and most preferably from about 0.1 to about 10.

A gel time regulator such as an acidic triacetine (glycerine triacetate), diacetine, Glauber's salt ($\text{NaSO}_4 \cdot 10\text{H}_2\text{O}$), hydrochloric acid, sulphuric acid, phosphoric acid or mixtures thereof may also be added to the composition to control the gelation thereof.

The term "gel time", as used herein, means the amount of time elapsed from the moment of mixing the ingredients making up the sealing composition to the time when the sealing composition becomes too viscous to move or be injected. The viscosity is suitably controlled in such a way that a homogeneous moving front of the composition is formed moving towards the point to be sealed. Suitably, the initial viscosity of the sealing composition is from about 3 to about 100, preferably from about 4 to about 30 mPas. The gel time may easily be controlled by adjustment of the amount of gelling agent. Sometimes, instantaneous gelling without substantial dilution of the composition is necessitated to provide a gelled pressure-resistant composition. A short gel time may be optimal in cracks that are relatively shallow and where the risk for rapid dilution of the composition exists. In some cases, it is essential that the gelling does not start before the front of the inserted composition has fully penetrated the crack. The gel time for sealing rock suitably ranges from about 1 to about 120, preferably from about 2 to about 60, more

preferably from about 5 to about 40, and most preferably from about 20 to about 40 minutes. The suitable gel time for soil sealing ranges from about 5 minutes to about 24 hours, preferably from about 10 minutes to about 6 hours, and most preferably from about 15 minutes to about 3 hours.

5 The components making up the composition may be mixed at ambient temperature. The gelling agent is suitably added to the premix of colloidal silica particles and silicate in an aqueous solution in an amount from about 1 to about 30 wt%, preferably from about 2 to about 15 wt% of the dry weight of the premix of silicate and silica particles.

10 The invention being thus described, it will be obvious that the same may be varied in many ways. Such variations are not to be regarded as a departure from the gist and scope of the present invention, and all such modifications as would be obvious to one skilled in the art are intended to be included within the scope of the claims. While the examples here below provide more specific details of the reactions, the following general
15 principles may here be disclosed. The following example will further illustrate how the described invention may be performed without limiting the scope of it.

All parts and percentages refer to part and percent by weight, if not otherwise stated.

20

Example 1

Samples comprising a potassium silicate solution available from Askania AB, with a SiO₂ content of 23.4 per cent by weight and a K₂O-content of 10.9 per cent by weight and silica sols of table 1 were premixed prior to adding 20 g of a 10 wt% sodium chloride solution. All samples were cured at 20 °C. The UCS (Unconfined Compressive
25 Strength) was tested in accordance with ASTM C403. All silica sol proportions are given by weight of the solutions as supplied. It can clearly be seen that an increased gel strength is achieved by the above-defined formulation compared to the control solutions (references) A and N.

30

Table 1

Silica sol	Particle Diameter	S-value (%)	Specific surface area (m ² /g)	Silica content (%)	Surface Modification
Cembinder ®20	12 nm	68	220	30	aluminate
Cembinder ®22	12 nm	76	220	40	none
Cembinder ®36	7 nm	71	360	30	none

Table 2

Composition/gel time (20 g 10 wt% NaCl solution was added to all samples A-S)	Gel strength after 24 hours (kPa)	Strength after 7 Days (kPa)
A) Control 100 g Cembinder ®20, gel time: 10 minutes	No measurable strength	No measurable strength
B) 97.5 g Cembinder ®20, 2.5g K silicate solution, gel time: 10 minutes	280	140
C) 95 g Cembinder ®20, 5 g K silicate solution Gel time: 10 minutes	690	410
D) 92.5g Cembinder ®20, 7.5g K silicate solution Forms a hard crust after addition of salt	480	690
E) 90g Cembinder ®20, 10 K Silicate solution Difficult to mix (gels quickly)	480	690
F) 85 g Cembinder ®20, 15g K Silicate solution Forms a hard brittle crust with a soft gel underneath	690	970
G) Control 100g Cembinder ®22 Gel time 10 minutes	480	759
H) 97.5g Cembinder ®22, 2.5g K Silicate solution	970	1241

Gel time: 3 minutes.		
I) 95 g Cembinder ®22, 5g K Silicate solution, very rapid gelling	1241	1655
J) 92.5g Cembinder ®22, 7.5g K Silicate solution	1517	2069
K) 90.0g Cembinder ®22, 10g K Silicate solution. Gel time: 10 minutes	2210	2621
L) 85g Cembinder ®22, 15.0g K Silicate solution Forms a hard brittle gel with some bleed liquid on top after 24 hours	1379	3034
M) 50g Cembinder ®22, 50 g K Silicate solution Does not gel	No measurable strength	No measurable strength
N) Control 100g Cembinder ®36, 20g Salt solution Gel time 10 minutes.	410	
O) 97.5 g Cembinder ®36, 2.5g K Silicate solution	690	
P) 95 g Cembinder ®36 , 5 g K Silicate solution	828	
Q) 90 g Cembinder ®36, 10 g K Silicate solution Gel time: 10 minutes	2138	
R) 85 g Cembinder ®36, 15 g K Silicate solution	1241	
S) 50g Cembinder ®36, 50 g K Silicate solution. Does not gel	No measurable strength	

Claims

1. Use of a composition for injection grouting obtainable by mixing an alkali metal silicate or an organic silicate, colloidal silica particles, and at least one gelling agent, wherein the weight ratio of silica to silicate is from about 2:1 to about 100:1.
2. Method of sealing a leaking part or cavity comprising injecting into said leaking part or cavity a grouting composition obtainable by mixing an alkali metal silicate or an organic silicate, colloidal silica particles, and at least one gelling agent, wherein the weight ratio of silica to silicate is from about 2:1 to about 100:1.
3. Method of cutting off a liquid flow in a leaking part or cavity by injecting into said leaking part or cavity a composition obtainable by mixing colloidal silica particles, an alkali metal silicate or organic silicate, and at least one gelling agent, wherein the weight ratio of silica to silicate is from about 2:1 to about 100:1.
4. Method for preparing a composition for injection grouting comprising mixing colloidal silica particles, an alkali metal silicate or an organic silicate, and at least one gelling agent, wherein the weight ratio of silica to silicate is from about 2:1 to about 100:1.
5. Grouting composition obtainable by mixing an alkali metal silicate or an organic silicate, colloidal silica particles, and at least one gelling agent, wherein the weight ratio of silica to silicate is from about 2:1 to about 100:1.
6. Composition for injection grouting comprising colloidal silica particles, an alkali metal silicate or an organic silicate, and at least one gelling agent, wherein the weight ratio of silica to silicate is from about 2:1 to about 100:1.
7. Use according to claim 1, method according to any of claims 2-4, composition according to claims 5 or 6, wherein the colloidal silica particles are present in an aqueous silica sol having an S-value from about 30 to about 90.
8. Use according to claims 1 or 7, method according to any of claims 2-4, 7, composition according to any of claims 5-7, wherein the colloidal silica particles are present in an aqueous silica sol having an S-value from about 60 to about 90.
9. Use according to any of claims 1, 7 or 8, method according to any of claim 2-4 or 7-8, composition according to any of claims 5-8, wherein the composition further comprises a hydraulic binder.
10. Use according to any of claims 1 or 7-9, method according to any of claims 2-4 or 6-9, composition according to any of claims 5-9, wherein the colloidal silica particles are present in a silica sol.
11. Use according to any of claims 1 or 7-10, method according to any of claims 2-4 or 7-10, composition according to any of claims 5-10, wherein the weight ratio of silica to silicate is from about 3:1 to about 70:1.

12. Use according to any of claims 1 or 7-11, method according to any of claims 2-4 or 7-11, composition according to any of claims 5-11, wherein the weight ratio of silica to silicate is from about 6:1 to about 20:1.

13. Use according to any of claims 1 or 7-12, method according to any of claims 5 2-4 or 7-12, composition according to any of claims 5-12, wherein the relative standard deviation of the particle size distribution is lower than about 15 % by numbers.

14. Use according to any of claims 1 or 7-13, method according to any of claims 2-4 or 7-13, composition according to any of claims 5-13, wherein the average particle diameter ranges from about 7 to about 50 nm.

INTERNATIONAL SEARCH REPORT

International Application No
PCT/SE 03/01250A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 C04B28/24 //C04B111:70

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 7 C04B C09K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	DE 100 01 831 A (KEIMFARBEN GMBH & CO KG) 16 August 2001 (2001-08-16) abstract page 2, line 40-43 page 2, line 58-67 page 4, line 21-62	1-8, 10-14
X	PATENT ABSTRACTS OF JAPAN vol. 2000, no. 16, 8 May 2001 (2001-05-08) -& JP 2001 003047 A (KYOKADO ENG CO LTD), 9 January 2001 (2001-01-09) abstract	1-12,14

 Further documents are listed in the continuation of box C. Patent family members are listed in annex.

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INTERNATIONAL SEARCH REPORT

International Application No

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C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	DE 199 25 412 A (MURJAHN AMPHIBOLIN WERKE) 7 December 2000 (2000-12-07) abstract column 3, line 57-66 column 4, line 2-15 column 5, line 17-23 ----	1-8, 10-12
X	US 4 102 400 A (CRINKELMEYER OLIVER W ET AL) 25 July 1978 (1978-07-25) column 2, line 5-13 column 3, line 27-52 column 4, line 64-68 column 7, line 8-20 ----	1-6,9,11
X	US 4 687 373 A (FALK CHARLES D ET AL) 18 August 1987 (1987-08-18) abstract column 1, line 57 -column 2, line 12 column 4, line 11-20 ----	1-6,11, 12
X	EP 0 495 336 A (HILTI AG) 22 July 1992 (1992-07-22) page 2, line 23-36 page 5, line 42-50 example 5 ----	1-10
A	ILER, R.K.; DALTON, R.L.: "Degree of hydration of particles of colloidal silica in aqueous solution" J. PHYS. CHEM., vol. 60, 1956, pages 955-957, XP002225258 cited in the application table 1 ----	7,8
A	US 5 569 323 A (FROUIN LAURENT ET AL) 29 October 1996 (1996-10-29) abstract column 4, line 5-15 example 1 ----	1-14
A	US 5 194 087 A (BERG JOHANNES G) 16 March 1993 (1993-03-16) abstract column 1, line 46-54 example 1 ----	1-14
A	DE 41 04 596 A (WOELLNER WERKE) 20 August 1992 (1992-08-20) the whole document -----	1-14

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/SE 03/01250

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
DE 10001831	A	16-08-2001	DE 10001831 A1	16-08-2001
			AT 236959 T	15-04-2003
			AU 2517401 A	31-07-2001
			CA 2397085 A1	26-07-2001
			CN 1395600 T	05-02-2003
			CZ 20022407 A3	15-01-2003
			DE 50100161 D1	15-05-2003
			DK 1222234 T3	28-07-2003
			WO 0153419 A1	26-07-2001
			EP 1222234 A1	17-07-2002
			NO 20023411 A	18-09-2002
			US 2003127024 A1	10-07-2003
JP 2001003047	A	09-01-2001	NONE	
DE 19925412	A	07-12-2000	DE 19925412 A1	07-12-2000
			AT 238248 T	15-05-2003
			AU 5672500 A	18-12-2000
			WO 0073237 A1	07-12-2000
			DE 50001881 D1	28-05-2003
			EP 1185486 A1	13-03-2002
US 4102400	A	25-07-1978	CA 1091438 A1	16-12-1980
US 4687373	A	18-08-1987	AT 83402 T	15-01-1993
			AU 588470 B2	14-09-1989
			AU 7480787 A	22-12-1987
			BR 8707311 A	13-09-1988
			CA 1280558 C	26-02-1991
			DE 3783124 D1	28-01-1993
			DE 3783124 T2	22-04-1993
			EP 0268644 A1	01-06-1988
			MX 165975 B	15-12-1992
			WO 8707189 A1	03-12-1987
			JP 1055908 B	28-11-1989
			JP 63502570 T	29-09-1988
			JP 64055908 B	28-11-1989
EP 0495336	A	22-07-1992	DE 4100386 A1	16-07-1992
			DE 59104619 D1	23-03-1995
			EP 0495336 A1	22-07-1992
			US 5565026 A	15-10-1996
US 5569323	A	29-10-1996	FR 2705338 A1	25-11-1994
			BR 9402026 A	27-12-1994
			CA 2124068 A1	22-11-1994
			CN 1104695 A	05-07-1995
			EP 0625488 A1	23-11-1994
			FI 942357 A	22-11-1994
			JP 7053959 A	28-02-1995
			NO 941893 A	22-11-1994
US 5194087	A	16-03-1993	NO 904639 A ,B,	19-11-1991
			AU 641247 B2	16-09-1993
			AU 7710691 A	21-11-1991
			CA 2042841 A1	19-11-1991
			DK 94391 A	19-11-1991
			EP 0457516 A1	21-11-1991

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/SE 03/01250

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 5194087	A	FI 912379 A	19-11-1991
DE 4104596	A	DE 4104596 A1	20-08-1992