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(54) Title: TEMPERATURE-ACTIVATED ADHESIVE ASSEMBLIES		
(57) Abstract <p>Temperature-activated adhesive compositions formulated with crystallizable polymers are provided. The compositions, which display a rapid transition from the tacky to the nontacky state, and/or from the nontacky to the tacky state, are useful in medical applications, where adhesion of a substrate to the skin is desired, as well as in a variety of nonmedical applications. Temperature-activated adhesive assemblies such as tapes, labels, and the like for use in a number of contexts are also provided.</p>		

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TEMPERATURE-ACTIVATED ADHESIVE ASSEMBLIES

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DescriptionTechnical Field

15 The present invention relates to adhesive compositions and assemblies, and more particularly concerns temperature-activated adhesive compositions and assemblies for medical and other purposes.

Background Art

20 Pressure-sensitive adhesives (PSA) are well known and are used for a variety of industrial, consumer and medical applications. Pressure-sensitive adhesives are characterized as being normally tacky and exhibiting instant tack when applied to a substrate. A variety of
25 polymers have been used to manufacture PSA, for example, acrylic and methacrylic ester homo- or copolymers, butyl rubber-based systems, silicones, urethanes, vinyl esters and amides, olefin copolymer materials, natural or synthetic rubbers, and the like.

30 Typically, the bond strength of a PSA remains the same or increases with time after application to a substrate. Increased bond strength may result from increased polymer flow or improved wetting over a period
35 of time, or it may result from hydrogen bonding to a

polar substrate or from covalent or ionic chemical interactions. Removal of an adhesive after application to a substrate is often desirable. For example, tape may be applied to a substrate to mask off a portion during a painting operation, at the completion of which the tape must be removed. Wallpaper can be applied with a PSA and removal at a future date is usually desirable. Adhesives applied to human skin, for example on a bandaid, wound dressing, transdermal drug delivery device, or monitoring or stimulating electrode, must be removed at a later date. In all of these instances and many others, it is desirable to be able to readily remove the adhesive without the need for harsh chemicals or special equipment.

The ideal performance characteristics of an adhesive intended for use on human skin, specifically, present difficult and conflicting technical requirements. The ideal medical adhesive should, first of all, be nonirritating yet bond quickly to skin at the intended site of use. At the same time, clearly, it should not stick to other objects until it is actually applied to the target site. The adhesive should maintain its bond for as long a period of time as necessary and be resistant to inadvertent removal. Furthermore, the adhesive should not be weakened or destroyed by exposure to moisture or high humidity. Finally, in order to provide protection to a wound or to maintain the integrity of placement of an electrode or other device the adhesive should resist skin movement and be able to transfer a mechanical load from the adhesive backing to the skin.

The invention is thus directed in one aspect to adhesive compositions and assemblies which are particularly useful in medical applications (i.e., for adhering to the skin), and which meet the

above-mentioned requirements. As will be discussed in detail hereinbelow, these medical adhesive assemblies are "temperature-activated" (or "temperature-sensitive") in that the medical adhesive assemblies are either: (1) "pre-positionable", i.e., they present a surface which is substantially nontacky below skin temperature and permits accurate placement on the skin before absorbing enough heat from the skin to become aggressively tacky. Thereafter, they adhere to the skin in the manner of a normal PSA; (2) removable by chilling, i.e., at skin temperatures, they exhibit strong adhesion to the skin but lose adhesion upon cooling; or (3) a combination of embodiments (1) and (2). All of these embodiments involve an adhesive composition which displays a rapid transition between a tacky state and a substantially nontacky state. For example, if the adhesive assembly is heated up above the normal skin transition temperature, the adhesive component will become aggressively tacky and stick to almost anything with which it comes in contact, i.e., it will form a stronger bond with a release liner if one is present. However, when the assembly is cooled to several degrees below that skin transition temperature, the adhesive will lose adhesivity, permitting its removal from the liner and rapidly revert to its nontacky state.

The invention is also directed to other types of assemblies and devices which make use of the temperature-activated adhesive compositions described herein. As may be readily inferred from the present disclosure, the temperature sensitivity of the adhesive compositions of the invention provides for a great deal of versatility and enables use of the assemblies in a number of different contexts. Examples of additional uses include: "chill off" labels or tape, i.e., which stick aggressively to a substrate during the desired

period of use but which are readily removable upon chilling; and heat-activated adhesive assemblies which are normally nontacky but which quickly become tacky when warmed to a temperature slightly above room
5 temperature.

The following references relate to one or more aspects of the present invention:

S.C. Temin, in the Encyclopedia of Polymer Science and Engineering, vol. 13 (New York: John Wiley & Sons, 1988), at pp. 345-368, and the Handbook of Pressure-Sensitive Adhesive Technology, ed. Donates Satas (New York: Van Nostrand Reinhold Co., Inc., 1982), both provide a comprehensive overview of medical and other adhesives. A.H. Landrock, Adhesives Technology Handbook (Park Ridge, New Jersey: Noyes Publications, 15 1985), pp. 154-156, and T. Flanagan, "Hot-melt Adhesives", in the Handbook of Adhesive Bonding, ed. C.V. Cagle (New York: McGraw-Hill, 1982), at pp. 8-1 to 8-17, describe hot-melt adhesives, i.e., adhesives which are applied to a substrate as a melt and which solidify and bond upon cooling. In contrast to the adhesive compositions of the present invention, hot-melt adhesives involve flowable polymers and do not allow for reversible adhesion.

25 U.S. Patent No. 3,635,754 to Beede describes a temporary pressure-sensitive polyolefin-based adhesive product which is activated by heating at least 15°C above its melting point, and which, after cooling, remains tacky for an extended period of time. (By contrast, applicant's adhesive products require no heat activation--with the exception of a few "warm-activated" compositions, which require only minimal "warming" to a temperature slightly above room temperature--and, in the embodiment wherein adhesion is temperature-reversible, 30 tack is very quickly lost upon cooling.) While not
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wishing to be bound by theory, it is postulated that the cause of gradation of tack of the composition described by Beede (see example 3c) from a nontacky to an aggressively tacky adhesive over more than a fifteen degree range--in contrast to the rapid transition seen with applicant's composition--is a result of the relative fraction of amorphous and crystalline components present in the composition. Not all the crystalline components in the olefin produced following the teaching of Beede exhibit melting transitions at precisely the same temperature. (That is, a sharp change from tacky to nontacky will result where there is rapid conversion of most of the amorphous to mostly crystalline material and vice versa over a narrow temperature range. By preparing a polymeric composition such that it has a very high level of crystallinity at one temperature but a very high level of amorphous material at a slightly higher temperature, applicant has found that it is then possible to achieve the desired adhesive switching over a relatively small temperature range.)

U.S. Patent No. 3,284,423 to Knapp describes a pressure-sensitive, cross-linked adhesive copolymer which is claimed to be storage stable in solution but readily cured when coated and heated. The copolymer consists of acrylic acid esters, lower alkyl acrylates (1-8 carbons, preferably 1-4 carbons), acrylic acid and glycidyl acrylate or methacrylate.

U.S. Patent No. 3,535,195 to Davis et al. describes a pressure-sensitive, amine-containing adhesive which is stated to exhibit good tack yet be easily removable from a substrate.

U.S. Patent No. 3,690,937 to Guse et al. relates to pressure-sensitive adhesives formulated from di-alkyl fumarates.

U.S. Patent No. 3,838,079 to Kosaka et al. describes copolymer resins prepared from alkyl acrylates (1-20C) and maleic anhydride.

5 U.S. Patent No. 3,299,010 to Samour describes a variety of adhesive compositions, some of which contain C-12 to C-24 residues. The patent teaches that the higher alkyl residues must be branched in order to avoid crystallinity. Some of the compositions contain acrylamido or other polar groups, and claims of adhesion
10 to moist skin are made.

U.K. Patent No. 870,022, inventors Dahlquist and Zenk, teaches the use of a copolymer of octadecyl acrylate, acrylonitrile, ethyl acrylate, and acrylic acid as a low adhesion backsize on a pressure-sensitive
15 adhesive tape.

PCT Publication No. WO84/03837 teaches the use of copolymers which contain a polyalkylene oxide monomer in addition to acrylate. The polyoxyalkylene moiety is stated to impart hydrophilic behavior to the adhesive
20 composition, thus facilitating adhesion to moist skin.

European Patent Application Publication No. 062682 describes the use of the monomer dodecyl methacrylate in a copolymer employed as an adhesive carrier for nitroglycerine. Small amounts of other
25 comonomers (acrylic acid and short chain acrylates) are added to the reaction mixture to improve the properties of the copolymer.

The following references relate to side-chain crystallizable polymers: J. Polymer Sci.:
30 Macromolecular Review 8:117 (1974) and J. Polymer Sci.:
Polymer Chemistry Edition 19:1871-1873 (1981).

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Disclosure of the Invention

Accordingly, it is a primary object of the invention to address the above-discussed deficiencies in the art, and to provide temperature-activated adhesive compositions and assemblies which display a rapid transition between tacky and substantially nontacky states.

It is also a primary object of the present invention to provide a temperature-activated adhesive assembly for application to the skin.

It is also an object of the invention to provide a medical adhesive assembly, specifically, which can be removed from the skin with less force, pulling and trauma than conventional adhesive tapes and bodies necessitate.

It is another object of the invention to provide a temperature-activated, pre-positionable medical adhesive assembly which is substantially nontacky at room temperature, becomes tacky upon application to skin, and retains adhesivity indefinitely.

It is still another object of the invention to provide a medical adhesive assembly which is aggressively tacky at skin temperature, and which rapidly loses tack upon cooling.

It is a further object of the invention to combine the advantages and characteristics of the latter two embodiments, in providing a pre-positionable medical adhesive assembly which is substantially nontacky at room temperature, becomes aggressively tacky upon contact with the skin, and loses adhesivity upon cooling.

Still a further object of the invention is to provide temperature-activated adhesive assemblies useful as labels.

Yet an additional object of the invention is to provide "warm-activated" adhesive assemblies formulated with an adhesive composition that is normally nontacky but which may be rendered tacky when warmed to a temperature just slightly above room temperature.

It is still a further object of the invention to provide methods of making and using the aforementioned adhesive assemblies.

Additional objects, advantages and novel features of the invention will be set forth in part in the description which follows, and in part will become apparent to those skilled in the art on examination of the following, or may be learned by practice of the invention.

In its broadest aspect, the invention is directed to a temperature-activated pressure-sensitive adhesive, wherein the adhesive comprises a polymeric composition containing a polymer having a first-order melting transition which occurs over a melting range less than about 15°C, the polymer being present in the composition in an amount effective to render the adhesive substantially nontacky at temperatures below the melting range and tacky at temperatures within or above the melting range. As noted above, the inventor herein has now found that a sharp transition from tacky to nontacky, or from nontacky to tacky, can be achieved by preparing a polymer composition which has a high level of crystallinity at one temperature and a high level of amorphous material at a slightly higher temperature.

The invention is also directed to temperature-activated adhesive assemblies for use in medical applications, i.e., adhesive assemblies which make use of the aforementioned adhesive and are particularly useful for adherence to the skin. The invention

encompasses the following three types of medical adhesive assemblies.

5 The first of these medical adhesive assemblies is a temperature-activated, "pre-positionable" medical adhesive assembly comprising a body member having a surface coated with an adhesive composition that is substantially nontacky at room temperature and tacky at skin temperature. By "pre-positionable" is meant an assembly which can be positioned and maneuvered about, at room temperature, or any other temperature below its application temperature, without unwanted adhesion until it contacts and is warmed by skin or some other substrate which causes the adhesive component to experience an elevation of its temperature above the transition point such that the adhesive component becomes aggressively tacky and adheres to the skin or other substrate.

10 The second medical adhesive assembly comprises a body member having a surface coated with an adhesive composition which, while tacky at, or slightly below skin temperature, loses adhesivity upon cooling 5 to 15 degrees Celsius below that temperature.

15 The third medical adhesive assembly combines the features of both of the aforementioned embodiments in that it is both pre-positionable (i.e., substantially nontacky at room temperature), aggressively tacky at skin temperature, and capable of losing adhesivity upon cooling.

20 These adhesive assemblies may be adhesive tapes in which a flexible backing is coated with a selected polymer, or they may comprise casts, splints or other immobilization devices which have been similarly coated. Where the composition provides temperature-reversible adhesion, strong bonding will be provided during use, i.e., between the immobilization device and

the skin, but after use, upon cooling, the adhesive becomes substantially less adhesive, allowing easy removal of the immobilization device.

5 Nonmedical adhesive assemblies are also within the scope of the present invention, and include temperature-sensitive labels or tape which stick aggressively to a substrate during the desired period of use but which are readily removable upon chilling; and
10 warm-activated adhesive assemblies which are normally nontacky but which become tacky when warmed to a temperature slightly above room temperature.

Modes for Carrying Out the Invention

15 The invention is thus directed to temperature-activated adhesive assemblies useful in a variety of medical and nonmedical applications. All of the temperature-activated adhesive assemblies of the invention make use of a polymeric composition comprising a polymer which has a first-order melting transition
20 between about 5°C and about 50°C, and wherein that transition occurs over a melting range of less than about 15°C, more preferably over a range of less than about 10°C. Such a narrow range for the melting transition ensures that the transition from the tacky
25 state to the substantially nontacky state (and/or vice versa) will be quite rapid. In all of the presently disclosed assemblies, in addition, the aforementioned polymer is present in an amount effective to render the adhesive composition substantially nontacky at
30 temperature below the melting range and tacky at temperatures within or above the melting range. These adhesive assemblies include medical as well as additional types of adhesive assemblies described and claimed herein.

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First turning to the adhesive assemblies useful in medical applications, those assemblies comprise body members having a surface which is coated with a temperature-activated adhesive composition. The
5 body member may be, for example, an EKG electrode, or other type of electrode, a flexible backing in the case of adhesive tape, a surgical dressing, bandaid, medicated bandaid, a transdermal drug delivery patch, or the like, or it may be a cast, splint, or similar
10 immobilization device. Other skin contact applications are intended to be encompassed by the present invention as well, as the focal point of the invention is not on particular uses of the adhesive assemblies described and claimed herein but on the incorporation of a
15 temperature-activated adhesive composition into different types of adhesive assemblies.

Three types of medical adhesive assemblies are intended to be encompassed by the subject invention:
20 (I) A pre-positionable adhesive assembly which is initially substantially nontacky, i.e., at or below room temperature, but which becomes tacky relatively quickly upon application to skin; (II) an adhesive assembly which may or may not be tacky to start with, but which strongly bonds to skin and is capable of losing adhesion
25 upon cooling; and (III) an adhesive assembly in which the characteristics of Embodiments (I) and (II) are combined, i.e., a temperature-reversible system which is initially substantially nontacky, strongly bonds to skin, and loses adhesivity upon cooling.

30 In Embodiment (I), the adhesive composition comprises a polymer which has a first-order transition temperature or melting point above room temperature (i.e., approximately 25°C in most cases) but at or below skin temperature. By the terms "melting point" or
35 "first order transition", as used herein is meant the

temperature at which an equilibrium process causes certain portions of the polymer, initially aligned in an ordered array, to become disordered. By the term "freezing point", is meant the temperature at which an equilibrium process causes those certain portions of the polymer, initially disordered, to become aligned in an ordered array. Preferably, the first-order transition temperature or melting point will be in the range of about 20°C to 35°C, more preferably in the range of about 25°C to 30°C. It is preferred that melting occur rapidly, i.e., over a relatively narrow temperature range, less than about 10°C, preferably less than about 5°C. In Embodiment (I), the adhesive assembly is substantially nontacky up until the moment of application, and becomes tacky upon contact with skin; no additional external activation of any sort is required. (Melting of most polymers can be monitored with the aid of a differential scanning calorimeter, or "DSC". Melting generally takes place over a 10°C range, and onset of tack occurs at the onset of melting; the melting "point" of the polymers as described herein is actually the temperature at which melting begins. The strongest bonding occurs when the temperature is high enough to effect complete melting.)

In this embodiment, the adhesive is typically completely nontacky at room temperature, so that a bandage, wound dressing, or the like formulated with the adhesive composition can be applied, positioned and maneuvered about, if necessary, within a short period of time and without unwanted adhesion. In some cases, however, it will be preferred that the adhesive be slightly tacky at room temperature, so that a release liner may adhere to and protect the adhesive coating prior to use. Slight tack at room temperature may be achieved by impregnation of the adhesive composition

with standard pressure-sensitive adhesives or incorporation of tackifiers into the selected polymer.

5 In Embodiment (II), the adhesive composition comprises a polymer which provides for loss of adhesion upon simple chilling, e.g., by application of ice, a cold pack, or the like. In this embodiment, the polymer should have a freezing (or "crystallization") point lower than skin temperature, preferably in the range of about 10°C to 28°C, more preferably in the range of
10 about 15°C to 25°C. It is also preferable that the polymer crystallize rapidly. To this end, seeding agents, or crystallization catalysts can be incorporated into the polymer which provide for rapid crystallization kinetics. In this embodiment, removal of the adhesive
15 assembly from the skin is greatly facilitated; after use, adhesive tapes, bodies and the like may be readily removed by simple chilling, only slightly below their use temperature, without imparting undue trauma to the skin.

20 In Embodiment (III), the characteristics of the adhesive assemblies of Embodiments (I) and (II) are combined. That is to say, adhesive assemblies in this class display temperature-reversible adhesion; they are substantially nontacky initially, prior to use (at room
25 temperature), they become tacky upon application to skin, and they rapidly lose tack and may thus be removed from skin by cooling. In this embodiment, it is preferred that the polymer of the adhesive assembly have a melting point or first-order transition temperature
30 within the range set forth for Embodiment (I), and a freezing or crystallization temperature within the range set forth for Embodiment (II).

35 Now turning to those embodiments which may have nonmedical as well as medical applications, the present invention also encompasses the use of a

"warm-activated" adhesive assembly (for purposes of convenience, designated "Embodiment (IV)" herein). As explained above, such an assembly is formulated with an adhesive composition that is normally nontacky but which
5 may be rendered tacky when warmed just slightly above room temperature, preferably less than about 20°C above room temperature, more preferably less than about 15°C above room temperature, most preferably less than about 10°C above room temperature. Such an assembly is
10 suitable for application to the skin, and is, in addition, useful in a number of other contexts.

In this case, again, it is desirable that the transition from nontacky to tacky occur over a relatively small temperature range, i.e., over a range
15 of less than about 15°C, and not require activation above about 45°C. Temperatures of up to 37°C may be obtained simply by contact with human skin while temperature of up to 45°C are readily attained (by using, for example, a hair dryer). Activation
20 temperatures greater than about 45°C are not desirable for medical applications used on skin because skin will not tolerate a temperature much above that temperature.

The warm-activated adhesive assembly is particularly useful for sticking permanently to porous
25 substrates after activation, even when the assembly is exposed to low temperatures for long periods of time. By "porous" substrates is meant, e.g., paper, cloth, wood, and the like, i.e., substrates that have a structure which allow for at least partial penetration
30 of the activated adhesive composition.

In Embodiment (V), temperature-activated adhesive compositions are used to formulate "chill-off" labels or tape, i.e., labels or tape which stick
35 aggressively to a substrate during a designated period of use but which may be readily removed upon chilling.

This type of adhesive assembly addresses several needs in the art. First, it is clearly desirable to have a normally tacky pressure-sensitive adhesive which bonds well to a variety of substrates yet can be easily removed at a later time. This will be readily appreciated by most if not all consumers who find virtually all labels impossible to remove completely without considerable time, scraping, and/or use of solvents or other chemicals. In addition, in the painting of substrates it is common to use tape to mask-off area that is not to be painted, and to later remove the tape. If the tape sticks poorly, paint may bleed into the masked area; if the tape sticks too strongly it may remove a portion of the substrate--also, clearly, an undesirable result. In still another example, it is desirable to affix labels or price tags to articles of commerce such that they cannot be readily removed. This is currently done by using adhesives and labels selected such that removal of the label results in cohesive failure of the label material. While this serves a "tamper-resistant" function, this type of label will be difficult to remove even when removal is proper, i.e., after purchase.

The "chill off" adhesive assembly of the invention, which can be rendered significantly less strongly bonded by simple chilling, thus addresses each of the aforementioned needs. An adhesive is used which is a normally tacky pressure-sensitive adhesive, but which loses a significant portion of its adhesion or bond strength to the substrate when cooled to a low temperature, thus allowing the adhesive assembly--i.e., the label, tape, or the like--to be readily removed from the substrate without leaving any residue and without need for scraping, solvents, or other chemicals. Such

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an adhesive composition can be used on many types of substrates, e.g., metal, glass, plastic, and the like.

The term "tack" as used hereinabove is intended to designate the tacky or sticky nature of the adhesive compositions. Tack can generally be determined by what is referred to as the thumb test in which the thumb makes rapid and reversible contact with the surface being considered to determine the tacky or sticky nature of the surface. A substantially more accurate and reproducible test for tack is designated D2979 by the American Society for Testing and Materials. In this test which employs a Polyken™ instrument, values are given in grams of force required to remove the end of a stainless steel rod, 5.0 mm in diameter, from the surface of an adhesive coated assembly. Prior to the removal, the rod approaches the surface at a speed of 10 mm per second and makes contact with the adhesive for a prescribed period of time. The actual time of contact is variable and discretionary. (In obtaining the values reported herein, a half-second dwell time was used.) The units of measure are given in grams•cm/sec and are the ones to which a plurality of the tack values used in the present specification refer. The Tack Rolling Ball Test (TRBT) designated as D3121 by the above Society and PSTC-6 (revision 8/85) by the Pressure Sensitive Tape Council is also ideally suited for accurately and reproducibly measuring tack over wide temperature ranges. This test is described in the Experimental section of this application, below.

The terms "tack" and "tacky" are qualitative. The tack of an adhesive film coating is affected by its thickness, especially when the coating is less than about .005 in thickness. Generally, tack increases linearly with thickness. In order to make a more quantitative evaluation between different samples made

with different coating thicknesses, tack values can be (and have been in this application) normalized to values which would be expected had that sample been cast or otherwise applied at a thickness of .001 inch.

5 The terms relating to the normalized value properties of an adhesive coating obtained by either the PKI or TRBT tack determination method can be verbally described as "substantially nontacky", "slightly tacky" and "tacky". These terms are intended to relate to
10 approximate values, respectively: (1) a tack value which is the minimum limit of the instrument or less than about 25 g•cm/sec of force; (2) a tack value between that described in (1) to about 100 g•cm/sec of force; and (3) a tack of at least about 100 g•cm/sec of
15 force.

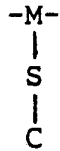
The term "bond strength" between an adhesive coated article and the substrate to which it is joined is defined as the force required to separate those components.

20 The term "peel" as used hereinabove is intended to define the bond strength between an adhesive coated article and the substrate to which it's joined when those components are separated at a rate of 12 inches/min and where the angle between separated
25 components approximates 180°. The peel strength of an adhesive film coating is also affected by its thickness, especially when the coating is less than .005 inches thick. Generally, peel strength of any particular
30 adhesive increases linearly with the square root of thickness. In order to make a more quantitative evaluation between different samples made with different coating thicknesses, peel strength values can be (and
35 have been in this application) normalized to values which would be expected had that sample been cast or otherwise applied at a thickness of .001 inch.

The polymer in each of Embodiments (I), (II) and (III), (IV) and (V) is preferably a crystallizable polymer or a functional equivalent of a crystallizable polymer having a weight average molecular weight in the range of about 20,000 to 2,300,000 Daltons, typically 100,000 to 1,300,000 Daltons, most typically 250,000 to 1,000,000 Daltons. By polymers which are "functionally equivalent" to crystallizable polymers for purposes of the present invention, applicant intends to include polymers which exhibit the temperature-dependent adhesion properties described above. Crystallizable polymers which may be used in the adhesive composition include both side-chain crystallizable and main-chain crystallizable polymers, the difference being that the former class of compounds contain crystallizable side-chain moieties, and the latter class are rendered crystallizable by their backbone structure. Depending on the embodiment, the polymer selected for incorporation into the adhesive assembly will comprise different monomers which provide the composition with the desired phase-transition temperature, bond strength, and tack. The adhesive composition may also be formulated so as to contain blends of two or more different polymers as described herein.

Side-chain crystallizable polymers, sometimes called "comb-like" polymers, are well-known and available commercially. These polymers are reviewed in J. Polymer Sci.: Macromol. Rev. 8:117-253 (1974), the disclosure of which is hereby incorporated by reference. In general, these polymers contain monomer units X of the formula:

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wherein M is a backbone atom, S is a spacer unit and C is a crystallizable group. These polymers generally have a heat of fusion (ΔH_f) of at least about 20 Joules/g, preferably at least about 40 Joules/g. The polymers will contain 50 to 100 wt.% monomer units represented by "X". If the polymer contains less than 100% X, it will in addition contain monomer units which may be represented by "Y" or "Z", or both, wherein Y is any polar or nonpolar monomer or mixture of polar or nonpolar monomers capable of polymerizing with X and/or Z, and wherein Z is a polar monomer or mixture of polar monomers. These polar monomers--e.g., polyoxyalkylenes, acrylates including hydroxyethylacrylate, acrylamides and methacrylamides--will typically increase adhesion to most substrates. If the polar species "Z" is acrylic acid, it is preferred that it comprise about 1-10 wt.% of the polymer.

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The backbone of the polymer (defined by "M") may be any organic structure (aliphatic or aromatic hydrocarbon, ester, ether, amide, etc.) or an inorganic structure (sulfide, phosphazine, silicone, etc.), and may include spacer linkages which can be any suitable organic or inorganic unit, for example ester, amide, hydrocarbon, phenyl, ether, or ionic salt (e.g., a carboxyl-alkyl ammonium or sulfonium or phosphonium ion pair or other known ionic salt pair).

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The side-chain (defined by "S" and "C") may be aliphatic or aromatic or a combination of aliphatic and aromatic, but must be capable of entering into a crystalline state. Common examples are: linear

aliphatic side-chains of at least 10 carbon atoms, e.g., C₁₄-C₂₂ acrylates or methacrylates, acrylamides or methacrylamides, vinyl ethers or esters, siloxanes or alpha olefins; fluorinated aliphatic side-chains of at least 6 carbons; and p-alkyl styrene side-chains wherein the alkyl is of 8 to 24 carbon atoms.

The length of the side-chain moiety is usually greater than 5 times the distance between side-chains in the case of acrylates, methacrylates, vinyl esters, acrylamides, methacrylamides, vinyl ethers and alpha olefins. In the extreme case of a fluoroacrylate alternate copolymer with butadiene, the side-chain can be as little as two times the length as the distance between the branches. In any case, the side-chain units should make up greater than 50% of the volume of the polymer, preferably greater than 65% of the volume.

Specific examples of side-chain crystallizable monomers are the acrylate, fluoroacrylate, methacrylate and vinyl ester polymers described in J. Poly. Sci. 10:3347 (1972); J. Poly. Sci. 10:1657 (1972); J. Poly. Sci. 9:3367 (1971); J. Poly. Sci. 9:3349 (1971); J. Poly. Sci. 9:1835 (1971); J.A.C.S. 76:6280 (1954); J. Poly. Sci. 7:3053 (1969); Polymer J. 17:991 (1985), corresponding acrylamides, substituted acrylamide and maleimide polymers (J. Poly. Sci.: Poly. Physics Ed. 18:2197 (1980)); poly(α -olefin) polymers such as those described in J. Poly. Sci.: Macromol. Rev. 8:117-253 (1974) and Macromolecules 13:12 (1980), polyalkyl-vinylethers, polyalkylethylene oxides such as those described in Macromolecules 13:15 (1980), alkylphosphazene polymers, polyamino acids such as those described in Poly. Sci. USSR 21:241, Macromolecules 18:2141, polyisocyanates such as those described in Macromolecules 12:94 (1979), polyurethanes made by reacting amine- or alcohol-containing monomers with

long-chain alkyl isocyanates, polyesters and polyethers, polysiloxanes and polysilanes such as those described in Macromolecules 19:611 (1986), and p-alkylstyrene polymers such as those described in J.A.C.S. 75:3326 (1953) and J. Poly. Sci. 60:19 (1962).

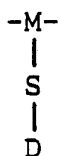
Of specific utility are polymers which are both relatively polar and capable of crystallization, but wherein the crystallizing portion is not affected by moisture. For example, incorporation of polyoxyethylene, polyoxypropylene, polyoxybutylene or copolyoxyalkylene units in the polymer will make the polymer more polar, improving adhesion to moist skin.

In a particularly preferred embodiment herein, in the above structure, -C is selected from the group consisting of $-(CH_2)_n-CH_3$ and $-(CF_2)_n-CF_2H$, where n is an integer in the range of 8 to 20 inclusive, -S- is selected from the group consisting of -O-, $-CH_2-$, $-(CO)-$, $-O(CO)-$ and $-NR-$ where R is hydrogen or lower alkyl (1-6C), and -M- is $-[(CH_2)_m-CH]-$ where m is 0 to 2.

Typical "Y" units include linear or branched alkyl or aryl acrylates or methacrylates, alpha olefins, linear or branched alkyl vinyl ether or vinyl esters, maleic esters or itaconic acid esters, acrylamides, styrenes or substituted styrenes, acrylic acid, methacrylic acid and hydrophilic monomers as detailed in WO84/0387, cited supra.

In addition to the above-described monomer units "M-S-C", monomer structures given by

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may in addition, or in the alternative, be present in the polymer. "D" is a hydrophilic polyether chain such as a polyoxyalkylene chain (e.g., polyoxyethylene) which, in contrast to "C", may or may not be
5 crystallizable. "D" preferably has a molecular weight higher than about 100 Daltons.

It is important in the case of polyolefins, which can exist in a plurality of tactic forms, that in order to effect the sharpness of transition between
10 tacky and nontacky states the tacticity of the polymer must be carefully selected. The polymer can be present in a singular configuration, i.e., either atactic, syndiotactic or isotactic, but not in a mixture of tacticities unless their melting points opportunis-
15 tically coincide. Having a mixture of various tactic polymers with different melting points will broaden the transition and cause the resultant polymer to exhibit sluggish adhesive property changes over a narrow temperature range.

20 Preferred main-chain crystallizable polymers include water-insoluble polyalkylene oxides, lower alkyl polyesters and polytetrahydrofuran.

The crystallizable polymer, whether side-chain or main-chain crystallizable, may or may not be
25 cross-linked. Cross-linking the adhesive composition as well as employing high molecular weight polymers will, in general, result in a material that exhibits decreased melt flow and greater cohesive strength than non-cross-linked and low molecular weight materials. Because the
30 adhesive composition may be used at temperatures above the melting point of the polymer, low melt flow is desirable so that the adhesive will not migrate, flow or transfer to the substrate surface (i.e., in contrast to conventional "hot-melt" adhesives). Adhesive
35 compositions with sufficient cohesive strength to

prevent cohesive failure is thus desirable. Low melt flow and suitable cohesive strength may be achieved by other means such as the addition of suitable co-monomers (e.g., high Tg monomers), by the use of block
5 copolymerization or other art-known methods, or the inducement of cross-linking before, during or after preparation of the adhesive assembly.

A variety of methods are available to produce cross-linked crystallizable materials. A network
10 copolymer can be prepared by polymerizing a crystallizable monomer and a multifunctional monomer either in one or two steps. A one-step process may be used to form an adhesive in place, while a two-step process is useful where an intermediate processing step
15 is necessary. A variety of multifunctional monomers (di-, tri- or multifunctional acrylic or methacrylic esters, vinyl ethers, esters or amides, isocyanates, aldehydes, epoxies and the like) are known in the art. These multifunctional monomers can be used in a one- or
20 two-step process depending on the desired result. Ionizing radiation, for example beta or gamma radiation, peroxides, silanes, or similar cure agents, can be used to cross-link a preformed crystallizable polymer with or without added comonomers. Ionic cross-links can be
25 formed by, for example, reacting an acidic polymer site with a di- or trivalent metal salt or oxide to produce a complex which serves as a cross-link site. Likewise, organic salts or complexes can be prepared by methods known in the art.

30 If the material is cross-linked to too great an extent, crystallinity and/or tack may be decreased to the point that the desirable temperature-activated properties are lost. To optimize the aforementioned factors, cross-linking should be in the range of about
35 0.01 percent to 5 mole percent and preferably .05 to 1

mole percent. The cross-linked polymers will normally have a heat of fusion of at least about 20 Joules/g, and preferably at least 30 Joules/g.

5 Effective cross-linking may also be obtained by physical methods. For example, a block copolymer containing a crystallizable portion and a second portion which exhibits a glass transition or melting point higher than the crystallizable polymer may be prepared wherein the entire mass exhibits mechanical stability
10 above the melting point of the crystallizable polymer but below the transition of the second polymer.

As noted earlier, it is also desirable that the adhesive composition be formulated with a polymer or mixture of polymers selected such that the first-order
15 melting transition of the composition occurs over a range narrower than about 15°C, preferably narrower than about 10°C. It is also preferred that the melting transition occur between about 5°C and about 50°C. In addition, the tack of the composition must develop
20 within less than a minute when it is raised above the melting range and then, without contacting any substrate, revert to the nontacky state when the temperature is lowered a few degrees below the melting range. This reversal should take less than about 5
25 minutes. The latter tack value (PKI) is preferably less than about 25 g•cm/sec or the minimum limit of the tack measuring instrument.

The adhesive compositions useful herein may include, in addition to one or more polymers as
30 described above, conventional additives such as tackifiers (wood rosin, polyesters, etc.), antioxidants, fibrous or nonfibrous fillers, colorants, and the like. It is also possible to include additional adhesives, providing that the overall temperature sensitivity
35 profile is not significantly affected. It is preferred

that the amount of crystallizable polymer in the adhesive composition be in the range of about 40 wt.% to about 100 wt.%.

5 Coating of the body members, or substrates, with the temperature-activated adhesive composition (i.e., to provide tapes, labels, or the like) may be done in any number of ways, e.g., by spray deposition, painting, dipping, gravure printing, rolling, or the like. The adhesive composition may also be applied by
10 transfer from a release sheet, i.e., in a manner similar to that involved in transfer printing. The composition may be applied neat, or in a suitable solvent or as an emulsion or latex. Alternatively, a mixture of the appropriate monomers and additives may be applied
15 directly to a substrate and cured in place by heat, irradiation, or other suitable art-known processes.

In adhesive tapes and sheets, specifically, the backing onto which the temperature-activated adhesive composition is coated may comprise any number
20 of backings which are well-known in the medical or surgical fields. Thus, the backing may be a woven or nonwoven fabric, paper, or a synthetic film. Depending on the specific medical application, the backing may or may not be occlusive.

25 It will be appreciated by those skilled in the art that the temperature-activated adhesive compositions and adhesive assemblies described herein are useful in a variety of medical applications, i.e., in binding adhesive tape, bandaids, immobilization devices,
30 transdermal drug delivery devices, surgical dressings, EKG electrodes, etc., to skin.

The foregoing description as well as the examples which follow are intended to illustrate and not
35 limit the scope of the invention.

ExamplesExperimental:

5 Melting temperatures and heats of fusion were determined using Differential Scanning Calorimetry (DSC) at a heating rate of 10°C/min. Heats of fusion (ΔH_f) are expressed in J/g.

10 Crystallization temperatures were also determined by DSC, at a cooling rate of 10°C/min.

10 Peel strength was measured by two different methods. One employed an Instron Materials Testing Instrument (IMTI), the other, an Instrumentors Slip/Peel Tester (ISPI).

15 IMTI Peel strength: Peel strength was measured as follows. A solution (50% solids) of adhesive formulation was prepared in tetrahydrofuran, cast onto a flexible PVC film, and air-dried at 70°C for 30 minutes. A 1" wide strip was covered with a 1" uncoated strip of the PVC film. The assembly was 20 lightly pressed together at a temperature of 37°C. Average peel strength was measured at 10"/min using an Instron materials testing instrument equipped with a variable temperature chamber.

25 ISPI Peel Strength: Peel strength was measured as follows. An Instrumentors, Inc. SP-102B Slip/Peel Tester was used according to the test procedure designated as PCTC-1 (revision 8/85) by the Pressure Sensitive Tape Council and was calibrated and operated according to the instruction manual 30 accompanying the instrument. In addition, sample preparation included: coating, generally, was cast from a heptane/methyl ethyl ketone solution (90:10) containing between 15% and 45% solute (adhesive). All coatings were dried in an oven for 18 to 24 hours @ 35 50°C. Dried coatings were cooled for 0.50 to 0.75 hours

at room temperature. Individual specimens were cut lengthwise (generally 0.5 in. x 5.0 in. strips) in the direction of extrusion (of the backing substrate). Each specimen was gauged for coating thickness to the nearest 0.1 mil. At least three (3) points on each specimen were measured and the average of these points were considered the thickness value. Each testing specimen did not have greater than 0.25 mil variation from the thinnest point to the thickest point and the variation between thickness values of all the testing specimens did not exceed 0.25 mils. The average of the specimen thickness values (to the nearest .001 inch) was the reported coating thickness. The peel test was conducted as soon as possible after measuring coating thicknesses; The testing substrate surface was a .012 inch thick polyurethane film of Tuftane 410 (Lord Corporation, Erie, PA) which was secured to the temperature controlled platen by means of double sided carpet tape. Each specimen remained on the testing surface for 5 minutes before testing at the prescribed temperature of testing. Ten specimens were tested at each specific temperature. Each specimen was rolled with a 4.5 pound rubber roller immediately after contact with testing surface. The roller traveled over the surface of the specimen at a rate of 12 inches/min in both directions. The average peel force & terminal surface temperature was recorded.

Tack values were measured by three different methods. One employs a Surface Texture Analyzer (STA); another, a Polyken™ Instrument (PKI) and the third, a Tack Rolling Ball Test (TRBT).

STA Tack Measurement: Tack was measured using a Surface Texture Analyzer (a Volland-Stevens-LFRA Texture Analyzer). The probe was allowed to contact the adhesive for 10 seconds and was then withdrawn at 0.2

mm/sec. Tack values reported herein are the average maximum reading.

PKI Tack Measurement: Tack was measured by the procedure designated D2979 by the American Society for Testing and Materials. In this test, which employs a Polyken™ instrument, values are given in grams of force required to remove the end of a stainless steel rod, 5.0 mm in diameter, from the surface of an adhesive coated assembly. Prior to the removal, the rod approaches the surface at a speed of 10 mm per second and makes contact with the adhesive for a prescribed period of time. The actual time of contact is variable and discretionary. We chose to use a half second dwell time. Testing was performed generally at room temperature. The units of measure are in grams·cm/sec.

TRBT Tack Measurement: Tack was measured by the Rolling Ball Test designated as D3121 by the American Society for Testing and Materials and PTSC-6 (revision 8/85) by the Pressure Sensitive Tape Council. The technique was ideally suited for accurately and reproducibly measuring tack over wide temperature ranges. This test uses a designated 21°30' incline, a 7/16 diameter steel ball and a temperature controlled level, hard and smooth plate upon which a layer of Scotch™ 3M brand double sided carpet tape (23-8049) is mounted. The test strips are placed on top of the tape with the adhesive side up, covered with release liner and rolled with a standard 4.5 lb rubber wheel. After the test strip comes into temperature equilibrium with the plate, the release liner is removed and the ball is released from the top of the inclined plane. The ball rolls down the plane, gathers momentum and encounters the horizontal test strip. At which point, the rolling ball decelerates (according to the degree of tack) travels a specific distance and stops. The average of

stopping distance measurements (between the end of the inclined plane and the center of the ball) are recorded. Five tests determine the average. In order to maintain a consistency of tack units, we have made the following empirical conversion from TRBT (cm of travel) to units of those produced by the Polyken™ test instrument.

$$\text{Tack(g}\cdot\text{cm/sec)} = k/\text{cm of travel} + 7.25 \text{ g}\cdot\text{cm/sec}$$
$$\text{where } k = 735 \text{ g}\cdot\text{cm}^2/\text{sec}$$

10

Measurement of tack temperature: A 1" X 1" test sample was bonded face up to a metal plate with double sided adhesive tape and the metal plate was placed in a temperature controlled oven and allowed to equilibrate at the selected temperature for 10 minutes. Tack was tested by lightly pressing a 1 cm diameter plastic rod onto the surface of the adhesive for 1 second and then removing. After testing at the lowest temperature, the oven temperature was increased by 2°C and test repeated. The tack temperature is defined as the minimum temperature at which a noticeable tack was first observed.

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Example 1

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A polymer was prepared by combining 10 g of hexadecyl acrylate, 2 g of ethyl acrylate, 15 ml of deoxygenated toluene, and 0.06 g of AIBN, and heating at 60°C under a nitrogen atmosphere for 12 hours. The resulting mass was extracted with ethanol and dried in vacuo to yield a rubbery mass. Thermal analysis showed that this material had a melting point of 34°C, a freezing point of 26°C, and a heat of fusion of 64 J/g. A sample of this material was heated to 70°C and pressed into a 0.001"-thick film. A sample of the film was placed onto the adhesive side of a commercial

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plastic-backed PSA tape and stored at 25°C. The resulting tape was nontacky to the touch and exhibited no tack or adhesion to paper at room temperature. When the tape was placed on the wrist of a human subject, however, it became tacky almost instantly and exhibited good adhesion. When removed from the skin and kept at room temperature, the tape quickly lost its tack and adhesive properties.

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Example 2

Five percent acrylic acid, 5% ethylacrylate-hexadecylacrylate copolymer (1 g) was mixed with 1 ml of toluene and 0.004 g XAMA2 (Virginia Chemicals, Portsmouth, VA) as cross-linking agent. The material was allowed to stand two days at 80°C at which point it was more viscous. More toluene was added to make the solution spreadable. The mixture was then spread onto clear PVC film, dried at 80°C for 1 hour, and allowed to cool. The composition displayed excellent adhesion to skin at body temperature and was easily removed with a cool, damp paper towel. This adhesive exhibited no tack at room temperature.

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Example 3

Sixteen grams of hexadecylacrylate, 3 grams of isodecylacrylate (Sartomer), 1 gram of acrylic acid, 0.100 gram of AIBN was combined with 30 ml of toluene, purged with nitrogen and reacted for 11 hours at 70°C. The resulting solution was precipitated into methanol and dried. Test samples were prepared by solution coating a 50% solids solution of the polymer in tetrahydrofuran onto a PVC backing and drying at 50°C. The dry thickness of the resulting adhesive was approximately 0.0025 cm. Crosslinked samples were

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prepared by addition of XAMA 2 to adhesive solution prior to coating and drying.

STA tack and IMTI adhesive strength were measured at 20°C and 39°C as shown in Table I.

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Table I

Cure (% w/w)	TACK(g/cm ²)		PEEL STRENGTH (g/cm)	
	(20°C)	(39°C)	(20°C)	(39°C)
0	0	>100	<4.5	7
0.75	0	12	<4.5	--
0.35	0	15	<4.5	21

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Example 3a

The following polymers were prepared using heptane as the solvent, a reduced amount of catalyst (AIBN) and longer reaction times. The symbols C16, C6 and AA, used in Table II below, refer to the monomers hexadecylacrylate, hexylacrylate and acrylic acid respectively.

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Table II

Sample	Monomer Content (%)			Mol. Weight Mw (K=1000)
	<u>C16</u>	<u>C6</u>	<u>AA</u>	
555	83.5	12.5	4	914 K
557	83.5	12.5	4	862 K
558	83.5	12.5	4	739 K
563	83.5	12.5	4	
639	84	13	3	401 K
999	25	72	3	999 K

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A DSC analysis of the first 5 samples revealed peak melting temperatures between about 29°C to 31.4°C. Of those samples, no sample began melting before about 27.5°C and all samples were totally disordered when the

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temperature reached 35°C. Enthalpy values (heats of fusion) ranged between 44 and 52 J/g. Sample 999 had no detectable first order transition above 5°C.

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Example 3b

A blend of polymers was used to make an adhesive composition, as follows:

Table III

10	<u>Ingredient or Sample Number</u>	<u>Parts by Weight (grams)</u>
	555	12.73
	557	13.26
	558	21.93
	563	9.02
	639	39.62
15	Lauric Acid	3.39

The adhesive properties of this blend are described in Example 3d.

Example 3c

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A dodecene polyolefin was prepared as follows: Into a dried, 2 liter reaction flask was added 100 ml of dry heptane under a blanket of dry nitrogen. Catalyst was prepared by adding 0.7 ml of $TiCl_4$ and 3 ml of $Al(C_2H_5)_3$, to the solvent. The system was stirred for
25 one hour at room temperature (a bath was used to afford a constant temperature) and purged with dry nitrogen to remove all traces of Oxygen. Next, 240 ml (180g) of dodecene was added to the flask in an anhydrous manner as practical. The polymerization reaction was run for 5
30 hours. At the end of this time, the reaction was quenched with ethanol. The polymer precipitated and the solution was decanted. The precipitate was washed with fresh alcohol and then dissolved in toluene at temperatures not exceeding 90° C. That solution was
35 again precipitated by adding an excess of ethanol. That

material was filtered and dried in a vacuum oven at room temperature. The final dried product weighed 89 grams. The viscosity of the polymer was measured in cyclohexane. The values of the inherent and reduced
5 specific viscosities were extrapolated to zero concentration and found to be 4.22 and 3.79 respectively. The intrinsic viscosity was taken as the average of these values. Therefore, $[\eta] = 4.01$ deciliters/g. The DSC analysis revealed that the
10 polymer sample exhibited a broad melting peak beginning at about 31° C and extending to about 52° C. Enthalpy (heat of fusion) value was 28 J/g.

Example 3d

15 The following experiment was run to demonstrate how the temperature of the adhesive coated article would affect the tack of various adhesives. A commercial 3M™ Steri Drape 1040 surgical incise drape was used as a control. The adhesive coating was measured to be .001
20 inch thick. The product of Example 3b was cast on a duPont Hytrel™ backing using a 90:10 mixture of heptane:methylethyl ketone as a solvent. The thickness of the dried adhesive coating was .00175 inch thick. In order to normalize and compare the tack values to the
25 standard 3M control, the values of tack were reduced by a value of 1/1.75. The product of Example 3c was cast on Mylar polyester film from a solution of heptane. The thickness of the dried adhesive coating was .00250 inch thick. In order to normalize and compare the tack
30 values with the standard 3M control, the values of tack were reduced by a factor of 1/2.5. Tack values were measured at various temperatures. The temperature of the samples were never brought above the temperature at which they were measured. Table IV below lists the tack
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values for the control and normalized tack values of the samples. (All values are reported in g·cm/sec units.)

Table IV

	<u>Temper- ature (°C)</u>	<u>3M Control</u>	<u>Example 3b</u>	<u>Example 3c</u>
5	23			14.66
	25	90.77		14.66
	26	123.92	21	14.66
	27	72.29	21	14.66
	28	52.62	26.63	15.25
10	29	71.16	211.26	29.07
	30	97.99	273.43	38.77
	31	75.31	297.71	
	32	57.94		59.05
	35		470.82	57
	40			84.04
	45			390.2

15 The data shows that the control is always tacky or slightly tacky at the temperature of the measurement and was not significantly effected by that temperature. The multitactic polyolefin of Example 3c was nontacky below 28°C, became slightly tacky when the temperature was

20 raised above 28°C and then became progressively more and more tacky as the temperature was raised to 45°C. In contrast, the blend in Example 3b was nontacky below 28°C and then became tacky above 28°C.

25 Nine grams of the adhesive material produced in example 3b was admixed with 1 gram of polymer sample 999 listed in example 3a in a 90:10 mixture of heptane:methylethylketone solvent. The solution was cast onto backing as described in example 3d. The final dried thickness was .001 inch. When tested for tack, a

30 value of 525 g · cm/sec was obtained.

Example 3e

An experiment was run to demonstrate the time decay of tackiness after the samples were heated to 70°C and

35 then quenched to 23°C. Fresh samples, prepared and

described in Examples 3b and 3c, were used for this study. Samples were conditioned at 70°C for 2 hours after mounting. All the tack values were measured at 23°C. Table V below lists the normalized tack values and the logarithm of the normalized tack values of the samples.

Table V

	Time (hours)	Example 3b	Log Example 3b	Example 3c	Log Example 3c
10	0.03	28.14	1.45	>400.00	2.60
	0.08	20.94	1.32	>400.00	2.60
	0.17	20.94	1.32	>400.00	2.60
	0.50	20.94	1.32	>400.00	2.60
	1.00	20.94	1.32	>400.00	2.60
15	2.00	20.94	1.32	>400.00	2.60
	5.00	20.94	1.32	>400.00	2.60
	8.00	20.94	1.32	172.84	2.24
	24.00	20.94	1.32	94.78	1.98
	48.00	20.94	1.32	90.40	1.96
	72.00	20.94	1.32	75.85	1.88
	120.00	20.94	1.32	55.78	1.75
20	126.00	20.94	1.32	53.07	1.72
	127.00	20.94	1.32	36.58	1.56
	144.00	20.94	1.32	32.30	1.51
	152.00	20.94	1.32	39.65	1.60
	169.00	20.94	1.32	33.98	1.53
	177.00	20.94	1.32	30.48	1.48
	192.00	20.94	1.32	40.12	1.60
	199.00	20.94	1.32	22.55	1.35
25	200.00	20.94	1.32	27.60	1.44
	216.00	20.94	1.32	22.90	1.36
	240.00	20.94	1.32	21.16	1.33
	247.00	20.94	1.32	15.60	1.19
	248.00	20.94	1.32	14.66	1.17

The data show that a high degree of tack is retained by sample 3c for several days. After 10 days, the 3c samples regain their non-tacky state. In contrast, the 3b samples regain their non-tacky state within .08 hours or approximately, 5 minutes.

Example 3f

An experiment was run to show that the adhesive compositions of the present invention will rapidly form good bonds to skin-like substrate (Tuftane 410), but only from a few degrees below skin temperature and above. In contrast, conventional PSA do not allow this temperature sensitive behavior. Peel strength measurements differentiate these behaviors. In the first set, the samples were never brought above the temperature at which they were affixed and measured. In a second set of experiments, a demonstration was achieved showing the reduction of bond strength after cooling the adhesive union (backing-adhesive-substrate) to a temperature several degrees below skin temperature (20°C). The adhesives were actually affixed to the substrate in the range of temperatures used in the first set of experiments before the unions were cooled to 20°C for 5 minutes. The peel strengths were then measured. Table VI lists the normalized peel strength values (grams/cm) for both sets of experiments including the commercial 3M™ Steri Drape 1040 surgical incise drape control. The terms "@ affix temp" and "@ 20°C" refer to the temperature of the union when the bond was made and the peel test was performed, respectively.

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Table VI

	<u>Affix Tempera- ature(C)</u>	<u>Sample 3b @ Affix Temp</u>	<u>Sample 3b @ 20°C</u>	<u>3M Control @ affix Temp</u>	<u>3M Control @ 20°C</u>
5	38.0		30.00		111.67
	37.1	79.28			
	37.0			87.36	
	36.0		28.73		107.74
	34.0	67.68			
10	33.8			89.55	
	33.0		11.34		107.85
	32.0	67.69		90.69	
	31.0		12.85		99.17
	30.2			89.88	
	30.1	52.95			
	28.0	28.45			
	27.9			100.16	
	26.0	0.00			
	25.7			101.81	
15	22.7	0.00		110.98	

The results show that peel values for the 3M controls are relatively flat across the temperature range used for affixation and removal. No significant loss in peel strength occurs in those unions when they were cooled after affixation. In fact, the values appear to increase. On the contrary, the 3b samples demonstrate a substantial discontinuity in peel strength value above and below 28°C transition point. In addition, the data reveal a significant reduction of peel strength when the unions are cooled to 20°C after being affixed at any temperature above the 28°C transition point.

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Example 4

Nineteen grams of pentadecylacrylate and 1 gram of acrylic acid were combined with 20 ml of heptane, 10 ml of ethylacetate and 0.070 grams of AIBN. Resulting solution was degassed and mixed at 70°C for 17 hours. Polymer was isolated and dried in vacuo.

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One gram of polymer and 0.003 grams of XAMA 2 was dissolved into 2 ml of 1:1 heptane:ethylacetate and coated onto a 0.005 cm thick polyurethane backing (2103 AE, Dow Chemical, Midland Michigan) and dried at 80°C for 15 minutes. The resulting tape was nontacky at room temperature but quickly adhered upon application to skin. During a 36 hour test on a human subject no disbondment of adhesive or film from skin was observed.

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Example 5

Two grams of a hexadecylacrylate, ethylacrylate, acrylic acid co-polymer (76.5:20:2.5) was combined with 4 ml of 1:1 ethylacrylate:heptane solution and 0.0043 grams of XAMA 2. The resulting solution was coated onto urethane film and dried at 110°C for 15 minutes. The obtained film was not tacky at 25°C but bonded readily to a human subject. Samples showed no loss in adhesion when subjected to washing. During a 36-hour test no delamination of adhesive was observed.

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Example 6

The polymer of Example 3 alone and with 0.3% XAMA 2 was coated onto urethane backing and dried at 110°C for 15 minutes. Both samples were nontacky at room temperature and quickly became tacky when contacted to human skin. Three 1" X 1" test samples were applied to a test subject for seven hours. At that time it was observed that the non-crosslinked samples had lifted off from the skin around the edges and left a tacky residue on the skin when removed. The crosslinked sample in contrast stayed well bonded and left no tacky residue on the skin upon removal.

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Example 7

5 A co-polymer of hexadecylacrylate, methylacrylate and acrylic acid (weight ratio 85:10:5) was prepared as in Example 3 above. A 1 gram sample was combined with
10 0.0024 grams of XAMA 2, 1 ml of heptane, 1 ml of ethyl acetate and coated onto a urethane backing and cured as in Example 6 above. Six 2.54 cm X 2.54 cm test samples were applied to a human test subject for 26 hours. Three of the samples were removed at ambient temperature
15 and three of the samples were chilled for 20 seconds with a cold Pepsi can prior to removal. All of the samples removed at ambient temperature resulted in visible amounts of skin being removed from the test subject. Two of the cooled samples showed no visible
20 skin removal, while the third showed a very small amount of skin removal.

Example 8

20 Forty-three grams of hexadecylacrylate, 6 grams of ethyl acrylate, 1 gram of acrylic acid and approximately 0.500 grams of AIBN were dissolved in 150 grams of ethyl acetate. The solution was purged with nitrogen and was heated at 45°C for 24 hours and then at 60°C for 2
25 hours. The viscous reaction product was precipitated into chilled ethanol as usual. After drying in a vacuum oven, the material was dissolved into heptane (30% solids). One weight percent of Zn stearate (based on polymer) was added to one half the solution.

30 The other half of the solution was used to cast film on a sheet of polyurethane (Tuftane). The resultant dried film was 0.00075 inches thick. Strips, 0.5 X 1.5 in² were cut from the sheet and mounted on the skin surface. They immediately were transformed from a non-tacky to an aggressively tacky strip. The strips
35 were worn for 5 hours before they were removed at a rate

of 1 cm/sec. A Chatillon DFG-2 Digital Gram Gauge was used to monitor the force needed for removal under normal conditions and after the sample was cooled with a cool damp cloth for several minutes. The uncooled
5 removal required a peak force of about 0.273 Kg while the chilled removal required a peak force of about 0.132. The ratio of work needed for removal between the two methods of removal was calculated. That calculation showed that 2.25 more energy was required to remove the
10 sample when it was at skin temperature than when it was chilled before removal.

The portion of the solution which contained the ionic cross-linking agent, Zn stearate, was cast on a similar backing as above. In this case, half of the
15 samples were chilled with an ice cold can of soda. The average force required to remove the skin temperature sample was 0.156 Kg while the force to remove the chilled sample was only 0.036 Kg.

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Example 9

The cross-linked adhesive coated urethane of Example 5 above was applied to a sample of Tuftane 410 polymer film (Lord Corporation, Erie, PA) and warmed to
25 35°C. Material bonded well at this temperature but exhibited no adhesion when cooled to 20°C.

Example 10

One gram of polyoctadecylmethacrylate acrylic acid co-polymer (97.5:2.5) was combined with 1 ml of
30 ethylacetate, and 0.0028 g of XAMA 2 and coated onto a urethane backing as in Example 6 above. Four 1" X 1" samples were applied to a human test subject. After 24 hours two of the samples were removed at room temperature and two were first cooled with a cold Pepsi
35 can and then removed. The two samples removed at

ambient temperature showed visible skin removal while the samples which were first cooled showed no visible skin removal.

5

Example 11

A polymer was prepared by polymerizing 4.25 g of hexadecyl acrylate, 4.24 g of tetradecyl acrylate 1.02 g of polyethyleneglycol monomethyl ether (DP=8) methacrylate and 0.50 g of acrylic acid in 20 ml of
10 toluene containing 0.033 g of AIBN at 60°C for 14 hours. A sample of the resultant polymer was combined with 0.25% XAMA 2 and coated onto a urethane backing as in Example 6 above.

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Example 12

Thirty-one grams of polytetrahydrofuran of M.W. 2900, (Scientific Polymer Products, Ontario, New York), 1.85 grams of hexamethylene diisocyanate, 1 drop of dibutyltin dilaurate and 200 ml of dry toluene were
20 combined and mixed for 24 hours and then 5 ml of ethanol was added with stirring. The resultant mixture was coated onto a glass slide, dried at 100°C for 1 hour and then allowed to cool overnight. The resulting film was
25 nontacky at 10°C but tacky at skin temperature yet showed no tendency to flow.

Example 13

One gram of polytetrahydrofuran (M.W. 2900) and 3.0 grams of the polymer solution of Example 12 were
30 combined and coated onto a glass microscope slide, dried at 100°C for 12 hours and allowed to cool.

Example 14

To 20 ml of toluene were added 8.5 grams of
35 hexadecylacrylate, 1.0 grams of polyethyleneglycol

monomethyl ether (DP=8) methacrylate, .5 grams of acrylic acid and .0667 grams of AIBN. The solution was purged with nitrogen and heated at 60°C for 14 hours. The product was precipitated into ethanol, filtered and dried under vacuum.

Test samples were prepared by solution casting. The solution contained 2 grams of the above polymer and enough ethyl acetate to bring the total weight up to 6 grams. To portions of this solution was added 0, 1, 2 and 3% of XAMA-2. Each solution was cast on polyvinyl chloride film using a 8.7 cm wide blade set at 3 mil. The films were air dried and then heated for 1 hour at 58°C. The width of the respective films were 8.1, 7.6, 7.3, 6.8 cm., indicating that crosslinking had occurred to various degrees.

Each of the samples became tacky above a temperature of 36°C. It was apparent that the order of tackiness decreased as the amount of XAMA-2 was increased.

Example 15

A polymer solution was prepared as in Example 13 with hexadecyl acrylate, ethyl acrylate and acrylic acid in an 80:15:5 ratio. Testing was carried out as described in Example 14.

Example 16

The tack temperature measured as above for the compositions of Examples 3, 4, 5, 7, 11, 14 and 15 are set forth in Table II:

Table VII

	<u>Composition</u>	<u>% XAMA 2 (w/w)</u>	<u>Tack temperature</u>
	Example 3	0	31
	Example 3	0.32	29
5	Example 4	0.3	31
	Example 5	0.22	29
	Example 7	0	33
	Example 7	0.30	33
	Example 11	0.25	27
	Example 14	--	33
10	Example 15	0.25	29

Example 17

A warm-activated pressure-sensitive adhesive assembly was prepared as follows. A solution containing (a) 26 parts by weight of polymer comprising 84% hexadecylacrylate, 13% hexylacrylate and 3% acrylic acid, and exhibiting a weight average molecular weight (in toluene) of 1.2×10^6 , (b) 66.6 parts by weight heptane, and (c) 7.4 parts by weight methyl ethyl ketone, was used to cast a film on a release liner backing of siliconized polypropylene paper. A #90 wire-wound coating rod (Paul N. Gardner Co.) was used such that the final wet coating was approximately 1.75 mils thick. After air drying, the film thickness was approximately 10 mils. The film was heated in an incubator (convection) oven for 18 hours at 50°C. The final dried adhesive film was approximately 2.5 mils thick and exhibited a melting temperature of 29°C.

The film assembly or the transfer device produced above was used to apply the hot melt film in the first of two operations. First, the device was placed with the adhesive face down on Whatman paper stock #4. The assembly was heated to about 35°C and rolled with a standard 4.5 lb roller. The adhesive film had formed a strong bond with, and had indeed

transferred to the paper during this process, as evidenced by the fact that the release liner could be separated from the adhesive film but the adhesive film was now inseparable from the paper.

5 Designs were drawn on the adhesive-backed paper while it was being conditioned at room temperature (approximately 25°C). The release liner was then removed. The adhesive coating appeared to be smooth and tack-free. Small shapes were cut from this paper and
10 positioned on the edges of another sheet of white bond paper to enhance its appearance.

The following step constitutes the second phase of the experiment. The cut-out designs could be repositioned easily because there was no tackiness at
15 all. When the arrangement was satisfactory, it was covered by a thin piece of cardboard and placed between two steel plates. The assembly was then placed in an oven maintained at 35°C for five minutes. The steel plate acted as a gentle press. The assembly was then
20 disassembled and the white bond paper with its designs were immediately examined in a room-temperature (approximately 24°C) environment.

It was apparent that the designs were permanently affixed to the bond paper. Even when the
25 paper was placed in the refrigerator, the designs remained in place and were stable.

Example 18

30 A "chill-off" label was prepared as follows. A solution containing (a) 25 parts by weight of a 40:60 mixture of polymers, the first polymer comprising 84% hexadecylacrylate, 13% hexylacrylate and 3% acrylic acid and exhibiting a weight average molecular weight (in
35 tetrahydrofuran) of 1.9×10^6 , the second polymer

comprising 74% hexadecylacrylate, 20% isodecylacrylate and 6% acrylic acid and exhibiting a weight average molecular weight (in toluene) of 50,000, and (b) containing 75 parts by weight of solvent composed of a 90:10 mixture of hexane:methylethylketone, was used to cast a film on a release liner backing of siliconized polypropylene paper. An appropriate wire-wound coating rod was used such that a final wet coating was approximately 1.25 mils thick. After air drying, the film was approximately 1.25 mils thick. The adhesive assembly resembled a normal PSA article. The film assembly was heated in an incubator (convection) oven for 18 hours at 50°C. The final dried adhesive film was 1.25 mils thick and exhibited a melting temperature of 21°C. The coating process and drying process was repeated to produce a final dried adhesive film thickness of 2.5 mils.

The transfer device produced above was placed with the adhesive face down on Whatman paper stock #4. The assembly was heated to about 40°C and rolled with a standard 4.5 lb roller to cause the adhesive polymer to flow into the interstices of the cellulosic fibers of the paper. The adhesive film had formed a strong bond with, and had indeed transferred to, the paper during this process, as suggested by the fact that the release liner could be separated from the adhesive film, but the adhesive film was now inseparable from the paper.

The assembly was cut into strips about 3/8" wide and 2" long. These strips and the substrates to which they were applied after removal of the release liner were conditioned at ambient temperature (approximately 24°C). The selected substrates were test tubes, synthetic leather, plastic and metal pens, painted-wood pencils and a small varnished article of wood.

The strips bonded well to each of the substrates. Attempts to remove them resulted in tearing of the paper label. This result demonstrates and substantiates the tamper-proof feature of this assembly.

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When the label-bonded substrates were placed in a cold chest maintained at 5°C for 15 minutes, the labels were easily removed without tearing the paper. The adhesive did not transfer to the substrate leaving the surface free of any marring.

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Claims

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1. A temperature-activated pressure-sensitive adhesive comprising a polymeric composition containing a polymer having a first-order melting transition which occurs over a temperature range narrower than about 15°C, the polymer being present in the composition in an amount effective to render the adhesive substantially nontacky at temperatures below the range and tacky at temperatures within or above the transition.
- 15
2. The adhesive of claim 1 wherein the polymer is a side-chain crystallizable polymer.
- 20
3. The adhesive of claim 2 wherein the polymer is selected from the group consisting of polyolefins of singular tacticity, alkyl acrylates and alkyl methacrylates.
- 25
4. An adhesive assembly comprising a body member coated with the polymeric composition of claim 1.
- 30
5. The adhesive assembly of claim 4 in the form of a tape.
- 35
6. The adhesive assembly of claim 4 in the form of a label.
7. A pre-positionable, temperature-activated adhesive assembly for application to the skin, comprising a body member having a surface coated with a polymeric adhesive composition which is substantially nontacky at or below about 25°C and which becomes tacky upon contact with the skin.

8. A temperature-activated adhesive assembly for application to the skin which is readily removable therefrom, comprising a body member having a surface coated with a polymeric adhesive composition which has a
5 freezing point in the range of about 10°C to 28°C.

9. A pre-positionable temperature-activated adhesive assembly for application to the skin which is readily removable therefrom, comprising a body member
10 having a surface coated with a polymeric adhesive composition which has a melting point or first-order transition temperature between about 20°C and 35°C and a freezing point between about 10° and 28°C.

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INTERNATIONAL SEARCH REPORT

International Application No PCT/US90/02223

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) ³		
According to International Patent Classification (IPC) or to both National Classification and IPC		
IPC(5): B32B 7/12		
U.S.C.I.: 428/355,343; 424/448		
II. FIELDS SEARCHED		
Minimum Documentation Searched ⁴		
Classification System	Classification Symbols	
U.S.	428/343,355; 424/448	
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched ⁵		
III. DOCUMENTS CONSIDERED TO BE RELEVANT ¹⁴		
Category *	Citation of Document, ¹⁶ with indication, where appropriate, of the relevant passages ¹⁷	Relevant to Claim No. ¹⁸
Y,E	US, A, 4,925,908 (BERNARD) 15 MAY 1990 (see col. 1, lines 48 et seq; col. 2, lines 24 et seq)	1-6
A	US, A, 4,675,009 (HYMES) 06 JUNE 1987.	1-9
Y,P	US, A, 4,880,683 (STOW) 14 NOVEMBER 1989.	1-6
Y	US, A, 4,199,646 (HORI) 22 APRIL 1980 see abstract.	1-6
Y	US, A, 4,323,557 (ROSSO) 06 APRIL 1982 see col. 3, lines 63 et seq.	7-9
<p>* Special categories of cited documents: ¹⁵</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&" document member of the same patent family</p>		
IV. CERTIFICATION		
Date of the Actual Completion of the International Search ²		Date of Mailing of this International Search Report ²
04 JUNE 1990		29 AUG 1990
International Searching Authority ¹		Signature of Authorized Officer ²
ISA/US		<i>Leon R. Horne</i> LEON R. HORNE