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(54) Titre : FILM HAUTE BARRIERE  
(54) Title: HIGH BARRIER FILM

(57) **Abrégé/Abstract:**

An oriented film combination of (a) a base layer of polypropylene at least one surface of which comprises a maleic acid anhydride modified polypropylene; and (b) on at least one surface, a skin layer of an amorphous polyamide or a blend of an amorphous polyamide and a semicrystalline polyamide.



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ABSTRACT

An oriented film combination of (a) a base layer of polypropylene at least one surface of which comprises a maleic acid anhydride modified polypropylene; and (b) on at least one surface, a skin layer of an amorphous polyamide or a blend of an amorphous polyamide and a semicrystalline polyamide.

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HIGH BARRIER FILM

This invention relates to a high barrier film. More particularly, the invention relates to a multi-layer film structure that may have a metallized surface.

The bonding of metals, such as, aluminum, silver, chromium, etc., to plastic films and the gas barrier developed, has allowed such films to replace metallic foils in many instances. The flexibility of the films necessitates the formation of a strong metal/plastic bond and a number of approaches have been developed for providing such bonding. In some cases a special primer layer must be applied to the base layer in order to achieve an effective bond between the metal and the substrate. In many cases a thermoplastic surface must be corona discharge treated in order for there to be an effective bond between the metal and the thermoplastic surface. Gas barrier properties will also depend upon the condition of the surface on which the metal is deposited.

It is an object of the present invention to present a metallized film having a far greater metal adhesion level than that obtainable by corona discharge treating or priming of a substrate layer. It is a further object of the invention to provide a metallized film which exhibits excellent bond strengths in both adhesive and extrusion lamination and a significantly improved oxygen gas barrier.

According to the present invention there is provided an oriented multi-layer film combination comprising;

- (a) a base layer of polypropylene homopolymer or copolymer at least one surface thereof comprising a maleic acid anhydride modified polypropylene homopolymer or copolymer; and
- (b) a skin layer on the or each surface, said skin layer comprising a member selected from the group consisting of an amorphous polyamide and a blend of an amorphous polyamide and a semicrystalline polyamide.

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This film combination presents a polyamide surface to which can be strongly bonded a metallized layer.

Preferably the or each surface comprises a member selected from the group consisting of:

- a) said maleic acid anhydride modified polypropylene homopolymer or copolymer intermixed with said polypropylene homopolymer or copolymer; and
- b) a separately applied skin layer comprising said maleic anhydride modified polypropylene homopolymer or copolymer on said substrate.

Desirably the skin layer is an amorphous polyamide; or the skin layer is a blend of an amorphous polyamide and a semicrystalline polyamide.

Preferably the base layer has a heat sealable layer on the surface opposite the surface carrying said polyamide skin layer. The heat sealable layer is preferably a member selected from the group consisting of an ethylene-propylene random copolymer and an ethylene-propylene-butene-1 terpolymer.

The polyamide surface may be metallized, preferably with aluminium.

In one embodiment the metallized film is laminated with another polyolefin film in contact with the metal layer.

The substrates contemplated herein include the homopolymers and copolymers of polypropylene; particularly preferred are isotactic polypropylenes containing at least 80% by weight of isotactic polypropylene. This material has a density ranging from about 0.90 to 0.91, a crystalline melting point of about 160-165°C, and a melt index of about 2-6, preferably 3-4, grams/10 minutes. This material is about 80-100% isotactic, and preferably 95-96% isotactic. Commercially available materials of this description are "Exxon 4252"\* and ARCO W472.\*\*

\* Trademark

\*\* Trademark



While the polyamide of the present invention can be applied directly to the surface of the polypropylene or the polypropylene copolymer, it is preferred that the surface be modified by a particular functionalized material. This can be mixed into or applied on top of the polypropylene substrate, then the polyamide will aggressively adhere to this so-modified material. If a maleic anhydride modified polypropylene homopolymer or copolymer becomes a part of the substrate surface either by being interblended therewith or by being extruded thereon as a surface skin layer, this then becomes an ideal surface for receiving the polyamide layer of the present invention. This modified surface layer or skin layer itself cannot be separated from the propylene homopolymer or copolymer substrate. By a "maleic anhydride modified polypropylene homopolymer or copolymer" is meant the product resulting from the reaction between maleic anhydride and the thermal degradation product of polypropylene or polypropylene copolymer. Examples of this material can be found disclosed in US-A-3480580; particular attention is directed to examples 3, 4 and 6 of this specification.

The maleic anhydride modified polypropylene or a copolymer thereof can be prepared by a fairly simple process. For example, about 125 grams of low viscosity polypropylene of 3300 centipoise at 190°C and 7.5 grams of maleic anhydride are placed in a 500 ml glass round-bottom flask. This flask has three necks and is fitted with a metal sweeper, a dropping funnel and a steam-jacketed condenser to return maleic anhydride which boils out of the reaction mixture. The flask is lowered into a Wood's metal bath at a temperature of about 200°C. After about 1 hour the product is poured from the flask onto a metal tray. The waxy product will have a melt viscosity of 3600 centipoise at 190°C and a saponification number of 16.

Low viscosity polymers can also be produced in the above manner from low molecular weight 98-96/2-4 propylene/ethylene copolymers, 50/50 propylene/butene-1 copolymer, 95/5 propylene/pentene-1 copolymer, 90/10 propylene/hexene-1 copolymer, 80/20 propylene/4-methyl-1-pentene copolymer, 75/25 propylene, dodecene-1 copolymer and 90/10

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butene-1/hexene copolymer, in general, propylene copolymerized with another alpha olefin, including ethylene.

The procedure for the use of a maleic anhydride modified polypropylene homopolymer or copolymer, when this material is interblended into the base layer is exemplified by the following technique. A maleic anhydride modified polypropylene, e.g., "Epolene E-43"\* , a maleic anhydride modified polypropylene obtained from Eastman Kodak Company, melt-blended with a polypropylene homopolymer, such as "Exxon 4252"\*(85 parts) to give a 15% master batch. 20 parts of the master batch is dry blended with 80 parts of the polypropylene copolymer to give a resin blend containing 3% of the maleic anhydride modified polypropylene. This blend will contain sufficient maleic anhydride-modified polypropylene at the surface of the substrate to aggressively bond the polyamide thereto. The alternative method of making available the maleic anhydride-modified polypropylene homopolymer or copolymer at the surface of the substrate is to coextrude the base polymer with a skin layer containing the maleic anhydride-modified polymer.

The contemplated polyamides broadly include the amorphous polyamides and the semicrystalline polyamides. The amorphous polyamides are described as those polyamides which are lacking in crystallinity as shown by the lack of endotherm crystalline melting peak in a Differential Scanning Calorimeter measurement (ASTM D 3417), 10°C./minute. Examples of the amorphous polyamides that can be used include those amorphous polymers prepared from the following diamines: hexamethylenediamine, 2-methylpentamethylenediamine, 2,2,4-trimethylhexamethylenediamine, 2,4,4-trimethylhexamethylenediamine, bis(4-aminocyclohexyl)methane, 2,2-bis(4-aminocyclohexyl)isopropylidene, 1,4-diaminocyclohexane, 1,3-diaminocyclohexane, meta-xylylenediamine, 1,5-diaminopentane, 1,4-diaminobutane, 1,3-diaminopropane, 2-ethyldiaminobutane, 1,4-diaminomethylcyclohexane, p-xylylenediamine, m-phenylenediamine, p-phenylenediamine, and alkyl substituted m-phenylenediamine and p-phenylenediamine.

\* Trademark (each instance).



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Examples of polyamides that can be used include those amorphous polymers prepared from the following dicarboxylic acids: isophthalic acid, terephthalic acid, alkyl substituted iso- and terephthalic acid, adipic acid, sebacic acid, butane dicarboxylic acid, and the like.

Specific examples of amorphous polyamides which are suitable for this invention include: hexamethylenediamine isophthalamide, hexamethylenediamine isophthalamide/ terephthalamide terpolymer, having iso/terephthalic moiety ratios of 100/0 to 60/40, mixtures of 2,2,4- and 2,4,4-trimethylhexamethylenediamine terephthalamide, copolymers of hexamethylene diamine and 2-methylpentamethylenediamine with iso- or terephthalic acids, or mixtures of these acids. Polyamides based on hexamethylenediamine iso/terephthalamide containing high levels of terephthalic acid moiety may also be useful provided a second diamine such as 2-methyldiaminopentane is incorporated to produce a processible amorphous polymer.

The above amorphous polyamides may contain as comonomers minor amounts of lactam species such as caprolactam or lauryllactam, even though polymers based on these monomers alone are not amorphous. The important feature is that the polyamide as a whole must be amorphous. Thus small amounts of these comonomers may be incorporated as long as they do not impart crystallinity to the polyamide. In addition, up to about 10 weight % of a liquid or solid plasticizer such as glycerol, sorbitol, or toluenesulfonamide ("Santicizer 8"\* from Monsanto) may be included with the amorphous polyamide.

For most applications the T<sub>g</sub> is likewise not clearly demarked. But amorphous polyamides with T<sub>g</sub> above about 160°C are not readily thermoformable when used as a barrier layer. Thus all-aromatic polyamides, having aromatic groups in both acid and amine moieties, tend to have a T<sub>g</sub> which is too high to permit thermoforming, and are thus normally unsuitable for the purposes of this invention.

As indicated above, the amorphous polyamide of the present invention can be blended with at least one semicrystalline polyamide.

\* Trademark

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This term refers to the traditional semicrystalline nylons which are generally prepared from lactams or amino acids, such as nylon 6 or nylon 11, or from condensation of diamines such as hexamethylene diamine with dibasic acids, such as succinic, adipic, or sebacic acids. Copolymers and terpolymers of these polyamides are also included, such as copolymers of hexamethylenediamine/adipic acid with caprolactam (nylon 6,66). Blends of two or more crystalline polyamides can also be used. The polyamide of the present invention, both semicrystalline and amorphous, are prepared by condensation polymerization which is well known to those skilled in the art.

As indicated above the polyamide skin comprises the surface for receipt of a vacuum metallization. A further advantage of employing such a polyamide is that it is not necessary to corona discharge treat the polyamide surface layer prior to metallization because it inherently has a wetting tension of 42 dynes/cm. Higher wetting tensions are, however, obtainable by employing corona or flame treatment processes. The inherent polar nature of the polyamide skin results in metal adhesion levels far exceeding those obtainable with a treated polypropylene homopolymer skin. The metallized OPP film also exhibits excellent bond strength in both adhesive and extrusion lamination. Further desirable properties of the polyamide/polypropylene base film for metallization, are that it displays an excellent oxygen gas barrier when metallized as well as optical clarity and metal uniformity. It also retains the other desirable properties of metallized oriented polypropylene film, such as, a good water vapor barrier.

It is to be understood that opposite the polyamide surface, the multi-layer film of the present invention can have a heat-seal layer applied thereto. Preferred heat-seal layers include an ethylene-propylene random copolymer, and an ethylene-propylene-butene-1 terpolymer, etc.



EXAMPLE

A three layer sheet was produced by coextrusion, employing the following materials:

A-layer duPont amorphous polyamide obtained from the condensation reaction of hexamethylenediamine and isophthalic anhydride;

B-layer 50% polypropylene homopolymer-50% Mitsui QF500 (maleic anhydride modified polypropylene); and

C-layer ethylene-propylene-butene-1 terpolymer containing 700 ppm erucamide antiblock.

The coextrudate was quenched at 150°F on a cast roll, reheated on rolls to 270°F and stretched 3.5 times its original length in the machine direction. The sheet was subsequently stretched eight times its original width in a tenter apparatus at 310-330°F and corona treated on the polyamide side before being wound on a roll. Line speed was 50 fpm. The overall film thickness was 80 gauge. The amorphous polyamide layer was approximately 10 gauge units and the terpolymer layer was approximately 4 gauge units. The off-line wetting tension of the polyamide side of the film was > 50 dynes/cm. The film was vacuum metallized with aluminum on the polyamide skin to an optical density 2.5 employing conventional vacuum metallization techniques. The water vapour transmission rate of the resulting metallized film was 0.02 g/100 in<sup>2</sup>/24hr at 100°F, 90% relative humidity. The oxygen transmission rate of the metallized film was 1.0cc/100 in<sup>2</sup>/24hr.

The metallized film was subsequently extrusion laminated to a low density polyethylene film. The metallized side of the film was buried in the lamination. The metallized film to low density polyethylene bond strengths were greater than 100g/in with no aluminum transfer from the metallized film to the low density polyethylene film on bond-pulls.

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Claims:

1. An oriented multi-layer film combination comprising;
  - (a) a base layer of polypropylene homopolymer or copolymer at least one surface thereof comprising a maleic acid anhydride modified polypropylene homopolymer or copolymer;
  - (b) a skin layer on the said at least one surface, said skin layer comprising a member selected from the group consisting of an amorphous polyamide and a blend of an amorphous polyamide and a semicrystalline polyamide; and
  - (c) a vacuum metallized layer on said polyamide skin layer.
2. A film combination according to claim 1, wherein said at least one surface comprises a member selected from the group consisting of:
  - a) said maleic acid anhydride modified polypropylene homopolymer or copolymer intermixed with said polypropylene homopolymer or copolymer; and
  - b) a separately applied skin layer comprising said maleic anhydride modified polypropylene homopolymer or copolymer on said substrate.
3. A film combination according to claim 1, wherein said skin layer is an amorphous polyamide.
4. A film combination according to claim 1, wherein said skin layer is a blend of an amorphous polyamide and a semicrystalline polyamide.
5. A film combination according to claim 1, wherein said base layer has a heat sealable layer on the surface opposite the surface carrying said polyamide skin layer.
6. A film combination according to claim 5, wherein said heat sealable layer is a member selected from the group consisting of an ethylene-propylene random copolymer and an ethylene-propylene-butene-1 terpolymer.



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7. A film combination according to claim 1, wherein the surface is metallized with aluminum.
8. A film combination according to claim 7, wherein the metallized film is laminated with another polyolefin film in contact with the metal layer.