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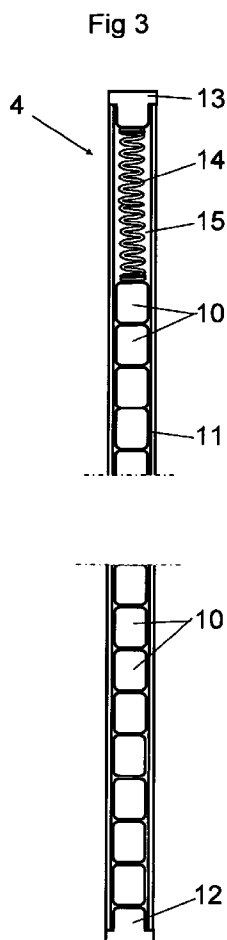
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[Continued on next page]

(54) Title: NUCLEAR FUEL, NUCLEAR FUEL ELEMENT, NUCLEAR FUEL ASSEMBLY AND A METHOD MANUFACTURING NUCLEAR FUEL



(57) Abstract: The invention refers to nuclear fuel (10), a fuel element (4), a fuel assembly (1) and a method of manufacturing nuclear fuel. The nuclear fuel is adapted for use in a water cooled nuclear reactor, including light water reactors LWR, such as Boiling Water Reactors BWR and Pressure Water Reactors PWR. The nuclear fuel comprises an uranium- containing compound consisting of UN. The uranium content of the uranium- containing compound comprises less than 10% by weight of the isotope ²³⁵U. The nuclear fuel comprises an additive substantially consisting of at least one element, in elementary form or as a compound, selected from the group consisting of Zr, Mo, Si, Al, Nb and U.

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NUCLEAR FUEL, NUCLEAR FUEL ELEMENT, NUCLEAR FUEL ASSEMBLY AND A METHOD
MANUFACTURING NUCLEAR FUEL

BACKGROUND OF THE INVENTION AND PRIOR ART

5

The present invention refers to a nuclear fuel pellet adapted for use in a water cooled nuclear reactor, including light water reactors LWR, such as Boiling Water Reactors BWR and Pressure Water Reactors PWR, the nuclear fuel pellet comprising an uranium-containing compound consisting of UN, wherein the uranium content of the uranium-containing compound comprises less than 10% by weight of the isotope ^{235}U .

15

The present invention also refers to a fuel element, a fuel assembly and a method of manufacturing a nuclear fuel adapted for use in a water cooled nuclear reactor, including light water reactors LWR, such as Boiling Water Reactors BWR and Pressure Water Reactors PWR.

20

In water-cooled nuclear reactors, including Light Water Reactors, LWR, and Heavy Water Reactors, HWR, a nuclear fuel including UO_2 is usually used. UO_2 is advantageous due to the fact that it has a high resistance to dissolution in water.

25

JP-11202072 discloses a nuclear fuel of the kind initially defined. The nuclear fuel is intended for use in a water-cooled reactor. The nuclear fuel comprises particles of oxides or nitrides and is contained in a fuel pellet. The fuel is enclosed in a protecting film or cover of, for instance, aluminium oxide, graphite, silicon carbide or a metal. The purpose of the protecting film is to prevent water from penetrating the fuel pellet and reach uranium nitride.

30

35

WO2007/011382 discloses a nuclear fuel of modified UN or modified PuN. The nuclear fuel has an additive of a further nitride, such as at least one of zirconium nitride, thorium nitride, haf-

5 nium nitride, titanium nitride, or rare earth nitrides or other acti-
nide nitrides. The nuclear fuel is not adapted for a water-cooled
reactor, but for a particular kind of reactor named Small Sealed
Transportable Autonomous Reactor, SSTAR, which is a breeder
reactor, and thus the nuclear fuel of UN has a very high content
of the isotope ^{235}U .

10 US-4,059,539 discloses a nuclear fuel consisting of (U,Zr)N
where ZrN is dissolved in a matrix of UN. The fuel is adapted for
a breeder reactor.

SUMMARY OF THE INVENTION

15 It is thus known to use a nuclear fuel based on uranium nitride
in breeder reactors, which are not water-cooled. UN has techni-
cal and economical advantages in relation to UO_2 . The object of
the present invention is therefore to provide a nuclear fuel
adapted for water-cooled reactors and based on UN.

20 This object is achieved by the nuclear fuel initially defined,
which is characterised in that the nuclear fuel comprises an ad-
ditive including at least one element, in elementary form or as a
compound form, selected from the group consisting of Zr, Mo,
Si, Al, Nb and U.

25 Pure UN contains approximately 40% more uranium atoms than
 UO_2 . A nuclear fuel based on UN will therefore result in a sig-
nificant improvement of the operating costs, and thus the costs
of generating electricity. Furthermore, UN has a higher thermal
30 conductivity, which, in addition and in contrast to UO_2 , increases
with the temperature. The thermal conductivity is approximately
3 – 8 higher for UN than for UO_2 depending on the temperature.
Consequently, a nuclear fuel based on UN will not be heated to
the same extent as UO_2 during operation of the nuclear reactor,
35 which is advantageous for several reasons, for instance less
thermal expansion, less release of fission gases and less stored

energy, the latter advantage being important in case of a LOCA, Loss Of Coolant Accident.

UN is disadvantageous in comparison with UO_2 since UN is more reactive with water than UO_2 . This is a potential limitation for use in water-cooled reactors, e.g., an LWR, where a leak in the fuel cladding cannot be excluded. While UO_2 reacts slowly with water at LWR conditions (250°C to 350°C), the reaction rate of UN is such that the gases produced expand and rupture the cladding. Consequently, it has up to now not been possible to use UN in water cooled reactors, where there is a risk for water penetrating the fuel cladding and contacting the nuclear fuel, see the article *XPS and XRD studies of corrosion of uranium nitride by water* of S. Sunder and N. H. Miller in *Journal of Alloys and Compounds*, pages 271-273 (1998). The writers conclude that UN can not be used in water cooled reactors.

By introducing an additive according to the present invention, the reaction rate of UN with water can be reduced to an acceptable level. UN with the defined additive or additives will be stable also in an environment containing water. The additives to be added to the uranium-containing compound will react with water to form a tight, water insoluble, protective layer over the UN content on all surfaces, including crack surfaces.

The defined additives meet the following important criteria. They do not react with the commonly used cladding made of a zirconium-based alloy, such as Zircaloy-2 and Zircaloy-4. They all have a relatively low neutron absorption cross-section. The preferred and here exemplified additives all have a neutron cross-section in the same range as Zr. Moreover, the additives are all stable in radioactive environments. The additives can be added in elementary form or as a compound, such as an oxide, a nitride, a hydride etc. They may also be present in the finished nuclear fuel in elementary form or as a compound, for instance an oxide, a nitride, a hydride etc., such as ZrN , Si_3N_4 , Al_2O_3 ,

ZrO₂, Mo_yO_x, SiO₂, AlN, etc. The additives will be accumulated at the grain boundaries, and they prevent penetration of water to the UN on all exposed surfaces, including crack surfaces.

- 5 The additives defined have very low corrosion rates in water. Zr in the form of ZrN has been shown in previous work to be effective in protecting PuN fuels from water at the 70 atom % level. Additives of Mo metal to U metal have been shown to eliminate U metal corrosion at the 5 to 19 volume % level.
- 10 Therefore, it is believed that the addition of nitride, oxide, or hydride compounds of Zr, Al, Mo, Si and U to UN should protect the majority of the UN in the fuel matrix. With regard to an additive comprising U or an uranium compound to protect UN, it is to be noted that, for instance, if U metal is added to UN, then upon
- 15 exposure to water, the U metal will oxidize to UO₂ which will protect the underlying UN.

According to an embodiment of the invention, the nitrogen content of the uranium-containing compound comprises at least 60% by weight of the isotope ¹⁵N, preferably at least 80% by

20 weight of the isotope ¹⁵N, and most preferably at least 90% by weight of the isotope ¹⁵N. Nitrogen is naturally present as 99,634% ¹⁴N (stable with 7 neutrons) and 0,366% ¹⁵N (stable with 8 neutrons). ¹⁴N has a high absorption cross-section. By enriching natural nitrogen in ¹⁵N, parasitic absorption of neutrons by ¹⁴N can be prevented or minimised.

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According to an embodiment of the invention, the additive includes at least one of ZrN, ZrH₂, Si₃N₄, Al₂O₃, ZrO₂, Mo_yO_x, SiO₂, AlN, ZrO₂, ZrH₃, SiO₂, U, Zr, Mo, Si, U₃Si₂, ZrUAl, ZrUSi,

30 ZrUH, UAl₂, U₃Si, U-5Nb-5Zr, U-3Nb-1.5Zr, U-9Mo, U-6Mo, U-1.5Mo-1.0Zr, U-10Zr, and U₃SiAl.

According to an embodiment of the invention, the additive includes at least one of ZrN, ZrH₂, Si₃N₄, Al₂O₃, ZrO₂, Mo_yO_x,

35 SiO₂, AlN, ZrO₂, ZrH₃, SiO₂, Zr, Mo, and Si, wherein the amount

of the additive is equal to or less than 30% by volume of the nuclear fuel. Consequently, in order to maintain high volumetric uranium densities, the amount of non-uranium containing additives should be less than 30% by volume. At this level, the overall uranium density will be higher than for UO_2 , i.e. good uranium volumetric densities will be maintained in the nuclear fuel.

According to an embodiment of the invention, the additive includes at least one of U, U_3Si_2 , ZrUAl , ZrUSi , ZrUH , UAl_2 , U_3Si , U-5Nb-5Zr , U-3Nb-1.5Zr , U-9Mo , U-6Mo , U-1.5Mo-1.0Zr , U-10Zr , and U_3SiAl , wherein the amount of the additive is equal to or less than 80% by volume of the nuclear fuel. Consequently, in order to maintain high volumetric uranium densities, the amount of uranium containing additives could be up to approximately 80% by volume. At this level, the overall uranium density will be higher than for UO_2 , i.e. good uranium volumetric densities will be maintained in the nuclear fuel.

According to an embodiment of the invention, the nuclear fuel is provided in the form of a nuclear fuel pellet. Advantageously, the nuclear fuel pellet may be formed through sintering of a powder of the uranium-containing compound and said at least one additive.

The object is also achieved by the fuel element defined in claim 10, and by the fuel assembly defined in claim 12.

Furthermore, the object is achieved by the method initially defined, comprising the steps of:
providing an uranium-containing compound consisting of UN, wherein the uranium content of the uranium-containing compound comprises less than 10% by weight of the isotope ^{235}U ,
adding to the uranium-containing compound an additive consisting of, or substantially consisting of, at least one element, in elementary form or as a compound, selected from the group consisting of Zr, Mo, Si, Al, Nb and U.

According to an embodiment, the method further comprises the steps of: providing a powder of the uranium-containing compound, providing a powder of the additive, mixing the uranium-containing compound and the additive to a powder mixture, and
5 sintering the mixture at sintering pressure and a sintering temperature to a nuclear fuel pellet. Advantageously, the sintering temperature may be at least 1800°C, at least 2000 °C, preferably at least 2100 °C, most preferably at least 2200 °C.

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BRIEF DESCRIPTION OF THE DRAWINGS

The present invention is now to be explained more closely by means of a description of various embodiments and examples, and with reference to the drawing attached hereto.

15

Fig. 1 discloses schematically a side view partly in section of a fuel assembly for a BWR.

Fig. 2 discloses schematically a side view of a fuel assembly for a PWR.

20 Fig. 3 discloses longitudinal sectional view of a fuel element of the fuel assembly in Fig 1 or 2.

DETAILED DESCRIPTION OF VARIOUS EMBODIMENTS AND EXAMPLES OF THE INVENTION

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Fig. 1 shows a fuel assembly 1 for use in water cooled light water reactors, LWR, and more precisely a Boiling Water Reactor, BWR. The fuel assembly 1 comprises known parts including bottom member 2, a top member 3 and a plurality of fuel elements in the form of elongated fuel rods 4 extending between the bottom member 2 and the top member 3. The fuel rods 4 are maintained in their positions by means of a plurality of spacers 5, one of which is shown in Fig. 1. Furthermore, the fuel assembly comprises a flow channel 6, or fuel box, surrounding and enclosing
35 the fuel rods 4.

Fig. 2 shows a fuel assembly 1 for use in water cooled light water reactors, LWR, and more precisely a Pressure Water Reactor, PWR. The fuel assembly 1 comprises known parts including
5 bottom member 2, a top member 3 and a plurality of fuel elements in the form of elongated fuel rods 4 extending between the bottom member 2 and the top member 3. The fuel rods 4 are maintained in their positions by means of a plurality of spacers 5.

10

Fig. 3 shows a single fuel element designed as a fuel rod 4 of the kind used in the fuel assemblies 1 of Figs 1 and 2. The fuel rod 4 comprises a nuclear fuel in the form of a plurality of fuel pellets 10 and a cladding in the form of a cladding tube 11, a
15 bottom plug 12 and a top plug 13. The fuel pellets 10 are arranged in a pile provided in the cladding tube 11. The cladding tube 11 thus encloses the fuel pellets 10 and a gas. A spring 14 is arranged in an upper plenum 15 and presses the fuel pellets towards the bottom plug 12.

20

The nuclear fuel of the fuel elements described above comprises an uranium-containing compound consisting of UN. The uranium content of the uranium-containing compound comprises at least the isotopes ^{238}U and ^{235}U . The uranium content is enriched with
25 respect to ^{235}U in relation to the natural composition of uranium, but the uranium content is less than 10, 9, 8, 7, 6 or 5% by weight of the isotope ^{235}U .

The nuclear fuel comprises, in addition to uranium and nitrogen,
30 an additive. The purpose of the additive is primarily to reduce the reaction rate of UN with water. The additive consists of, or substantially consists of, at least one element, in elementary form or as a compound, selected from the group consisting of Zr, Mo, Si, Al, Nb and U. The element(s) or compound(s) forming the additive is homogeneously distributed in the fuel element.
35

Natural nitrogen is composed of 99,634% ^{14}N (stable with 7 neutrons) and 0,366% ^{15}N (stable with 8 neutrons). ^{15}N has a significantly lower neutron absorption cross section than ^{14}N , which has a relatively high absorption cross-section. In order to minimise or reduce parasitic absorption of neutrons, the nitrogen content of the uranium-containing compound is therefore enriched with respect to ^{15}N . The nitrogen content of the uranium-containing compound may thus comprise at least 60, at least 70, at least 80, at least 90, at least 95 or at least 98% by weight of the isotope ^{15}N .

The additive may thus comprise or consist of one or several of the above mentioned elements in elementary form or as a compound. The compounds may for instance be oxides, nitrides, hydrides etc. Examples of suitable additives includes at least one of ZrN , ZrH_2 , Si_3N_4 , Al_2O_3 , ZrO_2 , Mo_yO_x , SiO_2 , AlN , ZrO_2 , ZrH_3 , SiO_2 , U , Zr , Mo , Si , U_3Si_2 , ZrUAl , ZrUSi , ZrUH , UAl_2 , U_3Si , U-5Nb-5Zr , U-3Nb-1.5Zr , U-9Mo , U-6Mo , U-1.5Mo-1.0Zr , U-10Zr , and U_3SiAl .

It is to be noted that the additive may comprise or consist of a single one of any of these elements or compounds. The additive may also comprise or consist of any combination of two or more of any of these elements or compounds.

A first group of these additives include at least one of the elements Zr , Mo and Si , and/or at least one of the compounds ZrN , ZrH_2 , Si_3N_4 , Al_2O_3 , ZrO_2 , Mo_yO_x , SiO_2 , AlN , ZrO_2 , ZrH_3 and SiO_2 .

In this first group the elements or compounds do not contain uranium or any other fissionable element, which limits the amount of the additive for maintaining a high volumetric uranium density in the nuclear fuel. Consequently, the amount of the additive for the first group should be equal to or less than 30% by volume of the nuclear fuel. Advantageously, the amount of the

additive for this group may be equal to or less than 25, 20, 15, or 10 % by volume of the nuclear fuel. The amount of the additive for this group is equal to or more than 2, 5, 7 or 10 % by volume of the nuclear fuel. The additives of the first group will
5 react with water to form a tight, water insoluble, protective layer over the UN content.

The amount of the additive, or more precisely the percentage by weight of the additive, may be smaller for the additive in elementary form than for the additive in the form of a compound.
10

A second group of the additives include the element U and/or at least one of the compounds U_3Si_2 , ZrUAl, ZrUSi, ZrUH, UAl_2 , U_3Si , U-5Nb-5Zr, U-3Nb-1.5Zr, U-9Mo, U-6Mo, U-1.5Mo-1.0Zr, U-10Zr, and U_3SiAl .
15

The additives of this second group includes uranium, which means that the amount of the additive may be higher than for the first group for maintaining a high volumetric uranium density in the nuclear fuel. Consequently, the amount of the additive for the second group should be equal to or less than 80% by volume of the nuclear fuel. Advantageously, the amount of the additive for the second group may be equal to or less than 70, 60, 50, 40, 30, 20 or 10% by volume of the nuclear fuel. The amount of the additive for this group is equal to or more than 2, 5, 7 or 10% by volume of the nuclear fuel. The uranium-containing compounds of the second group will react with water to form a tight, water insoluble, protective layer over the UN content. An additive comprising the element U will, upon exposure to water, oxidize to UO_2 , which will protect the underlying UN.
20
25
30

As mentioned above the nuclear fuel may be realised as a sintered solid body, such as said nuclear fuel pellet 10. The fuel pellet 10 may have a cylindrical shape, preferably a circular cylindrical shape, and may be annular.
35

The nuclear fuel may be manufactured through a suitable method comprising the following steps of:

5 Uranium, for instance in the form of a powder, enriched with respect to ^{235}U is provided.

Nitrogen, for instance in the form of a powder, enriched with respect to ^{15}N is provided.

10 Uranium and nitrogen are mixed to form a homogenous uranium-containing compound consisting of UN. The reaction may consist of, but is not limited to, direct nitridation of metallic uranium, or carbo-thermic nitridation of uranium oxide.

15 An additive, for instance in the form of a powder, consisting of, or substantially consisting of, at least one element, in elementary form or as a compound, selected from the group consisting of Zr, Mo, Si, Al, Nb and U is added to the uranium-containing compound.

20

The uranium-containing compound and the additive are mixed to form a homogeneous mixture, for instance in the form of a powder. As an alternative embodiment, the additive may be introduced in such a way that it covers and protects individual uranium-containing powder grains.

25

The mixture may also be sintered by means of a suitable sintering process at sintering pressure and a sintering temperature to a sintered solid body, for instance a cylindrical nuclear fuel pellet. The sintering temperature is at least 1800°C , at least 2000°C , preferably at least 2100°C , most preferably at least 2200°C .

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The invention is not limited to the embodiments and examples describe above, but may be varied and modified within the scope of the following claims.

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Claims

1. A nuclear fuel adapted for use in a water cooled nuclear reactor, including light water reactors LWR, such as Boiling Water Reactors BWR and Pressure Water Reactors PWR,
5 the nuclear fuel comprising an uranium-containing compound consisting of UN, wherein the uranium content of the uranium-containing compound comprises less than 10% by weight of the isotope ^{235}U ,
10 characterised in that the nuclear fuel comprises an additive consisting of, or substantially consisting of, at least one element, in elementary form or as a compound, selected from the group consisting of Zr, Mo, Si, Al, Nb and U.
- 15 2. A nuclear fuel according to claim 1, wherein the nitrogen content of the uranium-containing compound comprises at least 60% by weight of the isotope ^{15}N .
- 20 3. A nuclear fuel according to claim 1, wherein the nitrogen content of the uranium-containing compound comprises at least 70% by weight of the isotope ^{15}N .
- 25 4. A nuclear fuel according to claim 1, wherein the nitrogen content of the uranium-containing compound comprises at least 80% by weight of the isotope ^{15}N .
5. A nuclear fuel according to any one of claims 1 to 4, wherein the additive includes at least one of ZrN, ZrH₂, Si₃N₄, Al₂O₃, ZrO₂, Mo_yO_x, SiO₂, AlN, ZrO₂, ZrH₃, SiO₂, U, Zr, Mo, Si, U₃Si₂, ZrUAl, ZrUSi, ZrUH, UAl₂, U₃Si, U-5Nb-5Zr, U-3Nb-1.5Zr, U-9Mo, U-6Mo, U-1.5Mo-1.0Zr, U-10Zr, and U₃SiAl.
30
6. A nuclear fuel according to claim 5, wherein the additive includes at least one of ZrN, ZrH₂, Si₃N₄, Al₂O₃, ZrO₂, Mo_yO_x, SiO₂, AlN, ZrO₂, ZrH₃, SiO₂, Zr, Mo, and Si, and wherein the
35

amount of the additive is equal to or less than 30% by volume of the nuclear fuel.

7. A nuclear fuel according to claim 5, wherein the additive
5 includes at least one of U, U_3Si_2 , ZrUAl, ZrUSi, ZrUH, UAl_2 ,
 U_3Si , U-5Nb-5Zr, U-3Nb-1.5Zr, U-9Mo, U-6Mo, U-1.5Mo-1.0Zr,
U-10Zr, and U_3SiAl , and wherein the amount of the additive is
equal to or less than 80% by volume of the nuclear fuel.
- 10 8. A nuclear fuel according to any one of the preceding
claims, wherein the nuclear fuel is provided in the form of a nu-
clear fuel pellet (10).
9. A nuclear fuel according to claim 8, wherein the nuclear
15 fuel pellet is formed through sintering of a powder of the ura-
nium-containing compound and said at least one additive,
10. A fuel element comprising a cladding and a nuclear fuel
according to any one of claims 1 to 9.
- 20 11. A fuel element according to claim 10, wherein the fuel ele-
ment is designed as an elongated fuel rod (4).
12. A fuel assembly (1) comprising a plurality of fuel elements
25 according to any one of claims 10 and 11.
13. A method of manufacturing a nuclear fuel according to any
one of claims 1 to 9, the method comprising the step of:
providing an uranium-containing compound consisting of UN,
30 wherein the uranium content of the uranium-containing com-
pound comprises less than 10% by weight of the isotope ^{235}U ,
adding to the uranium-containing compound an additive consist-
ing of, or substantially consisting of, at least one element, in
elementary form or as a compound, selected from the group
35 consisting of Zr, Mo, Si, Al, Nb and U.

14. A method according to claim 13, comprising the steps of:
providing a powder of the uranium-containing compound,
providing a powder of the additive,
mixing the uranium-containing compound and the additive to a
5 powder mixture, and
sintering the mixture at sintering pressure and a sintering tem-
perature to a nuclear fuel pellet.
15. A method according to claim 14, wherein the sintering
10 temperature is at least 1800°C, at least 2000 °C, preferably at
least 2100 °C, most preferably at least 2200 °C.

Fig 1

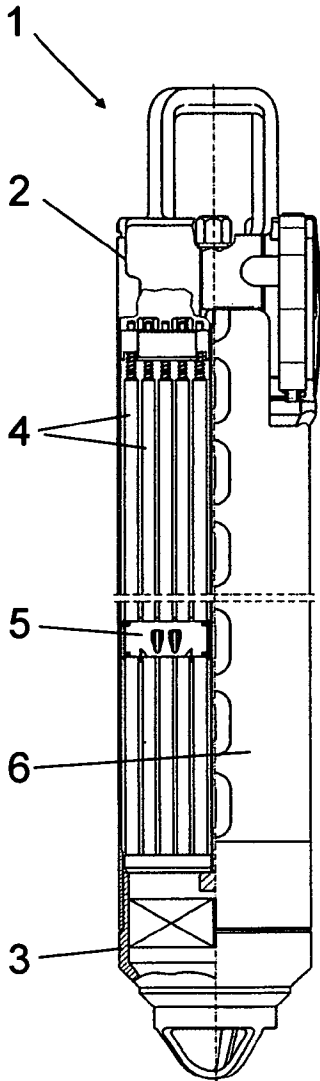


Fig 2

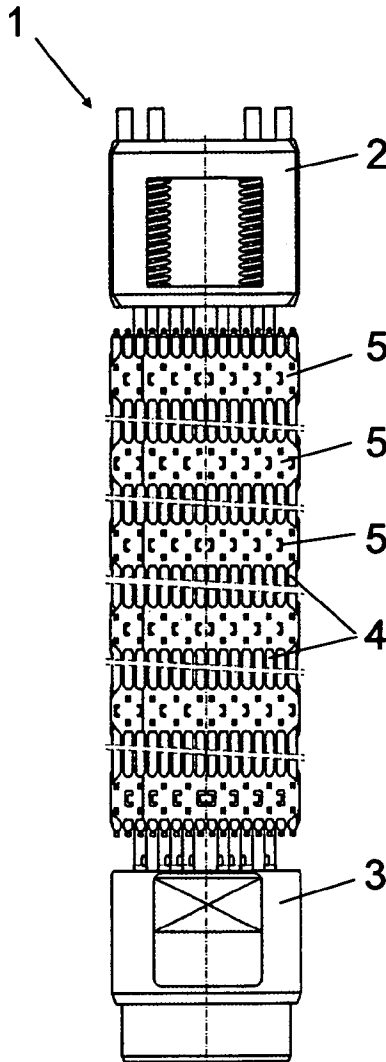
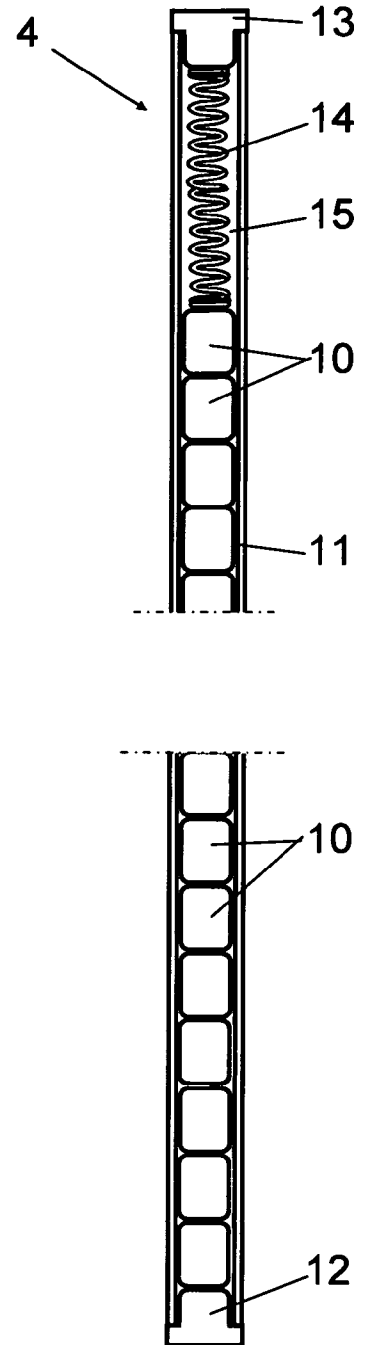


Fig 3



INTERNATIONAL SEARCH REPORT

International application No
PCT/EP2011/050983

A. CLASSIFICATION OF SUBJECT MATTER
INV. G21C3/00 G21C3/58 G21C3/62
ADD.
According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED
Minimum documentation searched (classification system followed by classification symbols)
G21C
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)
EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	JP 11 202072 A (TOSHIBA CORP) 30 July 1999 (1999-07-30)	1-8, 10-14
Y	paragraphs [0008], [0014] - [0017], [0032], [0033], [0038]; figures 5,6 abstract	9,15
Y	----- REEVE ET AL: "Ceramics as nuclear reactor fuels", CERAMURGIA INTERNATIONAL,, vol. 1, no. 2, 1 September 1975 (1975-09-01), pages 59-71, XP024165250, ISSN: 0390-5519, DOI: DOI:10.1016/0390-5519(75)90008-3 [retrieved on 1975-09-01]	9,15
A	----- page 64, paragraph 3 - page 65, paragraph 1 ----- -/--	1-8, 10-14

Further documents are listed in the continuation of Box C.

See patent family annex.

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Date of the actual completion of the international search 19 May 2011	Date of mailing of the international search report 30/05/2011
Name and mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Fax: (+31-70) 340-3016	Authorized officer Sewtz, Michael

INTERNATIONAL SEARCH REPORT

International application No
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