

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
27 November 2003 (27.11.2003)

PCT

(10) International Publication Number  
**WO 03/097555 A2**

(51) International Patent Classification<sup>7</sup>: **C04B 35/00**

(21) International Application Number: PCT/IB03/02395

(22) International Filing Date: 12 May 2003 (12.05.2003)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:  
P.10221866.8 15 May 2002 (15.05.2002) DE

(71) Applicant (for all designated States except US): **MARCONI COMMUNICATIONS, GMBH** [DE/DE]; Gerberstrasse 33, 71520 Backnang (DE).

(72) Inventors; and

(75) Inventors/Applicants (for US only): **SCHALLNER, Martin, Josef** [DE/DE]; Niedersachsenstrasse 45/1, 71640 Ludwigsburg (DE). **OPP, Andreas, Felix, Desire** [DE/DE]; Scherberger Feld 4, 52146 Wurselen (DE). **HOEPPE, Ulrich, Ernst, Ewald** [DE/DE]; Buttenfeld 2, 71522 Backnang (DE).

(74) Agent: **CAMP, Ronald**; Marconi Intellectual Property, Marrable House, The Vineyards, Great Baddow, Chelmsford, Essex CM2 7QS (GB).

(81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.

(84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

— without international search report and to be republished upon receipt of that report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.



**WO 03/097555 A2**

(54) Title: ALUMINA CERAMIC AND METHOD FOR ITS MANUFACTURE

(57) Abstract: A ceramic material made of alumina and titanium oxide, which is practically free from additives, achieves a high quality factor Q by annealing the material after sintering. Qualities of up to 17,900 at a measuring frequency of 10 GHz can be achieved.

MARCONI COMMUNICATIONS GMBH, 71522 BACKNANG

G. 81657

5

**Alumina ceramic and method for its manufacture**

The present invention relates to an alumina ceramic  
10 that is suitable for use as a dielectric material,  
in particular as a substrate material in RF and mi-  
crowave technology, and to a method for manufactur-  
ing such a ceramic. Applications of such a ceramic  
material are e.g. impedance matching in microwave  
15 circuits, dielectric microwave resonators, micro-  
wave filters, microwave transmission lines, micro-  
wave capacitors, circuit boards for microwave cir-  
cuits and the like. A ceramic material suitable for  
such applications should in particular have a low  
20 dielectric constant, a high quality factor Q in the  
microwave frequency range and, in general, a low  
temperature dependence of its dielectric proper-  
ties.

25 The applicability of alumina-based ceramic for the  
above mentioned purposes has been examined in a  
plurality of patent publications. Pure alumina is  
an attractive material due to its low dielectric  
constant in the RF range. A disadvantage of alumina  
30 is the rather strong dependence of its permittivity  
from temperature of approx. 110 ppm/°C. This tem-  
perature dependence causes e.g. a temperature de-  
pendence of the Eigenfrequencies of microwave reso-

-2-

nators based on such a material and thus restricts strongly the applicability of pure alumina as a dielectric in RF applications.

5 Therefore, a variety of alumina-based mixed ceramics have been examined for their usability in RF applications that contain, besides alumina, one or more additives that are to correct undesired properties of alumina. Various documents relate to  
10 mixed ceramics that contain titanium oxide  $TiO_2$  besides other additives. Titanium oxide has a negative temperature coefficient  $\tau_e$  of the permittivity, so that it is expected that by mixing alumina and titanium oxide in adequate proportions, it will be  
15 possible to produce a mixed ceramic having a low  $\tau_e$ .

However, it proves to be difficult to obtain ceramic materials with a quality factor  $Q$  sufficient for microwave applications by mixing only these two  
20 components. Therefore, the known alumina-titanium-oxide-mixed ceramics always contain further additives such as  $TaO_5$  and  $SnO_2$  in US-A-4 866 016 and  $CaO$  and  $La_2O_3$  in US-A-4 668 646. US-A-4 591 574 teaches the manufacture of a mixed ceramic material  
25 from the initial materials  $Al_2O_3$ ,  $CaO$  and  $TiO_2$ . According to this document, the  $TiO_2$  is first processed with the  $CaO$  into calcium titanate separately from the  $Al_2O_3$ , and the calcium titanate is then mixed with the  $Al_2O_3$  and sintered. I.e. in the material  
30 composition for sintering,  $TiO_2$  is practically not contained any more. Calcium titanate has a much more strongly negative value of  $\tau_e$  than that of  $TiO_2$ , so that small additions of this material are already sufficient in order to achieve a value

- 3 -

of  $\tau_e$  close to zero for the mixed ceramic material. It is disadvantageous, however, that small fluctuations of the quantity of added calcium titanate or in the course of the sintering process cause  $\tau_e$  to differ noticeably from the desired value.

From US 6 242 376 B1, a dielectric ceramic composition is known which comprises, besides alumina and titanium oxide, an addition of 0.1 to 3 weight percent Nb<sub>2</sub>O<sub>5</sub>. By sintering a mixture of these three initial materials during four hours at approx. 1,400 °C, a ceramic material is obtained which has a temperature dependence  $\tau_f$  of the resonance frequency between -30 and +30 ppm/°C and is claimed to have qualities Q between 10,000 and 55,000. There are no specific indications as to the individual ceramic material compositions, the qualities Q obtained with these and the frequencies at which they were measured. It is only stated that measurements were carried out in the frequency range above 2 GHz, and from the statements concerning the measuring device, it can be concluded that the measurement frequency was not above 6 GHz. From the fact that the quality Q is generally inversely proportional to the frequency at which it is measured, and that the quality measurements were apparently not conducted at a fixed frequency but in a frequency range, it can be concluded that the highest Q values cited in this document, if they were in fact measured and did not only define the upper limit of an interval which contained the actually measured values, were obtained at low measurement frequencies. If one adopts as a measure for the suitability of a material for RF applications not

- 4 -

the quality factor  $Q$  but the value of the product  $Q.f$  of quality factor and measuring frequency, which is largely independent from the measuring frequency,  $Q.f$ -factors of 110,000 at maximum, are  
5 obtained (for a measuring frequency in GHz).

Efforts to produce a ceramic material suitable for RF applications from a binary mixture of alumina and titanium oxide have up to now not led to satisf-  
10 fying results. Instead, from the article "Layered  $Al_2O_3$ - $TiO_2$  composite dielectric resonators with tuneable temperature coefficient for microwave ap- plications", N. Alford et. al., IEE proceedings- Science, measurement and technology, volume 147,  
15 no. 6, November 2000, pages 269 to 73, a dielectric body has become known that has a structure composed of alternating layers of alumina and titanium ox- ide. Such a layered structure is expensive to manu-  
20 facture and is therefore not suitable for mass pro- duction of moderately priced components.

From M. Ishitsuka, Synthesis and thermal stability of aluminium titanate solid solutions, J. Am. Ce-  
ram. Soc, volume 70, pages 69 to 71 (1987) it is  
25 known that at high temperatures aluminium titanate decomposes into alumina and titanium oxide.

The object of the present invention is to provide a ceramic material having an excellent quality and a  
30 low and selectively controllable temperature coef- ficient  $\tau_e$ , as well as a simple and economic proce- dure for its manufacture.

- 5 -

The object is achieved by a method according to claim 1 and a ceramic material according to claim 10.

5 The invention is based on the finding that unsatisfying qualities Q conventionally achieved with binary alumina-titanium-oxide-mixtures result from the formation of aluminium titanate during sintering of the raw components. While in the mixed alumina-titanium-oxide-ceramics of the prior art, the  
10 formation of aluminium titanate is apparently prevented from the beginning by suitable additives, the formation of aluminium titanate during sintering is voluntarily accepted according to the present invention, and instead, it is decomposed in  
15 the annealing phase after sintering. Surprisingly, in spite of the restructuration of the material associated with this decomposition, after annealing, high densities of the sintered body and excellent  
20 qualities Q are achieved.

Due to the decomposition of the aluminium titanate a posteriori by annealing, it is possible to avoid the use of sintering adjuvants. The finished ceramic is therefore very pure, it can contain 99.5 %  
25 or more of  $\text{Al}_2\text{O}_3$  and  $\text{TiO}_2$ .

The sintering temperature according to the invention is preferably between 1,390 and 1,450 °C. It  
30 has been shown that with a given composition of the ceramic, the coefficient  $\tau_s$  may be influenced by an appropriate choice of the sintering temperature. In this way, from one and the same raw material mixture, ceramic bodies having different temperature

- 6 -

coefficients  $\tau_e$  may be manufactured, and the temperature coefficient  $\tau_e$  may e.g. be chosen for a particular application such that the temperature coefficient of the ceramic material also compensates the temperature dependence of neighbouring components of a microwave circuit.

For decomposing the aluminium titanate, an annealing temperature below 1,280 °C is required; a speedy decomposition is achieved in a temperature interval between 1,000 °C and 1,200 °C, preferably between 1,075 °C and 1,125 °C.

The duration of the annealing phase of not more than five hours has proved sufficient for decreasing the aluminium titanate content of the ceramic material obtained by sintering below the detection limit of X-ray diffraction, i.e. below a proportion of approx. 1 %.

Further features and advantages of the invention become apparent from the subsequent description of embodiments.

For the manufacture of samples of the ceramic material according to the invention, the following raw materials were used:

Table 1:

30

	Alumina	Titanium oxide
Purity	> 99,9%	99,9 +%
Crystallization type	$\alpha$ -alumina	Rutile

-7-

Particle size	0,3 $\mu\text{m}$	1,17 $\mu\text{m}$
Si (ppm)	< 40	
Na (ppm)	< 10	
Mg (ppm)	< 10	
Cu (ppm)	< 10	
Fe (ppm)	< 20	
Surface according to BET	5-10 m <sup>2</sup> /g	

The raw materials were mixed in the following proportions:

5 Table 2:

Mixture	Alumina [mol %]	Titanium oxide [mol %]
1	89	11
2	90	10
3	91,5	8,5
4	93	7

In order to achieve homogeneity and to destroy powder agglomerates, 200 g of each of the mixtures defined in table 2 were mixed in an attritor (Netsch, PE-cup, zirconia grinding tool and 2 mm balls) with 130 g of purified water added, for twenty minutes at 800 revolutions per minute. A strong milling effect is not to be expected due to the refinement of the used powders and is also not necessary.

After finishing the kneading process, 2 to 2.5 weight percent of organic additives containing

- 8 -

binder, plastifier, lubricants and form adjuvants were added to the obtained slurry.

After mixing, the slurry was reduced to a ready to  
5 press granulate in a laboratory spray dryer (Büchi  
190, 0.7 mm nozzle, 190 °C inlet temperature, 115  
°C outlet temperature). This granulate was pressed  
in a metal mold having 11 mm in diameter to green  
bodies with a height of 8 mm under a pressure of  
10 1,500 kg/cm<sup>2</sup>.

The subsequent sintering of the shaped bodies began  
with a step of heating to up to 550 °C in order  
burn out all organic additives. Subsequently, the  
15 temperature was increased at a rather high rate of  
8 K/min to the sintering temperature. Tests were  
carried out with sintering temperatures between  
1,400 and 1,475 °C. After three hours of sintering,  
the temperature was decreased at a rate of 6 K/min  
20 to 1,100 °C, and a three hours annealing step at  
this temperature followed. Afterwards, the samples  
were cooled to room temperature.

The complete thermal processing was carried out in  
25 a pure oxygen atmosphere.

In order to remove surface impurities caused by the  
sintering process and to bring all samples into  
identical dimensions for the subsequent measure-  
30 ments, the finished sintered bodies were ground to  
a diameter of 7.5 mm ± 0.01 mm and a height of 5 mm  
± 0.01 mm. After grinding, the samples were cleaned  
and stored at normal atmospheric conditions.

### Measuring conditions

The microwave measurements were carried out in the  
 5 -50 °C to 120 °C temperature range by a resonant  
 cavity method using the TE<sub>01δ</sub> mode. The sintered  
 bodies were placed in a cylindrical, gold plated  
 copper cavity (diameter: 25.02 mm, height: 15.02  
 mm) on a 5 mm high, low loss sapphire spacer. Under  
 10 the cited conditions, the resonance frequency  $f_r$ ,  
 the quality Q, the relative permittivity  $\epsilon_r$  and the  
 temperature coefficient of the permittivity  $\tau_\epsilon$  were  
 measured. The influence of the resistance of the  
 cavity wall surface on the measured quality factor  
 15 Q of the sintered body was taken into account and  
 corrected.

### Results of measurements

20 Result of a first test series carried out with sin-  
 tered bodies manufactured under identical thermal  
 processing conditions are shown in subsequent table  
 3.

25 Table 3:

Mixture	Sintering temperature	Annealing temperature	$\epsilon_r$	$\tau_\epsilon$	Q
1	1.375	1.100/ 3 h	11,40	-44,47	15.900
2	1.375	1.100/ 3 h	11,36	-16,24	17.900
3	1.375	1.100/	11,17	-2,19	17.300

- 10 -

		3 h			
4	1.375	1.100/ 3 h	10,97	32,17	13.500

The indicated amounts of the quality factor  $Q$  relate to a measuring frequency of 10 GHz and a measuring temperature of 40 °C.

5

The temperature coefficient  $\tau_e$  may be set to positive and to negative values by selecting the composition of the raw mixture. Small, non-vanishing values of the temperature coefficient  $\tau_e$  in the shown range can be desirable in order to compensate the temperature dependence of adjacent circuit components by the temperature dependence of the ceramic material, so as to obtain as small as possible a temperature dependence of the behaviour of a complete circuit manufactured using the ceramic material of the invention. The measured quality factors  $Q$  correspond to Q.f factors of 130,00 to 179,000.

20 In a second test series, green bodies of mixture 2 were sintered at different temperatures. The other conditions of the thermal processing were the same as in the first test series. The obtained results are given in table 4.

25

Table 4:

Mixture	Sintering temperature	Annealing temperature	$\epsilon_r$	$\tau_e$	$Q$
2	1.450	1.100/	11,36	-34,67	11.500

		3 h			
2	1.440	1.100/ 3 h	11,36	-30,38	12.500
2	1.430	1.100/ 3 h	11,36	-27,77	14.200
2	1.420	1.100/ 3 h	11,36	-23,23	14.500
2	1.410	1.100/ 3 h	11,35	-21,87	15.600
2	1.400	1.100/ 3 h	11,35	-20,15	15.500

As the table shows, the sintering temperature also has an influence on the temperature coefficient  $\tau_e$  of the dielectric constant.

5

It is readily apparent that a modification of the composition of the mixture has a stronger effect on the dielectric properties of the sintered bodies than the sintering temperature. An adaptation of the sintering temperature might therefore be helpful for "fine tuning" the desired temperature coefficient  $\tau_e$  after coarsely defining it by the material composition.

15 It is to be assumed that similar results as given above for mixture of alumina and titanium oxide can be achieved if the titanium oxide is replaced by an earth alkali titanate such as  $\text{CaTiO}_3$  or  $\text{SrTiO}_3$  or a mixture of one or more earth alkali titanates and/or titanium oxide.  $\text{CaTiO}_3$  and  $\text{SrTiO}_3$  have a much more strongly negative temperature coefficient  $\tau_e$  than  $\text{TiO}_2$ . Therefore, when using these materials,

20

- 12 -

smaller proportions of titaniferous oxide in the mixture may be sufficient to obtain a temperature compensation in a desired extent than when using pure titanium oxide. However, it is to be expected  
5 that when using earth alkali titanates, precisely due to the strong temperature dependence of  $\tau_e$ , the properties of the finished sintered bodies will depend more strongly from small variations of the chemical composition or the sintering conditions  
10 than with the above described examples, so that exact control of the dielectric properties of the ceramic material may become more difficult.

G. 81657

**Claims**

5

1. A method for manufacturing alumina ceramic, wherein a mixture containing alumina powder and titaniferous oxide powder is sintered into a ceramic body, characterized in that the body obtained by sintering is annealed at an annealing temperature below 1,280 °C.  
10
2. The method of claim 1, characterized in that the titaniferous oxide powder is titanium oxide or an earth alkali titanate or a mixture of titanium oxide and at least one earth alkali titanate.  
15
3. The method of claim 1, characterized in that the titaniferous oxide powder is pure titanium dioxide.  
20
4. The method according to one of the preceding claims, characterized in that the sintering temperature is between 1,300 °C and 1,500 °C, preferably between 1,350 °C and 1,450 °C.  
25
5. The method according to one of the preceding claims, characterized in that the annealing temperature is between 1,000 °C and 1,200 °C, preferably between 1,075 °C and 1,125 °C.  
30
6. The method according to one of the preceding claims, characterized in that the duration of

the annealing is at least one hour and preferably less than five hours.

- 5 7. The method according to one of the preceding claims, characterized in that the body is annealed until an aluminium titanate content of less than 2 % is achieved.
- 10 8. The method according to one of the preceding claims, characterized in that at least 99.9 % of the mineral fraction of the mixture is formed of  $\text{Al}_2\text{O}_3$  and  $\text{TiO}_2$ .
- 15 9. The method according to one of the preceding claims, characterized in that the mineral fraction contains between 89 and 93 mol% of  $\text{Al}_2\text{O}_3$  and between 7 and 11 mol%  $\text{TiO}_2$ .
- 20 10. An alumina-based ceramic material having an alumina content between 89 and 93 mol% and a content of titaniferous oxide between 7 and 11 mol%, characterized by a total content of alumina and titaniferous oxide of at least 99.95 mol% and a Qf-factor of at least 100,000.
- 25 11. The ceramic material of claim 10, characterized in that the titaniferous oxide powder is titanium dioxide or an earth alkali titanate or a mixture of titanium dioxide and at least  
30 one earth alkali titanate.
12. The ceramic material of claim 10, characterized in that the titaniferous oxide powder is pure titanium dioxide.

13. The ceramic material according to one of claims 10 to 12, characterized in at maximum 2 mol%, preferably at maximum 1 mol% of the alumina and the titanium dioxide is present in the form of aluminium titanate.
- 5
14. The ceramic material according to one of claims 10 to 13, characterized in that it has a relative dielectric constant  $\epsilon$  between 10.5 and 12.0, preferably between 10.9 and 11.6, and a temperature coefficient  $\tau_\epsilon$  of the relative permittivity between -60 ppm and +40 ppm.
- 10
- 15