



(86) Date de dépôt PCT/PCT Filing Date: 2013/06/21

(87) Date publication PCT/PCT Publication Date: 2013/12/27

(45) Date de délivrance/Issue Date: 2016/12/13

(85) Entrée phase nationale/National Entry: 2014/12/15

(86) N° demande PCT/PCT Application No.: US 2013/046948

(87) N° publication PCT/PCT Publication No.: 2013/192480

(30) Priorités/Priorities: 2012/06/21 (US13/529,237);
2012/10/24 (US13/659,152)

(51) Cl.Int./Int.Cl. *C08G 75/045* (2016.01),
C08L 81/02 (2006.01)

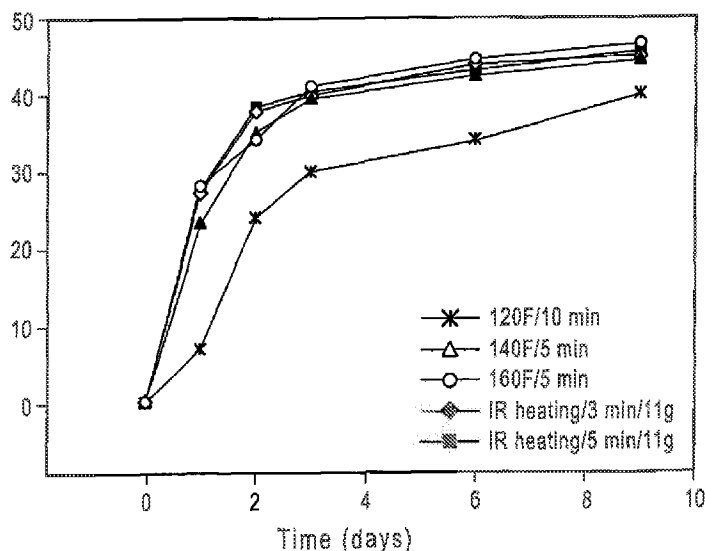
(72) Inventeurs/Inventors:
ANDERSON, LAWRENCE G., US;
CAI, JUEXIAO, US;
ITO, MARFI, US;
KELEDJIAN, RAQUEL, US;
LIN, RENHE, US

(73) Propriétaire/Owner:
PRC-DESOTO INTERNATIONAL, INC., US

(74) Agent: BORDEN LADNER GERVAIS LLP

(54) Titre : CHIMIES DE DURCISSEMENT PAR ADDITION DE MICHAEL POUR COMPOSITIONS COMPRENANT DES
POLYMERES SOUFRES

(54) Title: MICHAEL ADDITION CURING CHEMISTRIES FOR SULFUR-CONTAINING POLYMER COMPOSITIONS



(57) Abrégé/Abstract:

The use of Michael addition curing chemistries in compositions comprising sulfur-containing polymers such as polythioethers and polysulfides useful in aerospace sealant applications are disclosed. Sulfur-containing adducts comprising terminal Michael acceptor groups are also disclosed. Compositions including controlled release amine catalysts are also disclosed.

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property
Organization
International Bureau



(10) International Publication Number
WO 2013/192480 A3

(43) International Publication Date
27 December 2013 (27.12.2013)

(51) International Patent Classification:

C08G 75/02 (2006.01) *C09J 181/02* (2006.01)
C08L 81/02 (2006.01)

(21) International Application Number:

PCT/US2013/046948

(22) International Filing Date:

21 June 2013 (21.06.2013)

(25) Filing Language:

English

(26) Publication Language:

English

(30) Priority Data:

13/529,237 21 June 2012 (21.06.2012) US
13/659,152 24 October 2012 (24.10.2012) US

(71) Applicant: PRC-DESOTO INTERNATIONAL, INC.

[US/US]; 12780 San Fernando Road, Sylmar, California 91342 (US).

(72) Inventors: ANDERSON, Lawrence G.; 4119 Northampton Drive, Allison Park, Pennsylvania 15101 (US). CAI, Juexiao; 26504 Brooks Circle, Stevenson Ranch, California 91381 (US). ITO, Marfi; 3315 Sherbourne Drive, Culver City, California 90232 (US). KELEDJIAN, Raquel; 321 West Dryden Street #3, Glendale, California 91202 (US). LIN, Renhe; 25440 Wharton Drive, Stevenson Ranch, California 91381 (US).

(74) Agents: MEYERS, Diane R. et al.; PPG Industries, Inc., One PPG Place, 39th Floor, Pittsburgh, Pennsylvania 15272 (US).

(81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BN, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PA, PE, PG, PH, PL, PT, QA, RO, RS, RU, RW, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TH, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

(84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LR, LS, MW, MZ, NA, RW, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, RU, TJ, TM), European (AL, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, RS, SE, SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, KM, ML, MR, NE, SN, TD, TG).

Declarations under Rule 4.17:

- as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii))
- as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii))

Published:

- with international search report (Art. 21(3))

[Continued on next page]

(54) Title: MICHAEL ADDITION CURING CHEMISTRIES FOR SULFUR-CONTAINING POLYMER COMPOSITIONS

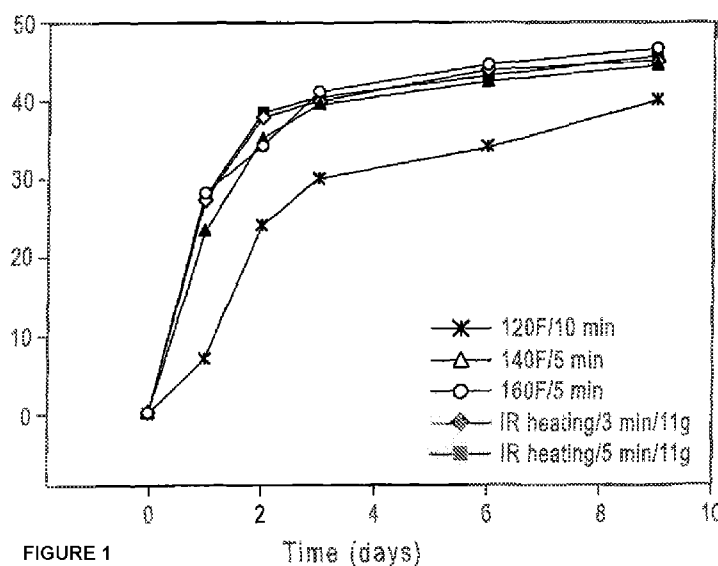


FIGURE 1

(57) Abstract: The use of Michael addition curing chemistries in compositions comprising sulfur-containing polymers such as polythioethers and polysulfides useful in aerospace sealant applications are disclosed. Sulfur-containing adducts comprising terminal Michael acceptor groups are also disclosed. Compositions including controlled release amine catalysts are also disclosed.

WO 2013/192480 A3



-
- *before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments (Rule 48.2(h))* (88) **Date of publication of the international search report:**
13 February 2014

MICHAEL ADDITION CURING CHEMISTRIES FOR SULFUR-CONTAINING POLYMER COMPOSITIONS

FIELD

[001] The present disclosure relates to the use of Michael addition curing chemistries in compositions comprising sulfur-containing polymers, such as polythioethers and polysulfides, useful in aerospace sealant applications. The disclosure also relates to sulfur-containing adducts having terminal Michael acceptor groups and compositions thereof.

BACKGROUND

[002] Sealants useful in aerospace and other applications must satisfy demanding mechanical, chemical, and environmental requirements. The sealants can be applied to a variety of surfaces including metal surfaces, primer coatings, intermediate coatings, finished coatings, and aged coatings. In sealants such as those described in U.S. Patent No. 6,123,179 an amine catalyst is used to provide a cured product. Such systems typically cure in over two hours and although exhibiting acceptable fuel resistance and thermal resistance for many applications, a faster curing rate and improved performance is desirable.

SUMMARY

[003] Michael addition curing chemistries are often used in acrylic-based polymer systems and as disclosed in U.S. Patent No. 3,138,573 have been adapted for use in polysulfide compositions. Application of Michael addition curing chemistries to sulfur-containing polymers not only results in cured sealants with faster cure rates and enhanced performance including fuel resistance and thermal resistance, but also provides a sealant with improved physical properties, such as elongation.

[004] Compositions having extended pot life and controlled curing rate are also realized by using a controlled release amine catalyst. In these systems, an amine catalyst such as a strong base or primary amine that produces a fast reaction rate is protected or encapsulated and dispersed in the composition. Upon exposure, for example, to ultraviolet radiation, moisture, or temperature, the catalytic amine is released and catalyzes the Michael addition reaction. In certain embodiments, systems provide a pot life greater than 2 hours to 12 hours and cure within 24 to 72 hours after the useful working time.

[005] In a first aspect, polythioether adducts comprising at least two terminal Michael acceptor groups are provided.

[006] In a second aspect, compositions are provided comprising a polythioether polymer comprising at least two terminal groups reactive with Michael acceptor groups; and a compound having at least two Michael acceptor groups.

[007] In a third aspect, compositions are provided comprising a polythioether adduct provided by the present disclosure and a curing agent comprising at least two terminal groups that are reactive with Michael acceptor groups.

[008] In a fourth aspect, compositions are provided comprising (a) the sulfur-containing adduct provided by the present disclosure; (b) a sulfur-containing polymer comprising at least two terminal groups reactive with Michael acceptor groups; and (c) a monomeric compound having at least two Michael acceptor groups.

[009] In a fifth aspect, hydroxyl-terminated sulfur-containing adducts are provided comprising the reaction products of reactants comprising (a) a sulfur-containing Michael acceptor adduct provided by the present disclosure; and (b) a compound having a hydroxyl group and a group that is reactive with the terminal groups of the sulfur-containing Michael acceptor adduct.

[010] In a sixth aspect, compositions are provided comprising (a) a hydroxyl-terminated sulfur-containing adduct provided by the present disclosure; and (b) a polyisocyanate curing agent.

[011] In a seventh aspect, amine-terminated sulfur-containing adducts are provided comprising the reaction products of reactants comprising (a) a sulfur-containing Michael acceptor adduct provided by the present disclosure; and (b) a compound having an amine group and a group that is reactive with the terminal groups of the sulfur-containing Michael acceptor adduct.

[012] In an eighth aspect, compositions are provided comprising (a) an amine-terminated sulfur-containing adduct provided by the present disclosure; and (b) a polyisocyanate curing agent.

[013] In a ninth aspect, cured sealants comprising a composition provided by the present disclosure are provided.

[014] In a tenth aspect, apertures sealed with a composition provided by present disclosure are provided.

[015] In an eleventh aspect, methods of sealing an aperture are provided comprising (a) applying a composition provided by the present disclosure formulated as a sealant to at least one

surface defining an aperture; (b) assembling the surfaces defining the aperture; and (c) curing the composition to provide a sealed aperture.

[016] In a twelfth aspect, compositions are provided comprising (a) a compound comprising at least two terminal groups reactive with Michael acceptor groups; (b) a compound having at least two Michael acceptor groups; and (c) a controlled release amine catalyst, wherein at least one of (a) and (b) comprises a polythioether polymer.

[017] In a thirteenth aspect, method of using a composition comprising (a) a compound comprising at least two terminal groups reactive with Michael acceptor groups; (b) a compound having at least two Michael acceptor groups; and (c) a controlled release amine catalyst, wherein at least one of (a) and (b) comprises a polythioether polymer, are provided.

DETAILED DESCRIPTION

Definitions

[018] For purposes of the following description, it is to be understood that embodiments provided by the present disclosure may assume various alternative variations and step sequences, except where expressly specified to the contrary. Moreover, other than in the examples, or where otherwise indicated, all numbers expressing, for example, quantities of ingredients used in the specification and claims are to be understood as being modified in all instances by the term “about.” Accordingly, unless indicated to the contrary, the numerical parameters set forth in the following specification and attached claims are approximations that may vary depending upon the desired properties to be obtained. At the very least, and not as an attempt to limit the application of the doctrine of equivalents to the scope of the claims, each numerical parameter should at least be construed in light of the number of reported significant digits and by applying ordinary rounding techniques.

[019] Notwithstanding that the numerical ranges and parameters setting forth the broad scope of the invention are approximations, the numerical values set forth in the specific examples are reported as precisely as possible. Any numerical value, however, inherently contains certain errors necessarily resulting from the standard variation found in their respective testing measurements.

[020] Also, it should be understood that any numerical range recited herein is intended to include all sub-ranges encompassed therein. For example, a range of “1 to 10” is intended to include all sub-ranges between (and including) the recited minimum value of about 1 and the

recited maximum value of about 10, that is, having a minimum value equal to or greater than about 1 and a maximum value of equal to or less than about 10. Also, in this application, the use of “or” means “and/or” unless specifically stated otherwise, even though “and/or” may be explicitly used in certain instances.

[021] A dash (“—”) that is not between two letters or symbols is used to indicate a point of bonding for a substituent or between two atoms. For example, —CONH₂ is bonded to another chemical moiety through the carbon atom.

[022] “Alkanediyl” refers to a diradical of a saturated, branched or straight-chain, acyclic hydrocarbon group, having, for example, from 1 to 18 carbon atoms (C₁₋₁₈), from 1 to 14 carbon atoms (C₁₋₁₄), from 1 to 6 carbon atoms (C₁₋₆), from 1 to 4 carbon atoms (C₁₋₄), or from 1 to 3 hydrocarbon atoms (C₁₋₃). It will be appreciated that a branched alkanediyl has a minimum of three carbon atoms. In certain embodiments, the alkanediyl is C₂₋₁₄ alkanediyl, C₂₋₁₀ alkanediyl, C₂₋₈ alkanediyl, C₂₋₆ alkanediyl, C₂₋₄ alkanediyl, and in certain embodiments, C₂₋₃ alkanediyl. Examples of alkanediyl groups include methane-diyl (—CH₂—), ethane-1,2-diyl (—CH₂CH₂—), propane-1,3-diyl and iso-propane-1,2-diyl (*e.g.*, —CH₂CH₂CH₂— and —CH(CH₃)CH₂—), butane-1,4-diyl (—CH₂CH₂CH₂CH₂—), pentane-1,5-diyl (—CH₂CH₂CH₂CH₂CH₂—), hexane-1,6-diyl (—CH₂CH₂CH₂CH₂CH₂CH₂—), heptane-1,7-diyl, octane-1,8-diyl, nonane-1,9-diyl, decane-1,10-diyl, dodecane-1,12-diyl, and the like.

[023] “Alkanecycloalkane” refers to a saturated hydrocarbon group having one or more cycloalkyl and/or cycloalkanediyl groups and one or more alkyl and/or alkanediyl groups, where cycloalkyl, cycloalkanediyl, alkyl, and alkanediyl are defined herein. In certain embodiments, each cycloalkyl and/or cycloalkanediyl group(s) is C₃₋₆, C₅₋₆, and in certain embodiments, cyclohexyl or cyclohexanediyl. In certain embodiments, each alkyl and/or alkanediyl group(s) is C₁₋₆, C₁₋₄, C₁₋₃, and in certain embodiments, methyl, methanediyl, ethyl, or ethane-1,2-diyl. In certain embodiments, the alkanecycloalkane group is C₄₋₁₈ alkanecycloalkane, C₄₋₁₆ alkanecycloalkane, C₄₋₁₂ alkanecycloalkane, C₄₋₈ alkanecycloalkane, C₆₋₁₂ alkanecycloalkane, C₆₋₁₀ alkanecycloalkane, and in certain embodiments, C₆₋₉ alkanecycloalkane. Examples of alkanecycloalkane groups include 1,1,3,3-tetramethylcyclohexane and cyclohexylmethane.

[024] “Alkanecycloalkanediyl” refers to a diradical of an alkanecycloalkane group. In certain embodiments, the alkanecycloalkanediyl group is C₄₋₁₈ alkanecycloalkanediyl, C₄₋₁₆ alkanecycloalkanediyl, C₄₋₁₂ alkanecycloalkanediyl, C₄₋₈ alkanecycloalkanediyl, C₆₋₁₂ alkanecycloalkanediyl, C₆₋₁₀ alkanecycloalkanediyl, and in certain embodiments, C₆₋₉

alkanecycloalkanediyl. Examples of alkanecycloalkanediyl groups include 1,1,3,3-tetramethylcyclohexane-1,5-diyl and cyclohexylmethane-4,4'-diyl.

[025] “Alkanearene” refers to a hydrocarbon group having one or more aryl and/or arenediyl groups and one or more alkyl and/or alkanediyl groups, where aryl, arenediyl, alkyl, and alkanediyl are defined here. In certain embodiments, each aryl and/or arenediyl group(s) is C₆₋₁₂, C₆₋₁₀, and in certain embodiments, phenyl or benzenediyl. In certain embodiments, each alkyl and/or alkanediyl group(s) is C₁₋₆, C₁₋₄, C₁₋₃, and in certain embodiments, methyl, methanediyl, ethyl, or ethane-1,2-diyl. In certain embodiments, the alkanearene group is C₄₋₁₈ alkanearene, C₄₋₁₆ alkanearene, C₄₋₁₂ alkanearene, C₄₋₈ alkanearene, C₆₋₁₂ alkanearene, C₆₋₁₀ alkanearene, and in certain embodiments, C₆₋₉ alkanearene. Examples of alkanearene groups include diphenyl methane.

[026] “Alkanearenediyl” refers to a diradical of an alkanearene group. In certain embodiments, the alkanearenediyl group is C₄₋₁₈ alkanearenediyl, C₄₋₁₆ alkanearenediyl, C₄₋₁₂ alkanearenediyl, C₄₋₈ alkanearenediyl, C₆₋₁₂ alkanearenediyl, C₆₋₁₀ alkanearenediyl, and in certain embodiments, C₆₋₉ alkanearenediyl. Examples of alkanearenediyl groups include diphenyl methane-4,4'-diyl.

[027] “Alkenyl” group refers to a group (R)₂C=C(R)₂. In certain embodiments, an alkenyl group has the structure –RC=C(R)₂ where the alkenyl group is a terminal group and is bonded to a larger molecule. In such embodiments, each R may be selected from, for example, hydrogen and C₁₋₃ alkyl. In certain embodiments, each R is hydrogen and an alkenyl group has the structure –CH=CH₂.

[028] “Alkoxy” refers to a –OR group where R is alkyl as defined herein. Examples of alkoxy groups include methoxy, ethoxy, n-propoxy, isopropoxy, and n-butoxy. In certain embodiments, the alkoxy group is C₁₋₈ alkoxy, C₁₋₆ alkoxy, C₁₋₄ alkoxy, and in certain embodiments, C₁₋₃ alkoxy.

[029] “Alkyl” refers to a monoradical of a saturated, branched or straight-chain, acyclic hydrocarbon group having, for example, from 1 to 20 carbon atoms, from 1 to 10 carbon atoms, from 1 to 6 carbon atoms, from 1 to 4 carbon atoms, or from 1 to 3 carbon atoms. It will be appreciated that a branched alkyl has a minimum of three carbon atoms. In certain embodiments, the alkyl group is C₂₋₆ alkyl, C₂₋₄ alkyl, and in certain embodiments, C₂₋₃ alkyl. Examples of alkyl groups include methyl, ethyl, n-propyl, iso-propyl, n-butyl, iso-butyl, tert-butyl, n-hexyl, n-decyl, tetradecyl, and the like. In certain embodiments, the alkyl group is C₂₋₆ alkyl, C₂₋₄ alkyl,

and in certain embodiments, C₂₋₃ alkyl. It will be appreciated that a branched alkyl has at least three carbon atoms.

[030] “Arenediyl” refers to diradical monocyclic or polycyclic aromatic group. Examples of arenediyl groups include benzene-diyl and naphthalene-diyl. In certain embodiments, the arenediyl group is C₆₋₁₂ arenediyl, C₆₋₁₀ arenediyl, C₆₋₉ arenediyl, and in certain embodiments, benzene-diyl.

[031] “Cycloalkanediyl” refers to a diradical saturated monocyclic or polycyclic hydrocarbon group. In certain embodiments, the cycloalkanediyl group is C₃₋₁₂ cycloalkanediyl, C₃₋₈ cycloalkanediyl, C₃₋₆ cycloalkanediyl, and in certain embodiments, C₅₋₆ cycloalkanediyl. Examples of cycloalkanediyl groups include cyclohexane-1,4-diyl, cyclohexane-1,3-diyl, and cyclohexane-1,2-diyl.

[032] “Cycloalkyl” refers to a saturated monocyclic or polycyclic hydrocarbon monoradical group. In certain embodiments, the cycloalkyl group is C₃₋₁₂ cycloalkyl, C₃₋₈ cycloalkyl, C₃₋₆ cycloalkyl, and in certain embodiments, C₅₋₆ cycloalkyl.

[033] “Heteroalkanediyl” refers to an alkanediyl group in which one or more of the carbon atoms are replaced with a heteroatom, such as N, O, S, or P. In certain embodiments of heteroalkanediyl, the heteroatom is selected from N and O.

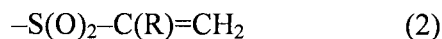
[034] “Heterocycloalkanediyl” refers to a cycloalkanediyl group in which one or more of the carbon atoms are replaced with a heteroatom, such as N, O, S, or P. In certain embodiments of heterocycloalkanediyl, the heteroatom is selected from N and O.

[035] “Heteroarenediyl” refers to an arenediyl group in which one or more of the carbon atoms are replaced with a heteroatom, such as N, O, S, or P. In certain embodiments of heteroarenediyl, the heteroatom is selected from N and O.

[036] “Heterocycloalkanediyl” refers to a cycloalkanediyl group in which one or more of the carbon atoms are replaced with a heteroatom, such as N, O, S, or P. In certain embodiments of heterocycloalkanediyl, the heteroatom is selected from N and O.

[037] A “Michael acceptor” refers to an activated alkene, such as an alkenyl group proximate to an electron-withdrawing group such as a ketone, nitro, halo, nitrile, carbonyl, or nitro group. Michael acceptors are well known in the art. A “Michael acceptor group” refers to an activated alkenyl group and an electron-withdrawing group. In certain embodiments, a Michael acceptor group is selected from a vinyl ketone, a vinyl sulfone, a quinone, an enamine, a ketimine, oxazolidine, and an acrylate. Other examples of Michael acceptors are disclosed in

Mather et al., *Prog. Polym. Sci.* 2006, 31, 487-531, and include acrylate esters, acrylonitrile, acrylamides, maleimides, alkyl methacrylates, cyanoacrylates. Other Michael acceptors include vinyl ketones, α,β -unsaturated aldehydes, vinyl phosphonates, acrylonitrile, vinyl pyridines, certain azo compounds, β -keto acetylenes and acetylene esters. In certain embodiments, a Michael acceptor group is derived from a vinyl ketone and has the structure of Formula (2):



where each R is independently selected from hydrogen, fluorine, and C_{1-3} alkyl. In certain embodiments, each R is hydrogen. In certain embodiments, a Michael acceptor or Michael acceptor group does not encompass acrylates. A “Michael acceptor compound” refers to a compound comprising at least one Michael acceptor. In certain embodiments, a Michael acceptor compound is divinyl sulfone, and a Michael acceptor group is vinylsulfonyl ($-\text{S}(\text{O})_2-\text{CH}=\text{CH}_2$).

[038] As used herein, “polymer” refers to oligomers, homopolymers, and copolymers. Unless stated otherwise, molecular weights are number average molecular weights for polymeric materials indicated as “Mn” as determined, for example, by gel permeation chromatography using a polystyrene standard in an art-recognized manner.

[039] “Substituted” refers to a group in which one or more hydrogen atoms are each independently replaced with the same or different substituent(s). In certain embodiments, the substituent is selected from halogen, $-\text{S}(\text{O})_2\text{OH}$, $-\text{S}(\text{O})_2$, $-\text{SH}$, $-\text{SR}$ where R is C_{1-6} alkyl, $-\text{COOH}$, $-\text{NO}_2$, $-\text{NR}_2$ where each R is independently selected from hydrogen and C_{1-3} alkyl, $-\text{CN}$, $=\text{O}$, C_{1-6} alkyl, $-\text{CF}_3$, $-\text{OH}$, phenyl, C_{2-6} heteroalkyl, C_{5-6} heteroaryl, C_{1-6} alkoxy, and $-\text{COR}$ where R is C_{1-6} alkyl. In certain embodiments, the substituent is chosen from $-\text{OH}$, $-\text{NH}_2$, and C_{1-3} alkyl.

[040] Reference is now made to certain embodiments of sulfur-containing adducts having terminal Michael acceptor groups, polymers, compositions, and methods. The disclosed embodiments are not intended to be limiting of the claims. To the contrary, the claims are intended to cover all alternatives, modifications, and equivalents.

Sulfur-containing adducts

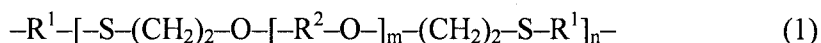
[041] Sulfur-containing adducts provided by the present disclosure comprise terminal Michael acceptor groups. Sulfur-containing polymers useful herein include, for example, polythioethers, polysulfides, and combinations thereof. Examples of suitable polythioethers are

disclosed in U.S. Patent No. 6,123,179. Examples of suitable polysulfides are disclosed in U.S. Patent No. 4,623,711. In certain embodiments, a sulfur-containing adduct may be difunctional, and in certain embodiments, may have a functionality greater than 2 such as 3, 4, 5, or 6. A sulfur-containing adduct may comprise a mixture of sulfur-containing adducts having different functionalities characterized by an average functionality from 2.05 to 6, from 2.1 to 4, from 2.1 to 3, from 2.2 to 2.8, and in certain embodiments, from 2.4 to 2.6. Sulfur-containing adducts have at least two terminal Michael acceptor groups, and in certain embodiments have two Michael acceptor groups, 3, 4, 5, or 6 Michael acceptor groups. A sulfur-containing adduct may comprise a combination of adducts having different numbers of terminal Michael acceptor groups characterized, for example, by an average Michael acceptor functionality of from 2.05 to 6, from 2.1 to 4, from 2.1 to 3, from 2.2 to 2.8, and in certain embodiments, from 2.4 to 2.6.

[042] In certain embodiments, a sulfur-containing adduct comprises a polythioether adduct characterized by a polythioether having at least two terminal Michael acceptor groups.

[043] In certain embodiments, a sulfur-containing adduct comprises a polythioether adduct comprising:

(a) a backbone comprising the structure of Formula (1):



where (i) each R^1 is independently selected from a C_{2-10} n-alkanediyl group, a C_{3-6} branched alkanediyl group, a C_{6-8} cycloalkanediyl group, a C_{6-10} alkanecycloalkanediyl group, a heterocyclic group, a $-[(CHR^3)_p-X]_q-(CHR^3)_r-$ group, wherein each R^3 is independently selected from hydrogen and methyl; (ii) each R^2 is independently selected from a C_{2-10} n-alkanediyl group, a C_{3-6} branched alkanediyl group, a C_{6-8} cycloalkanediyl group, a C_{6-14} alkanecycloalkanediyl group, a heterocyclic group, and a $-[(CH_2)_p-X]_q-(CH_2)_r-$ group; (iii) each X is independently selected from O, S, and a $-NR^6-$ group, in which R^6 is selected from H and a methyl group; (iv) m ranges from 0 to 50; (v) n is an integer ranging from 1 to 60; (vi) p is an integer ranging from 2 to 6; (vii) q is an integer ranging from 1 to 5; and (viii) r is an integer ranging from 2 to 10; and (b) at least two terminal Michael acceptor groups.

[044] In certain embodiments of a compound of Formula (1), R^1 is $-[(CHR^3)_s-X]_q-(CHR^3)_r-$ wherein each X is independently selected from $-O-$ and $-S-$. In certain embodiments wherein R^1 is $-[(CHR^3)_s-X]_q-(CHR^3)_r-$, each X is $-O-$ and in certain embodiments, each X is $-S-$.

[045] In certain embodiments of a compound of Formula (1), R^1 is $[-(CH_2)_s-X-]_q-(CH_2)_r-$ wherein each X is independently selected from $-O-$ and $-S-$. In certain embodiments wherein R^1 is $[-(CH_2)_s-X-]_q-(CH_2)_r-$, each X is $-O-$ and in certain embodiments, each X is $-S-$.

[046] In certain embodiments, R^1 in Formula (3a) is $[-(CH_2)_p-X-]_q-(CH_2)_r-$, where p is 2, X is O, q is 2, r is 2, R^2 is ethanediyl, m is 2, and n is 9.

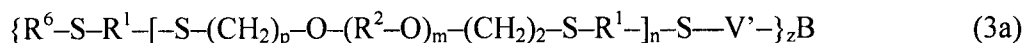
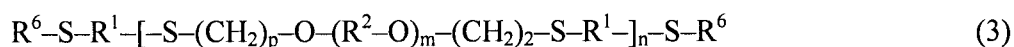
[047] Michael acceptor groups are well known in the art. In certain embodiments, a Michael acceptor group comprises an activated alkene, such as an alkenyl group proximate to an electron-withdrawing group such as an enone, nitro, halo, nitrile, carbonyl, or nitro group. In certain embodiments, a Michael acceptor group is selected from a vinyl ketone, a vinyl sulfone, a quinone, an enamine, a ketimine, an aldimine, and an oxazolidine. In certain embodiments, each of the Michael acceptor groups may be the same and in certain embodiments, at least some of the Michael acceptor groups are different.

[048] In certain embodiments, a Michael acceptor group is derived from a vinyl sulfone and has the structure of Formula (2):



wherein each R^4 is independently selected from hydrogen and C_{1-3} alkyl. In certain embodiments of Formula (2), each R^4 is hydrogen.

[049] In certain embodiments where the sulfur-containing adduct comprises a polythioether adduct, the polythioether adduct is selected from a polythioether adduct of Formula (3), a polythioether adduct of Formula (3a), and a combination thereof:



wherein:

each R^1 independently is selected from C_{2-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-10} alkanecycloalkanediyl, C_{5-8} heterocycloalkanediyl, and $[-(CHR^3)_s-X-]_q-(CHR^3)_r-$, wherein:

s is an integer from 2 to 6;

q is an integer from 1 to 5;

r is an integer from 2 to 10;

each R^3 is independently selected from hydrogen and methyl; and

each X is independently selected from $-O-$, $-S-$, and $-NR-$, wherein R is selected from hydrogen and methyl;

each R^2 is independently selected from C_{1-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-14} alkanecycloalkanediyl, and $-[(-CHR^3-)_s-X-]_q-(-CHR^3-)_r-$, wherein s, q, r, R^3 , and X are as defined for R^1 ;

m is an integer from 0 to 50;

n is an integer from 1 to 60;

p is an integer from 2 to 6;

B represents a core of a z-valent, vinyl-terminated polyfunctionalizing agent $B(-V)_z$ wherein:

z is an integer from 3 to 6; and

each V is a group comprising a terminal vinyl group; and

each $-V'$ is derived from the reaction of $-V$ with a thiol; and

each R^6 is independently a moiety comprising a terminal Michael acceptor group.

[050] In certain embodiments of Formula (3) and in Formula (3a), R^1 is $-[(-CH_2-)_p-X-]_q-(CH_2)_r-$, where p is 2, X is $-O-$, q is 2, r is 2, R^2 is ethanediyl, m is 2, and n is 9.

[051] In certain embodiments of Formula (3) and Formula (3a), R^1 is selected from C_{2-6} alkanediyl and $-[(-CHR^3)_s-X-]_q-(CHR^3)_r-$.

[052] In certain embodiments of Formula (3) and Formula (3a), R^1 is $-[(-CHR^3)_s-X-]_q-(CHR^3)_r-$, and in certain embodiments X is $-O-$ and in certain embodiments, X is $-S-$.

[053] In certain embodiments of Formula (3) and Formula (3a), where R^1 is $-[(-CHR^3)_s-X-]_q-(CHR^3)_r-$, p is 2, r is 2, q is 1, and X is $-S-$; in certain embodiments, wherein p is 2, q is 2, r is 2, and X is $-O-$; and in certain embodiments, p is 2, r is 2, q is 1, and X is $-O-$.

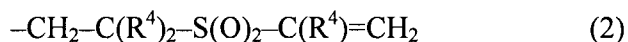
[054] In certain embodiments of Formula (3) and Formula (3a), where R^1 is $-[(-CHR^3)_s-X-]_q-(CHR^3)_r-$, each R^3 is hydrogen, and in certain embodiments, at least one R^3 is methyl.

[055] In certain embodiment of adducts of Formula (3) and Formula (3a), each R^1 is the same, and in certain embodiments, at least one R^1 is different.

[056] In certain embodiments of adducts of Formula (3) and Formula (3a), each R^6 is independently selected from a vinyl ketone, a vinyl sulfone, a quinone, an enamine, a ketimine, an aldimine, and an oxazolidine. In certain embodiments, each of the Michael acceptor groups

may be the same and in certain embodiments, at least some of the Michael acceptor groups are different.

[057] In certain embodiments of adducts of Formula (3) and Formula (3a), each R⁶ is independently derived from a vinyl sulfone and has the structure of Formula (2):



wherein each R⁴ is independently selected from hydrogen and C₁₋₃ alkyl. In certain embodiments of compounds of Formula (3) and Formula (3a) where each R⁶ is a moiety of Formula (2), each R⁴ is hydrogen.

[058] In certain embodiments, a sulfur-containing adduct comprises a polysulfide adduct comprising at least two terminal Michael acceptor groups.

[059] As used herein, the term polysulfide refers to a polymer that contains one or more disulfide linkages, i.e., $-\text{S}-\text{S}-$ linkages, in the polymer backbone and/or in pendant positions on the polymer chain. In certain embodiments, the polysulfide polymer will have two or more sulfur-sulfur linkages. Suitable polysulfides are commercially available, for example, from Akzo Nobel and Toray Fine Chemicals under the names Thiokol-LP and Thioplast®. Thioplast® products are available in a wide range of molecular weights ranging, for example, from less than 1,100 to over 8,000, with molecular weight being the average molecular weight in grams per mole. In some cases, the polysulfide has a number average molecular weight of 1,000 to 4,000. The crosslink density of these products also varies, depending on the amount of crosslinking agent used. The $-\text{SH}$ content, i.e., thiol or mercaptan content, of these products can also vary. The mercaptan content and molecular weight of the polysulfide can affect the cure speed of the polymer, with cure speed increasing with molecular weight.

[060] In certain embodiments provided by the present disclosure, a polysulfide composition comprises: (a) from 90 mole percent to 25 mole percent of mercaptan terminated disulfide polymer of the Formula $\text{HS}(\text{RSS})_m\text{R}-\text{SH}$; and (b) from 10 mole percent to 75 mole percent of diethyl formal mercaptan terminated polysulfide polymer of the Formula $\text{HS}(\text{RSS})_n\text{R}-\text{SH}$, wherein R is $-\text{C}_2\text{H}_4-\text{O}-\text{CH}_2-\text{O}-\text{C}_2\text{H}_4-$; R is a divalent member selected from alkyl of from 2 to 12 carbon atoms, alkyl thioether of from 4 to 20 carbon atoms, alkyl ether of from 4 to 20 carbon atoms and one oxygen atom, alkyl ether of from 4 to 20 carbon atoms and from 2 to 4 oxygen atoms each of which is separated from the other by at least 2 carbon atoms, alicyclic of from 6 to 12 carbon atoms, and aromatic lower alkyl; and the value of m and n is such that the

diethyl formal mercaptan terminated polysulfide polymer and the mercaptan terminated disulfide polymer have an average molecular weight of from 1,000 Daltons to 4,000 Daltons, such as 1,000 Daltons to 2,500 Daltons. Such polymeric mixtures are described in U.S. Patent No. 4,623,711 at col. 4, line 18 to col. 8, line 35. In some cases, R in the above formula is $-\text{CH}_2-\text{CH}_2-$; $-\text{C}_2\text{H}_4-\text{O}-\text{C}_2\text{H}_4-$; $-\text{C}_2\text{H}_4-\text{S}-\text{C}_2\text{H}_4-$; $-\text{C}_2\text{H}_4-\text{O}-\text{C}_2\text{H}_4-\text{O}-\text{C}_2\text{H}_4-$; or $-\text{CH}_2-\text{C}_6\text{H}_4-\text{CH}_2-$.

[061] In certain embodiments, a sulfur-containing adduct comprises a polythioether adduct comprising at least two terminal Michael acceptor groups, a polysulfide adduct comprising at least two terminal Michael acceptor groups, or a combination thereof.

[062] In certain embodiments, sulfur-containing Michael acceptor adducts provided by the present disclosure comprise the reaction products of reactants comprising: (a) a sulfur-containing polymer; and (b) a compound having a Michael acceptor group and a group that is reactive with a terminal group of the sulfur-containing polymer.

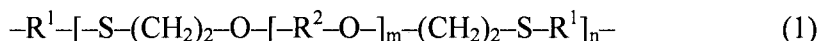
[063] In certain embodiments, the sulfur-containing polymer is selected from a polythioether and a polysulfide, and a combination thereof. In certain embodiments a sulfur-containing polymer comprises a polythioether, and in certain embodiments, a sulfur-containing polymer comprises a polysulfide. A sulfur-containing polymer may comprise a mixture of different polythioethers and/or polysulfides, and the polythioethers and/or polysulfides may have the same or different functionality. In certain embodiments, a sulfur-containing polymer has an average functionality from 2 to 6, from 2 to 4, from 2 to 3, and in certain embodiments, from 2.05 to 2.5. For example, a sulfur-containing polymer can be selected from a difunctional sulfur-containing polymer, a trifunctional sulfur-containing polymer, and a combination thereof.

[064] In certain embodiments, a sulfur-containing polymer is terminated with a group that is reactive with the terminal reactive group of the compound (b). In certain embodiments, the compound having a Michael acceptor group has two Michael acceptor groups, and the terminal groups of the sulfur-containing polymer are reactive with Michael acceptor groups such as a thiol group. A sulfur-containing polymer may comprise terminal thiol groups, terminal alkenyl groups, or terminal epoxy groups.

[065] In certain embodiments, a sulfur-containing polymer is thiol-terminated. Examples of thiol-functional polythioethers are disclosed, for example in U.S. Patent No. 6,172,179. In certain embodiments, a thiol-functional polythioether comprises Permapol® P3.1E, available from PRC-DeSoto International Inc., Sylmar, CA.

[066] In certain embodiments, a sulfur-containing polymer comprises a polythioether comprising:

(a) a backbone comprising the structure of Formula (1):



wherein:

(i) each R^1 is independently selected from a C_{2-10} n-alkanediyl group, a C_{3-6} branched alkanediyl group, a C_{6-8} cycloalkanediyl group, a C_{6-10} alkanecycloalkanediyl group, a heterocyclic group, a $-[(CHR^3)_p-X]_q-(CHR^3)_r-$ group, wherein each R^3 is selected from hydrogen and methyl;

(ii) each R^2 is independently selected from a C_{2-10} n-alkanediyl group, a C_{3-6} branched alkanediyl group, a C_{6-8} cycloalkanediyl group, a C_{6-14} alkanecycloalkanediyl group, a heterocyclic group, and a $-[(CH_2)_p-X]_q-(CH_2)_r-$ group;

(iii) each X is independently selected from O, S, and a $-NR^6-$ group, in which R^6 is selected from H and a methyl group;

(iv) m ranges from 0 to 50;

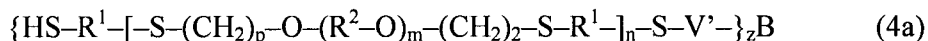
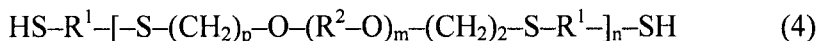
(v) n is an integer ranging from 1 to 60;

(vi) p is an integer ranging from 2 to 6;

(vii) q is an integer ranging from 1 to 5; and

(viii) r is an integer ranging from 2 to 10.

[067] In certain embodiments, a sulfur-containing polymer comprises a polythioether selected from a polythioether of Formula (4), a polythioether of Formula (4a), and a combination thereof:



wherein:

each R^1 independently is selected from C_{2-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-14} alkanecycloalkanediyl, C_{5-8} heterocycloalkanediyl, and $-[(CHR^3)_s-X]_q-(CHR^3)_r-$, wherein:

s is an integer from 2 to 6;

q is an integer from 1 to 5;

r is an integer from 2 to 10;

each R^3 is independently selected from hydrogen and methyl; and

each X is independently selected from $-O-$, $-S-$, and $-NR-$, wherein R is selected from hydrogen and methyl;

each R^2 is independently selected from C_{1-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-14} alkanecycloalkanediyl, and $-[(-CHR^3-)_s-X-]_q-(-CHR^3-)_r-$, wherein s, q, r, R^3 , and X are as defined as for R^1 ;

m is an integer from 0 to 50;

n is an integer from 1 to 60;

p is an integer from 2 to 6;

B represents a core of a z-valent, vinyl-terminated polyfunctionalizing agent $B(-V)_z$ wherein:

z is an integer from 3 to 6; and

each V is a group comprising a terminal vinyl group; and

each $-V'-$ is derived from the reaction of $-V$ with a thiol.

[068] In certain embodiments, Formula (4) and in Formula (4a), R^1 is $-[(-CH_2-)_p-X-]_q-(CH_2)_r-$, where p is 2, X is $-O-$, q is 2, r is 2, R^2 is ethanediyl, m is 2, and n is 9.

[069] In certain embodiments of Formula (4) and Formula (4a), R^1 is selected from C_{2-6} alkanediyl and $-[(-CHR^3-)_s-X-]_q-(CHR^3)_r-$.

[070] In certain embodiments of Formula (4) and Formula (4a), R^1 is $-[(-CHR^3-)_s-X-]_q-(CHR^3)_r-$, and in certain embodiments X is $-O-$ and in certain embodiments, X is $-S-$.

[071] In certain embodiments of Formula (4) and Formula (4a), where R^1 is $-[(-CHR^3-)_s-X-]_q-(CHR^3)_r-$, p is 2, r is 2, q is 1, and X is $-S-$; in certain embodiments, wherein p is 2, q is 2, r is 2, and X is $-O-$; and in certain embodiments, p is 2, r is 2, q is 1, and X is $-O-$.

[072] In certain embodiments of Formula (4) and Formula (4a), where R^1 is $-[(-CHR^3-)_s-X-]_q-(CHR^3)_r-$, each R^3 is hydrogen, and in certain embodiments, at least one R^3 is methyl.

[073] In certain embodiments of Formula (4) and Formula (4a), each R^1 is the same, and in certain embodiments, at least one R^1 is different.

[074] Various methods can be used to prepare such polythioethers. Examples of suitable thiol-functional polythioethers, and methods for their production, are described in U.S.

Patent No. 6,172,179 at col. 2, line 29 to col. 4, line 22; col. 6, line 39 to col. 10, line 50; and col. 11, lines 65 to col. 12, line 22. Such thiol-functional polythioethers may be difunctional, that is, linear polymers having two thiol terminal groups, or polyfunctional, that is, branched polymers have three or more thiol terminal groups. Suitable thiol-functional polythioethers are commercially available, for example, as Permapol® P3.1E, from PRC-DeSoto International Inc., Sylmar, CA.

[075] Suitable thiol-functional polythioethers may be produced by reacting a divinyl ether or mixtures of divinyl ethers with an excess of dithiol or a mixtures of dithiols. For example, dithiols suitable for use in preparing thiol-functional polythioethers include those having Formula (5), other dithiols disclosed herein, or combinations of any of the dithiols disclosed herein.

[076] In certain embodiments, a dithiol has the structure of Formula (5):



wherein:

R^1 is selected from C_{2-6} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-10} alkanecycloalkanediyl, C_{5-8} heterocycloalkanediyl, and $-\text{[(CHR}^3\text{)}_s\text{-X-]}_q\text{-(CHR}^3\text{)}_r\text{-}$;

wherein:

each R^3 is independently selected from hydrogen and methyl;
 each X is independently selected from $-\text{O}-$, $-\text{S}-$, and $-\text{NR}-$ wherein R is selected from hydrogen and methyl;
 s is an integer from 2 to 6;
 q is an integer from 1 to 5; and
 r is an integer from 2 to 10.

[077] In certain embodiments of a dithiol of Formula (5), R^1 is $-\text{[(CHR}^3\text{)}_s\text{-X-]}_q\text{-(CHR}^3\text{)}_r\text{-}$.

[078] In certain embodiments of a compound of Formula (5), X is selected from $-\text{O}-$ and $-\text{S}-$, and thus $-\text{[(CHR}^3\text{)}_s\text{-X-]}_q\text{-(CHR}^3\text{)}_r\text{-}$ in Formula (5) is $-\text{[(CHR}^3\text{)}_p\text{-O-]}_q\text{-(CHR}^3\text{)}_r\text{-}$ or $-\text{[(CHR}^3\text{)}_p\text{-S-]}_q\text{-(CHR}^3\text{)}_r\text{-}$. In certain embodiments, p and r are equal, such as where p and r are both two.

[079] In certain embodiments of a dithiol of Formula (5), R^1 is selected from C_{2-6} alkanediyl and $[-(CHR^3)_s-X-]_q-(CHR^3)_r-$.

[080] In certain embodiments, R^1 is $[-(CHR^3)_s-X-]_q-(CHR^3)_r-$, and in certain embodiments X is $-O-$, and in certain embodiments, X is $-S-$.

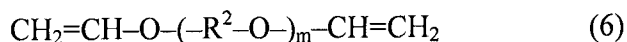
[081] In certain embodiments where R^1 is $[-(CHR^3)_s-X-]_q-(CHR^3)_r-$, p is 2, r is 2, q is 1, and X is $-S-$; in certain embodiments, wherein p is 2, q is 2, r is 2, and X is $-O-$; and in certain embodiments, p is 2, r is 2, q is 1, and X is $-O-$.

[082] In certain embodiments where R^1 is $[-(CHR^3)_s-X-]_q-(CHR^3)_r-$, each R^3 is hydrogen, and in certain embodiments, at least one R^3 is methyl.

[083] Examples of suitable dithiols include, for example, 1,2-ethanedithiol, 1,2-propanedithiol, 1,3-propanedithiol, 1,3-butanedithiol, 1,4-butanedithiol, 2,3-butanedithiol, 1,3-pentanedithiol, 1,5-pentanedithiol, 1,6-hexanedithiol, 1,3-dimercapto-3-methylbutane, dipentenedimercaptan, ethylcyclohexyldithiol (ECHDT), dimercaptodiethylsulfide, methyl-substituted dimercaptodiethylsulfide, dimethyl-substituted dimercaptodiethylsulfide, dimercaptodioxaoctane, 1,5-dimercapto-3-oxapentane, and a combination of any of the foregoing. A polythiol may have one or more pendant groups selected from a lower (e.g., C_{1-6}) alkyl group, a lower alkoxy group, and a hydroxyl group. Suitable alkyl pendant groups include, for example, C_{1-6} linear alkyl, C_{3-6} branched alkyl, cyclopentyl, and cyclohexyl.

[084] Other examples of suitable dithiols include dimercaptodiethylsulfide (DMDS) (in Formula (5), R^1 is $[-(CH_2)_p-X-]_q-(CH_2)_r-$, wherein p is 2, r is 2, q is 1, and X is $-S-$); dimercaptodioxaoctane (DMDO) (in Formula (5), R^1 is $[-(CH_2)_p-X-]_q-(CH_2)_r-$, wherein p is 2, q is 2, r is 2, and X is $-O-$); and 1,5-dimercapto-3-oxapentane (in Formula (5), R^1 is $[-(CH_2)_p-X-]_q-(CH_2)_r-$, wherein p is 2, r is 2, q is 1, and X is $-O-$). It is also possible to use dithiols that include both heteroatoms in the carbon backbone and pendant alkyl groups, such as methyl groups. Such compounds include, for example, methyl-substituted DMDS, such as $HS-CH_2CH(CH_3)-S-CH_2CH_2-SH$, $HS-CH(CH_3)CH_2-S-CH_2CH_2-SH$ and dimethyl substituted DMDS, such as $HS-CH_2CH(CH_3)-S-CH(CH_3)CH_2-SH$ and $HS-CH(CH_3)CH_2-S-CH_2CH(CH_3)-SH$.

[085] Suitable divinyl ethers for preparing polythioethers and polythioether adducts include, for example, divinyl ethers of Formula (6):



where R^2 in Formula (6) is selected from a C_{2-6} n-alkanediyl group, a C_{3-6} branched alkanediyl group, a C_{6-8} cycloalkanediyl group, a C_{6-10} alkanecycloalkanediyl group, and $-[(-CH_2-)_p-O-]_q-(-CH_2-)_r-$, where p is an integer ranging from 2 to 6, q is an integer from 1 to 5, and r is an integer from 2 to 10. In certain embodiments of a divinyl ether of Formula (6), R^2 is a C_{2-6} n-alkanediyl group, a C_{3-6} branched alkanediyl group, a C_{6-8} cycloalkanediyl group, a C_{6-10} alkanecycloalkanediyl group, and in certain embodiments, $-[(-CH_2-)_p-O-]_q-(-CH_2-)_r-$.

[086] Suitable divinyl ethers include, for example, compounds having at least one oxyalkanediyl group, such as from 1 to 4 oxyalkanediyl groups, i.e., compounds in which m in Formula (6) is an integer ranging from 1 to 4. In certain embodiments, m in Formula (6) is an integer ranging from 2 to 4. It is also possible to employ commercially available divinyl ether mixtures that are characterized by a non-integral average value for the number of oxyalkanediyl units per molecule. Thus, m in Formula (6) can also take on rational number values ranging from 0 to 10.0, such as from 1.0 to 10.0, from 1.0 to 4.0, or from 2.0 to 4.0.

[087] Examples of suitable divinyl ethers include, for example, divinyl ether, ethylene glycol divinyl ether (EG-DVE) (R^2 in Formula (6) is ethanediyl and m is 1), butanediol divinyl ether (BD-DVE) (R^2 in Formula (6) is butanediyl and m is 1), hexanediol divinyl ether (HD-DVE) (R^2 in Formula (6) is hexanediyl and m is 1), diethylene glycol divinyl ether (DEG-DVE) (R^2 in Formula (4) is ethanediyl and m is 2), triethylene glycol divinyl ether (R^2 in Formula (14) is ethanediyl and m is 3), tetraethylene glycol divinyl ether (R^2 in Formula (6) is ethanediyl and m is 4), cyclohexanedimethanol divinyl ether, polytetrahydrofuryl divinyl ether; trivinyl ether monomers, such as trimethylolpropane trivinyl ether; tetrafunctional ether monomers, such as pentaerythritol tetravinyl ether; and combinations of two or more such polyvinyl ether monomers. A polyvinyl ether may have one or more pendant groups selected from alkyl groups, hydroxyl groups, alkoxy groups, and amine groups.

[088] In certain embodiments, divinyl ethers in which R^2 in Formula (6) is C_{3-6} branched alkanediyl may be prepared by reacting a polyhydroxy compound with acetylene. Examples of divinyl ethers of this type include compounds in which R^2 in Formula (6) is an alkyl-substituted methanediyl group such as $-CH(CH_3)-$ (for example Pluriol® blends such as Pluriol®E-200 divinyl ether (BASF Corp., Parsippany, NJ), for which R^2 in Formula (6) is ethanediyl and m is 3.8) or an alkyl-substituted ethanediyl (for example $-CH_2CH(CH_3)-$ such as DPE polymeric blends including DPE-2 and DPE-3 (International Specialty Products, Wayne, NJ)).

[089] Other useful divinyl ethers include compounds in which R^2 in Formula (6) is polytetrahydrofuryl (poly-THF) or polyoxyalkanediyl, such as those having an average of about 3 monomer units.

[090] Two or more types of polyvinyl ether monomers of Formula (6) may be used. Thus, in certain embodiments, two dithiols of Formula (5) and one polyvinyl ether monomer of Formula (6), one dithiol of Formula (5) and two polyvinyl ether monomers of Formula (6), two dithiols of Formula (5) and two divinyl ether monomers of Formula (6), and more than two compounds of one or both Formula (5) and Formula (6), may be used to produce a variety of thiol-functional polythioethers.

[091] In certain embodiments, a polyvinyl ether monomer comprises 20 to less than 50 mole percent of the reactants used to prepare a thiol-functional polythioether, and in certain embodiments, 30 to less than 50 mole percent.

[092] In certain embodiments provided by the present disclosure, relative amounts of dithiols and divinyl ethers are selected to yield polythioethers having terminal thiol groups. Thus, a dithiol of Formula (5) or a mixture of at least two different dithiols of Formula (5), are reacted with of a divinyl ether of Formula (6) or a mixture of at least two different divinyl ethers of Formula (6) in relative amounts such that the molar ratio of thiol groups to vinyl groups is greater than 1:1, such as 1.1 to 2.0:1.0.

[093] The reaction between compounds of dithiols and divinyl ethers may be catalyzed by a free radical catalyst. Suitable free radical catalysts include, for example, azo compounds, for example azobisnitriles such as azo(bis)isobutyronitrile (AIBN); organic peroxides such as benzoyl peroxide and t-butyl peroxide; and inorganic peroxides such as hydrogen peroxide. The catalyst may be a free-radical catalyst, an ionic catalyst, or ultraviolet radiation. In certain embodiments, the catalyst does not comprise acidic or basic compounds, and does not produce acidic or basic compounds upon decomposition. Examples of free-radical catalysts include azo-type catalyst, such as Vazo®-57 (Du Pont), Vazo®-64 (Du Pont), Vazo®-67 (Du Pont), V-70® (Wako Specialty Chemicals), and V-65B® (Wako Specialty Chemicals). Examples of other free-radical catalysts are alkyl peroxides, such as t-butyl peroxide. The reaction may also be effected by irradiation with ultraviolet light either with or without a cationic photoinitiating moiety.

[094] Thiol-functional polythioethers provided by the present disclosure may be prepared by combining at least one compound of Formula (5) and at least one compound of

Formula (6) followed by addition of an appropriate catalyst, and carrying out the reaction at a temperature from 30°C to 120°C, such as 70°C to 90°C, for a time from 2 to 24 hours, such as 2 to 6 hours.

[095] As disclosed herein, thiol-terminated polythioethers may comprise a polyfunctional polythioether, i.e., may have an average functionality of greater than 2.0. Suitable polyfunctional thiol-terminated polythioethers include, for example, those having the structure of Formula (7):



wherein: (i) A comprises, for example, a structure of Formula (1), (ii) B denotes a z-valent residue of a polyfunctionalizing agent; and (iii) z has an average value of greater than 2.0, and, in certain embodiments, a value between 2 and 3, a value between 2 and 4, a value between 3 and 6, and in certain embodiments, is an integer from 3 to 6.

[096] Polyfunctionalizing agents suitable for use in preparing such polyfunctional thiol-functional polymers include trifunctionalizing agents, that is, compounds where z is 3. Suitable trifunctionalizing agents include, for example, triallyl cyanurate (TAC), 1,2,3-propanetrithiol, isocyanurate-containing trithiols, and combinations thereof, as disclosed in U.S. Publication No. 2010/0010133 at paragraphs [0102]-[0105]. Other useful polyfunctionalizing agents include trimethylolpropane trivinyl ether, and the polythiols described in U.S. Patent. Nos. 4,366,307; 4,609,762; and 5,225,472. Mixtures of polyfunctionalizing agents may also be used.

[097] As a result, thiol-functional polythioethers suitable for use in embodiments provided by the present disclosure may have a wide range of average functionality. For example, trifunctionalizing agents may afford average functionalities from 2.05 to 3.0, such as from 2.1 to 2.6. Wider ranges of average functionality may be achieved by using tetrafunctional or higher functionality polyfunctionalizing agents. Functionality may also be affected by factors such as stoichiometry, as will be understood by those skilled in the art.

[098] Thiol-functional polythioethers having a functionality greater than 2.0 may be prepared in a manner similar to the difunctional thiol-functional polythioethers described in U.S. Publication No. 2010/0010133. In certain embodiments, polythioethers may be prepared by combining (i) one or more dithiols described herein, with (ii) one or more divinyl ethers described herein, and (iii) one or more polyfunctionalizing agents. The mixture may then be

reacted, optionally in the presence of a suitable catalyst, to afford a thiol-functional polythioether having a functionality greater than 2.0.

[099] Thus, in certain embodiments, a thiol-terminated polythioether comprises the reaction product of reactants comprising:

- (a) a dithiol of Formula (5):



wherein:

R^1 is selected from C_{2-6} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-10} alkanecycloalkanediyl, C_{5-8} heterocycloalkanediyl, and $[-(\text{CHR}^3)_s\text{-X-}]_q\text{-(CHR}^3)_r\text{-}$; wherein:

each R^3 is independently selected from hydrogen and methyl;

each X is independently selected from $-\text{O}-$, $-\text{S}-$, $-\text{NH}-$, and $-\text{NR}-$

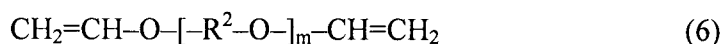
wherein R is selected from hydrogen and methyl;

s is an integer from 2 to 6;

q is an integer from 1 to 5; and

r is an integer from 2 to 10; and

- (b) a divinyl ether of Formula (6):



wherein:

each R^2 is independently selected from C_{1-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-14} alkanecycloalkanediyl, and $[-(\text{CHR}^3)_s\text{-X-}]_q\text{-(CHR}^3)_r\text{-}$, wherein s, q, r, R^3 , and X are as defined above;

m is an integer from 0 to 50;

n is an integer from 1 to 60; and

p is an integer from 2 to 6.

And, in certain embodiments, the reactants comprise (c) a polyfunctional compound such as a polyfunctional compound B(-V)_z , where B, $-\text{V}$, and z are as defined herein.

[0100] Thiol-terminated polythioethers provided by the present disclosure represent thiol-terminated polythioethers having a molecular weight distribution. In certain embodiments, useful thiol-terminated polythioethers can exhibit a number average molecular weight ranging

from 500 Daltons to 20,000 Daltons, in certain embodiments, from 2,000 Daltons to 5,000 Daltons, and in certain embodiments, from 3,000 Daltons to 4,000 Daltons. In certain embodiments, useful thiol-terminated polythioethers exhibit a polydispersity (M_w/M_n ; weight average molecular weight/number average molecular weight) ranging from 1 to 20, and in certain embodiments, from 1 to 5. The molecular weight distribution of thiol-terminated polythioethers may be characterized by gel permeation chromatography.

[0101] In certain embodiments, thiol-functional polythioethers provided by the present disclosure are essentially free, or free, of sulfone, ester and/or disulfide linkages. As used herein, “essentially free of sulfone, ester, and/or disulfide linkages” means that less than 2 mole percent of the linkages in the thiol-functional polymer are sulfone, ester, and/or disulfide linkages. As a result, in certain embodiments, the resulting thiol-functional polythioethers are also essentially free, or free, of sulfone, ester, and/or disulfide linkages.

[0102] To prepare a sulfur-containing Michael acceptor adduct, a sulfur-containing polymer such as those disclosed herein may be reacted with (b) a compound having a group that is reactive with the terminal groups of the sulfur-containing polymer and a Michael acceptor group.

[0103] In certain embodiments, a Michael acceptor group is selected from a vinyl ketone, a vinyl sulfone, a quinone, an enamine, a ketimine, an aldimine, and an oxazolidine. In certain embodiments, a Michael acceptor group is a vinyl ketone, and in certain embodiments, a vinyl sulfone such as divinyl sulfone. In embodiments in which the compound having a Michael acceptor group is divinyl sulfone, the sulfur-containing polymer may be thiol-terminated such as a thiol-terminated polythioether, a thiol-terminated polysulfide, or a combination thereof.

[0104] The reaction between a sulfur-containing polymer and a compound having a Michael acceptor group and a group that is reactive with a terminal group of the sulfur-containing polymer can be performed in the presence of an appropriate catalyst.

[0105] In certain embodiments, compositions provided by the present disclosure comprise a catalyst such as an amine catalyst. For example, in embodiments in which the sulfur-containing polymer is thiol-terminated and the compound is a difunctional Michael acceptor, the reaction may take place in the presence of an amine catalyst. Examples of suitable amine catalysts include, for example, triethylenediamine (1,4-diazabicyclo[2.2.2]octane, DABCO), dimethylcyclohexylamine (DMCHA), dimethylethanolamine (DMEA), bis-(2-dimethylaminoethyl)ether, N-ethylmorpholine, triethylamine, 1,8-diazabicyclo[5.4.0]undecene-7

(DBU), pentamethyldiethylenetriamine (PMDETA), benzyldimethylamine (BDMA), N,N,N'-trimethyl-N'-hydroxyethyl-bis(aminoethyl)ether, and N'-(3-(dimethylamino)propyl)-N,N-dimethyl-1,3-propanediamine.

Compositions

[0106] Michael addition chemistries may be employed in a variety of ways in conjunction with sulfur-containing polymers to provide curable compositions. For example, a curable composition provided by the present disclosure may comprise (a) a sulfur-containing polymer and a Michael acceptor curing agent; (b) a sulfur-containing Michael acceptor adduct and a curing agent comprising at least two terminal groups that are reactive with Michael acceptor groups; or (c) a sulfur-containing polymer and a curing agent comprising a combination of a monomeric Michael acceptor and a sulfur-containing Michael acceptor adduct.

Sulfur-containing polymer and Michael acceptor curing agent

[0107] In certain embodiments, compositions provided by the present disclosure comprise a sulfur-containing polymer and a Michael acceptor curing agent. A sulfur-containing polymer may be a polythioether or combination of polythioethers having terminal groups reactive with the Michael acceptor; a polysulfide or combination of polysulfides having terminal groups reactive with the Michael acceptor; or a combination of any of the foregoing. In certain embodiments, a sulfur-containing polymer is thiol-terminated. In such embodiments, a Michael acceptor will be polyfunctional and have Michael acceptor groups reactive with the terminal groups of the sulfur-containing polymer.

[0108] In certain embodiments, a sulfur-containing polymer comprises a thiol-terminated polythioether, including any of the thiol-terminated polythioethers disclosed herein, such as a thiol-terminated polythioether of Formula (1). In certain embodiments, a sulfur-containing polymer comprises a thiol-terminated polythioether, such as a thiol-terminated polythioether of Formula (4), Formula (4a), or a combination thereof. In certain embodiments, a sulfur-containing polymer is selected from a difunctional sulfur-containing polymer, a trifunctional-containing polymer and a combination of thereof. In certain embodiments, a thiol-terminated polymer comprises a mixture of sulfur-containing polymers having an average functionality from 2 to 3, and in certain embodiments, from 2.2 to 2.8. In certain embodiments, a thiol-terminated polythioether comprises Permapol® 3.1E, available from PRC-DeSoto International.

[0109] A polyfunctional Michael acceptor has at least two Michael acceptor groups. A polyfunctional Michael acceptor may have an average Michael acceptor functionality from 2 to 6, from 2 to 4, from 2 to 3, and in certain embodiments, from 2.05 to 2.5. In certain embodiments, a polyfunctional Michael acceptor is difunctional, such as, divinyl ketone and divinyl sulfone. A Michael acceptor having a functionality greater than two may be prepared by reacting a compound having a Michael acceptor group and a group reactive with terminal groups of a polyfunctionalizing agent such as those disclosed herein, using appropriate reaction conditions.

[0110] In certain embodiments where a Michael acceptor is used as a curing agent, the molecular weight of the Michael acceptor is less than 600 Daltons, less than 400 Daltons, and in certain embodiments, less than 200 Daltons.

[0111] In certain embodiments, a Michael acceptor comprises from about 0.5 wt% to about 20 wt% of the composition, from about 1 wt % to about 10 wt%, from about 2 wt% to about 8 wt%, from about 2 wt% to about 6 wt%, and in certain embodiments, from about 3 wt% to about 5 wt%, where wt% is based on the total dry solids weight of the composition.

Sulfur-containing Michael acceptor adduct and a curing agent

[0112] In certain embodiments, a composition comprises a sulfur-containing Michael acceptor adduct provided by the present disclosure and a sulfur-containing polymer curing agent.

[0113] In such compositions a sulfur-containing adduct comprises any of those disclosed herein. In certain embodiments, a sulfur-containing adduct comprises a polythioether adduct, and in certain embodiments a polythioether adduct has an average functionality from 2 to 3, from 2.2 to 2.8, and in certain embodiments, from 2.4 to 2.6. In certain embodiments, a sulfur-containing adduct has an average functionality of 2.

[0114] In certain embodiments, a sulfur-containing Michael acceptor adduct comprises a compound of Formula (3), Formula (3a), or a combination thereof, and the sulfur-containing polymer curing agent comprises a polythioether of Formula (4), Formula (4a), or a combination thereof. In certain embodiments, the sulfur-containing adduct comprises the Michael acceptor adduct of Permapol® 3.1E. In certain embodiments, the sulfur-containing polymer curing agent comprises Permapol® 3.1E.

[0115] In certain embodiments, a sulfur-containing Michael acceptor adduct comprises a compound of Formula (3), Formula (3a), or a combination thereof, and the sulfur-containing

polymer curing agent comprises a polysulfide. In certain embodiments, the sulfur-containing adduct comprises the Michael acceptor adduct of Permapol® 3.1E. In certain embodiments, the sulfur-containing polymer comprises a polysulfide selected from a Thiokol-LP® polysulfide, a Thioplast® polysulfide, and a combination thereof.

[0116] In such compositions the Michael acceptor groups of the adduct are reactive with the terminal groups of the sulfur-containing polymer. For example, the Michael acceptor groups may be activated alkenyl groups, e.g., Michael acceptor groups, and the sulfur-containing polymer comprises terminal thiol groups.

[0117] A sulfur-containing polymer used as a curing agent comprises at least two terminal groups reactive with Michael acceptor groups. A sulfur-containing polymer used as a curing agent in such compositions may comprise a polythioether including any of those disclosed herein, a polysulfide including any of those disclosed herein, or a combination thereof. The sulfur-containing polymer may have an average functionality of about 2 or any functionality from about 2 and about 6, such as from about 2 to about 4, or from about 2 to about 3.

[0118] In certain embodiments, the sulfur-containing polymer curing agent comprises a thiol-terminated polythioether such as, for example, Permapol® 3.1E. In certain embodiments, the sulfur-containing polymer comprises a thiol-terminated polysulfide such as, for example, a Thiokol-LP® polysulfide, a Thioplast® polysulfide, or a combination thereof.

[0119] In such embodiments, when used as a curing agent, a sulfur-containing polymer, comprises from about 20 wt% to about 90 wt% of the composition, from about 30 wt% to about 80 wt%, from about 40 wt% to about 60 wt%, and in certain embodiments, about 50 wt%, where wt% is based on the total dry weight of the composition.

[0120] In such embodiments, a sulfur-containing Michael acceptor adduct comprises from about 20 wt% to about 90 wt% of the composition, from about 30 wt% to about 80 wt%, from about 40 wt% to about 60 wt%, and in certain embodiments, about 50 wt%, where wt% is based on the total dry weight of the composition.

[0121] Compositions comprising a sulfur-containing Michael acceptor adduct and a sulfur-containing polymer curing agent may comprise a catalyst such as an amine catalyst including any of those disclosed herein.

[0122] In certain embodiments, a composition comprises a polythioether adduct and a curing agent. A polythioether adduct includes any of those disclosed herein, such as polythioether adducts of Formula (3), Formula (3a), and combinations thereof.

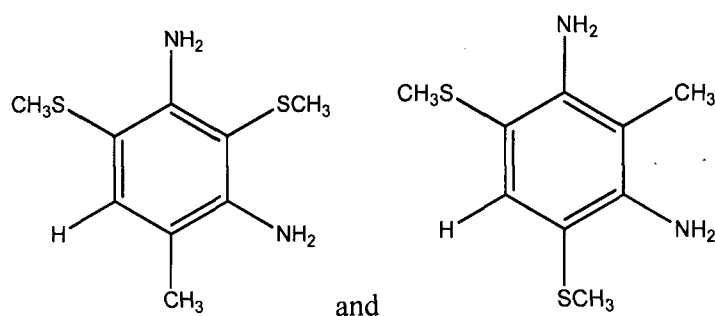
[0123] In certain embodiments of such compositions, the composition comprises a sulfur-containing Michael acceptor adduct provided by the present disclosure and a curing agent selected from a sulfur-containing polymer comprising at least two terminal groups reactive with Michael acceptor groups, a monomeric thiol, a polythiol, a polyamine, a blocked polyamine, and a combination of any of the foregoing. In certain embodiments, a curing agent comprises a sulfur-containing polymer comprising at least two terminal groups reactive with Michael acceptor groups; in certain embodiments a monomeric thiol; in certain embodiments a polythiol; in certain embodiments a polyamine; and in certain embodiments, a blocked polyamine. In certain embodiments of such compositions, a curing agent comprises a sulfur-containing polymer comprising at least two terminal groups reactive with Michael acceptor groups and a compound having at least two terminal groups reactive with Michael acceptor groups selected from a monomeric thiol, a polythiol, a polyamine, a blocked polyamine, and a combination of any of the foregoing.

[0124] In certain embodiments, a sulfur-containing polymer comprising at least two terminal groups reactive with Michael acceptor groups is selected from a polythioether polymer comprising at least two terminal groups reactive with Michael acceptor groups, a polysulfide polymer comprising at least two terminal groups reactive with Michael acceptor groups, and a combination thereof. In certain embodiments, the terminal groups reactive with Michael acceptor groups are terminal thiol groups. In such embodiments, a thiol-terminated polythioether may be selected from a polythioether of Formula (4), a polythioether of Formula (4a), and a combination thereof. In certain embodiments, the sulfur-containing polymer curing agent comprises a thiol-terminated polysulfide such as, for example, Thiokol-LP® and Thioplast® polysulfide polymers.

[0125] In certain compositions, the curing agent comprises a monomeric thiol. A monomeric thiol refers to a compound having at least two terminal thiol groups. Examples of monomeric thiols include dithiols of Formula (5). Polythiols refer to higher molecular weight compounds having terminal thiol groups and thiol groups in the backbone.

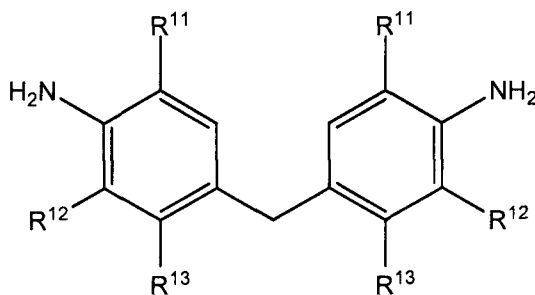
[0126] Examples of polyamines include, for example, aliphatic polyamines, cycloaliphatic polyamines, aromatic polyamines and mixtures thereof. In certain embodiments, the polyamine can include a polyamine having at least two functional groups independently chosen from primary amine ($-NH_2$), secondary amine ($-NH-$) and combinations thereof. In certain embodiments, the polyamine has at least two primary amine groups.

[0127] In certain embodiments, a polyamine is a sulfur-containing polyamine. Examples of suitable sulfur-containing polyamines are isomers of benzenediamine-bis(methylthio)-, such as 1,3-benzenediamine-4-methyl-2,6-bis(methylthio)- and 1,3-benzenediamine-2-methyl-4,6-bis(methylthio)-, having the structure:



[0128] Such sulfur-containing polyamines are commercially available, for example, from Albemarle Corporation under the tradename Ethacure® 300.

[0129] Suitable polyamines also include, for example, polyamines having the following structure:



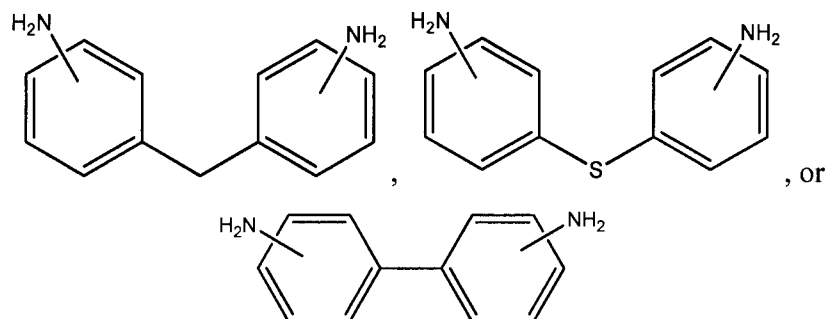
wherein each R^{11} and each R^{12} are independently selected from methyl, ethyl, propyl, and isopropyl groups, and each R^{13} is independently selected from hydrogen and chlorine. Examples of suitable amine-containing curing agents include the following compounds available from Lonza Ltd. (Basel, Switzerland): Lonzacure® M-DIPA, Lonzacure® M-DMA, Lonzacure® M-MEA, Lonzacure® M-DEA, Lonzacure® M-MIPA, Lonzacure® M-CDEA.

[0130] In certain embodiments, a polyamine comprises a diamine, such as 4,4'-methylenebis(3-chloro-2,6-diethylaniline) (Lonzacure® M-CDEA), 2,4-diamino-3,5-diethyltoluene, 2,6-diamino-3,5-diethyltoluene and mixtures thereof (collectively diethyltoluenediamine or DETDA), a sulfur-containing diamine, such as Ethacure® 300, 4,4'-methylene-bis-(2-chloroaniline) and mixtures thereof. Other suitable diamines include 4,4'-methylene-bis(dialkylaniline), 4,4'-methylene-bis(2,6-dimethylaniline), 4,4'-methylene-bis(2,6-

diethylaniline), 4,4'-methylene-bis(2-ethyl-6-methylaniline), 4,4'-methylene-bis(2,6-diisopropylaniline), 4,4'-methylene-bis(2-isopropyl-6-methylaniline), 4,4'-methylene-bis(2,6-diethyl-3-chloroaniline), and combinations of any of the foregoing.

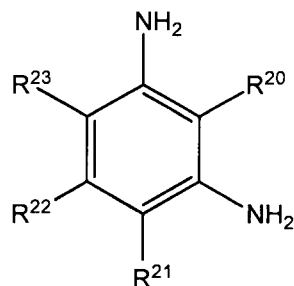
[0131] Further, examples of suitable polyamines include ethyleneamines, such as, ethylenediamine (EDA), diethylenetriamine (DETA), triethylenetetramine (TETA), tetraethylenepentamine (TEPA), pentaethylenehexamine (PEHA), piperazine, morpholine, substituted morpholine, piperidine, substituted piperidine, diethylenediamine (DEDA), 2-amino-1-ethylpiperazine, and combinations thereof. In certain embodiments, a polyamine may be selected from one or more isomers of C₁₋₃ dialkyl toluenediamine, such as, 3,5-dimethyl-2,4-toluenediamine, 3,5-dimethyl-2,6-toluenediamine, 3,5-diethyl-2,4-toluenediamine, 3,5-diethyl-2,6-toluenediamine, 3,5-diisopropyl-2,4-toluenediamine, 3,5-diisopropyl-2,6-toluenediamine, and combinations thereof. In certain embodiments, a polyamine may be selected from methylene dianiline, trimethyleneglycol di(para-aminobenzoate), and combinations thereof.

[0132] In certain embodiments, a polyamine includes a compound having the structure:



[0133] In certain embodiments, a polyamine includes one or more methylene bis anilines, one or more aniline sulfides, and/or one or more biphenylamines which can be represented by the general structures disclosed, for example, in paragraph [0072] of U.S. Publication No. 2011/0092639.

[0134] In certain embodiments, a polyamine includes compounds represented by the general structure:



where R²⁰, R²¹, R²², and R²³ are independently selected from C₁₋₃ alkyl, CH₃-S- and halogen, such as but not limited to chlorine or bromine. In certain embodiments, a polyamine represented by the immediately preceding structure can be diethyl toluene diamine (DETDA) wherein R²³ is methyl, R²⁰ and R²¹ are each ethyl, and R²² is hydrogen. In certain embodiments, the polyamine is 4,4'-methylenedianiline.

[0135] Examples of blocked polyamines include ketimines, enamines, oxazolidines, aldimines, and imidazolidines. In certain embodiments, the blocked polyamine is Vestamin® A139.

**Sulfur-containing polymer adduct, sulfur-containing polymer,
and a compound having at least two Michael acceptor groups**

[0136] In certain embodiments, a composition comprises a sulfur-containing polymer, and a sulfur-containing Michael acceptor adduct. In certain embodiments, a composition comprises a sulfur-containing polymer, a polyfunctional Michael acceptor, and a sulfur-containing Michael acceptor adduct.

[0137] In such compositions, a sulfur-containing polymer comprises at least two terminal groups reactive with Michael acceptor groups. In such compositions, the sulfur-containing polymer may be selected from a polythioether polymer, a polysulfide polymer, or a combination thereof, including a suitable polythioether polymer or polysulfide polymer provided by the present disclosure.

[0138] In certain embodiments, a sulfur-containing polymer is selected such that the terminal groups are reactive with the polyfunctional Michael acceptor and the sulfur-containing Michael acceptor adduct. In certain embodiments, a sulfur-containing polymer comprises terminal thiol groups including any of the thiol-terminated polythioethers, thiol-terminated polysulfides, and combinations thereof disclosed herein.

[0139] In certain embodiments of such compositions, a sulfur-containing polymer adduct comprises a polythioether polymer adduct provided by the present disclosure, a polysulfide polymer adduct provided by the present disclosure, or a combination thereof.

[0140] When a composition comprises a polyfunctional monomeric Michael acceptor, any suitable monomeric Michael acceptor having at least two Michael acceptor groups such as, for example, divinyl sulfone or other Michael acceptors including any of those disclosed herein may be used.

[0141] In certain embodiments, a sulfur-containing polymer is selected from a polythioether of Formula (3), Formula (3a), and a combination thereof; a polyfunctional Michael acceptor adduct is selected from an adduct of Formula (4), Formula (4a), and a combination thereof; and a polyfunctional monomeric Michael acceptor is selected from a compound having two or more activated alkenyl groups such as a vinyl ketone or a vinyl sulfone, such as divinyl sulfone.

[0142] In such embodiments, the polyfunctional Michael acceptor and Michael acceptor adduct comprise 10 wt% to 90 wt% of the composition, from 20 wt% to 80 wt%, from 30 wt% to 70 wt%, and in certain embodiments, from 40 wt% to 60 wt%, where wt% is based on the total dry solids weight of the composition.

[0143] Compositions comprising a sulfur-containing polymer, a polyfunctional Michael acceptor, and a sulfur-containing polymer adduct may comprise a catalyst such as an amine catalyst including any of those disclosed herein.

Epoxy blend

[0144] In certain embodiments, compositions provided by the present disclosure comprise an epoxy curing agent. Thus, in addition to a Michael acceptor curing agent, a sulfur-containing polymer curing agent, and/or a sulfur-containing Michael acceptor adduct curing agent, a composition may comprise one or more polyepoxy curing agents. Examples of suitable epoxies include, for example, polyepoxide resins such as hydantoin diepoxide, diglycidyl ether of bisphenol-A, diglycidyl ether of bisphenol-F, Novolac® type epoxides such as DENT™ 438 (available from Dow), certain epoxidized unsaturated resins, and combinations of any of the foregoing. A polyepoxide refers to a compound having two or more reactive epoxy groups.

[0145] In certain embodiments, a polyepoxy curing agent comprises an epoxy-functional polymer. Examples of suitable epoxy-functional polymers include the epoxy-functional

polyformal polymers disclosed in U.S. Patent No. 8,541,513 and epoxy-functional polythioether polymers disclosed in U.S. Patent No. 7,671,145. In general, when used as a curing agent, an epoxy-functional polymer has a molecular weight less than about 2,000 Daltons, less than about 1,500, Daltons, less than about 1,000 Daltons, and in certain embodiments, less than about 500 Daltons.

[0146] In such compositions, an epoxy may comprise about 0.5 wt% to about 20 wt% of the composition, from about 1 wt % to about 10 wt%, from about 2 wt% to about 8 wt%, from about 2 wt% to about 6 wt%, and in certain embodiments, from about 3 wt% to about 5 wt%, where wt% is based on the total solids weight of the composition.

Isocyanate blend

[0147] In certain embodiments, compositions provided by the present disclosure comprise an isocyanate curing agent. Thus, in addition to a Michael acceptor curing agent, a sulfur-containing polymer curing agent, and/or a sulfur-containing Michael acceptor adduct curing agent, a composition may comprise one or more polyisocyanate curing agents that are reactive with thiol groups but not reactive with Michael acceptor groups such as vinyl sulfone groups. Examples of suitable isocyanate curing agents include allyl isocyanate, 3-isopropenyl- α,α -dimethylbenzyl isocyanate, toluene diisocyanate, and combinations of any of the foregoing. Isocyanate curing agents are commercially available and include, for example, products under the tradenames Baydur® (Bayer MaterialScience), Desmodur® (Bayer MaterialScience), Solubond® (DSM), ECCO (ECCO), Vestanat® (Evonik), Irodur® (Huntsman), Rhodocoat™ (Perstorp), and Vanchem® (V.T. Vanderbilt). In certain embodiments, a polyisocyanate curing agent comprises isocyanate groups that are reactive with thiol groups and that are less reactive with Michael acceptor groups.

[0148] In certain embodiments, an isocyanate curing agent comprises an isocyanate-functional polymer. Examples of suitable isocyanate-functional polymers include the isocyanate-functional polyformal polymers disclosed in U.S. Patent No. 8,507,617. In general, when used as a curing agent, an isocyanate-functional polymer has a molecular weight less than about 2,000 Daltons, less than about 1,500, Daltons, less than about 1,000 Daltons, and in certain embodiments, less than about 500 Daltons.

[0149] In such compositions, an epoxy may comprise about 0.5 wt% to about 20 wt% of the composition, from about 1 wt % to about 10 wt%, from about 2 wt% to about 8 wt%, from

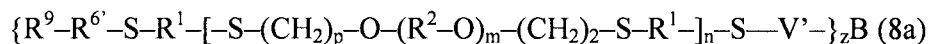
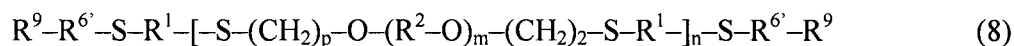
about 2 wt% to about 6 wt%, and in certain embodiments, from about 3 wt% to about 5 wt% of the composition, where wt% is based on the total solids weight of the composition.

Hydroxyl and amine curing

[0150] Sulfur-containing Michael acceptor adducts provided by the present disclosure may also be modified for use in particular applications and curing chemistries. For example, spray seal applications require rapid curing without heating. Amine-based systems using epoxy curing agents are well suited for such applications. Accordingly, sulfur-containing Michael acceptor adducts may be adapted to other curing chemistries by modifying or capping the terminal Michael acceptor groups with, for example, hydroxyl groups or amine groups.

[0151] Hydroxyl-terminated sulfur-containing adducts may be prepared by reacting a sulfur-containing Michael acceptor adduct provided by the present disclosure such as an adduct of Formulae (1), Formula (3), or Formula (3a), and a compound having a terminal thiol group and a terminal hydroxyl group. In certain embodiments, a compound having a terminal thiol group and a terminal hydroxyl group has the structure $\text{HS-R}^{11}\text{-OH}$, where R^{11} is selected from C_{2-6} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-10} alkanecycloalkanediyl, C_{5-8} heterocycloalkanediyl, C_{6-8} arenediyl, C_{6-10} alkanearenediyl, C_{5-8} heteroarenediyl, and $-\text{[(CHR}^3\text{)}_s\text{-X-]}_q\text{-(CHR}^3\text{)}_r\text{-}$, where q , r , s , X , and R^3 are defined as for Formula (5). In certain embodiments, a sulfur-containing adduct is derived from Permapol® 3.1E. The reaction may take place in the presence of a catalyst at a temperature from about 25°C to about 50°C.

[0152] In certain embodiments, a hydroxyl-terminated sulfur-containing adduct comprises a hydroxyl-terminated polythioether adduct of Formula (8), a hydroxyl-terminated polythioether adduct of Formula (8a), and a combination thereof:



wherein:

each R^1 independently is selected from C_{2-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-10} alkanecycloalkanediyl, C_{5-8} heterocycloalkanediyl, and $-\text{[(CHR}^3\text{)}_s\text{-X-]}_q\text{-(CHR}^3\text{)}_r\text{-}$,

wherein:

s is an integer from 2 to 6;

q is an integer from 1 to 5;

r is an integer from 2 to 10;

each R^3 is independently selected from hydrogen and methyl; and

each X is independently selected from $-O-$, $-S-$, and $-NR-$, wherein R is selected from hydrogen and methyl;

each R^2 is independently selected from C_{1-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-14} alkanecycloalkanediyl, and $-[(-CHR^3-)_s-X]_q-(-CHR^3-)_r-$, wherein s , q , r , R^3 , and X are as defined for R^1 ;

m is an integer from 0 to 50;

n is an integer from 1 to 60;

p is an integer from 2 to 6;

B represents a core of a z -valent, vinyl-terminated polyfunctionalizing agent $B(-V)_z$ wherein:

z is an integer from 3 to 6; and

each V is a group comprising a terminal vinyl group;

each $-V'$ is derived from the reaction of $-V$ with a thiol;

each $-R^6-$ is $-CH_2-C(R^4)_2-S(O)_2-C(R^4)_2-CH_2-$, wherein each R^4 is independently selected from hydrogen and C_{1-3} alkyl; and

each R^9- is a moiety having a terminal hydroxyl group.

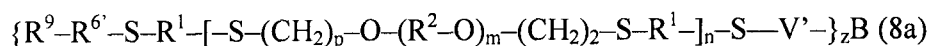
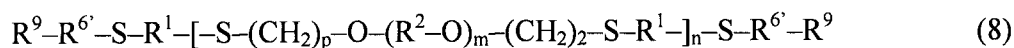
[0153] In certain embodiments of Formula (8) and Formula (8a), R^9 is $-S-R^{11}-OH$, wherein R^{11} is defined herein.

[0154] In certain embodiments, compositions comprise one or more hydroxyl-terminated sulfur-containing adducts and one or more polyisocyanate curing agents. Examples of suitable isocyanate curing agents include toluene diisocyanate, and combinations of any of the foregoing. Isocyanate curing agents are commercially available and include, for example, products under the tradenames Baydur® (Bayer MaterialScience), Desmodur® (Bayer MaterialScience), Solubond® (DSM), ECCO (ECCO), Vestanat® (Evonik), Irodur® (Huntsman), Rhodocoat™ (Perstorp), and Vanchem® (V.T. Vanderbilt).

[0155] Amine-terminated sulfur-containing adducts may be prepared by reacting a sulfur-containing Michael accepter adduct provided by the present disclosure such as an adduct of Formulae (1), (3), or (3a), and a compound having a terminal thiol group and a terminal amine group. In certain embodiments, a compound having a terminal thiol group and a terminal hydroxyl group has the structure $HS-R^{11}-N(R^{12})H$, where R^{11} is selected from C_{2-6} alkanediyl, C_{6-8}

cycloalkanediyl, C₆₋₁₀ alkanecycloalkanediyl, C₅₋₈ heterocycloalkanediyl, C₆₋₈ arenediyl, C₆₋₁₀ alkanerenediyl, C₅₋₈ heteroarenediyl, and $[-(\text{CHR}^3)_s\text{-X-}]_q\text{-(CHR}^3)_r\text{-}$, where q, r, s, X, and R³ are defined as for Formula (5). In certain embodiments, R¹² is selected from hydrogen and C₁₋₃ alkyl, and in certain embodiments, R¹² is hydrogen. In certain embodiments, a sulfur containing adduct is derived from Permapol® 3.1E. The reaction may take place in the presence of a catalyst at a temperature from about 25°C to about 50°C.

[0156] In certain embodiments, an amine-terminated sulfur-containing adduct comprises an amine-terminated polythioether adduct of Formula (8), an amine-terminated polythioether adduct of Formula (8a), and a combination thereof:



wherein:

each R¹ independently is selected from C₂₋₁₀ alkanediyl, C₆₋₈ cycloalkanediyl, C₆₋₁₀ alkanecycloalkanediyl, C₅₋₈ heterocycloalkanediyl, and $[-(\text{CHR}^3)_s\text{-X-}]_q\text{-(CHR}^3)_r\text{-}$, wherein:

s is an integer from 2 to 6;

q is an integer from 1 to 5;

r is an integer from 2 to 10;

each R³ is independently selected from hydrogen and methyl; and

each X is independently selected from -O-, -S-, and -NR-, wherein R is selected from hydrogen and methyl;

each R² is independently selected from C₁₋₁₀ alkanediyl, C₆₋₈ cycloalkanediyl, C₆₋₁₄ alkanecycloalkanediyl, and $[-(\text{CHR}^3)_s\text{-X-}]_q\text{-(CHR}^3)_r\text{-}$, wherein s, q, r, R³, and X are as defined for R¹;

m is an integer from 0 to 50;

n is an integer from 1 to 60;

p is an integer from 2 to 6;

B represents a core of a z-valent, vinyl-terminated polyfunctionalizing agent B(-V)_z wherein:

z is an integer from 3 to 6; and

each V is a group comprising a terminal vinyl group;
 each $-V'$ is derived from the reaction of $-V$ with a thiol;
 each $-R^{6'}$ is $-\text{CH}_2-\text{C}(\text{R}^4)_2-\text{S}(\text{O})_2-\text{C}(\text{R}^4)_2-\text{CH}_2-$, wherein each R^4 is independently selected from hydrogen and C_{1-3} alkyl; and
 each R^9 is a moiety having a terminal amine group.

[0157] In certain embodiments, R^9 is $\text{HS}-\text{R}^{11}-\text{N}(\text{R}^{12})\text{H}$, and in certain embodiments of Formula (8) and Formula (8a), R^9 is $-\text{S}-\text{R}^{11}-\text{NH}_2$.

[0158] In certain embodiments, compositions comprise one or more amine-terminated sulfur-containing adducts and one or more polyisocyanate curing agents such as any of those disclosed herein.

Compositions

[0159] Compositions provided by the present disclosure may include one or more catalysts. Catalysts appropriate for use in reactions between Michael acceptors such as activated alkenyl groups and thiol groups include base catalysts such as amines. Examples of suitable amine catalysts include, for example, triethylenediamine (1,4-diazabicyclo[2.2.2]octane, DABCO), dimethylcyclohexylamine (DMCHA), dimethylethanolamine (DMEA), bis-(2-dimethylaminoethyl)ether, N-ethylmorpholine, triethylamine, 1,8-diazabicyclo[5.4.0]undecene-7 (DBU), pentamethyldiethylenetriamine (PMDETA), benzyldimethylamine (BDMA), N,N,N'-trimethyl-N'-hydroxyethyl-bis(aminoethyl)ether, and N'-(3-(dimethylamino)propyl)-N,N-dimethyl-1,3-propanediamine.

[0160] In compositions comprising epoxies, the composition may comprise a base catalyst, including amine catalysts such as any of those disclosed herein.

[0161] In certain embodiments, compositions provided by the present disclosure comprise one or more than one adhesion promoters. A one or more additional adhesion promoter may be present in amount from 0.1 wt% to 15 wt% of a composition, less than 5 wt%, less than 2 wt%, and in certain embodiments, less than 1 wt%, based on the total dry weight of the composition. Examples of adhesion promoters include phenolics, such as Methylon[®] phenolic resin, and organosilanes, such as epoxy, mercapto or amino functional silanes, such as Silquest[®] A-187 and Silquest[®] A-1100. Other useful adhesion promoters are known in the art.

[0162] Compositions provided by the present disclosure may comprise one or more different types of filler. Suitable fillers include those commonly known in the art, including

inorganic fillers, such as carbon black and calcium carbonate (CaCO_3), silica, polymer powders, and lightweight fillers. Suitable lightweight fillers include, for example, those described in U.S. Patent No. 6,525,168. In certain embodiments, a composition includes 5 wt% to 60 wt% of the filler or combination of fillers, 10 wt% to 50 wt%, and in certain embodiments, from 20 wt% to 40 wt%, based on the total dry weight of the composition. Compositions provided by the present disclosure may further include one or more colorants, thixotropic agents, accelerators, fire retardants, adhesion promoters, solvents, masking agents, or a combination of any of the foregoing. As can be appreciated, fillers and additives employed in a composition may be selected so as to be compatible with each other as well as the polymeric component, curing agent, and or catalyst.

[0163] In certain embodiments, compositions provided by the present disclosure include low density filler particles. As used herein, low density, when used with reference to such particles means that the particles have a specific gravity of no more than 0.7, in certain embodiments no more than 0.25, and in certain embodiments, no more than 0.1. Suitable lightweight filler particles often fall within two categories – microspheres and amorphous particles. The specific gravity of microspheres may range from 0.1 to 0.7 and include, for example, polystyrene foam, microspheres of polyacrylates and polyolefins, and silica microspheres having particle sizes ranging from 5 to 100 microns and a specific gravity of 0.25 (Eccospheres®). Other examples include alumina/silica microspheres having particle sizes in the range of 5 to 300 microns and a specific gravity of 0.7 (Fillite®), aluminum silicate microspheres having a specific gravity of from about 0.45 to about 0.7 (Z-Light®), calcium carbonate-coated polyvinylidene copolymer microspheres having a specific gravity of 0.13 (Dualite® 6001AE), and calcium carbonate coated acrylonitrile copolymer microspheres such as Dualite® E135, having an average particle size of about 40 μm and a density of 0.135 g/cc (Henkel). Suitable fillers for decreasing the specific gravity of the composition include, for example, hollow microspheres such as Expancel® microspheres (available from AkzoNobel) or Dualite® low density polymer microspheres (available from Henkel). In certain embodiments, compositions provided by the present disclosure include lightweight filler particles comprising an exterior surface coated with a thin coating, such as those described in U.S. Publication No. 2010/0041839 at paragraphs [0015]-[0051].

[0164] In certain embodiments, a low density filler comprises less than 2 wt% of a composition, less than 1.5 wt%, less than 1.0 wt%, less than 0.8 wt%, less than 0.75 wt%, less

than 0.7 wt% and in certain embodiments, less than 0.5 wt% of a composition, where wt% is based on the total dry solids weight of the composition.

[0165] In certain embodiments, compositions provided by the present disclosure comprise at least one filler that is effective in reducing the specific gravity of the composition. In certain embodiments, the specific gravity of a composition is from 0.8 to 1, 0.7 to 0.9, from 0.75 to 0.85, and in certain embodiments, is 0.8. In certain embodiments, the specific gravity of a composition is less than about 0.9, less than about 0.8, less than about 0.75, less than about 0.7, less than about 0.65, less than about 0.6, and in certain embodiments, less than about 0.55.

[0166] In certain embodiments, compositions provided by the present disclosure comprise an electrically conductive filler. Electrical conductivity and EMI/RFI shielding effectiveness can be imparted to composition by incorporating conductive materials within the polymer. The conductive elements can include, for example, metal or metal-plated particles, fabrics, meshes, fibers, and combinations thereof. The metal can be in the form of, for example, filaments, particles, flakes, or spheres. Examples of metals include copper, nickel, silver, aluminum, tin, and steel. Other conductive materials that can be used to impart EMI/RFI shielding effectiveness to polymer compositions include conductive particles or fibers comprising carbon or graphite. Conductive polymers such as polythiophenes, polypyrroles, polyaniline, poly(p-phenylene) vinylene, polyphenylene sulfide, polyphenylene, and polyacetylene can also be used.

[0167] Examples of electrically non-conductive fillers include materials such as, but not limited to, calcium carbonate, mica, polyamide, fumed silica, molecular sieve powder, microspheres, titanium dioxide, chalks, alkaline blacks, cellulose, zinc sulfide, heavy spar, alkaline earth oxides, alkaline earth hydroxides, and the like. Fillers also include high band gap materials such as zinc sulfide and inorganic barium compounds. In certain embodiments, an electrically conductive base composition can comprise an amount of electrically non-conductive filler ranging from 2 wt% to 10 wt% based on the total weight of the base composition, and in certain embodiments, can range from 3 wt% to 7 wt%. In certain embodiments, a curing agent composition can comprise an amount of electrically non-conductive filler ranging from less than 6 wt% and in certain embodiments ranging from 0.5% to 4% by weight, based on the total weight of the curing agent composition.

[0168] Fillers used to impart electrical conductivity and EMI/RFI shielding effectiveness to polymer compositions are well known in the art. Examples of electrically conductive fillers

include electrically conductive noble metal-based fillers such as pure silver; noble metal-plated noble metals such as silver-plated gold; noble metal-plated non-noble metals such as silver plated copper, nickel or aluminum, for example, silver-plated aluminum core particles or platinum-plated copper particles; noble-metal plated glass, plastic or ceramics such as silver-plated glass microspheres, noble-metal plated aluminum or noble-metal plated plastic microspheres; noble-metal plated mica; and other such noble-metal conductive fillers. Non-noble metal-based materials can also be used and include, for example, non-noble metal-plated non-noble metals such as copper-coated iron particles or nickel plated copper; non-noble metals, e.g., copper, aluminum, nickel, cobalt; non-noble-metal-plated-non-metals, e.g., nickel-plated graphite and non-metal materials such as carbon black and graphite. Combinations of electrically conductive fillers can also be used to meet the desired conductivity, EMI/RFI shielding effectiveness, hardness, and other properties suitable for a particular application.

[0169] The shape and size of the electrically conductive fillers used in the compositions of the present disclosure can be any appropriate shape and size to impart EMI/RFI shielding effectiveness to the cured composition. For example, fillers can be of any shape that is generally used in the manufacture of electrically conductive fillers, including spherical, flake, platelet, particle, powder, irregular, fiber, and the like. In certain sealant compositions of the disclosure, a base composition can comprise Ni-coated graphite as a particle, powder or flake. In certain embodiments, the amount of Ni-coated graphite in a base composition can range from 40 wt% to 80 wt%, and in certain embodiments can range from 50 wt% to 70 wt%, based on the total weight of the base composition. In certain embodiments, an electrically conductive filler can comprise Ni fiber. Ni fiber can have a diameter ranging from 10 μm to 50 μm and have a length ranging from 250 μm to 750 μm . A base composition can comprise, for example, an amount of Ni fiber ranging from 2 wt% to 10 wt%, and in certain embodiments, from 4 wt% to 8 wt%, based on the total weight of the base composition.

[0170] Carbon fibers, particularly graphitized carbon fibers, can also be used to impart electrical conductivity to compositions of the present disclosure. Carbon fibers formed by vapor phase pyrolysis methods and graphitized by heat treatment and which are hollow or solid with a fiber diameter ranging from 0.1 micron to several microns, have high electrical conductivity. As disclosed in U.S. Patent No. 6,184,280, carbon microfibers, nanotubes or carbon fibrils having an outer diameter of less than 0.1 μm to tens of nanometers can be used as electrically conductive fillers. An example of graphitized carbon fiber suitable for conductive compositions of the

present disclosure include PANEX 3OMF (Zoltek Companies, Inc., St. Louis, MO), a 0.921 μm diameter round fiber having an electrical resistivity of 0.00055 $\Omega\text{-cm}$.

[0171] The average particle size of an electrically conductive filler can be within a range useful for imparting electrical conductivity to a polymer-based composition. For example, in certain embodiments, the particle size of the one or more fillers can range from 0.25 μm to 250 μm , in certain embodiments can range from 0.25 μm to 75 μm , and in certain embodiments can range from 0.25 μm to 60 μm . In certain embodiments, composition of the present disclosure can comprise Ketjen Black EC-600 JD (Akzo Nobel, Inc., Chicago, IL), an electrically conductive carbon black characterized by an iodine absorption of 1000-11500 mg/g (J0/84-5 test method), and a pore volume of 480-510 $\text{cm}^3/100\text{ gm}$ (DBP absorption, KTM 81-3504). In certain embodiments, an electrically conductive carbon black filler is Black Pearls 2000 (Cabot Corporation, Boston, MA).

[0172] In certain embodiments, electrically conductive polymers can be used to impart or modify the electrical conductivity of compositions of the present disclosure. Polymers having sulfur atoms incorporated into aromatic groups or adjacent to double bonds, such as in polyphenylene sulfide, and polythiophene, are known to be electrically conductive. Other electrically conductive polymers include, for example, polypyrroles, polyaniline, poly(p-phenylene) vinylene, and polyacetylene. In certain embodiments, the sulfur-containing polymers forming a base composition can be polysulfides and/or polythioethers. As such, the sulfur-containing polymers can comprise aromatic sulfur groups and sulfur atoms adjacent to conjugated double bonds such as vinylcyclohexene-dimercaptodioxaoctane groups, to enhance the electrical conductivity of the compositions of the present disclosure.

[0173] Compositions of the present disclosure can comprise more than one electrically conductive filler and the more than one electrically conductive filler can be of the same or different materials and/or shapes. For example, a sealant composition can comprise electrically conductive Ni fibers, and electrically conductive Ni-coated graphite in the form of powder, particles or flakes. The amount and type of electrically conductive filler can be selected to produce a sealant composition which, when cured, exhibits a sheet resistance (four-point resistance) of less than 0.50 Ω/cm^2 , and in certain embodiments, a sheet resistance less than 0.15 Ω/cm^2 . The amount and type of filler can also be selected to provide effective EMI/RFI shielding over a frequency range of from 1 MHz to 18 GHz for an aperture sealed using a sealant composition of the present disclosure.

[0174] Galvanic corrosion of dissimilar metal surfaces and the conductive compositions of the present disclosure can be minimized or prevented by adding corrosion inhibitors to the composition, and/or by selecting appropriate conductive fillers. In certain embodiments, corrosion inhibitors include strontium chromate, calcium chromate, magnesium chromate, and combinations thereof. U.S. Patent No. 5,284,888 and U.S. Patent No. 5,270,364 disclose the use of aromatic triazoles to inhibit corrosion of aluminum and steel surfaces. In certain embodiments, a sacrificial oxygen scavenger such as Zn can be used as a corrosion inhibitor. In certain embodiments, the corrosion inhibitor can comprise less than 10% by weight of the total weight of the electrically conductive composition. In certain embodiments, the corrosion inhibitor can comprise an amount ranging from 2% by weight to 8% by weight of the total weight of the electrically conductive composition. Corrosion between dissimilar metal surfaces can also be minimized or prevented by the selection of the type, amount, and properties of the conductive fillers comprising the composition.

[0175] In certain embodiments, a sulfur-containing polymer and/or sulfur-containing polymer adduct comprises from about 50 wt% to about 90 wt% of a composition, from about 60 wt% to about 90 wt%, from about 70 wt% to about 90 wt%, and in certain embodiments, from about 80 wt% to about 90 wt% of the composition, where wt% is based on the total dry solids weight of the composition.

[0176] A composition may also include any number of additives as desired. Examples of suitable additives include plasticizers, pigments, surfactants, adhesion promoters, thixotropic agents, fire retardants, masking agents, and accelerators (such as amines, including 1,4-diazabicyclo[2.2.2] octane, DABCO®), and combinations of any of the foregoing. When used, the additives may be present in a composition in an amount ranging, for example, from about 0% to 60% by weight. In certain embodiments, additives may be present in a composition in an amount ranging from about 25% to 60% by weight.

Compositions Containing a Controlled Release Catalyst

[0177] As a further improvement to such compositions it is desirable to extend the pot life and to control the curing rate. These and other properties can be achieved by using a controlled release amine catalyst. Accordingly, compositions provided by the present disclosure comprise (a) a compound comprising at least two terminal groups reactive with Michael acceptor

groups; (b) a compound having at least two Michael acceptor groups; and (c) a controlled release amine catalyst, wherein at least one of (a) and (b) comprises a polythioether polymer.

[0178] Systems in which a controlled release amine catalyst is released by photolytic, hydrolytic, thermal, or ultrasonic mechanisms are disclosed. Upon release of the amine catalyst by any one of the mechanisms, the catalytic amine catalyzes the Michael addition reaction between a compound terminated with groups reactive with Michael acceptor groups, such as terminal thiol groups, and a compound having at least two Michael acceptor groups. The compound having at least two terminal groups reactive with Michael acceptor groups may be a small molecule such as a molecule having a molecular weight less than 400 Dalton, a sulfur-containing polymer such as a polythioether, or a combination thereof. The compound having at least two Michael acceptor groups may be a small molecule and/or may be a Michael acceptor adduct. One suitable Michael acceptor adduct comprises a sulfur-containing compound such as a polythioether that is terminated with Michael acceptor groups. In certain embodiments, at least one of the compound terminated with groups reactive with Michael acceptor groups and the compound having at least two Michael acceptor groups comprises a polythioether.

[0179] In certain embodiments, compositions having extended pot life and controlled curing rate can be realized by using a controlled release amine catalyst. In these systems, an amine catalyst such as a strong base or primary amine that produces a fast reaction rate is protected or encapsulated and dispersed in the composition. Upon exposure, for example, to ultraviolet radiation, moisture, or temperature, the catalytic amine is released and catalyzes the Michael addition reaction. In certain embodiments, systems provide a pot life greater than 2 hours to 12 hours and cure within 24 to 72 hours after the useful working time.

[0180] In certain embodiments, compositions comprise: (a) a compound comprising at least two terminal groups reactive with Michael acceptor groups; (b) a compound having at least two Michael acceptor groups; and (c) a controlled release amine catalyst, wherein at least one of (a) and (b) comprises a polythioether polymer.

[0181] In certain embodiments, methods of using a composition comprise (a) a compound comprising at least two terminal groups reactive with Michael acceptor groups; (b) a compound having at least two Michael acceptor groups; and (c) a controlled release amine catalyst, wherein at least one of (a) and (b) comprises a polythioether polymer, are provided.

Controlled release amine catalyst

[0182] Controlled release amine catalysts have little or no activity until released, either chemically or physically. In certain embodiments, a controlled release amine catalyst may be released upon exposure to ultraviolet radiation, heat, ultrasonication, or moisture.

[0183] In the case of controlled release amine catalysts that are released by ultraviolet radiation or moisture, the amine catalyst comprises a blocking group that reacts upon exposure to the ultraviolet radiation or moisture to release or unblock a reactive amine catalyst. In matrix encapsulant systems, the amine catalyst is trapped among side chains of a crystalline or semi-crystalline polymer. At elevated temperature, the polymer melts allowing the amine catalyst to diffuse into the composition to catalyze the reaction.

[0184] In certain embodiments, a controlled release amine catalyst comprises a controlled release amine catalyst. In certain embodiment, a controlled release amine catalyst may be a controlled release primary amine catalyst, a controlled release secondary amine catalyst, or a controlled release tertiary amine catalyst. Examples of suitable primary amine catalysts include, for example, C₃₋₁₀ aliphatic primary amines, such as heptane amine, hexylamine, and octamine. Examples of suitable secondary amine catalysts include, for example, cycloaliphatic diamines such as Jefflink[®] 754 and aliphatic diamines such as Clearlink[®] 1000. Examples of suitable tertiary amine catalysts include, for example, N,N-dimethylethanolamine (DMEA), diaminobicyclooctane (DABCO), triethylene diamine (TEDA), bis(2-dimethylaminoethyl)ether (BDMAEE), N-ethylmorpholine, N,N'-dimethylpiperazine, N, N, N', N', N''-pentamethyldiethylene-triamine (PMDETA), N, N'-dimethylcyclohexylamine (DMCHA), N,N-dimethylbenzylamine (DMBA), N, N-dimethylcethylamine, N, N, N', N'', N''-pentamethyldipropylene-triamine (PMDPTA), triethylamine, and 1-(2-hydroxypropyl) imidazole. Other suitable amine catalysts include amidine catalysts such as tetramethyguanidine (TMG), dizabicyclononene (DBN), diazabicyclo undecene (DBU) and imidazoles; and bicyclic guanidines such as 1,5,7,-triazabicyclo[4.4.0]dec-5-ene (TBD) and 1,5,7,-triazabicyclo[4.4.0]dec-5-ene, 7-methyl (MTBD).

[0185] In certain embodiments, an amine catalyst is selected from DBU, DABCO, IPDA, a C₆₋₁₀ primary amine, and a combination of any of the foregoing.

[0186] Compositions may comprise one or more different types of amine catalyst.

[0187] When released, controlled release amine catalysts provided by the present disclosure catalyze the reaction between a compound containing at least two terminal groups that

are reactive with Michael acceptor groups and a compound comprising at least two Michael acceptor groups.

[0188] In controlled release compositions provided by the present disclosure, the pot life of a composition can be greater than 2 weeks if the catalyst is not released. When the catalyst is released, either by chemical, photochemical, or physical mechanisms, the cure time can be less than 72 hours, less than 60 hours, less than 48 hours, less than 36 hours, and in certain embodiments less than 24 hours. The cure time without heating and in the presence of ambient moisture, can be several days such as, for example, 7 days.

Photolabile controlled release amine catalyst

[0189] Certain compositions provided by the present disclosure comprise a photolabile catalyst. In such systems, UV radiation unblocks a blocked amine catalyst, which catalyzes the Michael addition reaction between a compound comprising at least two terminal groups that are reactive with Michael acceptor groups and a compound comprising at least two Michael acceptor groups. In certain embodiments, the UV radiation initiates the reaction, which takes place over time, such as for example several hours. The slow curing can be useful to improve surface adhesion and to extend the pot life to provide a longer working time.

[0190] Photolabile amines comprise a photolabile moiety bonded to an amine.

[0191] In certain embodiments, a photolabile catalyst comprises CGI 90 (BASF), which following UV activation, generates the tertiary amine, 1,5-diazabicyclo(4.3.0)non-5-ene (DBN). Other suitable photolabile amines are disclosed in International Publication No. WO 2003/033500 and in the documents cited therein.

[0192] In compositions comprising a photolabile amine catalyst, the photolabile amine catalyst may comprise from 0.1 wt% to 5 wt% of the composition, from 0.3 wt% to 2 wt% of the composition, and in certain embodiments, from 0.5 wt% of the composition to 1 wt% of the composition.

Moisture-release amine catalyst

[0193] In certain embodiments, a controlled release amine catalyst comprises a moisture-released blocked amine catalyst. In such systems, the blocked amine catalyst can be unblocked in the presence of moisture to release an amine catalyst capable of catalyzing a Michael addition reaction. Examples of moisture-release blocked amine catalysts include ketimines, enamines,

oxazolidines, aldimines, and imidazolidines. In the presence of moisture, the blocking group, e.g., the ketamine, enamine, oxazolidine, aldimine, or imidazolidine blocking group or groups reacts with water to provide a catalytic amine catalyst and a ketone or alcohol.

[0194] In certain embodiments, a composition comprising a moisture-released amine catalyst comprises from 0.1 wt% to 2 wt% water, from 0.2 wt% to 1.5 wt% water, and in certain embodiments, from 0.5 wt% to 1 wt% water. The composition is stored a low temperature, such as at a temperature less than 0°C, less than -20°C, or in certain embodiments, less than -40°C. When the composition is heated before and/or during application, the water unblocks the blocked amine to catalyze the Michael addition reaction.

[0195] In certain embodiments, a moisture-released blocked amine catalyst releases a primary amine, a secondary amine, and in certain embodiments a tertiary amine. In certain embodiments, a moisture-released blocked amine catalyst is Vestamin[®] A139, which is a blocked cycloaliphatic diamine. In certain embodiments, the unblocked amine is isophorone diamine (IPDA).

[0196] In compositions comprising a moisture-released amine catalyst, the moisture released amine catalyst may comprise from 0.1 wt% to 4 wt% of the composition, from 0.5 wt% to 3 wt% of the composition, and in certain embodiments, from 1 wt% of the composition to 2 wt% of the composition.

[0197] In certain embodiments, the ratio (wt%/wt%) of wt% water to moisture-released amine catalyst (wt%) in compositions provided by the present disclosure can be from 1 to 4, from 1 to 2, and in certain embodiment, from 1 to 1.

[0198] Compositions comprising a moisture-released blocked amine catalyst may, in addition to being stored a low temperature, may be stored such as to prevent exposure to ambient moisture.

Matrix encapsulation

[0199] Matrix encapsulation is a process by which droplets or particles of liquid or solid material are trapped among side chains of a crystalline polymer. With increased temperature, the crystalline polymer becomes amorphous and releases the droplets or particles into the medium. Matrix encapsulants provided by the present disclosure comprise a crystalline matrix material incorporating droplets or particles comprising an amine catalyst. Thus, the rate of reaction is to some extent controlled by thermally dependent diffusion of the amine catalyst from the

crystalline polymer. The crystalline polymers may have a sharp well-defined melting point or may exhibit a melting point range. The use of waxy polymers for encapsulation of amine catalysts used in Michael addition compositions is disclosed in U.S. Application Publication no. 2007/0173602.

[0200] Examples of suitable matrix encapsulants include Intelimer[®] polymers (Air Products), such as Intelimer[®] 13-1 and Intelimer[®] 13-6. The properties of Intelimer[®] polymers is disclosed in Lowry et al., Cure evaluation of Intelimer[®] latent curing agents for thermoset resin applications, presented at the Thermoset Resin Formulators Association Meeting, Chicago, IL, September 15-16, 2008.

[0201] A matrix encapsulant may be selected to release the amine catalyst following a brief high-temperature exposure such as for less than 10 minutes, less than 5 minutes, or less than 2 minutes. During this brief temperature excursion, amine catalyst is released from the matrix and diffuses into the reactive polymer components. The composition may be heated during the curing process or may be left at ambient temperature. When left at ambient temperature, the released amine catalyst composition may cure in less than 2 hours, in less than 4 hours, and in certain embodiments, in less than 6 hours.

[0202] Amine catalysts may be incorporated into a matrix encapsulant by blending at a temperature above the melt temperature of the matrix encapsulant, rapidly cooling the mixture, and grinding the solid to a powder. In certain embodiments, the average particle size is less than 200 μm , less than 150 μm , less than 100 μm , less than 50 μm , and in certain embodiments, less than 25 μm .

[0203] In certain embodiments, a composition may comprise from 0.1 wt% to 25 wt%, from 1 wt% to 15 wt%, and in certain embodiments, from 5 wt% to 10 wt% of a matrix encapsulant comprising an amine catalyst. This correlates to about 0.01 wt% to 2 wt%, from 0.05 wt% to 1.5 wt%, and in certain embodiments, from 0.5 wt% to 1 wt% of an amine catalyst.

[0204] In certain embodiments, a matrix encapsulant suitable for use in compositions provided by the present disclosure comprises a ratio (wt%/wt%) of wt% amine catalyst to wt% matrix polymer from 1 to 15, from 2 to 10, and in certain embodiments, from 5 to 8.

[0205] Compositions comprising a sulfur-containing compound, a polyfunctional Michael acceptor, and a sulfur-containing adduct comprise a controlled release catalyst including any of those disclosed herein.

Uses

[0206] Compositions provided by the present disclosure may be used, for example, in sealants, coatings, encapsulants, and potting compositions. A sealant includes a composition capable of producing a film that has the ability to resist operational conditions, such as moisture and temperature, and at least partially block the transmission of materials, such as water, fuel, and other liquid and gases. A coating composition includes a covering that is applied to the surface of a substrate to, for example, improve the properties of the substrate such as the appearance, adhesion, wettability, corrosion resistance, wear resistance, fuel resistance, and/or abrasion resistance. A potting composition includes a material useful in an electronic assembly to provide resistance to shock and vibration and to exclude moisture and corrosive agents. In certain embodiments, sealant compositions provided by the present disclosure are useful, e.g., as aerospace sealants and as linings for fuel tanks.

[0207] In certain embodiments, compositions, such as sealants, may be provided as multi-pack compositions, such as two-pack compositions, wherein one package comprises one or more thiol-terminated polythioethers provided by the present disclosure and a second package comprises one or more polyfunctional sulfur-containing epoxies provided by the present disclosure. Additives and/or other materials may be added to either package as desired or necessary. The two packages may be combined and mixed prior to use. In certain embodiments, the pot life of the one or more mixed thiol-terminated polythioethers and epoxies is at least 30 minutes, at least 1 hour, at least 2 hours, and in certain embodiments, more than 2 hours, where pot life refers to the period of time the mixed composition remains suitable for use as a sealant after mixing.

[0208] In two- part compositions, one or more controlled release amine catalysts may be included in either component or in both components. In certain embodiments, the controlled release catalyst may be a third component that is mixed with a polythioether and Michael acceptor components prior to use. In certain embodiments, the compositions are provided as a one-part composition. Such one-part compositions are maintained and stored under conditions such as the controlled release catalyst is not substantially released. For example, a composition comprising a photolabile catalyst may be shielded from UV radiation, a moisture release catalyst may be sealed from moisture and frozen, and a composition comprising a matrix encapsulant may be stored at temperatures below the melting temperature of the matrix encapsulant polymer.

[0209] Compositions, including sealants, provided by the present disclosure may be applied to any of a variety of substrates. Examples of substrates to which a composition may be applied include metals such as titanium, stainless steel, and aluminum, any of which may be anodized, primed, organic-coated or chromate-coated; epoxy; urethane; graphite; fiberglass composite; Kevlar®; acrylics; and polycarbonates. In certain embodiments, compositions provided by the present disclosure may be applied to a coating on a substrate, such as a polyurethane coating.

[0210] Compositions provided by the present disclosure may be applied directly onto the surface of a substrate or over an underlayer by any suitable coating process known to those of ordinary skill in the art.

[0211] Furthermore, methods are provided for sealing an aperture utilizing a composition provided by the present disclosure. These methods comprise, for example, applying a composition provided by the present disclosure to a surface to seal an aperture, and curing the composition. In certain embodiments, a method for sealing an aperture comprises (a) applying a sealant composition provided by the present disclosure to one or more surfaces defining an aperture, (b) assembling the surfaces defining the aperture, and (c) curing the sealant, to provide a sealed aperture.

[0212] In certain embodiments, a composition may be cured under ambient conditions, where ambient conditions refers to a temperature from 20°C to 25°C, and atmospheric humidity. In certain embodiments, a composition may be cured under conditions encompassing a temperature from a 0°C to 100°C and humidity from 0% relative humidity to 100% relative humidity. In certain embodiments, a composition may be cured at a higher temperature such as at least 30°C, at least 40°C, and in certain embodiments, at least 50°C. In certain embodiments, a composition may be cured at room temperature, e.g., 25°C. In certain embodiments, a composition may be cured upon exposure to actinic radiation, such as ultraviolet radiation. As will also be appreciated, the methods may be used to seal apertures on aerospace vehicles including aircraft and aerospace vehicles.

[0213] In certain embodiments, the composition achieves a tack-free cure in less than about 2 hours, less than about 4 hours, less than about 6 hours, less than about 8 hours, and in certain embodiments, less than about 10 hours, at a temperature of less than about 200°F.

[0214] The time to form a viable seal using curable compositions of the present disclosure can depend on several factors as can be appreciated by those skilled in the art, and as

defined by the requirements of applicable standards and specifications. In general, curable compositions of the present disclosure develop adhesion strength within 24 hours to 30 hours, and 90% of full adhesion strength develops from 2 days to 3 days, following mixing and application to a surface. In general, full adhesion strength as well as other properties of cured compositions of the present disclosure becomes fully developed within 7 days following mixing and application of a curable composition to a surface.

[0215] Cured compositions disclosed herein, such as cured sealants, exhibit properties acceptable for use in aerospace applications. In general, it is desirable that sealants used in aviation and aerospace applications exhibit the following properties: peel strength greater than 20 pounds per linear inch (pli) on Aerospace Material Specification (AMS) 3265B substrates determined under dry conditions, following immersion in JRF for 7 days, and following immersion in a solution of 3% NaCl according to AMS 3265B test specifications; tensile strength between 300 pounds per square inch (psi) and 400 psi; tear strength greater than 50 pounds per linear inch (pli); elongation between 250% and 300%; and hardness greater than 40 Durometer A. These and other cured sealant properties appropriate for aviation and aerospace applications are disclosed in AMS 3265B. It is also desirable that, when cured, compositions of the present disclosure used in aviation and aircraft applications exhibit a percent volume swell not greater than 25% following immersion for one week at 60°C (140°F) and ambient pressure in JRF type 1. Other properties, ranges, and/or thresholds may be appropriate for other sealant applications.

[0216] In certain embodiments, therefore, compositions provided by the present disclosure are fuel-resistant. As used herein, the term “fuel resistant” means that a composition, when applied to a substrate and cured, can provide a cured product, such as a sealant, that exhibits a percent volume swell of not greater than 40%, in some cases not greater than 25%, in some cases not greater than 20%, in yet other cases not more than 10%, after immersion for one week at 140°F (60°C) and ambient pressure in Jet Reference Fluid (JRF) Type I according to methods similar to those described in ASTM D792 (American Society for Testing and Materials) or AMS 3269 (Aerospace Material Specification). Jet Reference Fluid JRF Type I, as employed for determination of fuel resistance, has the following composition: toluene: $28 \pm 1\%$ by volume; cyclohexane (technical): $34 \pm 1\%$ by volume; isooctane: $38 \pm 1\%$ by volume; and tertiary dibutyl disulfide: $1 \pm 0.005\%$ by volume (see AMS 2629, issued July 1, 1989, § 3.1.1 etc., available from SAE (Society of Automotive Engineers)).

[0217] In certain embodiments, compositions provided herein provide a cured product, such as a sealant, exhibiting a tensile elongation of at least 100% and a tensile strength of at least 400 psi when measured in accordance with the procedure described in AMS 3279, § 3.3.17.1, test procedure AS5127/1, § 7.7.

[0218] In certain embodiments, compositions provide a cured product, such as a sealant, that exhibits a lap shear strength of greater than 200 psi, such as at least 220 psi, at least 250 psi, and, in some cases, at least 400 psi, when measured according to the procedure described in SAE AS5127/1 paragraph 7.8.

[0219] In certain embodiments, a cured sealant comprising a composition provided by the present disclosure meets or exceeds the requirements for aerospace sealants as set forth in AMS 3277.

[0220] Apertures, including apertures of aerospace vehicles, sealed with compositions provided by the present disclosure are also disclosed.

[0221] In certain embodiments, an electrically conductive sealant composition provided by the present disclosure exhibits the following properties measured at room temperature following exposure at 500°F for 24 hours: a surface resistivity of less than 1 ohms/square, a tensile strength greater than 200 psi, an elongation greater than 100%, and a cohesive failure of 100% measured according to MIL-C-27725.

[0222] In certain embodiments, a cured sealant provided by the present disclosure exhibits the following properties when cured for 2 days at room temperature, 1 day at 140°F, and 1 day at 200°F: a dry hardness of 49, a tensile strength of 428 psi, and an elongation of 266%; and after 7 days in JRF, a hardness of 36, a tensile strength of 312 psi, and an elongation of 247%.

[0223] In certain embodiments, compositions provided by the present disclosure exhibit a Shore A hardness (7-day cure) greater than 10, greater than 20, greater than 30, and in certain embodiments, greater than 40; a tensile strength greater than 10 psi, greater than 100 psi, greater than 200 psi, and in certain embodiments, greater than 500 psi; an elongation greater than 100%, greater than 200%, greater than 500%, and in certain embodiments, greater than 1,000%; and a swell following exposure to JRF (7 days) less than 20%.

EXAMPLES

[0224] Embodiments provided by the present disclosure are further illustrated by reference to the following examples, which describe the synthesis, properties, and uses of certain sulfur-containing polymers, Michael acceptor adducts, and compositions comprising sulfur-containing polymers, Michael acceptor adducts, and Michael acceptors. It will be apparent to those skilled in the art that many modifications, both to materials, and methods, may be practiced without departing from the scope of the disclosure.

Example 1

Polythioether cured with Monomeric Divinyl Sulfone

[0225] To prepare Resin Mixture A, thiol-terminated polythioethers of the type described in U.S. Patent No. 6,172,179, average thiol functionality: 2.05-2.95, commercially available from PRC-Desoto International, Inc., Sylmar, CA, HB-40 plasticizer (Solutia Inc.), DABCO® 33LV (Huntsman), Winnofil® SPM (Solvay), Sipernat® D13 (Evonik) and tung oil (Alnor Oil Company, Inc.) were added to a Max 300 (FlackTek) jar in the order and amounts listed in Table 1. The materials were mixed with a DAC 600.1 FVZ mixer (FlackTek) for 45 seconds. Vinyl sulfone (Aldrich) (4.99 g) was then add to Resin Mixture A and mixed for 1 minute. The mixture was immediately poured onto polyethylene sheets and pressed out flat to form a 1/8" sheets. Samples were cured for two weeks at room temperature. The sheet material was then tested for hardness, tensile strength, elongation, and fluid resistance. The results are provided in Table 2.

Table 1. Components of Resin Mixture A.

Resin Mixture A	
Material	Amount (g)
Polythioethers*	140.05
HB-40	1.18
DABCO® 33LV	1..58
Winnofil® SPM	43.08
Sipernat® D13	9.11

* *Thiol-terminated polythioethers of the type described in U.S. Patent No. 6,172,179, average thiol functionality: 2.05-2.95, commercially available from PRC-Desoto International, Inc., Sylmar, CA.

Table 2. Tests properties, methods and results.

Description	Test	Result
Durometer (Shore A)	ASTM D2240	53
Elongation	ASTM D412	390%
Tensile Strength	ASTM D412	3006 Kpas
Swell, distilled water	SAE AS5271/1 7.4	8.3%
Swell, 3% NaCl	SAE AS5271/1 7.4	3%
Swell, JRF	SAE AS5271/1 7.4	16.7%

Example 2

Polysulfide polymer cured with monomeric divinyl sulfone

[0226] Mixing was performed in a 60-g plastic container with a lid. Divinyl sulfone (1.22 g), triethylenediamine (0.17 g) and Thiokol LP-32 (33.01 g, a liquid polysulfide polymer available from Toray Fine Chemicals) were added to the 60-g container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm. The mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, the hardness of the cured material was 14 Shore A, measured according to ASTM D 2240.

Example 3

Polythioether cured with monomeric divinyl sulfone

[0227] Divinyl sulfone (3.05 g), triethylenediamine (0.39 g) and Permapol® P3.1E (74.7 g, a thiol-terminated polythioether polymer available from PRC-Desoto International, Inc., Sylmar, CA) were added to a plastic container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm.

[0228] A portion of the mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, the hardness of the cured material was 42 Shore A, measured according to ASTM D 2240.

[0229] A second portion of the mixed material was poured onto a 12"×18"×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 696 psi and an elongation of 933%. The tensile strength and elongation were measured according to ASTM D412.

Example 4

Polythioether cured with divinyl sulfone

[0230] Divinyl sulfone (3.05 g), triethylenediamine (0.62 g), Permapol® P3.1E (74.70 g, a thiol-terminated polythioether polymer available from PRC-Desoto International, Inc., Sylmar, CA), and calcium carbonate (48.50 g) were added to a 100-gram plastic container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm.

[0231] A portion of the mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, the hardness of cured material was 25 Shore A, measured according to ASTM D 2240.

[0232] A second portion of the mixed material was poured onto a 12"×18"×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 546 psi and an elongation of 1,077%. The tensile strength and elongation were measured according to ASTM D412.

Example 5

Synthesis of divinyl sulfone-terminated polythioether adduct

[0233] In a 300 mL, 3-necked, round bottom flask fitted with a mechanical stirrer, thiol-terminated polythioether polymer Permapol® P3.1E (149.40 g, available from PRC-Desoto International, Inc., Sylmar, CA), divinyl sulfone (12.18 g), and triethylenediamine (0.81 g) were added at room temperature. The mixture was stirred for 10 minutes, resulting a vinyl sulfone-terminated polythioether adduct that had a viscosity of 309.0 poise. The viscosity was measured by CAP2000 viscometer with spindle #6, 50 RPM.

Example 6

Polythioether Michael acceptor adduct cured with a thiol-terminated polysulfide polymer

[0234] Mixing was performed in a 60-gram plastic container with lid. The adduct from Example 5 (9.27 g) and Thiokol LP-980 (5.90 g, a liquid polysulfide polymer, available from Toray Fine Chemicals) were added to the 60-gram container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm. The mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, hardness of cured

material was 11 Shore A and volume swell percentage in jet reference fluid type I (JRF Type I) of cured material was 19.20%. Hardness and volume swell percentage in jet reference fluid type I (JRF Type I) were measured according to ASTM D 2240 and SAE AS5127/1 Section 7.4, respectively.

Example 7

Polythioether Michael acceptor adduct cured with a thiol-terminated polysulfide polymer

[0235] Mixing was performed in a 60-g plastic container with lid. The adduct from Example 5 (9.27 g) and Thiokol LP-32 (9.17 g, a liquid polysulfide polymer, available from Toray Fine Chemicals) were added to the 60-g container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm. The mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, hardness of cured material is 24 Shore A and volume swell percentage in jet reference fluid type I (JRF Type I) of cured material is 18.81%. Hardness and volume swell percentage in jet reference fluid type I (JRF Type I) were measured according to ASTM D 2240 and SAE AS5127/1 Section 7.4, respectively.

Example 8

Polythioether Michael acceptor adduct cured with a thiol-terminated polysulfide polymer

[0236] Mixing was performed in a 60-gm plastic container with lid. The adduct from Example 5 (9.27 g) and Thiokol LP-12 (9.17 g, a liquid polysulfide polymer, available from Toray Fine Chemicals) were added to the 60-gm container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm. The mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, hardness of cured material is 25 Shore A and volume swell percentage in jet reference fluid type I (JRF Type I) of cured material is 19.41%. Hardness and volume swell percentage in jet reference fluid type I (JRF Type I) were measured according to ASTM D 2240 and SAE AS5127/1 Section 7.4, respectively.

Example 9

Polythioether Michael acceptor adduct cured with a thiol-terminated polysulfide polymer

[0237] Mixing was performed in a plastic container with lid. The adduct from Example 5 (74.13 g) and Thioplast® G4 (19.12 g, a liquid polysulfide polymer, available from Akzo Nobel) were added to the container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm. A portion of the mixed material was cured inside the plastic

container for 7 days at room temperature. After 7 days, hardness of cured material is 25 Shore A and volume swell percentage in jet reference fluid type I (JRF Type I) of cured material is 18.70%. Hardness and volume swell percentage in jet reference fluid type I (JRF Type I) were measured according to ASTM D 2240 and SAE AS5127/1 Section 7.4, respectively.

[0238] A second portion of the mixed material was poured onto a 12"×18" ×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 92 psi and an elongation of 181%. The tensile strength and elongation were measured according to ASTM D412.

Example 10

Polythioether Michael acceptor adduct cured with a thiol-terminated polysulfide polymer

[0239] The mixing was performed in a plastic container with lid. The adduct from Example 5 (74.13 g) and Thioplast® G21 (48.80 g, a liquid polysulfide polymer, available from Akzo Nobel) were added to the container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm. A portion of the mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, hardness of cured material is 32 Shore A and volume swell percentage in jet reference fluid type I (JRF Type I) of cured material is 18.48%. Hardness and volume swell percentage in jet reference fluid type I (JRF Type I) were measured according to ASTM D 2240 and SAE AS5127/1 Section 7.4, respectively.

[0240] A second portion of the mixed material was poured onto a 12"×18" ×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 88 psi and an elongation of 107%. The tensile strength and elongation were measured according to ASTM D412.

Example 11

Polythioether Michael acceptor adduct cured with a thiol-terminated polysulfide polymer

[0241] The mixing was performed in a plastic container with lid. The adduct from Example 5 (55.60 g) and Thiokol LP-2 (57.48 g, a liquid polysulfide polymer, available from Toray Fine Chemicals) were added to the container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm. A portion of the mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, hardness of cured material is 33 Shore A and volume swell percentage in jet reference fluid type I (JRF Type I) of cured material is 18.06%. Hardness and volume swell percentage in jet reference fluid type I

(JRF Type I) were measured according to ASTM D 2240 and SAE AS5127/1 Section 7.4, respectively.

[0242] A second portion of the mixed material was poured onto a 12"×18" ×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 108 psi and an elongation of 113%. The tensile strength and elongation were measured according to ASTM D412.

Example 12

Polythioether Michael acceptor adduct cured with a thiol-terminated polythioether polymer

[0243] The mixing was performed in a plastic container with lid. The adduct from Example 5 (32.56 g) and Permapol® P3.1E (29.96 g, a thiol-terminated polythioether polymer available from PRC-Desoto International Inc., Sylmar, CA), and triethylenediamine (0.31 g) were added to the container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm. A portion of the mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, hardness of cured material is 31 Shore A. Hardness was measured according to ASTM D 2240.

[0244] A second portion of the mixed material was poured onto a 12"×18" ×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 446 psi and an elongation of 504%. The tensile strength and elongation were measured according to ASTM D412.

Example 13

Polythioether Michael acceptor adduct cured with a thiol-terminated polythioether polymer

[0245] A sealant was produced according to the composition shown in Table 1.

Table 1: Formulation of Example 13.

Composition	Charge weight, g
Example 5 adduct	34.17
Permapol® P3.1E	29.96
Carbon Black	20.00
Triethylenediamine	0.32

[0246] The mixing was performed in a 100-gram plastic container with lid. The adduct from Example 5 (34.17 g), Permapol® P3.1E (29.96 g, a thiol-terminated polythioether polymer, available from PRC-Desoto International Inc., Sylmar, CA), carbon black (20.00 g), and triethylenediamine (0.32 g) were added to the 100-gram container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm. A portion of the mixed material was cured inside the plastic container for 7 days at room temperature. After 7-day cure, hardness of cured material is 43 Shore A, measured according to ASTM D 2240.

[0247] A second portion of the mixed material was poured onto a 12"×18" ×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 1810 psi and an elongation of 950%. The tensile strength and elongation were measured according to ASTM D412.

Example 14

Polythioether Michael acceptor adduct cured with a thiol-terminated polythioether polymer, low density

[0248] A sealant was produced according to the composition shown in Table 2:

Table 2: Formulation of Example 14.

Composition	Charge weight, g
Example 5 adduct	34.17
Permapol® P3.1E	29.96
Carbon Black	7.20
Triethylenediamine	0.32
Dualite® E135-040D	7.20

[0249] The mixing was performed in a 100-gram plastic container with lid. The adduct from Example 5 (34.17 g), Permapol® P3.1E (29.96 g, a thiol-terminated polythioether polymer, available from PRC-Desoto International Inc., Sylmar, CA), carbon black (7.20 g), triethylenediamine (0.32 g) and Dualite® E135-040D (7.20 g, available from Henkel) were added to the 100-gram container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm. A portion of the mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, hardness of cured material is 35 Shore A, measured according to ASTM D 2240.

[0250] A second portion of the mixed material was poured onto a 12"×18" ×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 252 psi and an elongation of 772%. The tensile strength and elongation were measured according to ASTM D412. The estimated specific gravity was 0.706.

Example 15

Polythioether Michael acceptor adduct cured with a thiol-terminated polythioether and epoxy blend

[0251] The mixing was performed in a 60-gm plastic container with lid. The adduct from Example 5 (16.28 g) and Permapol® P3.1E (29.96 g, a thiol-terminated polythioether polymer, available from PRC-Desoto International Inc., Sylmar, CA), triethylenediamine (0.23 g), and Novalac® DEN™ 431 (1.75 g, an epoxy resin available from Dow Chemical, Midland, MI) were added to the 60-gram container. The container was placed in a speed mixer (DAC 600 FVZ) and mixed for 60 seconds for 2,300 rpm. A portion of the mixed material was cured inside

the plastic container for 7 days at room temperature. After 7-day cure, hardness of cured material is 35 Shore A, measured according to ASTM D 2240.

[0252] A second portion of the mixed material was poured onto a 12"×18" ×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 228 psi and an elongation of 276%. The tensile strength and elongation were measured according to ASTM D412.

Example 16

Polythioether Michael acceptor adduct cured with a thiol-terminated polythioether and isocyanate blend

[0253] The mixing was performed in a 60-gram plastic container with lid. The adduct from Example 5 (33.04 g), Permapol® P3.1E (38.05 g, a thiol-terminated polythioether polymer, available from PRC-Desoto International Inc., Sylmar, CA), and an isocyanate-terminated prepolymer (5.0 g, Example 5 of U.S. Patent No. 8,541,513). The container was placed in a speed mixer (DAC 600 FVZ) and mixed for 60 seconds for 2,300 rpm. A portion of the mixed material was allowed to cure inside the plastic container for 7 days at room temperature. After 7-day cure, hardness of cured material was 35 Shore A, measured according to ASTM D 2240.

[0254] A second portion of the mixed material was poured onto a 12"×18" ×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet has a tensile strength of 309 psi and elongation of 576%. The tensile strength and elongation were measured per ASTM D412.

Example 17

Summary Results

[0255] The properties of the cured compositions presented in Examples 1-16 are summarized in Table 3. In general, for aerospace sealant applications, it is desirable that a cured composition exhibit a hardness greater than about 10 Shore A, a tensile strength greater than about 10 psi, an elongation greater than about 100%, and a JRF swell less than about 20 volume%. Note that Examples 13 and 14 include filler, whereas the other compositions contain polymer only. For compositions containing polymer only, it is generally desirable that the composition exhibit a tensile strength greater than 80 psi and elongation greater than 100%.

Table 3. Summary of composition properties.

Example	Polymer	Curing Agent	Hardness Shore A 7-day cure	Tensile Strength (psi)	Elongation (%)	JRF swell (vol%)
1	PTE*	DVS**	—†	—	—	—
2	PTE	DVS	14	—	—	—
3	PTE	DVS	42	696	933	—
4	PTE	DVS	25	546	1077	—
5				—	—	—
6	PTE Adduct	PS [§] LP-980	11	—	—	19.2
7	PTE Adduct	PS LP-32	24	—	—	18.8
8	PTE Adduct	PS LP-12	25	—	—	19.4
9	PTE Adduct	PS G4	25	92	181	18.7
10	PTE Adduct	PS G21	32	88	107	18.5
11	PTE Adduct	PS LP-2	33	108	113	18.1
12	PTE Adduct	PTE	31	446	504	—
13	PTE Adduct	PTE	43	1810	950	—
14	PTE Adduct	PTE (low density)	35	252	772	—
15	PTE Adduct	PTE epoxy blend	35	228	276	—
16	PTE Adduct	PTE isocyanate blend	35	309	576	—

* Polythioether.

** Divinyl sulfone.

§ Polysulfide.

† Not measured.

Example 18**Polythioether Michael acceptor adduct cured with a diamine**

[0256] The divinyl sulfone-terminated polythioether adduct from Example 5 (83.99 g), isophorone diamine (4.26 g), Cab-O-Sil® M5 (3.68 g) and triethylenediamine (0.69 g) were

added to a plastic container. The container was placed in a mixer (DAC 600 FVZ) and the contents mixed for 60 seconds at 2,300 rpm. A portion of the mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, the hardness of cured material measured according to ASTM D 2240 was 24 Shore A.

[0257] A second portion of the mixed material was poured onto a 12"×18" ×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 562 psi and an elongation of 1170%. The tensile strength and elongation were measured according to ASTM D412.

Example 19

Polythioether Michael acceptor adduct cured with a diamine

[0258] The divinyl sulfone-terminated polythioether adduct from Example 5 (83.99 g), isophorone diamine (4.26 g) and Cab-O-Sil M5 (3.68 g) were added to a plastic container. The container was placed in a mixer (DAC 600 FVZ) and the contents mixed for 60 seconds at 2,300 rpm. A portion of the mixed material was cured inside the plastic container for 7 days at room temperature. After 7 days, the hardness of cured material measured according to ASTM D 2240 was 20 Shore A.

[0259] A second portion of the mixed material was poured onto a 12"×18" ×1/4" flat glass substrate and pressed to form a uniform 1/8"-thick sheet. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 420 psi and an elongation of 1209%. The tensile strength and elongation were measured according to ASTM D412.

Example 20

Polythioether Michael acceptor adduct cured with a blocked diamine

[0260] The adduct from Example 5 (16.80 g), Vestamin® A 139 (1.39 g, available from Evonik) and triethylenediamine (0.27 g) were added to a plastic container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 60 seconds at 2,300 rpm. A portion of the mixed material was cured in the plastic container for 5 weeks at ambient conditions. After 5 weeks, the mixed material cured, forming a solid elastomer.

Example 21

Encapsulated Catalyst Preparation

[0261] 9.3 grams of Intelimer® 13-6 (from Air Products and Chemicals, Allentown, PA) and 0.7 gram of isophorone diamine (3-aminomethyl-3,5,5-trimethylcyclohexylamine, Vestamin® IPD, Evonik Industries) were blended at 80°C for 30 minutes. The mixture was

rapidly cooled to room temperature and then ground to powders with an average particle size of 25 microns.

Example 22

Encapsulated Catalyst Preparation

[0262] 9.0 grams of Intelimer[®] 13-1 (from Air Products and Chemicals, Allentown, PA) and 1.0 gram of isophorone diamine were blended at 80°C for 30 minutes. The mixture was rapidly cooled to room temperature and then ground to a powder.

Example 23

Encapsulated Catalyst Preparation

[0263] 9.5 grams of Intelimer[®] 13-1 (from Air Products and Chemicals, Allentown, PA) and 0.5 gram of isophorone diamine were blended at 80°C for 30 minutes. The mixture was rapidly cooled to room temperature and then ground to a powder.

Example 24

Synthesis of Vinyl Sulfone-capped Polythioether Adduct

[0264] In a 300 mL, 3-necked, round bottom flask fitted with a mechanical stirrer, thiol-terminated polythioether polymer Permapol[®] P3.1E (149.40 g, available from PRC-Desoto International, Inc., Sylmar, CA), divinyl sulfone (12.18 g), and triethylenediamine (0.81 g) were added at room temperature. The mixture was stirred for 10 minutes, resulting a vinyl sulfone-terminated polythioether adduct that had a viscosity of 309.0 poise. The viscosity was measured using a CAP2000 viscometer with spindle #6, 50 RPM.

Example 25

Synthesis of Polythioether Polymer

[0265] In a 2 L flask, 524.8 g (3.32 mol) of diethylene glycol divinyl ether (DEG-DVE) and 706.7 g (3.87 mol) of dimercaptodioxaoctane (DMDO) were mixed with 19.7 g (0.08 mol) of triallylcyanurate (TAC) and heated to 77°C. To the heated reaction mixture was added 4.6 g (0.024 mol) of an azobisnitrile free radical catalyst (Vazo[®] 67, 2,2'-azobis(2-methylbutyronitrile), commercially available from DuPont). The reaction proceeded substantially to completion after 2 hours to afford 1,250 g (0.39 mol, yield 100%) of a liquid polythioether resin having a T_g of -68°C and a viscosity of 65 poise. The resin was faintly yellow and had low odor.

[0266] The synthesis of the polythioether and other suitable polythioethers are disclosed in U.S. Patent No. 6,172,179.

Example 26**Encapsulated Catalyst Preparation**

[0267] 9.5 grams of Intelimer[®] 13-1 (from Air Products and Chemicals, Allentown, PA) and 0.5 gram of isophorone diamine were blended at 80°C for 30 minutes. The mixture was rapidly cooled to room temperature and then ground to a powder.

Example 27**Heat-triggered Release – Matrix Encapsulation**

[0268] Mixing was performed in a plastic container with a lid. The polythioether adduct of Example 24 (20 g), T-5314 (32 g, a thiol-terminated intermediate comprising the polymer described in Example 25, available from PRC-Desoto International, Inc., Sylmar, CA), and the matrix-encapsulated catalyst (1.86 g) of Example 21 were added to the container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 10 seconds at 2,300 rpm. A portion of the mixture was heated according to the heating schedule described in Table 1 and another portion of the mixture was kept at ambient conditions. The physical state of the samples is indicated in Table 1.

Table 1

Example 27 Composition	
Heating Schedule	Physical State of Samples
No Heating	Remained liquid for 48 hours
180°F/5min.	Cured in 4 hours after heating
200°F/5min.	Cured in 4 hours after heating

Example 28**Matrix Encapsulation**

[0269] Mixing was performed in a plastic container with a lid. The polythioether adduct of Example 24 (20 g), T-5314 (32 g, a thiol-terminated intermediate comprising the polymer described in Example 25, available from PRC-Desoto International, Inc., Sylmar, CA), and the encapsulated catalyst of Example 22 (1.86 g) were added to the container. The container was

placed in a mixer (DAC 600 FVZ) and the materials mixed for 10 seconds at 2,300 rpm. A portion of the mixture was heated according to the heating scheduled described in Table 2 and another portion of the mixture was maintained at ambient conditions. The physical state of the samples is indicated in Table 2.

Table 2

Example 28 Composition	
Heating Schedule	Physical State of Samples
No Heating	Remained liquid for 24 hours
160°F/5min.	Cured in 4 hours after heating
170°F/5min.	Cured in 4 hours after heating

Example 29

Matrix Encapsulant

[0270] Mixing was performed in a plastic container with a lid. The polythioether adduct of Example 24 (53.79 g), Permapol[®] P3.1E (53.27 g, a thiol-terminated polymer described in Example 25, available from PRC-Desoto International, Inc., Sylmar, CA), silica (7.28 g), and the encapsulated catalyst of Example 23 (13.45 g) were added to the container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 30 seconds at 2,300 rpm. A portion of the mixed material was transferred to five separate metal cans. Each can held about 11 grams of the mixed material. Three of the five cans were heated in an oven at varying temperatures and times as shown in Figure 1. Two of the five cans were heated by an infrared heater for 3 minutes and 5 minutes, respectively. The hardness values were measured on the material after it was exposed to heat. Figure 1 shows the results.

[0271] A second portion of the mixed material was exposed to ambient conditions for 4 days. After 4 days, the mixed material still remained workable.

Example 30

Matrix Encapsulant – Ultrasonic release

[0272] Mixing was performed in a plastic container with a lid. The polythioether adduct of Example 24 (11.76 g), Permapol® P3.1E (1.65 g, a thiol-terminated polymer described in Example 25, available from PRC-Desoto International, Inc., Sylmar, CA), Cab-O-Sil® M5 (1.46 g), and the encapsulated catalyst of Example 23 (2.69 g) were added to the container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 30 seconds at 2,300 rpm. A portion of the mixed material was placed in between two pieces of aluminum plate, each plate having a dimension of 3 in. × 3 in. × 0.001 in. The distance between the two plates was 0.002 in. The set of aluminum plates having the mixed material between was placed in contact with the horn of an ultrasonicator (Model 2000X, available from Emerson Industrial Automation, Danbury, CT) for 3 seconds at 20 KHz. After this, the mixture between two aluminum plates cured in 2 days.

[0273] A second portion of the mixed material was exposed to ambient conditions for 4 days. After 4 days, the mixed material still remained pasty.

Example 31

Photolabile Catalyst

[0274] Mixing was performed in a plastic container with a lid. The polythioether adduct of Example 24 (50.39 g), Permapol® P3.1E (46.74 g, a thiol-terminated polymer described in Example 25, available from PRC-DeSoto International, Inc., Sylmar, CA), and CGI 90 catalyst (1.86 g, a photo-labile amine, available from BASF) were added to the container. The container was placed in a high speed mixer (DAC 600 FVZ) and the materials mixed for 10 seconds at 2,300 rpm. A portion of the mixed material was left inside the plastic container for 4 days at room temperature. After 4 days, the mixture still remained liquid and no curing was observed.

[0275] A second portion of the mixed material was poured onto a 12 in. × 18 in. × ¼ in. flat glass substrate and pressed to form a uniform sheet with approximately 1/8 in. thickness. The sheet was exposed to a UV energy for 60 seconds using a Phoseon Firefly UV light available from Phoseon Technology, Hillsboro, OR. The sheet was cured for 7 days at ambient conditions. The cured sheet had a tensile strength of 605 psi, an elongation of 987%, and a hardness of 35 A. The tensile strength and elongation were measured according to ASTM D412, and the hardness was measured according to ASTM D 2240.

Example 32

Moisture-Release Catalyst

[0276] Mixing was performed in a plastic container with a lid. The polythioether adduct of Example 24 (16.80 g), Permapol[®] P3.1E (15.22 g, a thiol-terminated polymer described in Example 25 available from PRC-Desoto International, Inc., Sylmar, CA), and Vestamin[®] A 139 (0.32 g, a blocked isophorone diamine, commercially available from Evonik) were added to the container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 30 seconds at 2,300 rpm. A portion of the mixed material was allowed to remain inside the plastic container (no moisture exposure) for 1 day at room temperature. After 1 day, the mixture still remained liquid and no curing was observed.

[0277] A second portion of the mixed material was exposed to ambient conditions and allowed to cure for 8 hours at ambient conditions. The mixed material cured as a solid elastomer.

Comparative Example 33

Michael Addition Using Un-encapsulated Triethylamine as Catalyst

[0278] Mixing was performed in a plastic container with a lid. The polythioether adduct of Example 24 (8.40 g), Permapol[®] P3.1E (7.61 g, a thiol-terminated polymer described in Example 25, available from PRC-Desoto International, Inc., Sylmar, CA), and Cab-O-Sil[®] M5 (0.66 g) and triethylamine (0.16 g) were added to the container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 30 seconds at 2,300 rpm.

[0279] The mixed material was allowed to cure at ambient conditions. The mixed material cured as a solid elastomer in 5 weeks. However, the surface of the cured polymer was tacky.

Comparative Example 34

Michael Addition Using Un-encapsulated IPDA as a Catalyst

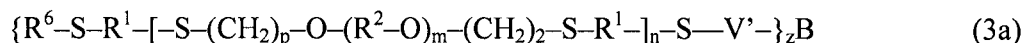
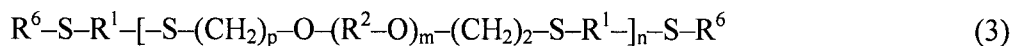
[0280] Mixing was performed in a plastic container with a lid. The polythioether adduct of Example 24 (8.40 g), Permapol[®] P3.1E (7.61 g, a thiol-terminated polymer described in Example 25, available from PRC-Desoto International, Inc., Sylmar, CA), and isophorone diamine (0.16 g) were added to the container. The container was placed in a mixer (DAC 600 FVZ) and mixed for 30 seconds at 2,300 rpm.

[0281] The mixed material was allowed to cure at ambient conditions. The mixed material cured as a solid elastomer within 2 hours.

[0282] Finally, it should be noted that there are alternative ways of implementing the embodiments disclosed herein. Accordingly, the present embodiments are to be considered as illustrative and not restrictive. Furthermore, the claims are not to be limited to the details given herein, and are entitled their full scope and equivalents thereof.

CLAIMS:

1. A polythioether adduct comprising at least two terminal Michael acceptor groups, wherein the polythioether adduct comprises a polythioether adduct of Formula (3), a polythioether adduct of Formula (3a), or a combination thereof:



wherein:

each R^1 independently is selected from C_{2-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-10} alkanecycloalkanediyl, C_{5-8} heterocycloalkanediyl, and $-[(-CHR^3-)_s-X-]_q-(-CHR^3-)_r-$, wherein:

s is an integer from 2 to 6;

q is an integer from 1 to 5;

r is an integer from 2 to 10;

each R^3 is independently selected from hydrogen and methyl; and

each X is independently selected from -O-, -S-, and -NR-, wherein R is selected from hydrogen and methyl;

each R^2 is independently selected from C_{1-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-14} alkanecycloalkanediyl, and $-[(-CHR^3-)_s-X-]_q-(-CHR^3-)_r-$, wherein s, q, r, R^3 , and X are as defined for R^1 ;

m is an integer from 0 to 50;

n is an integer from 1 to 60;

p is an integer from 2 to 6;

B represents a core of a z-valent, vinyl-terminated polyfunctionalizing agent

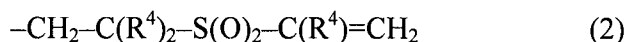
$B(-V)_z$ wherein:

z is an integer from 3 to 6; and

each V is a group comprising a terminal vinyl group; and

each $-V'$ is derived from the reaction of $-V$ with a thiol; and each R^6 is independently a moiety comprising a terminal Michael acceptor group.

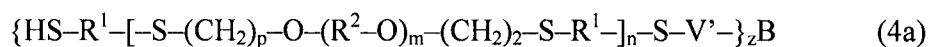
2. The polythioether adduct of claim 1, wherein each R⁶ comprises a vinyl sulfone.
3. The polythioether adduct of claim 1, wherein each R⁶ has the structure of Formula (2):



wherein each R⁴ is independently selected from hydrogen and C₁₋₃ alkyl.

4. The polythioether adduct of claim 1, wherein the polythioether adduct comprises the reaction products of reactants comprising:

(a) a polythioether polymer comprising a polythioether polymer of Formula (4), a polythioether polymer of Formula (4a), or a combination thereof:



wherein:

each R¹ independently is selected from C₂₋₁₀ alkanediyl, C₆₋₈ cycloalkanediyl, C₆₋₁₄ alkanecycloalkanediyl, C₅₋₈ heterocycloalkanediyl, and $[-(\text{CHR}^3-)_s-\text{X}-]_q-(\text{CHR}^3-)_r$, wherein:

s is an integer from 2 to 6;

q is an integer from 1 to 5;

r is an integer from 2 to 10;

each R³ is independently selected from hydrogen and methyl; and

each X is independently selected from -O-, -S-, and -NR-, wherein R is selected from hydrogen and methyl;

each R² is independently selected from C₁₋₁₀ alkanediyl, C₆₋₈ cycloalkanediyl, C₆₋₁₄ alkanecycloalkanediyl, and $[-(\text{CHR}^3-)_s-\text{X}-]_q-(\text{CHR}^3-)_r$, wherein s, q, r, R³, and X are as defined for R¹;

m is an integer from 0 to 50;

n is an integer from 1 to 60;

p is an integer from 2 to 6;

B represents a core of a z-valent, vinyl-terminated polyfunctionalizing agent B(–V)_z wherein:

z is an integer from 3 to 6; and

each –V is a group comprising a terminal vinyl group; and

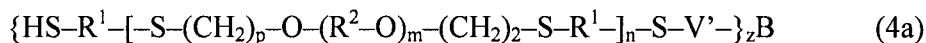
each –V'– is derived from the reaction of –V with a thiol; and

(b) a compound having a Michael acceptor group and a group that is reactive with a terminal group of the polythioether polymer.

5. The polythioether adduct of claim 4, wherein the compound having a Michael acceptor group and a group that is reactive with a terminal group of the polythioether polymer comprises divinyl sulfone.

6. A composition comprising:

(a) a polythioether polymer comprising at least two terminal groups reactive with Michael acceptor groups, wherein the polythioether polymer comprises a polythioether polymer of Formula (4), a polythioether polymer of Formula (4a), or a combination thereof:



wherein:

each R¹ independently is selected from C₂₋₁₀ alkanediyl, C₆₋₈ cycloalkanediyl, C₆₋₁₄ alkanecycloalkanediyl, C₅₋₈ heterocycloalkanediyl, and –[(–CHR³)_s–X]_q–(–CHR³)_r–, wherein:

s is an integer from 2 to 6;

q is an integer from 1 to 5;

r is an integer from 2 to 10;

each R³ is independently selected from hydrogen and methyl; and

each X is independently selected from –O–, –S–, and –NR–, wherein R is selected from hydrogen and methyl;

each R^2 is independently selected from C_{1-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-14} alkanecycloalkanediyl, and $-[(-CHR^3-)_s-X-]_q-(-CHR^3-)_r-$, wherein s, q, r, R^3 , and X are as defined for R^1 ;

m is an integer from 0 to 50;

n is an integer from 1 to 60;

p is an integer from 2 to 6;

B represents a core of a z-valent, vinyl-terminated polyfunctionalizing agent $B(-V)_z$ wherein:

z is an integer from 3 to 6; and

each $-V$ is a group comprising a terminal vinyl group; and

each $-V'$ is derived from the reaction of $-V$ with a thiol; and

(b) a compound having at least two Michael acceptor groups.

7. The composition of claim 6, wherein the compound having at least two Michael acceptor groups has a molecular weight less than 400 Daltons.

8. The composition of claim 6, wherein the compound having at least two Michael acceptor groups comprises the polythioether adduct of claim 1.

9. A composition comprising:

(a) the polythioether adduct of claim 1; and

(b) a curing agent comprising at least two terminal groups that are reactive with Michael acceptor groups.

10. The composition of claim 9, wherein the curing agent comprises a sulfur-containing polymer comprising at least two terminal groups reactive with Michael acceptor groups.

11. The composition of claim 9, comprising a compound selected from a polyepoxy, a polyisocyanate having isocyanate groups that are reactive with thiol groups and that are less reactive with Michael acceptor groups, and a polysulfide adduct.

12. A composition comprising:

(a) a sulfur-containing adduct containing at least two Michael acceptor groups, wherein the sulfur-containing adduct comprises the polythioether adduct of claim 1, a polysulfide adduct having at least two terminal Michael acceptor groups, or a combination thereof;

(b) a sulfur-containing polymer comprising at least two terminal groups reactive with Michael acceptor groups; and

(c) a monomeric compound having at least two Michael acceptor groups.

13. The composition of any one of claims 6, 9 or 12, comprising a controlled release amine catalyst.

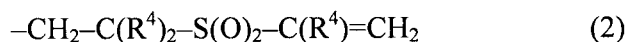
14. The composition of claim 13, wherein the controlled release amine catalyst is selected from a blocked amine catalyst and a matrix encapsulant comprising an amine catalyst.

15. The polythioether adduct of claim 1, wherein each R^6 independently comprises a vinyl ketone, a vinyl sulfone, a quinone, an enamine, a ketamine, an aldimine, or an oxazolidine.

16. The polythioether adduct of claim 4, wherein each Michael acceptor group independently comprises a vinyl ketone, a vinyl sulfone, a quinone, an enamine, a ketamine, an aldimine, or an oxazolidine.

17. The polythioether adduct of claim 4, wherein the Michael acceptor group comprises a vinyl sulfone.

18. The polythioether adduct of claim 4, wherein the Michael acceptor group has the structure of Formula (2):



wherein each R^4 is independently selected from hydrogen and C_{1-3} alkyl.

19. The composition of claim 6, wherein the compound having at least two Michael acceptor groups comprises divinyl sulfone.

20. The composition of claim 9, wherein the curing agent comprises a polythioether polymer selected from a polythioether polymer of Formula (4), a polythioether polymer of Formula (4a), and a combination thereof:



wherein:

each R^1 independently is selected from C_{2-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-14} alkanecycloalkanediyl, C_{5-8} heterocycloalkanediyl, and $\text{--[}(\text{--CHR}^3\text{--})_s\text{--X--}]_q\text{--}(\text{--CHR}^3\text{--})_r\text{--}$, wherein:

s is an integer from 2 to 6;

q is an integer from 1 to 5;

r is an integer from 2 to 10;

each R^3 is independently selected from hydrogen and methyl; and

each X is independently selected from --O-- , --S-- , and --NR-- , wherein R is selected from hydrogen and methyl;

each R^2 independently is selected from C_{1-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-14} alkanecycloalkanediyl, and $\text{--[}(\text{--CHR}^3\text{--})_s\text{--X--}]_q\text{--}(\text{--CHR}^3\text{--})_r\text{--}$, wherein s , q , r , R^3 , and X are as defined for R^1 ;

m is an integer from 0 to 50;

n is an integer from 1 to 60;

p is an integer from 2 to 6;

B represents a core of a z -valent, vinyl-terminated polyfunctionalizing agent $\text{B}(\text{--V})_z$ wherein:

z is an integer from 3 to 6; and

each --V is a group comprising a terminal vinyl group; and

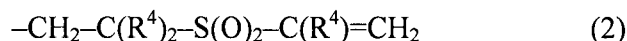
each --V'-- is derived from the reaction of --V with a thiol.

21. The composition of claim 9, wherein the curing agent comprises a thiol-terminated polythioether, a thiol-terminated polysulfide, or a combination thereof.

22. The composition of claim 9, wherein each R^6 independently comprises a vinyl ketone, a vinyl sulfone, a quinone, an enamine, a ketamine, an aldimine, or an oxazolidine.

23. The polythioether adduct of claim 9, wherein each R^6 comprises a vinyl sulfone.

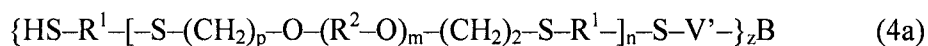
24. The polythioether adduct of claim 9, wherein each R^6 has the structure of Formula (2):



wherein each R^4 is independently selected from hydrogen and C_{1-3} alkyl.

25. The composition of claim 9, wherein,

(b) the sulfur-containing polymer comprises a thiol-terminated polythioether polymer selected from a polythioether polymer of Formula (4), a polythioether polymer of Formula (4a), and a combination thereof:



wherein:

each R^1 independently is selected from C_{2-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-14} alkanecycloalkanediyl, C_{5-8} heterocycloalkanediyl, and $-[(\text{-CHR}^3\text{-})_s\text{-X-}]_q\text{-}(\text{-CHR}^3\text{-})_r\text{-}$, wherein:

s is an integer from 2 to 6;

q is an integer from 1 to 5;

r is an integer from 2 to 10;

each R^3 is independently selected from hydrogen and methyl; and

each X is independently selected from -O-, -S-, and -NR-, wherein R is selected from hydrogen and methyl;

each R^2 is independently selected from C_{1-10} alkanediyl, C_{6-8} cycloalkanediyl, C_{6-14} alkanecycloalkanediyl, and $-[(\text{-CHR}^3\text{-})_s\text{-X-}]_q\text{-}(\text{-CHR}^3\text{-})_r\text{-}$, wherein s, q, r, R^3 , and X are as defined for R^1 ;

m is an integer from 0 to 50;

n is an integer from 1 to 60;

p is an integer from 2 to 6;

B represents a core of a z-valent, vinyl-terminated polyfunctionalizing agent

$B(-V)_z$ wherein:

z is an integer from 3 to 6; and

each -V is a group comprising a terminal vinyl group; and

each -V' is derived from the reaction of -V with a thiol; and

(c) the compound having at least two Michael acceptor groups comprises divinyl sulfone.

1/1

