

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
3 May 2007 (03.05.2007)

PCT

(10) International Publication Number
WO 2007/048745 A1

(51) International Patent Classification:

C09D 4/00 (2006.01) **C09D 183/08** (2006.01)
C04B 41/49 (2006.01) **C08G 77/26** (2006.01)

(74) Agent: **JÖNSSON, Hans-Peter**; von Kreisler Selting
Werner (224), Bahnhofsvorplatz 1, Deichmannhaus Am
Dom, 50667 Cologne (DE).

(21) International Application Number:

PCT/EP2006/067549

(22) International Filing Date: 18 October 2006 (18.10.2006)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:

05110033.7 26 October 2005 (26.10.2005) EP

(71) Applicants (for all designated States except US):
NANOGATE AG [DE/DE]; Gewerbepark Eschberger-
weg, 66121 Saarbrücken (DE). **DOW CORNING LTD.**
[GB/GB]; Cardiff Road, Barry South Glamorgan CF63
2YL (GB).

(72) Inventors; and

(75) Inventors/Applicants (for US only): **DANZEBRINK,**
Rolf [DE/DE]; Römerstrasse 39, 66386 St. Ingbert (DE).
ZIMMERMANN, Lucia [DE/DE]; Karlstrasse 64, 66578
Heiligenwald (DE). **BUTLER, Derek, William** [GB/GB];
"sunny Side" 246, Gladstone Road, Barry Vale Of Glam-
organ CF62 8NG (GB). **HUPFIELD, Peter, Cheshire**
[GB/GB]; Parc-y-gelly,, Henfwich Road, Trevaughan,
Carmarthen SA33 6AA (GB).

(81) Designated States (unless otherwise indicated, for every
kind of national protection available): AE, AG, AL, AM,
AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN,
CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,
GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS,
JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS,
LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MY,
MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS,
RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN,
TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

(84) Designated States (unless otherwise indicated, for every
kind of regional protection available): ARIPO (BW, GH,
GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM,
ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM),
European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI,
FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT,
RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA,
GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

— with international search report

For two-letter codes and other abbreviations, refer to the "Guid-
ance Notes on Codes and Abbreviations" appearing at the begin-
ning of each regular issue of the PCT Gazette.

(54) Title: MIXED SILANES

(57) Abstract: The present invention relates to a composition comprising at least one fluorosilane and at least one aminosilane, a condensation product of said fluorosilane and said aminosilane, and a surface protective agent made thereof.

WO 2007/048745 A1

Mixed Silanes

The present invention relates to a composition comprising a fluorosilane and an aminosilane, a condensation product of said fluorosilane and said aminosilane, and a surface protective agent made thereof.

Silanes are used for building protection as anti-corrosives, anti-graffiti-agents and water-repellents on substrates such as marble, sandstone, concrete, granite, sand-limestone, terracotta, clinker, split-face block or bricks. For such applications, the treatment products need to be preferably water-based and slightly acidic.

Fluorinated silanes exhibit the best performance with regard to simultaneous water-repellence and oil-repellence. Such fluorinated silanes so far possess several drawbacks. First of all, they do not easily form stable solutions, emulsions or dispersions with solvents having a dielectric constant greater than 30 at 20 °C. Secondly, most of the fluorinated silanes used for building protection may release perfluorooctanoic acid (PFOA), which has been found to persist and bioaccumulate in animal and human tissue and to accumulate in the liver where it inhibits glutathione peroxidase, a selenoprotein essential for thyroid hormone conversion, thereby also causing cancer (Occup Environ Med 60(10):722-9 (2003); Int J Cancer 78(4):491-5 (1998)).

US 6,054,601 A discloses compositions of long-chain perfluorinated silanes and aminosilanes that undergo a reaction in aqueous media.

EP 0738771 A1 discloses aqueous compositions comprising long-chain perfluorinated silanes and aminosilanes. Compositions comprising less than 90% water are described to possess shelf instability.

US 5,442,011 A discloses compositions of long-chain perfluorinated silanes and aminosilanes that undergo a reaction in aqueous media.

Therefore, the problem underlying the present invention finally is to provide a stable and non-toxic surface protective agent resulting in good water- and oil-repellence deliverable in a solvent system with a high dielectric constant.

In a first embodiment, the problem underlying the present invention is solved by a reactive composition comprising, in particular consisting of

(a) at least one fluorosilane of the general formula I

$R_{tf}-SiX_3$ (formula I),

wherein

X is selected from the group of alkoxy, halide, oxime, carboxyl, phenoxide and polyether, and

- 3 -

R_{tf} is a straight, branched or cyclic residue of the general formula II or III

-Y- R_f (formula (II)) or

-Y-(SiR_1R_2O) $_x$ SiR_1R_2 -Y- R_f (formula (III)),

wherein

Y is a divalent organic moiety selected from the group of $-(CH_2)_n-$, $-CO_2-$, $-O-$, $-CONH-$, $-Ph-$, $-SO_2-$, and $-SO_2NH-$, wherein n is an integer from 1 to 30,

R_f is a C1 to C30 linear or branched perfluoralkylene group,

R_1 and R_2 are independently selected from monovalent organic residues, x is an integer from 0 to 5, and

(b) and at least one aminosilane of the general formula IV

$R_a-SiR_3R_4R_5$ (formula IV),

where

R_a is a straight, branched or cyclic alkyl residue comprising 1 to 7 carbon atoms and at least one primary, secondary, ternary or quaternary, preferably being protonated, amino-group,

R_3 and R_4 are independently selected from $-R_a$, $-OR_6$ and/or $-R_6$, and

R₅ is -OR₆, and

R₆ is a straight, branched or cyclic alkyl residue comprising 1 to 3 carbon atoms.

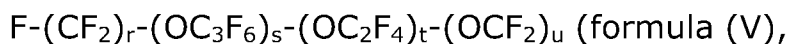
Protonated in the sense of the invention does not necessarily mean a positive charge on the nitrogen atom. It just means that at least one hydrogen atom is connected to the nitrogen atom.

The composition according to the present invention is preferably non-aqueous. "non-aqueous" in the sense of the present invention means that no additional water is added. This does not exclude usual water traces in the starting materials, but excludes the addition of water to the reaction system. Preferably, the composition comprises less than 1 wt.% water, particularly preferred less than 0.1 wt.% water. This is of particular advantage, since it has been found that a composition comprising water will lead to mostly non-hydrolysable condensation products and surface protective agents. Compared to the findings of EP 0738771 A1, surprisingly it has been found that compared to the stable solutions in water with fluorosilanes with at least 8 carbon atom in the fluorinated chain such as the compositions in EP 0738771 A1, the non-aqueous compositions according to the present invention exhibit a high stability and shelf-life with fluorinated alkylsilanes with carbon chain lengths of less than 8 carbon atoms due to their low water content.

Preferably, R_f comprises 1 to 8 carbon atoms. Particularly, R_f is

(a) selected from the group of CF₃-, CF₃CF₂-, CF₃(CF₂)₃-, C₃F₇-, (CF₃)₂CF-, C₄F₉-, C₅F₁₁- or C₆F₁₃- or

(b) a perfluorinated polyether of the general formula V



wherein r is an integer in the range from 1 to 3, and

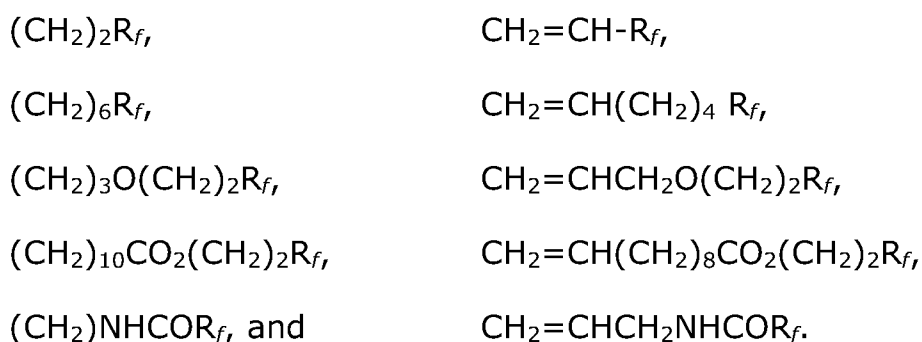
s , t , and u are independently integers in the range from 0 to 200. These perfluorinated polyethers may preferably be homopolymers or block-copolymers comprising units selected from the group $-(\text{CF}(\text{CF}_3)-\text{CF}_2-\text{O})-$, $-(\text{CF}_2-\text{CF}_2-\text{CF}_2-\text{O})-$, $-(\text{CF}_2-\text{CF}_2)-$. These polyether residues are preferably terminated by R_f as defined above.

R_5 and R_6 are preferably the same or different. Examples of such groups are C1 to C30 linear or branched alkylene group, an aromatic containing group, an aminoalkyl containing group, and a fluoroalkyl containing group.

Advantageously, X is a halide selected from the group of F, Br, Cl and I, an alkoxide OR_7 wherein R_7 is a C1 to C22 linear or branched alkylene group, an oxime $\text{R}_8\text{R}_9\text{C}=\text{N}-\text{O}$, wherein R_8 and R_9 are independently selected from C1 to C30 linear or branched alkylene groups, wherein R_8 and R_9 may be same or different, a carboxyl residue R_{10}CO_2 wherein R_{10} is a C1 to C30 linear or branched alkylene group, a phenoxide $\text{M}-\text{Ph}-\text{O}-$, where M is hydrogen or a monovalent organic group, or a polyether selected from the group of polyalkylene oxides containing one or more of the following repeating structural units $(\text{CH}_2\text{CH}_2\text{O})_q$, or $(\text{CH}_3\text{CHCH}_2\text{O})_q$ wherein q is a value in the range from 1 to 100, terminated by a C1 to C30 linear or branched alkylene group.

Preferably, Y is a moiety selected from the group of $-(CH_2)_o-$, $-CO_2-$, $-(CH_2)_o-CO_2-(CH_2)_m-$, $-(CH_2)_o-O-(CH_2)_m-$, $-(CH_2)_o-CONH-(CH_2)_m-$, $-(CH_2)_o-Ph-(CH_2)_m-$, $-(CH_2)_o-SO_2-(CH_2)_m-$, and $-(CH_2)_o-SO_2NH-(CH_2)_m-$, $-SO_2-O-$, $-SO_2NH-$, $-CH_2=CH-$, and $-CH_2=CH-(CH_2)_o-$, wherein o is a number in the range from 1 to 30 and m is a number in the range from 0 to 30, in particular wherein the divalent organic group may also contain branched alkylene groups.

$Y-R_f$ may preferably comprise a unit of a starting olefin and preferably is a residue selected from the group



R_f is preferably a C1 to C30 linear or branched perfluoroalkylene group, in particular selected from the group CF_3- , CF_3CF_2 , $CF_3(CF_2)_3-$, and $(CF_3)_2CF-$.

R_f may preferably also be a perfluorinated polyether of the general formula V

- 7 -

$F-(CF_2)_q-(OC_3F_6)_m-(OC_2F_4)_n-(OCF_2)_o$ (formula (V),

wherein q is an integer in the range from 1 to 3; m, n, and o are independently integers in the range from 0 to 200.

Advantageously, R_f is a straight, branched or cyclic perfluorinated alkyl residue comprising 1 to 30 carbon atoms, R_1 and R_2 are independently selected from $-R_f$, $-CH_2-CH_2-R_f$, $-OR_6$ and/or $-R_6$, and where R_3 is $-OR_6$.

Preferably, R_1 or R_2 are independently selected from the group of C1 to C30 linear or branched alkylene groups, aromatic containing groups, aminoalkyl containing groups, and fluoroalkyl containing groups.

Preferably, in the composition the molar ratio of R_f -groups of formula I to amino-groups present in formula IV is in a range from 2:1 to 6:1, particularly in a range from 2,5:1 to 4:1. This ratio has been found to be particularly stable in solution with solvents having a dielectric constant of at least 30 measured at 20 °C in case of such molar ratios.

The invention is of particular advantage, if R_f comprises 1 to 6 carbon atoms, in particular 3 to 6, more preferred 4 to 6, even more preferred 3 to 5 carbon atoms, since then the resulting surface protective agent will definitely not release PFOA, and since oil-repellence was found to be best in this range.

R_a favorably comprises at least as many carbon atoms as the longest residue of said fluorosilane, since this has been found to yield the most stable solutions, emulsions or dispersions.

Preferably, at most one, in particular none of residues X is R_f or $-\text{CH}_2-\text{CH}_2-R_f$ and/or at most one, in particular none of R_3 , R_4 and R_5 is R_a , since then high hydrophobicity of the treated surface material could be achieved together with good stability of the treatment solution, emulsion or dispersion.

Said fluorosilane may advantageously be present in the composition in a range from 40 to 75 weight % and said aminosilane may advantageously be present in the composition in a range from 10 to 30 weight %.

Preferably, the composition according to the present invention comprises at least one acid in a range between 1 to 90 wt.%, even more preferred in a range between 20 and 50 wt.%, most preferred in a range between 30 and 40 wt.%.

The composition according to the present invention may comprise an additional solvent system comprising a single solvent or a mixture of solvents, where the solvent system has a dielectric constant of at least 30 measured at 20 °C. Of particular advantage is a solvent or solvent mixture selected from the group of alcohols, acetone, water, ethers or N-methylformamide. Said solvent system may preferably be present in the composition in a range from 4 to 20 weight %.

X, R₃, R₄, and/or R₅ are preferably alkoxy groups, in particular ethoxy or methoxy groups, since then the resulting condensation product exhibits a higher stability due to better crosslinking between said fluorosilane and said aminosilane.

Said amino group is preferably a terminal group, in case the residue is not cyclic, i.e. the amino group is attached to a primary carbon atom with only one bond to another carbon atom. The amino group may preferably be -NH₂ or substituted, in particular with one or two -CH₂CH₂NH₂, phenyl groups or cyclohexyl groups. Preferably, the amino group is attached to a straight alkyl chain. These features result in particularly stable solutions, emulsions or dispersions.

Said aminosilane according to the present invention preferably comprises in the complete molecule 4 to 17 carbon atoms, 1 to 4 nitrogen atoms, 2 to 5 oxygen atoms, and 13 to 37 hydrogen atoms. The boiling point thereof is preferably in a range between 100 and 280 °C, whereas the molecular weight thereof is preferably in a range from 170 to 270 g/mol. The flash point thereof is preferably in a range between 70 to 120 °C. Such aminosilane is of advantage, since it poses no fire hazard during normal handling and at the same time results in optimal hydrophobicity of the resulting coating in combination with the fluorosilane.

In a further embodiment, the problem underlying the present invention is solved by a process preparing a reactive composition by combining a

fluorosilane and an aminosilane each according to the composition of the present invention followed by an acid treatment. The reaction time for protonation is preferably in a range from 1 to 20 min, even more preferred in a range from 5 to 15 min. The reaction temperature is preferably in a range of from 40 to 80 °C, even more preferred in a range of from 60 to 75 °C.

In a further embodiment, the problem underlying the present invention is solved by a condensation product of an fluorosilane of the general formula II and/or III and an aminosilane of the general formula IV (master batch), obtainable by a catalytic promoted treatment of a mixture of said fluorosilane and said aminosilane, in particular by an acid treatment.

Preferably, this condensation product is non-aqueous. Preferably, the condensation product only exists in a chemical system comprising less than 1 wt.% water, particularly preferred less than 0,1 wt.% water. Surprisingly, it has been found that only a non-aqueous reaction product is water hydrolysable at a later stage (e.g. as part of a surface protective agent). Also, such reaction product has been found to be much more stable (shelf-life) compared to condensation products produced in an aqueous system.

Preferably, said condensation product is clear and exhibits a haze value of at most 10 %. The haze can be measured according to ASTM D 1003 using 10 mm thick samples of solution, e.g. in cuvettes.

It is of particular advantage, if the fluorosilane undergoes the condensation reaction in the presence of a further hydrophilic silane. It is of also particular advantage, if the fluorosilane undergoes the condensation reaction in the presence of no additional water, i.e. a non-aqueous system. Preferably this hydrophilic silane is a polar material with a dielectric constant of at least 5. Preferably this silane may also comprise monovalent organic groups Z such as epoxide groups. The molar ratio of the fluorosilane to the hydrophilic silane is preferably in a range from 20 to 1. The hydrophilic silane is conforming with the general formula

$R_{11}R_{12}MeSi-Y-Z$ (formula VI),

where R_{11} and R_{12} are independently selected from the group of R_3 or Me, and

Z is a polar monovalent organic group.

Said condensation product according to the present invention exhibits high stability toward the chemical environment due to the fluorosilane and the aminosilane crosslinked with each other, and at the same time results in highly stable solutions, emulsions or dispersions thereof, while still providing highly hydrophobic and oleophobic surface materials treated with such condensation product.

Preferably, the acid used exhibits a pK_a value in a range from 3 to 7, particularly in a range from 3,5 to 5,5. If the pK_a value is too low, the degree of crosslinkage is too high, insoluble or indispersible particles of

too large size are built. If the pK_a value is too high, the degree of crosslinkage is insufficient to form stable solutions, emulsions and dispersions.

The acid is a Lewis acid or a Bronsted acid preferably selected from the group of boric acid, aceto acetic acid, citric acid, crotonic acid, formic acid, fumaric acid, glyceric acid, glycolic acid, lactic acid, malic acid, tartaric acid, and/or acetic acid.

The shape of the condensation product is preferably particles, in particular particles with a medium particle size in a range from 1 to 1000 nm, in particular in a range from 5 to 100 nm. The monodispersity of the condensation product is preferably in a range from 1 to 15 nm. In case the particle size is too large, the penetration into a substrate to be coated becomes worse. Also the stability of a dispersion, e.g. in form of a surface protective agent, containing such larger than preferred particles suffers.

The condensation product preferably exists within a solvent system. This solvent system preferably exhibits a pH in the range from 4 to 5. This pH is preferably accomplished by addition of a Lewis acid or a Bronsted acid.

In a further embodiment, the problem underlying the present invention is solved by a process for obtaining a condensation product according to the invention, characterized in that it comprises at least the step of adding an acid to the composition according to the invention.

Preferably, this process is a non-aqueous process, i.e. a process, in which no additional water is added.

Preferably, the weight ratio of the acid to be added to the composition according to the invention is in a range between 1:1 to 1:4, in particular in a range between 1:1.5 to 1:2.5. In view of the exothermic crosslinkage reaction, preferably no extra heat is provided during or before addition of the acid to prevent degradation of sensitive ingredients also present in the composition.

In a further embodiment, the problem underlying the present invention is solved by a process for the preparation of a fluorosilane according to formula I, II and/or III, characterized by the steps of hydrosilylation of an unsaturated C-C or C-O bond and following alkoxylation substitution of residues attached to the silicon atom after hydrosilylation.

In a further embodiment, the problem underlying the present invention is solved by a surface protective agent comprising said composition according to the invention and/or the condensation product according to the invention further comprising common additives for surface protective agents.

For the first time, a surface protective agent is provided comprising fluorosilanes suitable for a highly polar solvent system.

Preferably, the surface protective agent according to the present invention comprises active components in a range from 20 to 40 wt.%. Active component is preferably the condensation product according to the present invention.

Said surface protective agent according to the present invention may preferably comprise a solvent or a mixture of solvents, where the solvent or the mixture of solvents has a dielectric constant of at least 30 measured at 20 °C. Of particular advantage is a solvent or solvent mixture selected from the group of alcohols, acetone, water, ethers or N-methylformamide. Such high dielectric constant solvent system has been found to adsorb to and infiltrate best polar surface materials such as for example concrete or limestone. Preferably this solvent or mixture of solvents is present in the surface protective agent in an amount in the range from 60 to 80 wt.%.

For this purpose, the surface protective agent according to the present invention preferably comprises at most 5 weight % of solvents with a dielectric constant up to 29 measured at 20 °C and at the same time comprises at least 10, in particular at least 90 weight % of solvents with a dielectric constant of at least 30 measured at 20 °C.

Said surface protective agent preferably comprises 0,1 to 10 weight % of known additives such as a compound or mixture of compounds selected from the group of silicones/siloxanes, acrylic compounds, melamine derivatives, and waxes for better adhesion to the surface material as well as improved hydrophobicity and oleophobicity of the impregnated surface material.

Preferably, the surface protective agent according to the present invention exhibits a pH value in a range from 3 to 6,5 to have best compatibility with and efficiency on the substrate such as for example sandstone, limestone or concrete.

The invention is of particular advantage, if the additives are selected from the group acrylics, waxes, silicones, extenders, and polyurethanes. Preferably, the additives are present in an amount from 0,5 to 5 weight % to improve the overall performance.

Said surface protective agent preferably comprises a diluted solution, emulsion, or dispersion of the composition and/or the condensation product according to the invention for best hydrophobicity and oleophobicity of the impregnated surface materials while at the same time providing the surface protective agent as a stable solution, emulsion or dispersion. Preferably, said surface protective agent comprises an amount of 0,1 to 15, preferably 1 to 7 % by weight of fluorinated compounds namely of said fluorosilanes or of the master batch or the composition or the condensation product according to the present invention.

A stable solution, emulsion or dispersion in the sense of the present invention refers to a solution, emulsion or dispersion exhibiting no significant precipitation or phase separation during storage at room temperature and normal pressure for seven days, preferably for 5 weeks (shelf-life).

In a further embodiment, the problem underlying the present invention is solved by a method for obtaining the surface protective agent according to the present invention comprising at least the step of mixing said composition or condensation product with a solvent system having a dielectric constant of at least 30 measured at 20 °C and additional additives.

In a last embodiment, the problem underlying the present invention is solved by a surface material treated with said surface protective agent according to the present invention.

Said surface material is preferably selected from the group natural stone, marble, sandstone, concrete, granite, sand limestone, terracotta, clinker, split-face block or brick.

The composition according to the present invention may be used for coatings (with e.g. greaseproof, food-release, easy-clean, anti-stain, oil- or water-repellent effect) on natural or artificial stone, on decorative elements such as walls, on household furnishing, on textile material such as woven, non-woven, or carpets, on leather, on plastics, on glass, on metal (e.g. as a mould release coating), on ceramic, on wood or on paper. The surface protective agent according to the present invention is particularly useful as a building protective agent.

The oleophobicity and hydrophobicity of the treated surface material is evaluated using contact angle measurement. The contact angle of

linseed oil against air on the treated surface material is at least 50°, whereas the contact angle of water against air on the treated surface material is at least 100°. The contact angle can be measured at room temperature and normal pressure using sessile drop measurement of drops with a volume of 0,5 ml using a DSA 100 (Krüss GmbH).

Examples:

Abbreviations:

FTS= 3,3,4,4,5,5,6,6,7,7,8,8,8-Tridecafluorooctyltriethoxysilane

B3958 = Nonafluorohexyl-1,1,2,2-H-trimethoxysilan

AMMO = 3-Aminopropyltrimethoxysilane

AMEO = 3-Aminopropyltriethoxysilane

HAC = acetic acid (100% (glacial acetic acid)

HOOC = Formic acid (100%)

HCl = hydrochloric acid (37%)

B2858 = heptafluoroisohexyl-1,1,2,2,3,3-H-trimethoxysilane

Example 1: Surface protective Agent for Stone

To 6g Nonafluorohexyl-1,1,2,2-H-trimethoxysilane (B3958), 12g 3-Amino-propyl-trimethoxysilane (AMMO) was added and stirred at ambient temperature for 5 min. to form the composition. 4 g concentrated acetic acid was added and stirred for 5 min., whereafter 1,3 g deionized water was added under reflux to form the condensation product. This condensation product was diluted with 400 g of methanol. 20 g of the silicone emulsion Z 66-83 (Dow Corning) was added to form the surface protective agent. 500 g of this surface protective agent was applied to 1 m² sand-limestone and dried for 5 days. After drying, olive-oil, green-tea, coffee, red wine, coca cola and ketchup stain could be easily wiped off using water and towel.

Water pickup after 24 h was approximately 260 g/m².

Example 2: Surface protective Agent for Wood

3 g Nonafluorohexyl-1,1,2,2-H-trimethoxysilane were stirred with 3 g glycidyloxypropyl-3-triethoxysilane at ambient temperature for 1 minute. To this mixture were added 100 g methanol and stirred for 1 minute. 10 g acetic acid (25%) were added under reflux and the resulting mixture was stirred for 30 minutes. After subsequent addition of 1 g 3-aminopropyltrimethoxysilane the mixture was stirred for an additional 10 minutes to form the condensation product. The mixture was applied to a wooden surface. After curing for 5 days, the treated wood surface exhibits a strongly diminished water take-up behavior. Water is repelled by the treated wood surface.

Example 3: Surface protective Agent for Textiles

To 6 g Nonafluorohexyl-1,1,2,2-H-trimethoxysilane 1 g 3-aminopropyl-trimethoxysilane is added dropwise, with subsequent addition of 4 g acetic acid (100% (glacial acetic acid)) under reflux to form the condensation product. After stirring for 1 minute, 1.5 g dematerialized water was added and the mixture was stirred for 4 hours. Demineralized water is added to the resulting clear solution to give 3000 g of treatment agent. Textiles were immersed in this protective agent and dried in air. Quality of the resulting protective coating is determined to be better than the quality of known protective coatings according to DIN EN24920 and ISO2 (70).

Example 4: Surface protective Agent for Stone

A 250 ml three necked flask equipped with a stirrer, condenser, thermometer and dropping funnel was charged with 16.0 g of AMMO. Dropwise 36 g of acetic acid (glacial acetic acid) were added to form the quaternary amine. White fume was forming. After stirring for 10 minutes the white fume is drained. The reaction was exothermic and the temperature rose up to 58°C. Then 48 g of FTS was added to form a composition. The temperature was adjusted to 65°C and was maintained for 48 hours by heating. After cooling down to ambient temperature, this intermediate condensation product was stable for more than 1 year when storing at ambient temperature and was soluble with water in all concentrations.

2.5% of this intermediate in water and some commercial additives like silicones gave best performance on sandstone and concrete. The diluted final product was stable (shelf-life) for more than 1 year.

Example 5: Surface protective Agent for Stone

The procedure of Example 4 was repeated except that B3958 was used for the FTS. The diluted final product was stable (shelf-life) for more than 1 year.

Example 6: Surface protective Agent for Stone

The procedure of Example 1 was repeated except that B2858 was used for the FTS. The diluted final product was stable (shelf-life) for more than 1 year.

Comparative Example 1 (according to EP 0738771 A1):

A 250 ml three necked flask equipped with a stirrer, condenser, thermometer and dropping funnel was charged with 17.07 g of AMEO. Then 35.53 g of FTS was added. 42.1 g IPA is given to the mixture to form one phase. 1.58 g of HOOC and 0.05 g of HCl and 3,67 g of water were added dropwise. The temperature was adjusted to 80°C and maintained for 3 hours by heating. After cooling down to ambient temperature this intermediate was not stable for more than 4 weeks when storing at ambient temperature and was soluble with water only in low concentrations such as 1 wt.%.

2.5% of this intermediate in water and some commercial additives like silicones gave satisfactory performance on sandstone and concrete. The diluted final product was not stable for more than 4 weeks (shelf-life).

Comparative Example 2 (according to EP 0738771 A1):

A 250 ml three necked flask equipped with a stirrer, condenser, thermometer and dropping funnel was charged with 46.5 g of B3958. Then 15.5 g of AMMO were added. 34.9 g of HAC were added dropwise. A two phase mixture was obtained. By dropping 3.1 g of water to the

mixture, a hydrolysis was initiated and the temperature rose up to 60°C. After cooling down to ambient temperature, the formed intermediate was soluble with water in low concentrations such as 1 wt.%. The stability (shelf-life) of the diluted and the intermediate is restricted to at most 4 weeks.

Comparative Example 3 (according to EP 0738771 A1):

The procedure of Comparative Example 1 was repeated except that the amount of water used was 9.74 g. Like in Comparative Example 2, the formed intermediate was also soluble with water in low concentrations such as 1 wt.%. The stability (shelf-life) of the diluted and the intermediate was restricted to at most 4 weeks.

Claims

1. Reactive composition comprising

(a) at least one fluorosilane of the general formula I

$R_{tf}-SiX_3$ (formula I),

wherein

X is selected from the group of alkoxy, halide, oxime, carboxyl, phenoxide and polyether, and

R_{tf} is a straight, branched or cyclic residue of the general formula II or III

$-Y-R_f$ (formula (II)) or

$-Y-(SiR_1R_2O)_xSiR_1R_2-Y-R_f$ (formula (III)),

wherein

Y is a divalent organic moiety selected from the group of $-(CH_2)_n-$, $-CO_2-$, $-O-$, $-CONH-$, $-Ph-$, $-SO_2-$, and $-SO_2NH-$, wherein n is an integer from 1 to 30,

R_f is a C1 to C30 linear or branched perfluoralkylene group,

R_1 and R_2 are independently selected from monovalent organic residues,

x is an integer from 0 to 5, and

(b) and at least one aminosilane of the general formula IV

$R_a\text{-SiR}_3\text{R}_4\text{R}_5$ (formula IV),

where

R_a is a straight, branched or cyclic alkyl residue comprising 1 to 7 carbon atoms and at least one primary, secondary, ternary or quaternary, preferably protonated, amino-group,

R_3 and R_4 are independently selected from $-R_a$, $-\text{OR}_6$ and/or $-R_6$, and

R_5 is $-\text{OR}_6$, and

R_6 is a straight, branched or cyclic alkyl residue comprising 1 to 3 carbon atoms.

2. Composition according to claim 1, characterized in that X is

a halide selected from the group of F, Br, Cl and I,

an alkoxide OR_7 wherein R_7 is a C1 to C22 linear or branched alkylene group,

an oxime $R_8R_9\text{C}=\text{N}-\text{O}$, wherein R_8 and R_9 are indepently selected from C1 to C30 linear or branched alkylene groups, wherein R_8 and R_9 may be same or different,

a carboxyl residue $R_{10}\text{CO}_2$ wherein R_{10} is a C1 to C30 linear or branched alkylene group,

a phenoxide M-Ph-O-, where M is hydrogen or a monovalent organic group, or

a polyether selected from the group of polyalkylene oxides containing one or more of the following repeating structural units $(\text{CH}_2\text{CH}_2\text{O})_q$, or $(\text{CH}_3\text{CHCH}_2\text{O})_q$ wherein q is a value in the range from 1 to 100, terminated by a C1 to C30 linear or branched alkylene group.

3. Composition according to claim 1, characterized in that Y is a moiety selected from the group of $-(\text{CH}_2)_o-$, $-\text{CO}_2-$, $-(\text{CH}_2)_o-\text{CO}_2-(\text{CH}_2)_m-$, $-(\text{CH}_2)_o-\text{O}-(\text{CH}_2)_m-$, $-(\text{CH}_2)_o-\text{CONH}-(\text{CH}_2)_m-$, $-(\text{CH}_2)_o-\text{Ph}-(\text{CH}_2)_m-$, $-(\text{CH}_2)_o-\text{SO}_2-(\text{CH}_2)_m-$, and $-(\text{CH}_2)_o-\text{SO}_2\text{NH}-(\text{CH}_2)_m-$, $-\text{SO}_2-\text{O}-$, $-\text{SO}_2\text{NH}-$, $-\text{CH}_2=\text{CH}-$, and $-\text{CH}_2=\text{CH}-(\text{CH}_2)_o-$, wherein o is a number in the range from 1 to 30 and m is a number in the range from 0 to 30, in particular wherein the divalent organic group may also contain branched alkylene groups.

4. Composition according to claim 1, characterized in that R_f is

(a) selected from the group of CF_3- , CF_3CF_2- , $\text{CF}_3(\text{CF}_2)_3-$, C_3F_7- , $(\text{CF}_3)_2\text{CF}-$, C_4F_9- , $\text{C}_5\text{F}_{11}-$ or $\text{C}_6\text{F}_{13}-$ or

(b) a perfluorinated polyether of the general formula V

$\text{F}-(\text{CF}_2)_r-(\text{OC}_3\text{F}_6)_s-(\text{OC}_2\text{F}_4)_t-(\text{OCF}_2)_u$ (formula (V)),

wherein r is an integer in the range from 1 to 3, and

s, t, and u are independently integers in the range from 0 to 200.

5. Composition according to claim 1, characterized in that R_f comprises 1 to 6 carbon atoms.
6. Composition according to claim 1, characterized in that R_1 or R_2 are independently selected from the group of C1 to C30 linear or branched alkylene groups, aromatic containing groups, aminoalkyl containing groups, and fluoroalkyl containing groups.
7. Composition according to claim 1, characterized in that X, R_3 , R_4 , and/or R_5 independently are selected from the group of alkoxy groups, in particular ethoxy or methoxy groups.
8. Composition according to claim 1, characterized in that the molar ratio of R_f -groups of formula I to amino-groups present in formula IV is in a range from 2:1 to 6:1, particularly in a range from 2,5:1 to 4:1.
9. Composition according to claim 1, characterized in that it further comprises a hydrophilic silane conforming with the general formula

$R_{11}R_{12}MeSi-Y-Z$ (formula VI),

where R_{11} and R_{12} are independently selected from the group of R_3 or Me, and

Z is a polar monovalent organic group.

10. Composition according to any of claims 1 to 9, characterized in that it comprises an acid in a range between 20 and 50 wt.%, preferably in a range between 30 and 40 wt.%.
11. Non-aqueous composition according to any of claims 1 to 10.
12. Process for preparing a reactive composition by combining a fluorosilane and an aminosilane each according to claim 1 followed by an acid treatment.
13. Process according to claim 12 characterized in that said fluorosilane and/or the aminosilane are reacted in protonated form.
14. Process according to claim 12 or 13 in an non-aqueous reaction system.
15. Condensation product of an fluorosilane of the general formula II and/or III and an aminosilane of the general formula IV, obtainable by a catalytic promoted treatment of a mixture of said fluorosilane and said aminosilane, in particular by an acid treatment.
16. Condensation product according to claim 15, characterized in that the acid exhibits a pK_a value in a range from 3 to 7, particularly in a range from 4 to 5,5.

17. Surface protective agent comprising the composition according to claim 1 and/or the condensation product of claim 15 further comprising common additives for surface protective agents.

18. Surface protective agent according to claim 17, characterized in that it comprises 0,5 to 5 weight % of a compound or mixture of compounds selected from the group of silicones/siloxanes, acrylic compounds, melamine derivatives, and waxes.

INTERNATIONAL SEARCH REPORT

International application No

PCT/EP2006/067549

A. CLASSIFICATION OF SUBJECT MATTER

INV. C09D4/00 C04B41/49 C09D183/08 C08G77/26

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

C09D C04B C08G

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, PAJ, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 6 054 601 A (STANDKE ET AL) 25 April 2000 (2000-04-25) claims 1-18; examples 1,5	1-18
X	EP 0 738 771 A (SHIN-ETSU CHEMICAL CO., LTD) 23 October 1996 (1996-10-23) claims 1-8; example 1	1-18
X	US 5 442 011 A (HALLING ET AL) 15 August 1995 (1995-08-15) claims 1-19; examples 2,4	1-18



Further documents are listed in the continuation of Box C.



See patent family annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search

8 January 2007

Date of mailing of the international search report

15/01/2007

Name and mailing address of the ISA/

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Buestrich, Ralf

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

PCT/EP2006/067549

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
US 6054601	A	25-04-2000	NONE	
EP 0738771	A	23-10-1996	DE 69613623 D1	09-08-2001
			DE 69613623 T2	08-05-2002
			JP 3196621 B2	06-08-2001
			JP 9003403 A	07-01-1997
			US 5739369 A	14-04-1998
US 5442011	A	15-08-1995	AU 693170 B2	25-06-1998
			AU 1937595 A	18-09-1995
			CA 2182866 A1	08-09-1995
			DE 69510334 D1	22-07-1999
			DE 69510334 T2	04-11-1999
			EP 0748357 A1	18-12-1996
			JP 3597864 B2	08-12-2004
			JP 9509972 T	07-10-1997
			WO 9523830 A1	08-09-1995