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**YAMAUCHI, K. ET AL., J. Chem. Soc., Perkin Trans. I, 1973, vol. 21, pages 2506-2508.**  
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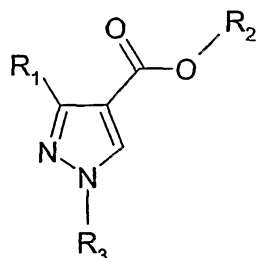
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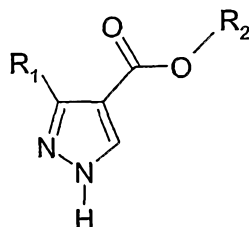
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(54) Title: PROCESS FOR THE PREPARATION OF PYRAZOLES

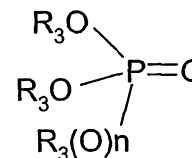
(I)



(II)



(III)



(57) Abstract: The present invention relates to a process for the preparation of compounds of formula (I), wherein R<sub>1</sub> is C<sub>1</sub>-C<sub>4</sub>haloalkyl; R<sub>2</sub> is C<sub>1</sub>-C<sub>6</sub>alkyl and R<sub>3</sub> is methyl or ethyl, by reaction of compounds of formula (II), wherein the substituents are as defined for formula (I), with compounds of formula (III), wherein R<sub>3</sub> is as defined for formula (I) and n is 0 or 1.

### Process for the preparation of pyrazoles

The present invention relates to a process for the regioselective N-alkylation of substituted pyrazoles and to the use of trialkyl phosphates or trialkyl phosphonates in the regioselective N-alkylation of substituted pyrazoles.

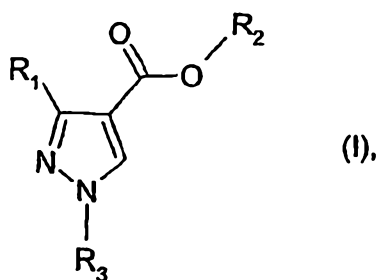
N-alkylated substituted pyrazoles, for example 3-difluoromethyl-1-methyl-1H-pyrazole-4-carboxylic acid ethyl ester, are valuable intermediates in the preparation of fungicides, as described, for example, in WO 03/074491.

According to WO 95/25099, N-alkylated substituted pyrazoles can be prepared by reacting the corresponding substituted pyrazoles with alkyl halides under basic conditions. The use of alkyl halides in the N-alkylation of substituted pyrazoles is problematic, however, on account of their toxic properties. Furthermore, those compounds are expensive and, in addition, exhibit only a low degree of regioselectivity – in respect of the two nitrogen atoms of the pyrazole ring. For those reasons, such processes are particularly unsuitable for large-scale preparation of N-alkylated substituted pyrazoles.

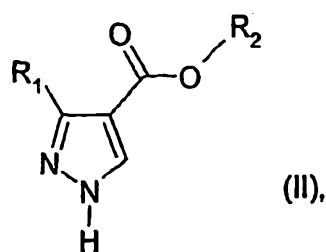
According to JP-2000-044541, N-alkylated substituted pyrazoles can be prepared by reacting the corresponding substituted pyrazoles with carboxylic acid dialkyl esters, with addition of a base. The use of carboxylic acid dialkyl esters is not desirable, because those compounds are of low reactivity and it is therefore generally necessary to increase the reactivity of the substituted pyrazoles by addition of a base. Furthermore, the regioselectivity of such N-alkylation is generally dependent upon the chemical nature of the substituents on the pyrazole ring, so that N-alkylations using carboxylic acid dialkyl esters in some cases exhibit unsatisfactory regioselectivity.

The present invention therefore advantageously provides a novel process for the preparation of N-alkylated substituted pyrazoles that avoids the disadvantages of the known processes mentioned above and makes it possible to prepare those compounds in high yields and good quality in an economically advantageous and easily handled way.

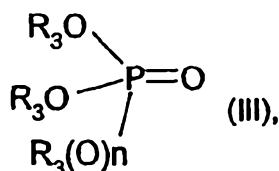
The present invention accordingly relates to a process for the preparation of compounds of formula I



wherein  $R_1$  is  $C_1$ - $C_4$ haloalkyl;  $R_2$  is  $C_1$ - $C_6$ alkyl and  $R_3$  is methyl or ethyl,  
by reaction of a compound of formula II



wherein  $R_1$  and  $R_2$  are as defined for formula I,  
with a compound of formula III



wherein  $R_3$  is as defined for formula I and  $n$  is 0 or 1.

The alkyl groups appearing in the above substituent definitions may be straight-chain or branched and are, for example, methyl, ethyl, n-propyl, isopropyl, n-butyl, sec-butyl, isobutyl or tert-butyl, preferably methyl or ethyl. Halogen is generally fluorine, chlorine, bromine or iodine, preferably fluorine.  $C_1$ - $C_4$ Haloalkyl groups are derived from the mentioned  $C_1$ - $C_4$ alkyl groups and are preferably difluoromethyl or trifluoromethyl.

The process according to the invention is suitable preferably for the preparation of compounds of formula I wherein

$R_1$  is difluoromethyl or trifluoromethyl;

$R_2$  is methyl or ethyl and/or

$R_3$  is methyl.

The process according to the invention is especially suitable for the preparation of compounds of formula I wherein  $R_1$  is difluoromethyl.

The process according to the invention is very especially suitable for the preparation of compounds of formula I wherein  $R_1$  is difluoromethyl;  $R_2$  is ethyl and  $R_3$  is methyl.

The process according to the invention is also very especially suitable for the preparation of compounds of formula I wherein  $R_1$  is trifluoromethyl;  $R_2$  is ethyl and  $R_3$  is methyl.

In preferred processes, compounds of formula II are reacted with compounds of formula III wherein  $n$  is 1.

In especially preferred processes, compounds of formula II are reacted with compounds of formula III wherein  $n$  is 1 and  $R_3$  is methyl.

The reaction according to the invention is preferably carried out in a temperature range of from 100°C to 200°C, especially from 150°C to 200°C.

The reaction according to the invention can be carried out in an anhydrous, inert solvent. Suitable solvents are, for example, xylene, mesitylene, tert-butyl benzene, chlorobenzene, 1,2-dichlorobenzene, Decalin, dibutyl ether, dipentyl ether, diphenyl ether and anisole. The reaction according to the invention is preferably carried out without a solvent.

In the reactions according to the invention, compounds of formula III are used in equimolar amounts or in excess relative to compounds of formula II, preferably in an up to 30-fold excess, especially in an up to 10-fold excess, more especially in a 2-fold to 8-fold excess.

The process according to the invention is very especially suitable for the preparation of compounds of formula I wherein  $R_1$  is difluoromethyl,  $R_2$  is ethyl and  $R_3$  is methyl, by reaction of a compound of formula II wherein  $R_1$  is difluoromethyl and  $R_2$  is ethyl with a compound of formula III wherein  $R_3$  is methyl and  $n$  is 1, in a temperature range of from 150°C to 200°C, without a solvent, the compound of formula III being used in a 2-fold to 8-fold excess relative to the compound of formula II.

The compounds of formula II are known or can be prepared analogously to processes known in the literature. For example, such compounds can be prepared from the 3-oxo-carboxylic acid esters on which they are based by means of a two-step synthesis by reaction with trimethyl orthoformate and subsequent reaction with hydrazine. Such reactions are described, for example, in JP-2000-044541. A further synthesis route for the preparation of compounds of formula II is described in JP-2001-322983, wherein, for example, 3-trifluoromethyl-1H-pyrazole-4-carboxylic acid ethyl ester is prepared starting from 3-chloro-4,4,4-trifluoro-2-formyl-2-butenoic acid ethyl ester by reaction with hydrazine.

Compounds of formula III are known as alkylating agents and are commercially available. For example, the N-alkylation of unsubstituted nitrogen-containing heterocycles is described in *Journal of the Chemical Society, Perkin Transactions 1*, 21, 2506-2508 (1973) and in *Bulletin of the Chemical Society of Japan*, 50, 1510-1512 (1977). There is no mention of such alkylating agents having regioselective properties in the N-alkylation of pyrazoles.

The present invention relates also to compounds of formula I when prepared by the process according to the present invention.

The present invention relates also to the use of compounds of formula III in the regioselective alkylation of compounds of formula II.

The present invention relates also to a process for the regioselective alkylation of compounds of formula II, wherein a compound of formula III is used as alkylating agent.

The present invention is illustrated with the aid of the following Examples:

**Example P1: Preparation of 3-difluoromethyl-1-methyl-1H-pyrazole-4-carboxylic acid ethyl ester.**

A mixture of 5.7 g of 3-difluoromethyl-1H-pyrazole-4-carboxylic acid ethyl ester (30 mmol) and 25 ml of trimethyl phosphate (214 mmol) is stirred at a temperature of 180°C for 18 hours. 250 ml of an ice-water mixture are then added. The resulting reaction product is filtered, washed with water and dissolved in 50 ml of ethyl acetate. The organic phase is washed with 50 ml of saturated sodium chloride solution and dried over sodium sulfate and concentrated by evaporation. 3.9 g (64 % of theory) of 3-difluoromethyl-1-methyl-1H-pyrazole-4-carboxylic acid ethyl ester are obtained in the form of crystals (m.p. 59-60°C).

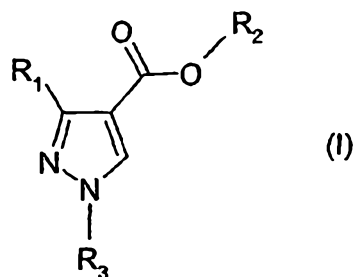
Example P2: Preparation of 3-trifluoromethyl-1-methyl-1H-pyrazole-4-carboxylic acid ethyl ester:

A mixture of 4.16 g of 3-trifluoromethyl-1H-pyrazole-4-carboxylic acid ethyl ester (20 mmol) and 10 ml of trimethyl phosphate (86.4 mmol) is stirred at a temperature of 180°C for 16 hours. 200 ml of an ice-water mixture are then added. The resulting reaction product is filtered, washed with water and dissolved in 50 ml of ethyl acetate. The organic phase is washed twice with 50 ml of saturated sodium chloride solution each time and dried over sodium sulfate and concentrated by evaporation. 4.0 g (90 % of theory) of 3-trifluoromethyl-1-methyl-1H-pyrazole-4-carboxylic acid ethyl ester are obtained in the form of crystals (m.p. 55-57°C).

Example P3: Preparation of 3-trifluoromethyl-1-methyl-1H-pyrazole-4-carboxylic acid ethyl ester:

A mixture of 2.08 g of 3-trifluoromethyl-1H-pyrazole-4-carboxylic acid ethyl ester (10 mmol) and 2.3 ml of trimethyl phosphate (20 mmol) is stirred at a temperature of 180°C for 16 hours. 200 ml of an ice-water mixture are then added. The resulting reaction product is filtered, washed with water and dissolved in 50 ml of ethyl acetate. The organic phase is washed twice with 50 ml of saturated sodium chloride solution each time and dried over sodium sulfate and concentrated by evaporation. 1.9 g (86 % of theory) of 3-trifluoromethyl-1-methyl-1H-pyrazole-4-carboxylic acid ethyl ester are obtained in the form of crystals (m.p. 55-57°C).

The following compounds of formula I can be prepared on the basis of the above Examples:

**Table 1:** Compounds of formula I

Comp. No.	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>
A1	CF <sub>2</sub> H	CH <sub>2</sub> CH <sub>3</sub>	CH <sub>3</sub>
A2	CF <sub>2</sub> H	CH <sub>3</sub>	CH <sub>3</sub>
A3	CF <sub>2</sub> H	CH <sub>3</sub>	CH <sub>2</sub> CH <sub>3</sub>
A4	CF <sub>2</sub> H	CH <sub>2</sub> CH <sub>3</sub>	CH <sub>2</sub> CH <sub>3</sub>
A5	CF <sub>3</sub>	CH <sub>2</sub> CH <sub>3</sub>	CH <sub>3</sub>
A6	CF <sub>3</sub>	CH <sub>3</sub>	CH <sub>3</sub>
A7	CF <sub>3</sub>	CH <sub>3</sub>	CH <sub>2</sub> CH <sub>3</sub>
A8	CF <sub>3</sub>	CH <sub>2</sub> CH <sub>3</sub>	CH <sub>2</sub> CH <sub>3</sub>

The present invention makes it possible for substituted pyrazoles to be alkylated in a controlled manner in a high yield, with a high degree of regioselectivity and at low cost.

A further advantage of the present invention is that substituted pyrazoles can be alkylated without addition of bases.

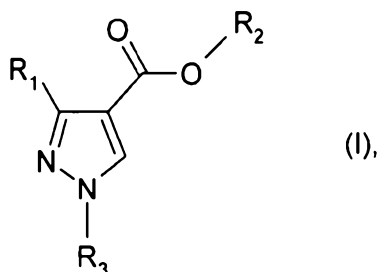
The starting materials for the process of the present invention are distinguished by ready accessibility and ease of handling and are also inexpensive.

In a preferred embodiment of the invention, the process is carried out without a solvent, such an embodiment constituting an especially inexpensive variant of the process according to the invention.

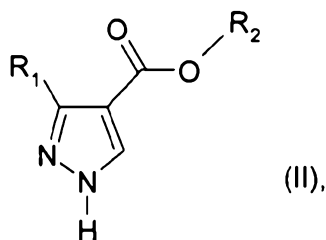
The reference in this specification to any prior publication (or information derived from it), or to any matter which is known, is not, and should not be taken as an acknowledgment or admission or any form of suggestion that that prior publication (or information derived from it) or known matter forms part of the common general knowledge in the field of endeavour to which this specification relates.

THE CLAIMS DEFINING THE INVENTION ARE AS FOLLOWS:

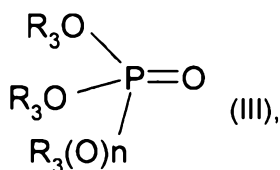
1. A process for the preparation of a compound of formula I



5 wherein R<sub>1</sub> is C<sub>1</sub>-C<sub>4</sub>haloalkyl; R<sub>2</sub> is C<sub>1</sub>-C<sub>6</sub>alkyl and R<sub>3</sub> is methyl or ethyl,  
wherein a compound of formula II



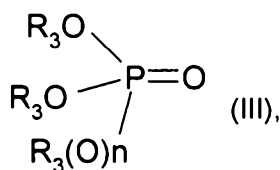
wherein R<sub>1</sub> and R<sub>2</sub> are as defined for formula I, is reacted with a compound of formula III



10 wherein R<sub>3</sub> is as defined for formula I and n is 0 or 1.

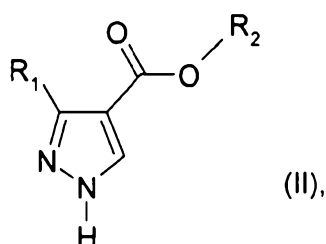
2. A process according to claim 1, wherein the reaction is carried out without addition of a solvent.

15 3. Use of a compound of formula III



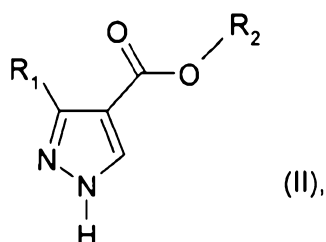
wherein R<sub>3</sub> and n are as defined in claim 1, in the regioselective alkylation of a compound of formula II

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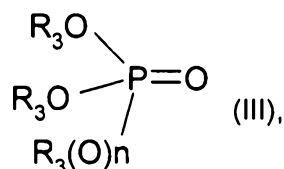
wherein R<sub>1</sub> and R<sub>2</sub> are as defined in claim 1.

4. A process for the regioselective alkylation of a compound of formula II



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wherein R<sub>1</sub> and R<sub>2</sub> are as defined in claim 1, wherein a compound of formula III



wherein R<sub>3</sub> and n are as defined in claim 1, is used as alkylating agent .

10 5. A process according to claim 1 or 2, wherein R<sub>1</sub> is difluoromethyl.

6. A use according to claim 3, wherein R<sub>1</sub> is difluoromethyl.

7. A process according to claim 4, wherein R<sub>1</sub> is difluoromethyl.

15

8. A process according to claim 1 or 2, wherein R<sub>1</sub> is difluoromethyl; R<sub>2</sub> is ethyl and R<sub>3</sub> is methyl.

9. A use according to claim 3, wherein R<sub>1</sub> is difluoromethyl; R<sub>2</sub> is ethyl and R<sub>3</sub> is  
20 methyl.

10. A process according to claim 4, wherein R<sub>1</sub> is difluoromethyl; R<sub>2</sub> is ethyl and R<sub>3</sub> is methyl.

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11. A compound of formula I when prepared by the process of any one of claims 1, 2, 5 and 8.
12. A process according to claim 1 or 4, or a use according to claim 3, substantially as  
5 hereinbefore described with reference to any one of the examples.
13. A compound according to claim 11, substantially as hereinbefore described with reference to any one of the examples.