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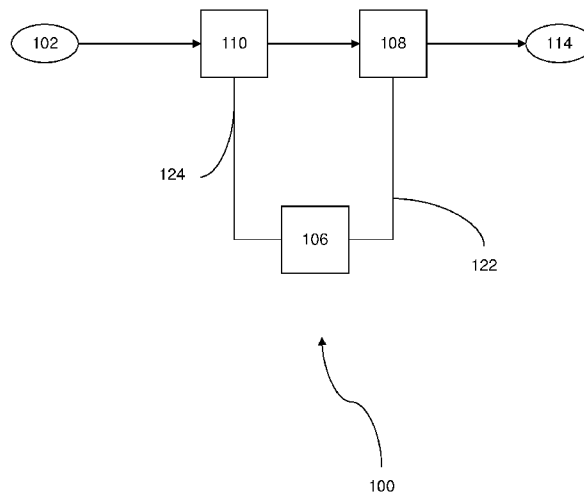


FIG. 1

(57) **Abstract:** Water treatment systems including electrically-driven and pressure-driven separation apparatus configured to produce a first treated water suitable for use as irrigation water and a second treated water suitable for use as potable water from one of brackish water and saline water and methods of operation of same.



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TREATMENT OF SALINE WATER FOR AGRICULTURAL AND POTABLE USE

CROSS-REFERENCE TO RELATED APPLICATIONS

5 This application claims priority under 35 U.S.C. §119(e) to U.S. Provisional Patent Application No. 62/547,971, titled “TREATMENT OF SALINE WATER FOR AGRICULTURAL AND POTABLE USE”, filed August 21, 2017, which is incorporated by reference herein in its entirety for all purposes.

10 BACKGROUND OF INVENTION

1. Field of Invention

This invention relates to systems and methods of providing crop irrigation water as well as potable water and, more particularly, to systems and methods of providing irrigation water and/or potable water from water having unacceptable
15 dissolved solids content.

2. Discussion of Related Art

Climate change is expected to pose an increasing threat to agriculture as global temperatures increase and farms require more water to satisfy the needs of their crops.
20 A constraint to providing increased water by irrigation of farmlands is the availability of freshwater. Agriculture consumes more freshwater than any other industry – an estimated 70% of total global freshwater withdrawal. Many areas are already water stressed. Continued global warming may cause additional problems regarding freshwater availability by, for example, altering the geographical distribution of
25 freshwater and changing the distribution, quantity, and annual timing of rainfall. Accordingly, the production of water suitable for irrigation purposes from otherwise unsuitable water sources may be of increasing importance.

Desalting or desalination refers to a water treatment process that removes salt from, for example, water. In some cases, the water source is brackish water or
30 seawater and desalting techniques thereof provide at least a portion of municipal requirements for potable, drinking water. Desalination techniques typically include

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those based on distillation as well as reverse osmosis techniques. The desalted water can also be consumed in commercial and industrial applications as, for example, process feed water, boiler feed water, and irrigation water. Particular examples of industries that may utilize desalted water include the pharmaceutical, mining, paper and pulp, and agricultural industries.

SUMMARY

In accordance with one aspect, there is provided a water treatment system for producing potable water and water for irrigation from saline water. The system comprises an electro dialysis apparatus including one or more monovalent selective membranes and having an inlet fluidly connectable to a source of water to be treated having a concentration of dissolved salts of between 500 mg/L and 10,000 mg/L, a diluate outlet, and a concentrate outlet, a low-pressure nanofiltration apparatus positioned downstream of the electro dialysis apparatus and having an inlet fluidly connectable to the diluate outlet of the electro dialysis apparatus, a permeate outlet, and a retentate outlet, an irrigation distribution system fluidly connectable to the retentate outlet of the nanofiltration apparatus and with the diluate outlet of the electro dialysis apparatus, and a diverting system configured to vary an amount of diluate from the electro dialysis apparatus directed to the irrigation distribution system and an amount of diluate from the electro dialysis apparatus directed to the nanofiltration apparatus.

In some embodiments, one or both of diluate of the electro dialysis system and permeate of the nanofiltration system is post-treated to remove excess boron.

In some embodiments, the system further comprises a pre-filtration system in fluid communication between the source of water to be treated and the inlet of the electro dialysis apparatus. The pre-filtration system may include at least one of a microfilter, a settler, a screen filter, a microsand filter, and a coarse particle filter. The pre-filtration system may be preceded by a pH adjusting system and an air stripping system.

In some embodiments, the system further comprises a conduit fluidly connecting the permeate outlet of the nanofiltration apparatus to at least one of an

inlet of a concentrating compartment or inlet of an electrode stream compartment of the electro dialysis apparatus. The conduit may directly connect the permeate outlet of the nanofiltration apparatus to at least one of an inlet of a concentrating compartment or inlet of an electrode stream compartment of the electro dialysis apparatus.

5 In some embodiments, the concentrate outlet of the electro dialysis apparatus is fluidly connectable to an inlet of an electrochemical generation system for production of one or more of chlorine gas, hypochlorite ion, sodium hydroxide, sulfuric acid, or hydrochloric acid. The electrochemical generation system may be fluidly connectable to a source of pH adjustment agent configured to introduce pH adjustment agent into
10 the water to be treated.

In some embodiments, the system further comprises a water softening apparatus upstream of the inlet to the electrochemical generation system.

In some embodiments, the system further comprises a system for treating permeate of the nanofiltration apparatus to achieve at least one of upward pH
15 adjustment, lowered pH adjustment, addition of chlorine gas or addition of hypochlorite ion.

In some embodiments, the electro dialysis apparatus further includes one or more monovalent cation selective membranes.

In some embodiments, the one or more monovalent cation selective
20 membranes have a selectivity coefficient for sodium over calcium of at least about 2.

In some embodiments, the electro dialysis apparatus further includes one or more monovalent anion selective membranes. The one or more monovalent anion selective membranes may have a selectivity coefficient for chloride over sulfate of at least about 2.

25 In some embodiments, the electro dialysis apparatus is configured to produce diluate having a sodium adsorption ratio (SAR) value of less than about 10. The electro dialysis apparatus may be configured to produce diluate having a SAR value of less than about 5. The nanofiltration apparatus may be configured to produce retentate having a SAR value less than the SAR value of the diluate. The
30 nanofiltration apparatus may be configured to produce retentate and permeate, the retentate having a SAR value less than the SAR value of the permeate.

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In some embodiments, the electro dialysis apparatus is configured to produce diluate having a higher ratio of divalent cations including magnesium and calcium to monovalent cations including sodium and a lower total concentration of ions than in the water to be treated.

5 In some embodiments, wherein the nanofiltration apparatus is configured to produce retentate having a higher concentration of the divalent cations than the concentration of the divalent cations in the diluate and a lower concentration of the monovalent cations than the concentration of the monovalent cations in the diluate.

In some embodiments, the system further comprises a controller configured to
10 adjust the amount of diluate from the electro dialysis apparatus directed to the irrigation distribution system and the amount of diluate from the electro dialysis apparatus directed to the nanofiltration apparatus with the diverting system in any amount from 0% to 100% of an amount of diluate produced in the electro dialysis apparatus. The controller may be configured to adjust the amount of diluate directed
15 to the irrigation distribution system and the amount of diluate directed to the nanofiltration apparatus depending on a relative demand for irrigation water and potable water.

In some embodiments, the system further comprises a mixer configured to blend diluate produced in the electro dialysis apparatus with any retentate produced in
20 the nanofiltration apparatus and produce irrigation water having a SAR value intermediate of a SAR value of the diluate and a SAR value of the retentate.

In some embodiments, the system further comprises a sensor in communication with the controller, the sensor configured to provide the controller with an indication of one or more parameters of one of diluate from the electro dialysis
25 apparatus and retentate from the nanofiltration apparatus, the controller configured to adjust the diverting system responsive to the indication of the one or more parameters.

In some embodiments, the system further comprises a sensor in communication with a controller, the sensor configured to provide the controller with an indication of one or more parameters of one of diluate from the electro dialysis
30 apparatus and retentate from the nanofiltration apparatus, the controller configured to

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adjust one or more operating parameters of the electro dialysis apparatus responsive to the indication of the one or more parameters.

In accordance with another aspect, there is provided a method of treating saline water to produce potable water and irrigation water. The method comprises directing water to be treated having a concentration of dissolved salts of between 500 mg/L and 10,000 mg/L to an inlet of an electro dialysis apparatus including one or more monovalent cation selective membranes, treating the water to be treated in the electro dialysis apparatus to produce a diluate, determining an amount of the diluate to direct to an irrigation distribution system and an amount of the diluate to direct to an inlet of a low-pressure nanofiltration apparatus, treating any diluate directed to the inlet of the low-pressure nanofiltration apparatus in the nanofiltration apparatus to produce a permeate and a retentate, and combining any retentate with any of the diluate directed to the irrigation distribution system at the irrigation distribution system.

In some embodiments, the method further comprises directing at least a portion of any permeate produced in the nanofiltration apparatus to a concentrating compartment of the electro dialysis apparatus.

In some embodiments, producing the diluate in the electro dialysis apparatus comprises producing treated water having a SAR value of less than about 10.

In some embodiments, producing the diluate in the electro dialysis apparatus comprises producing treated water having a SAR value of less than about 5.

In some embodiments, producing the retentate in the nanofiltration apparatus comprises producing treated water having a SAR value of less than the SAR value of the diluate.

In some embodiments, the method further comprises directing at least a portion of the diluate to the irrigation distribution system.

In some embodiments, combining any retentate with any of the diluate directed to the irrigation distribution system comprises producing irrigation water with a SAR value intermediate of a SAR value of the diluate and a SAR value of the retentate.

In some embodiments, combining any retentate with any of the diluate directed to the irrigation distribution system comprises producing irrigation water with a SAR value of less than about 8 and a total dissolved solids level of greater than about 750 ppm.

5 In some embodiments, the method further comprises adjusting an amount of diluate from the electro dialysis apparatus directed to the irrigation distribution system and the amount of diluate from electro dialysis apparatus directed to the nanofiltration apparatus based on one of one or more parameters of the diluate from the electro dialysis apparatus and flow rate of retentate from the nanofiltration apparatus.

10 In some embodiments, the method further comprises adjusting one or more operating parameters of the electro dialysis apparatus based on one or more parameters of one of the diluate from the electro dialysis apparatus and any retentate from the nanofiltration apparatus.

In some embodiments, the method further comprises pre-filtering the water to be treated prior to directing the water to be treated to the inlet of the electro dialysis apparatus.

In some embodiments, the method further comprises directing permeate suitable for use as potable water from the permeate outlet of the nanofiltration apparatus to a potable water point of use.

20 In some embodiments, a pH of the water to be treated is reduced prior to introduction into the electro dialysis apparatus such that a concentration of bicarbonate ions in the water to be treated is reduced. The pH may be reduced using a controlled amount of one of either hydrochloric or sulfuric acid. An amount of hydrochloric or sulfuric acid added to the water to be treated may be determined based on a
25 relationship between the pH of the water to be treated, and one of either of a concentrate water recovery of the electro dialysis apparatus or a SAR value of diluate of the electro dialysis apparatus. The reduced pH level water to be treated may be treated in an air stripping apparatus to remove a portion of dissolved or gaseous carbon dioxide prior to introduction of the water to be treated into the electro dialysis
30 apparatus.

In some embodiments, the water to be treated is subjected to aeration followed by filtration such that a portion of one or more of elements or compounds of iron, manganese, arsenic, or sulfur are reduced in concentration prior to introduction of the water to be treated into the electro dialysis apparatus.

5 In some embodiments, a ratio of retentate outlet flow to permeate flow through the nanofiltration apparatus is controlled so as to limit a scaling potential of dissolved constituents of the nanofiltration retentate.

In some embodiments, effluent from one of more of the nanofiltration permeate stream or the electro dialysis diluate outlet stream is further treated to reduce dissolved concentration of boron compounds.

10 In some embodiments, a system including the nanofiltration apparatus and the electro dialysis apparatus is controlled and operated such that permeate from the nanofiltration apparatus meets water quality standards for potable drinking water.

In some embodiments, concentrations of one or both of dissolved boron or silica compounds in the water to be treated remain essentially the same in concentrating and diluting streams of the electro dialysis apparatus, and one or more of retentate and permeate of the nanofiltration apparatus.

15 In some embodiments, a system including the nanofiltration apparatus and the electro dialysis apparatus is controlled and operated such that retentate from the nanofiltration apparatus, diluate from the electro dialysis apparatus, or a combination of both meet water quality requirements for agricultural water.

In some embodiments, a system including the nanofiltration apparatus and the electro dialysis apparatus is controlled and operated such that concentrate from the electro dialysis apparatus meets water quality standards for use in produced oil recovery operations. The concentrate may be treated with one of ion exchange softening to reduce a concentration of divalent cations or pH adjustment prior to use in produced oil recovery operations.

25 In some embodiments, a ratio of divalent selenate or selenite ion concentration to chloride ion concentration in the concentrate from the electro dialysis apparatus is less than that of the water to be treated.

30

In some embodiments, a ratio of divalent selenate or selenite ion concentration to chloride concentration in diluate from the electro dialysis apparatus is less than that of the water to be treated and a ratio of calcium ion concentration to sodium ion concentration in the diluate from the electro dialysis apparatus is greater than that of the water to be treated.

In some embodiments, the method further comprises removing at least about 50% of nitrite present in the water to be treated in the electro dialysis apparatus. The method may further comprise removing at least about 50% of nitrate present in diluate of the electro dialysis apparatus directed to the nanofiltration apparatus in the nanofiltration apparatus.

BRIEF DESCRIPTION OF DRAWINGS

The accompanying drawings are not intended to be drawn to scale. In the drawings, each identical or nearly identical component that is illustrated in various figures is represented by a like numeral. For purposes of clarity, not every component may be labeled in every drawing. In the drawings:

FIG. 1 is a schematic illustration of a system in accordance with one or more features of the invention;

FIG. 2 is a schematic illustration of an irrigation system in accordance with further features of the invention;

FIG. 3 is a schematic illustration of another system in accordance with further features of the invention;

FIG. 4 is a schematic illustration of another system in accordance with further features of the invention;

FIG. 5 is a schematic illustration of another system in accordance with further features of the invention; and

FIG. 6 is a graph showing representative ranges of acceptable levels of water characteristics in accordance with some aspects of the invention.

DETAILED DESCRIPTION

This invention is not limited in its application to the details of construction and the arrangement of components set forth in the following description or illustrated in the drawings. The invention is capable of embodiments and of being practiced or of
5 being carried out in various ways beyond those exemplarily presented herein.

Scarcity of irrigation water of sufficient quality is deleterious to crop yields and may require choice of crop species that are of less demand. Newer methods of irrigation that reduce the amount of water used, using techniques such as drip
10 irrigation, may also cause a non-sustainable condition due to salt and impurity buildup in the soil from the water used for irrigation. The soil salinity may rise to much higher concentrations than in the irrigation water due to use of most of the water by the crops, and by evaporation. Conditions of irrigation and soil with inadequate
15 source water for leaching the soil or insufficient rainfall may result in soil salinities 4 to 5 times higher than in the irrigation water itself. Further, should the land consist of relatively shallow impermeable ground layers, the irrigation water may raise the water table. When highly saline ground water reaches crop root levels, the water may be harmful to crop growth. Also, saline soils may damage leafy crops due to water splash off the soil surface. Furthermore, if the agricultural land is drained of the saline water, trace impurities in the soil such as selenium or boron, or residual
20 contaminants from fertilizer use such as nitrate may cause contamination of the drainage water and cause difficulties in safe effluent control.

Irrigation water needs also are in competition with potable drinking water for humans, and water free of contaminants for livestock, and wildlife. Thus it is commonly the case that a source of a combination of irrigation water and potable
25 water are needed in agricultural regions.

One or more aspects of the invention can involve systems and techniques for providing water suitable for agricultural facilities. Other aspects of the invention can provide potable water or water suitable for human use or consumption as well as for livestock and poultry. Some systems and techniques of the invention can convert or
30 otherwise render non-potable water suitable for agricultural, livestock, poultry, and/or human consumption. Still further aspects of the invention can involve systems and

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techniques that preferentially or selectively remove some species over other species from a fluid to be treated to provide a product having one or more desirable characteristics. In contrast with non-selective techniques, some selective removal aspects of the invention can be more cost effective by avoiding or reducing additional post-treatment processes, for example, blending. Thus, the systems and techniques of the invention economically provide treated water that is more suitable for an intended use.

In some embodiments of the invention, some types of species are retained in a treated fluid stream while other types of species are preferentially removed. The resultant product fluid can be utilized in various applications and/or otherwise satisfy one or more objectives. Other aspects of the invention can involve systems and techniques that provide water having one or more properties or characteristics tailored to satisfy a particular purpose. Some embodiments of the invention can thus involve systems and techniques that provide one or more water streams or bodies that have one or more attributes that have been adjusted based on one or more parameters of the point of use or facility in which the stream or body is to be utilized.

Even further aspects of the invention can involve systems and techniques that economically provide water for agricultural, industrial, commercial, and/or residential service. Further, some particular aspects of the invention can involve providing water to serve a plurality of requirements or levels of purity or quality. Thus in some embodiments, the systems and techniques of the invention can provide one or more water streams or bodies in a mixed use facility. Particularly advantageous aspects of the invention can involve providing the plurality of water streams or bodies, each of which may have differing water quality levels, from a source of water having high solids content, to a plurality of points of use, each of which may have differing requirements. Such aspects of the invention can provide systems and techniques that treat, for example, non-potable water to render it potable and/or suitable for irrigation, for livestock and/or poultry consumption, and/or for human consumption or use.

In some aspects of the invention, water having a high level of one or more objectionable species dissolved therein can be treated to remove or at least reduce the concentration of such species to an acceptable level. The one or more objectionable

species can be any species that render the untreated water unsuitable for a particular application. For example, the water may contain a high level or undesirable concentration of monovalent cations and/or anions which adversely or undesirably hinders retention of water in soil or adsorption of other species, including, for
5 example, divalent or even multivalent species. If the requirement is pertinent to crop irrigation, the undesirable condition or characteristic can involve water that contains one or more species that affects the permeability and/or infiltration properties of the soil being irrigated. For example, some aspects of the invention can involve rendering or treating water to preferentially remove monovalent species over non-
10 monovalent species.

In accordance with one or more particular aspects, the invention can involve embodiments directed to systems and/or methods comprising providing or introducing water to be treated into an electrically-driven separation apparatus. Some
15 embodiments of the invention can involve an irrigation system comprising an electrically-driven separation apparatus fluidly connected, or at least connectable, to one or more sources of water to be treated and at least one irrigation water distribution system.

Other aspects of the invention may involve a method of providing potable water. Some aspects of the invention can provide irrigation water and/or potable
20 water without thermally-driven separation techniques, high pressure operation, or unit operations. For example, a method in accordance with some embodiments of the invention can comprise one or more acts or steps of providing water to be treated and treating at least a portion of the water to be treated in an electrically-driven separation apparatus to produce a first treated water. The method can further comprise one or
25 more acts of treating a portion of the water to be treated, typically a separate portion, in one or more pressure-driven separation apparatus to produce a second treated water. In some cases, the method can further comprise a step of using the first treated water as feed to the second apparatus to provide a second treated water to produce the potable water. The potable water typically has a target or desired total dissolved
30 solids (TDS) content, for example, a TDS content of less than about 500 ppm.

Aspects of the invention directed to systems that provide potable water can comprise a source of water to be treated and a pressure-driven separation apparatus having an inlet that is fluidly connected, or at least connectable, to the source of water to be treated. The pressure-driven apparatus can also have one or more outlets, typically at least one product outlet as a treated water outlet. The pressure-driven separation apparatus typically also has at least one reject outlet as an outlet for a stream containing one or more species, typically an undesirable species, removed from the treated water. The system for providing potable water can further comprise one or more electrically-driven separation apparatus which can be fluidly connected, or connectable, to the source of water to be treated, to the pressure-driven separation apparatus, or both. For example, as described in further detail below, one or more electrically-driven separation apparatus can be fluidly connected to a reject outlet of the pressure-driven separation apparatus. In accordance with particular embodiments of the invention, the system for providing potable water can further comprise one or more mixers having one or more inlets fluidly connected, or connectable, to the treated water outlet of the pressure-driven apparatus and the product water outlet of the electrically-driven separation apparatus. The mixer can comprise any mixing unit operation that facilitates at least partially blending or combining one or more product streams including, in some cases, a stream from the source of water to be treated to form a final product stream having one or more desirable characteristics.

The water to be treated can comprise brackish water having a TDS of between about 1,500 mg/L (1,500 ppm) and about 5,000 mg/L (5,000 ppm), and/or saline water containing high concentrations of dissolved solids or salts, for example, a concentration of dissolved salts of between about 500 mg/L and about 10,000 mg/L. Other sources of water to be treated can comprise water that would be unsuitable for use in agricultural facilities because of infiltration and/or toxicity considerations.

The systems and techniques of the invention can comprise, where appropriate, pre-treatment subsystems to facilitate one or more operating principles thereof. One or more pre-treatment and/or post-treatment unit operations may be utilized in one or more embodiments of the invention. For example, the systems and techniques of the invention may comprise a pre-treatment subsystem including one or a plurality of

filters or screens that separate or remove at least a portion of any suspended solids from the water to be treated. Such pre-treatment subsystems typically remove particulate material that would damage any downstream unit operation of the systems of the invention, or, for example, irrigation equipment. Other pre-treatment unit operations may include, for example, microfilters and/or sedimentary-based systems that can remove suspended solids having characteristic dimensions of, for example, one micron or greater, or 10 microns or greater.

Further pre-treatment operations may be utilized to improve the effectiveness of one or more unit operations of the invention. For example, a pre-treatment subsystem can comprise coolers or heaters that, respectively, cool or heat the water to be treated prior to separation operations. Cooling of the raw feed stream, or any intermediate process stream may be performed to, for example, facilitate the transport of an undesirable species, or to hinder the transport of a desirable species, from the stream to be treated. Likewise, heating may be performed to raise the temperature of the raw feed stream, or one or more intermediate process streams, to a desired temperature that, for example, facilitates economical or efficient operation of the one or more separation apparatus. Non-limiting examples of heating processes may involve heaters, furnaces, or heat exchangers that may be associated with or be a unit operation of a process or system of the invention. For example, heating may be provided through a heat exchanger of a power plant that is not necessarily associated with the treatment systems of the invention.

Pre-treatment systems may also or alternatively be used to remove or reduce the concentration one or more species in the water to be treated to improve the effectiveness of one or more treatment unit operations, reduce the potential for scaling, or remove undesirable species that may be difficult to remove in the one or more treatment unit operations. For example, the pH of the water to be treated may be adjusted to cause one or more dissolved species to precipitate or to be converted into a different form so that it may be removed from the water to be treated by, for example, filtration, settling, or air stripping. pH adjustment agents that may be used include, for example, hydrochloric acid, sulfuric acid, sodium hydroxide, or any other pH adjustment agent known in the art. Species that may be removed from the water

to be treated may include, for example, elemental or ionic forms or compounds of any one or more of bicarbonate, iron, manganese, arsenic, sulfur, selenium (e.g., divalent selenate or selenite), boron, or silica. In other embodiments, a pre-treatment system may include a water softening apparatus, for example, an ion exchange system that
5 may remove hardness components, for example, calcium from the water to be treated.

Post-treatment unit operations may polish, remove, or reduce the concentration one or more species in the treated water. For example, one or more ion exchange columns may be utilized to remove species that are not readily removed in the electrically-driven separation apparatus and/or the pressure-driven separation
10 apparatus. Non-limiting examples of species that would typically be removed or at least have a reduction in concentration to, preferably, non-toxic and/or non-objectionable levels, in post-treatment operations include those that may affect soil aggregation, water infiltration, and/or would be toxic to plant growth such as aluminum, arsenic, beryllium, boron, cadmium, cobalt, chromium, copper, iron,
15 fluoride, lithium, manganese, molybdenum, nickel, lead, selenium, tin, titanium, tungsten, vanadium, boron, and zinc. An example of an ion exchange resin that may be utilized to remove species such as boron is the AMBERLITE™ PWA10 boron selective ion exchange resin available from the Dow Chemical Company. Other species that may be addressed by one or more post-treatment operations include those
20 that may be toxic or objectionable to humans, poultry, and/or livestock in drinking water such as, but not limited to, nitrates, nitrites, selenium, vanadium, and sulfides. Disinfecting processes may also be performed to at least partially inactivate or reduce the concentration of colony-forming microorganisms that may be harmful to human and/or livestock or that create biofilms that may clog irrigation nozzles.

Alternatively, or in combination with the one or more polishing unit
25 operations, the systems and techniques of the invention may involve adding one or more species to at least a portion of the treated water. For example, gypsum may be added to adjust the concentration of one or more desirable species or adjust a characteristic of the water. Other additives may include fertilizers, disinfectants, or
30 other supplements that facilitate crop growth when the water is used for irrigation.

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An electrically-driven apparatus typically utilizes a potential field to create a motive force that induces one or more species, typically the target species, which can include desirable as well as undesirable species, to migrate from the carrier or fluid. The electrically-driven apparatus can utilize one or more components that segregate the target species during migration and/or inhibit the return or reverse process. Non-limiting examples of such devices include electrodialysis (ED) devices, including current reversing electrodialysis (EDR) devices, as well as electrodeionization (EDI) devices. “Electrodialysis” is a broader term and includes “electrodeionization.” Another form of electrically-driven separation apparatus that may be utilized in various aspects and embodiments disclosed herein is a capacitive deionization (CapDI) apparatus which deionizes water by applying an electrical potential difference over two porous carbon electrodes. Anions are removed from the water and are stored in the positively polarized electrode; likewise, cations are stored in the cathode, which is the negatively polarized electrode. Aspects and embodiments disclosed herein are described as including ED apparatus, however, it is to be appreciated that in alternate embodiments other forms of electrically-driven separation apparatus may be used in addition to or instead of ED apparatus. Aspects and embodiments disclosed herein are not limited to one or a combination of particular electrically-driven separation apparatus and may be practiced with other apparatus that provide a motive force that facilitates the preferential migration of one or more target species over other species in the fluid to be treated.

The electrically-driven separation apparatus of the invention typically utilize ion selective membranes to facilitate separation phenomena. In some cases, the selectively permeable membrane can preferentially or selectively allow transport of some species relative to other species. For example, cation selective membranes may be utilized in some compartments of the electrically-driven separation apparatus. In other embodiments, anion selective membranes may be utilized in one or more compartments in addition to or instead of the cation selective membranes. In still other embodiments, the electrically-driven separation apparatus of the invention may comprise one or more monovalent ion selective membranes to selectively promote transfer of the monovalent cationic or anionic species. Indeed, in some embodiments

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of the invention, the separation apparatus of the invention may comprise monovalent cation selective membranes and one or more monovalent anion selective membranes. Non-limiting examples of commercially available monovalent selective membranes include NEOSEPTA® cation and anion selective membranes from ASTOM

5 Corporation, Tokyo, Japan or Tokuyama Corporation, Tokyo, Japan. The monovalent cation or monovalent anion selective membranes may be selected to have a selectivity of, for example, greater than about 2, greater than about 5, or greater than about 10. In some embodiments, the monovalent cation selective membranes may have a selectivity coefficient for sodium over calcium of at least about 2, at least about 5, or

10 at least about 10. In some embodiments, the monovalent anion selective membranes may have a selectivity coefficient for chloride over sulfate of at least about 2, at least about 5, or at least about 10. As the term is used herein, selectivity may be calculated according to the following formula which illustrates a calculation of selectivity of sodium (Na) over magnesium (Mg) and calcium (Ca), although the selectivity of

15 separation apparatus for other elements or compounds over one another may be similarly calculated:

$$Selectivity = \frac{\frac{\Delta v_{Na}}{v_{Na}}}{2 \left[\frac{\Delta v_{Ca} + \Delta v_{Mg}}{v_{Ca} + v_{Mg}} \right]}$$

where v is the molarity of ionic species i and Δv is the change in the molarity of ionic species i.

20 The monovalent cation and or anion selective membranes may provide for the embodiments of electrically-driven separation apparatus to be operated to produce treated water (diluate) having a sodium adsorption ratio (SAR) of less than about 20, less than about 9, or less than about 3 from brackish water having a TDS of between about 1,500 mg/L (1,500 ppm) and about 5,000 mg/L (5,000 ppm), and/or saline

25 water containing high concentrations of dissolved solids or salts, for example, a concentration of dissolved salts of between about 500 mg/L and about 10,000 mg/L. These SAR values may be achievable due to the preferential transfer of monovalent cations vs. divalent or multivalent cations into concentrating compartments of the

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electrically-driven separation apparatus, thus increasing a ratio of the divalent or multivalent cations to monovalent cations in the diluting compartments of the electrically-driven separation apparatus.

5 A pressure-driven separation apparatus typically utilizes one or more barriers to inhibit migration of a first species therethrough while allowing penetration of another. The motive force facilitating the separation phenomena typically involve pressurizing fluid to be treated. Non-limiting examples of pressure-driven separation apparatus include microfiltration and nanofiltration (NF) apparatus as well as reverse osmosis (RO) systems. NF apparatus typically selectively pass monovalent ions as compared to divalent or multivalent ions. Accordingly, the permeate produced in a 10 NF apparatus may have a higher concentration ratio of monovalent ions than divalent or multivalent ions than in the retentate produced in a NF apparatus. The retentate produced in a NF apparatus may thus have a lower SAR value than the permeate or the water influent to be treated in the NF apparatus. Examples of nanofilters that have 15 a high selectivity for passing monovalent as compared to divalent ions and may be utilized in various embodiments disclosed herein may include SWSR nanofiltration membranes available from the General Electric Company, NF245 nanofiltration membranes, NF 270 nanofiltration membranes, or NF345 nanofiltration membranes, available from the Dow Chemical Company or ESNA1-LF-LD nanofiltration 20 membranes or HYDRACoRe10 nanofiltration membranes available from Hydranautics.

One or more embodiments of the invention can be directed to a water treatment system 100 as exemplarily shown in FIG. 1. System 100 can be a system for providing potable water, irrigation water, or both, to, for example, a point of use 25 114. The treatment system 100 can comprise at least one separation unit operation or separation apparatus 110 that, in some cases, selectively removes one or more species or types of species from water from the source 102 of water to be treated. The system can optionally comprise one or more monitoring subsystems that provide an indication of one or more operating characteristics of the treatment system. As 30 illustrated, system 100 can have one or more monitoring sensors 108 that typically provide an indication of water quality produced, or otherwise treated, from the

separation apparatus 110. In some aspects of the present invention, system 100 can utilize a control system or controller configured or constructed and arranged to regulate one or more parameters of one or more unit operations in the systems of the invention. Referring again to FIG. 1, system 100 can thus have one or more
5 controllers 106 that adjust at least one operating parameter of separation apparatus 110 typically to at least one desired condition. The one or more monitoring sensors 108 may communicate with the one or more controllers 106 through a communications network including, for example, a signal line 122, or additionally or alternatively, through a wireless communication link. The one or more controllers 106 may
10 communicate with the at least one separation unit operation or separation apparatus 110 through a communications network including, for example, a signal line 124, or additionally or alternatively, through a wireless communication link. Any suitable control technique may be utilized to adjust the at least one operating parameter of any unit operation in system 100 to provide treated water having the one or more desired
15 characteristics. For example, in some embodiments, a sensor/transmitter (not shown) located between the source 102 of water to be treated and the separation unit operation or separation apparatus 110 may be utilized to provide measurements of one or more parameters of the water to be treated to the controller 106 to control the removal of contaminants or water recovery from the treatment unit 110.

20 The systems and techniques of the invention may include one or more water distribution systems that facilitate delivery of the treated water to one or more points of use. For example, the distribution system may include an irrigation distribution system that delivers irrigation water to various points of use in an agricultural facility. To facilitate the delivery of the treated water, the distribution system can include one
25 or more storage systems, such as reservoirs, tanks, wells, or other vessels and containers. The irrigation systems of the invention may utilize overhead and/or surface or sub-surface irrigation techniques to convey water to a designated area. The irrigation system components can thus employ non-movable as well as mobile devices.

30 The one or more storage systems may be considered as part of the distribution system or be an ancillary subsystem of the treatment system. The one or more storage

systems may further facilitate providing treated water having desired characteristics. Treated water having a first condition or characteristic may be stored in one or more storage systems prior to further treatment or processing, for example, blending, with another treated or untreated water body or stream.

5 FIG. 2 is a schematic diagram exemplarily showing some features of the invention pertinent to an irrigation system 200. Irrigation system 200 can comprise a separation apparatus 220 fluidly connected and, as illustrated, disposed to receive water to be treated from source 202, for example, through conduit 222. Separation apparatus 220 can treat water from source 202 and provide treated water to a first
10 point of use 228, illustrated herein as a first type of crop through irrigation water distribution system 224. Point of use 228 can be a portion of a crop that, for example, is at a stage of growth different from at least one portion of the entire crop. System 200 can further comprise one or more second separation apparatus 230. Separation apparatus 230 can also treat water from source 202 and provide treated water to a
15 second point of use 238, illustrated as a second type of crop, through second irrigation distribution system 234. Second point of use 238 may be a portion the same type of crop to be irrigated as, for example, first point of use 228 or a portion of a second crop at a different stage of growth. In accordance with some embodiments of the invention, separation apparatus 230 can optionally provide treated water to first point
20 of use 228, instead of and/or to supplement treated water from separation apparatus 220, through conduit or connection 244. Some embodiments of the invention contemplate, at least partially, a staged treatment scheme. For example, first separation apparatus 220 may provide treated water having a first water quality or characteristic which can further be treated in second separation apparatus 230 through
25 conduit or distribution system 242. A plurality of second separation apparatus 230 may be utilized with one or more first separation apparatus 220 to provide treated water to one or more points of use. Some embodiments of the invention may involve serial arrangement of separation apparatus and other embodiments may utilize separation apparatus in parallel configurations to provide treated water so as to satisfy
30 the volumetric requirements of the one or more points of use. For example, water to be treated from source 202 may be provided in parallel to both first separation

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apparatus 220 and second separation apparatus 230 through conduits or distribution systems 222 and 232, respectively. In some cases, however, a combination of serial and parallel treatment paths may be implemented to provide treated water at a rate or a plurality of rates, wherein each of the one or more treated water streams have one or more desired characteristics.

System 200 can include one or more controllers (not shown) to control one or more operating parameters of any component or subsystem of system 200. Like the system exemplarily illustrated in FIG. 1, system 200 can have one or more controllers that can adjust one or more operating parameters. For example, one or more controllers of system 200 can have adjust the current, potential, or both, of the applied electric field in any of the separation apparatus. Other parameters that may be adjusted include, for example, TDS content, pressure, temperature, pH, flow ratio or any combination, of any stream of the system.

In accordance with some aspects of the invention, the one or more characteristics of the treated water stream can be any measured or derived attribute of the product stream so as to render it suitable for its intended use at point 114. However, the invention is not limited as such; for example, the characteristic of the water may be an attribute of the treated or product water stream in terms relative to the water stream to be treated. The attribute or parameter can be a singular or a composite or aggregate characteristic of the water. Specific, non-limiting examples of such attributes can include the conductivity or resistivity of the water, the presence, absence, or concentration of one more particular species or kinds of species in the water, as well as combinations thereof.

In accordance with one or more embodiments of the invention, the systems and techniques of the invention provide water having a desired water attribute can be represented or quantified as a composite character. The composite character can provide an indication of suitability of the treated water for a particular purpose. Consequently, the systems and techniques of the invention can involve operations that seek or at least promote providing water having one or more desired composite characteristics. In irrigation applications, the treated water attribute can be related to its suitability as irrigation water. Thus, some aspects of the invention can be directed

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to treating non-potable water and rendering the water, as treated water, suitable for irrigation in one or more agricultural facilities by adjusting one or more characteristics thereof. Some aspects of the invention can provide irrigation water tailored to one or more crops grown or cultivated in one or more agricultural facilities. For example, with reference again to FIG. 2, the systems and techniques of the invention can provide a first treated water, having a first composite characteristic, to a first type of crop 228 and a second treated water, having a second composite characteristic, to a second type of crop 238. The second treated water can be used to supplement and/or adjust the characteristic of the first treated water and, conversely, the first treated water can be used to adjust one or more characteristics of the second treated water. The one or more characteristics can be adjusted to meet a particular requirement by, for example, mixing together or blending the one or more treated water streams or portions thereof. The particular target characteristic can be achieved by regulating the ratios or relative amounts or rates of the treated water streams to be mixed. The treated water streams may be mixed in any ratio desired, for example, by mixing from 0% (no mixing) to 100%, from about 10% to about 90%, from about 25% to about 75%, or from about 40% to about 60% of a the first treated water with the second treated water or by mixing from 0% (no mixing) to 100%, from about 10% to about 90%, from about 25% to about 75%, or from about 40% to about 60% of the second treated water with the first treated water.

During typical operation, each of the one or more separation apparatus 220 and 230 typically generates one or more secondary streams. Typically, the one or more secondary streams contain an unacceptable level of one or more undesirable species. Any one or more secondary streams can be discharged as waste streams. For example, the waste stream typically containing the one or more species transferred from the stream treated in separation apparatus 230 can be discharged or transferred to the source of water to be treated 202 through conduit or distribution system 236. The waste stream typically containing the one or more species transferred from the stream treated in separation apparatus 220 can be discharged or transferred to the source of water to be treated 202 through conduit or distribution system 226. Likewise, other embodiments of the invention contemplate combining one or more secondary streams,

typically from one or more downstream separation apparatus, with a water stream to be treated in one or more upstream separation apparatus. For example, one or more streams output from a downstream electrically-driven or pressure-drive separation apparatus may be returned to an upstream electrically-driven separation apparatus as concentrate makeup or diluate makeup. The waste stream can also be discharged with other streams that may or may not be directly associated with the treatment system. For example, the stream to be discharged may be returned to the source of water to be treated after being mixed with one or more blow down streams from, for example, a cooling tower, which may not be a unit operation of the treatment system. In other cases, however, the one or more waste streams may be stored and combined with water having very low salinity to mitigate water infiltration problems that could result in leaching soluble minerals, and salts such as calcium from surface soils.

In some embodiments of the invention, the secondary stream from second separation apparatus 230 contained in conduit 236 can be introduced into first separation apparatus 220, alone or combined, as shown in FIG. 2, with water to be treated from source 202 as delivered through conduit 222.

The schematically illustrated systems depicted in FIGS. 1 and 2 may further comprise unit operations that facilitate the treatment of water. For example, an optional system may be utilized upstream of separation apparatus 220 and 230 to filter or otherwise remove at least a portion of suspended solids in the water from source 202. Non-limiting examples of pre-treatment unit operations that may be utilized to reduce the concentration of at least one suspended solid entrained in the water to be treated include microfilters, settlers, screen filters, and coarse particle filters.

Further, one or more unit operations may be utilized to further process one or more of the treated water streams. For example, a polishing bed may further remove one or more species from one or more of the treated streams in distribution systems 224 and 234. Non-limiting examples of such unit operations that can be utilized to remove at least a portion of weakly ionized or ionizable species, such as, but not limited to, boron, selenium, and arsenic, include ion exchange columns.

Further unit operations that facilitate post-treatment of one or more treated water streams of the invention include those that add or otherwise adjust a

concentration of one or more desirable species or characteristics of the water stream. Post-treatment operations may be employed to render the one or more waste streams suitable for discharge to the environment.

Accordingly, a mixer may be disposed downstream of one or more separation
5 apparatus of the invention that facilitates incorporation of another treated or untreated water stream, disinfectants, nutrients, and/or desirable salts from one or more sources of such. In accordance with some embodiments of the invention, one or more sources of a salt can be disposed to be introduced into the treated water stream. For example, a separation apparatus may be utilized in the treatment or irrigation system of the
10 invention that selectively removes or reduces the concentration of divalent or other non-monovalent species from a water stream to be treated. Such an optional apparatus would typically provide at least one product stream having a relatively high concentration of non-monovalent species which can be introduced to the treated stream to adjust at least one characteristic thereof so as to provide a stream or body of
15 water with a target or desirable condition. Examples of systems and techniques that advantageously provide beneficial species-rich streams include those disclosed in co-owned U.S. Patent No. 7,820,024, titled "Electrically-Driven Separation Apparatus," the substance of which is incorporated herein by reference. In some cases, however, one or more otherwise unconnected or distinct sources of, for example, calcium
20 and/or magnesium salts, may be utilized to adjust one or more characteristics of the treated water stream prior to its use. Additionally, one or more intrinsic and/or extrinsic properties of the water stream may be further adjusted. For example, the water stream may be cooled or heated to adjust the temperature thereof. The pH of the water stream or body may also be adjusted by, for example, adding one or more
25 acids or bases, to achieve a desired pH value. The desired property or characteristic may be dependent on a plurality of factors including, for example, the pH of the soil to be irrigated, the salt tolerance the crops to be irrigated and, in some cases, the moisture content of the soil. Thus, some features of the invention provide further capabilities directed to achieving one or more desired composite characteristics.

The further adjustment of the one or more properties or characteristics may be performed after treatment in the separation apparatus, prior to use or introduction to the point of use, or during storage of the treated water in one or more reservoirs.

However, some aspects of the invention contemplate beneficial or economically attractive attributes of such secondary streams containing high concentrations of one or more dissolved species, relative to the first or treated product stream and/or the stream introduced into the separation apparatus. For example, the secondary product stream may contain high dissolved solids and can serve as a feed stream that may be further processed to obtain additional products or at least provide a product stream having a high concentration of a desirable species.

One or more characteristics of the water utilized in some systems and techniques of the invention can provide an indication of the suitability of the water for agricultural use. For example, the one or more characteristics of the water can be represented as the salinity, as total dissolved salts or solids content, and/or electrical conductivity, as well as or in conjunction with any of the alkalinity, iron content, and pH of the water. In some cases, the level of salinity of the water can become a selective parameter when considered relative to the type of crops to be irrigated by the at least partially treated water. Thus, in accordance with some aspects of the invention, the salinity of the water may be used as a factor to consider when making a determination whether to adjust at least one operating parameter of the systems of the invention. In other embodiments of the system and techniques of the invention, the characteristic value can be represented as a ratio of the concentration of species that tends to render soil as water-impermeable relative to the concentration of species that tends to render soil as aggregating or water-adsorbing.

In accordance with some aspects of the invention, the characteristic value can provide an indication of the suitability of the water for irrigation purposes, for human consumption, and/or for livestock or poultry use. In some embodiments, the characteristic value of a water stream or body can be represented as a ratio of the concentration of monovalent species relative to the concentration of divalent species in the water. For example, the characteristic value can be at least partially expressed as the sodium adsorption ratio (SAR) or exchangeable sodium percentage. Preferably,

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the SAR value of a stream or body of water can provide an indication as to whether the water may be suitable to irrigate a type or kind of crop. Thus, in accordance with some aspects of the invention, some embodiments thereof relate to systems and techniques that can involve controlling one or more operating parameters based at least partially on a desired characteristic value that is at least partially derived from at least one requirement of a point of use. Where the point of use is, for example, a crop to be irrigated, the desired characteristic value can be based on the salt tolerance of the crop and/or one or more attributes or characteristics of the soil.

The sodium adsorption ratio value is typically determined according to the following formula:

$$SAR = \frac{[Na]}{\sqrt{\frac{1}{2}([Ca] + [Mg])}}$$

where [Na] is the sodium species concentration, in milliequivalents/liter in the water, [Ca] is the calcium species concentration, in milliequivalents/liter in the water, and [Mg] is the magnesium species concentration, in milliequivalents/liter in the water.

Other characteristic values of the water may be utilized, alone or in conjunction with the SAR value. Thus, in some cases, the characteristic value of the water that can serve as indication of water quality or suitability for its intended purpose involves the total dissolved solids concentration in the water, the pH, and/or the concentration of one or more toxic or hazardous species.

Adjusting the SAR value of the, for example, irrigation water, may be effected by adjusting one or more operating parameters of the water system. For example, the relative ratio of treated water from different treatment units having various associated SAR values may be adjusted to provide a composite or blended mixture of product water having the desired SAR value. Other techniques including reducing the flow rate of the water stream through the one or more separation apparatus or increasing the residence or treatment period can facilitate achieving the desired SAR value. In addition or in conjunction with such techniques, adjusting the applied potential or pressure level through, for example, the electrically-driven or pressure-driven

separation apparatus can also facilitate providing treated water having the one or more desired characteristics.

Embodiments of systems of the invention may desalinate brackish water to provide irrigation water that avoids or reduces the extent of any soil permeability and/or infiltration problems.

The one or more characteristic values of the treated water may be a relative correlation between species contained in the water. For example, the characteristic value may be a ratio of dissolved sodium species to dissolved calcium. A preferred desirable sodium to calcium ratio of not more than about 3:1 may avoid or reduce the likelihood of water infiltration problems due to soil dispersion and plugging and soil surface pore sealing. Further, some embodiments of the invention can selectively reduce the concentration of monovalent sodium in irrigation water, so that a source of relatively calcium-rich water can be provided to counteract any sodium-dispersing phenomena in irrigation.

The product water can have an SAR value in a range from about 1 to about 8. The target or desirable SAR value may, however, depend on one or more factors in the agricultural facility. For example, the target SAR value depend on the type of crops grown in the facility, the stage of growth of one or more crops in the facility, and the soil conditions including the water infiltration rate, sodicity, and/or alkalinity of the soil. Particular guidelines that may be used to provide one or more target characteristics of irrigation water include those provided by The Food and Agriculture Organization of the United Nations (FAO) in the publication by Ayers, R.S. and Westcot, D.W., titled "Water Quality for Agriculture," FAO Irrigation and Drainage Paper 29 rev. 1, Food and Agriculture Organization of the United Nations, 1989, 1994. For example, the exchangeable sodium percentage ($ESP = \text{Exchangeable } \{(\text{Na})/(\text{Ca} + \text{Mg} + \text{K} + \text{Na})\} \times 100$), which can be correlated to the SAR value, can serve as a desirable characteristic value of water utilized for irrigation purposes, and a measure of the likelihood that a soil, particularly for ESP values greater than 15, will be rendered sodic with that likelihood increasing as the ESP increases. In particular, sensitive crops such as, but not limited to fruits, nuts, and citrus typically require irrigation water having an SAR value of up to about 8; other sensitive crops such as

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beans may tolerate irrigation water having an SAR value of up to about 18; moderately tolerant crops such as clover, oats, and rice may tolerate irrigation water having an SAR value of up to about 18 to 46; and tolerant crops such as, but not limited to wheat, barley, tomato, beets, and tall wheat grass, may tolerate irrigation water having an SAR value of up to about 46 to 102. Embodiments of systems disclosed herein may include treatment units or combinations of same configured to produce irrigation water having SAR values in any of the ranges disclosed above as desired for a particular use. SAR and ESP values even lower than the ones mentioned may further be beneficial, since although a given species may tolerate the SAR values above, crop yields may be deleteriously affected nonetheless. Further, elevated SAR values may affect soil aggregation and cause irrigation water to either run off the soil surface and not penetrate to the root systems of the crops, or, run through the soil levels rapidly and cause the irrigation water to run below the root systems prior to being absorbed by the crops.

Infiltration issues typically arise when irrigation water does not enter the soil and becomes unavailable to crops. In contrast to salinity issues, which reduce the availability of water, infiltration problems can effectively reduce the quantity of water available for crop use. Water infiltration can increase with increasing salinity and can decrease with decreasing salinity or increasing sodium content relative to calcium and magnesium. Further, low salinity water having, for example, a conductivity (which can be measured either in the water or from a paste extract of the soil) of less than about 0.5 dS/m ("EC"), is typically corrosive and tends to leach surface soil of soluble minerals and salts, such as calcium, which in turn can reduce soil aggregation and structure. Soil without or having low salt content tends to be dispersive as fine soil particles which fill pore spaces, effectively sealing the soil surface and reducing the rate of water infiltration. The soil would tend to form a crust which reduces the amount of water entering the subsurface and can also prevent crop emergence. Thus, in some embodiments of the invention, the desired water quality may be further based on the salinity of the irrigation water. For example, FIG. 6, which is based on the publication by Ayers, R.S. and Westcot, D.W., titled "Water Quality for Agriculture," FAO Irrigation and Drainage Paper 29 rev. 1, Food and Agriculture Organization of

the United Nations, 1989, 1994, and which shows the influence of salinity, as represented by TDS concentration, and SAR on infiltration, can conjunctively provide desirable salinity levels and SAR values of irrigation water that reduces or avoids infiltration problems. In FIG. 6, seawater properties were used to derive TDS concentration values from electrical conductivity data from the above reference. In particular, the correlations between the density and salinity and between the salinity and electrical conductivity of seawater at 20 °C were determined based on published physical properties. These correlations were then used to convert the electrical conductivity values of seawater from the above-identified reference into the corresponding TDS concentration, which were then mapped relative to the corresponding SAR values to obtain the infiltration guidelines presented in FIG. 6.

Further embodiments of the invention may also provide suitable irrigation water when it has a composite characteristic value such as having an SAR value of less than about 15 while having a TDS level of about 1,500 ppm or more.

Some embodiments of the invention can provide desalination systems and techniques that selectively remove undesirable species which contrasts to non-selective desalination techniques such as those based on thermal and pressure-driven processes. Further, some systems and techniques of the invention can provide a product water stream without requiring the further addition of preferred species. For example, embodiments of the invention can provide irrigation water that does not involve further adjusting characteristic values by the addition of supplemental species.

Further features and aspects of the invention are illustrated in FIG. 3. The treatment system 300 exemplarily illustrated can comprise a first separation apparatus 304 and a second separation apparatus 306. Separation apparatus 304 and 306 typically treat a fluid from one or more sources 302. The water to be treated from source 302 typically contains a high or unacceptable level of dissolved species. The one or more separation apparatus can thus be utilized to at least partially remove or reduce the concentration of one or more undesirable species from the water. The water to be treated from source 302 may be supplied in parallel to both the first separation apparatus 304, and the second separation apparatus 306 through, for example, distribution system or conduit 320. As exemplarily illustrated, treated water

from separation apparatus 304 can be combined with treated water from separation apparatus 306 delivered through, for example, distribution system or conduit 318, in one or more mixing operations or mixer 308 to provide a treated water stream having desired properties and/or characteristics to point of use 314. In accordance with some
5 embodiments of the invention, the treated water may be rendered suitable to be used as potable and/or bathing water in one or more points of use 314.

First separation apparatus 304 may be an electrically-driven separation apparatus or a pressure-driven separation apparatus. Likewise, second separation apparatus 306 may be a pressure-driven separation apparatus. In accordance with
10 some aspects of the invention, separation apparatus 304 removes at least a portion of a plurality of undesirable species in water to be treated from source 302.

The second separation apparatus can remove one or more undesirable species from the water stream to be treated. In some cases, the separation apparatus selectively removes at least a portion of one or more undesirable species from the
15 water to produce a product water stream. If the product water stream from the second separation apparatus fails to meet or exceed potable water quality requirements, a portion of the treated water from the first separation apparatus that exceeds the potable water quality requirements may be incorporated or blended therewith. For example, where the first separation apparatus provides product water having a TDS
20 level of about 250 mg/L and the second separation apparatus provides product water having a TDS level of about 1,000 mg/L, the product water streams can be combined in a volumetric ratio of about 2:1 to produce a blended product having a TDS level of about 500 mg/L. The target level may be a concentration that meets or exceeds one or more guidelines suggested by the World Health Organization for potable water in, for
25 example, the WHO Guidelines for drinking-water quality, fourth edition, World Health Organization 2011. Other water streams may also be blended with one or more products streams of the separation apparatus of the invention to provide drinking and/or bathing water that meet or exceed guidelines or requirements typically set by government regulatory organizations.

30 One or more reject streams from the first separation apparatus, typically containing relative high levels of species removed from the first treated product

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stream may be discharged to drain, directed to one or more ancillary points of use 310, or returned to source 302. Further embodiments of the invention contemplate combining the reject water stream with water from source 302 through conduit 322 so as to be treated in the second separation apparatus. A secondary or reject water stream from second separation apparatus may also be discharged to a drain, directed to one or more ancillary points of use 310 and/or 312, returned to source 302 as shown through conduit 316.

In another embodiment, indicated generally at 400 in FIG. 4, a treatment system may include a source of water to be treated 402 which may include, for example, brackish water in fluid communication through a distribution system or conduit 420 to an optional pre-filtration system 404. The water to be treated may be brackish water having a TDS of between about 1,500 mg/L (1,500 ppm) and about 10,000 mg/L (10,000 ppm), and/or saline water containing high concentrations of dissolved solids or salts, for example, a concentration of dissolved salts of between about 500 mg/L and about 5,000 or about 10,000 mg/L and one or more other contaminants, for example, about 40 ppm of nitrate, and undesirable levels of any one or more of boron, silica, bicarbonate, or divalent selenate or selenite. The pre-filtration system 404 may be in fluid communication through a distribution system or conduit 422 to a first separation apparatus 406. Non-limiting examples of pre-treatment unit operations that may be utilized in pre-filtration system 404 which may reduce the concentration of at least one suspended solid entrained in the water to be treated include softening systems, microfilters, settlers, screen filters, microsand filters, and coarse particle filters. In some embodiments, the pre-treatment subsystem may include a VACLEEN® stainless steel self-cleaning filtration system available from Evoqua Water Technologies LLC or its affiliates. In other embodiments, the pre-treatment subsystem may alternatively or additionally include a Vortisand® cross flow microsand submicron filtration system available from Evoqua Water Technologies LLC or its affiliates.

The first separation apparatus 406 may be an electrically-driven separation apparatus, for example, an electro dialysis apparatus. Distribution system or conduit 422 may split to provide water to be treated to diluting compartments 406A and

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concentrating compartments 406B of the first separation apparatus 406. The first separation apparatus 406 may include one or more monovalent ion selective membranes to selectively promote transfer of one or more monovalent cationic and/or anionic species. In some embodiments, the first separation apparatus 406 may
5 comprise one or more monovalent cation selective membranes and one or more monovalent anion selective membranes.

Because the nature of electro dialysis is that power costs are inversely related to the level of salinity in water to be treated, high concentrations of ionic contaminants in the water to be treated may reduce power losses in the electro dialysis
10 equipment and also recover more of the incoming water as usable product (albeit at some expense of thermodynamic efficiency.) To promote tangential flow in the concentrating compartments 406B, while increasing the concentration of salts in compartments 406B, the system 406 may optionally include a pump and valve (fixed or controllable based on a set point) comprising a recirculation system (not shown)
15 that recirculates all or a portion of side stream 444 back to the inlet feed stream to 406B and combines with the feed entering from 422. As a result the concentrate compartments may be operated at a controllable very low water usage from line 422 while maintaining optimum hydrodynamic conditions within compartments 406B. The lower water feed from line 422 can thus result, in a controllable way, in a very
20 large fraction (water recovery) of the feed water being sent through the diluting compartments as product water for irrigation or potable use, and less to effluent stream 430. As described in further embodiments, the inlet feed to compartments 406B may also be sourced all or part not from line 422, but instead from line 442. As a result, since the outlet from line 442 has very low contaminant levels as compared to
25 the inlet 422, the potential for operation at high water recoveries is even further increased.

The first separation apparatus 406 may remove a large fraction of contaminants from the water to be treated to produce a first diluate stream. The contaminants removed may include one or more ionic species. The first separation
30 apparatus 406 may remove, for example, about 80% of sodium chloride and potassium chloride and about 8% of the nitrate from the water to be treated supplied

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through distribution system or conduit 422. The first separation apparatus 406 may produce a diluate which may be high in divalent ions as compared to monovalent ions and may be acceptable for agricultural use without further treatment. To be suitable for agricultural use, the diluate from the first separation apparatus 406 may have a SAR value of, for example, less than about 20, less than about 18, less than about 9, less than about 8, less than about 3, or less than about 2, and may have a TDS value of less than about 3,500 ppm or less than 750 ppm.

A first portion of the diluate from the first separation apparatus 406 may be directed through a distribution system or conduit 424 and a distribution system or conduit 432 to a point of use 414 for agricultural use, for example, an irrigation water system or a holding system for irrigation water. A second portion of the diluate from the first separation apparatus 406 may be directed through the distribution system or conduit 424 to a second separation apparatus 408. The relative amounts of the diluate from the first separation apparatus 406 directed to point of use 414 and to the second separation apparatus 408 may be selected in any amount in a range of from about 0% to about 100%, from about 10% to about 90%, from about 25% to about 75%, or from about 40% to about 60% as desired. A diverting valve or other form of diverting system 410 may be utilized to vary the amount of diluate from the first separation apparatus 406 directed to point of use 414 and to the second separation apparatus 408 as desired. The diverting valve or system 410 may be operated by a controller 106. The controller 106 may receive an indication of one or more parameters of water from one or more of distribution systems or conduits 432 or 436 or from the point of use 414 from one or more sensors 108. The one or more parameters of water may include, for example, salinity, conductivity, pH, SAR value, TDS value, or any other parameter or parameters of interest. The controller 106 may adjust the diverting valve or system 410 and thus the ratio of diluate from the first separation apparatus 406 directed to point of use 414 and to the second separation apparatus 408 based on the indication of the one or more parameters of water from the sensor(s) 108. Controller 106 may also adjust one or more operating parameters of the first separation apparatus 406 based on the indication of the one or more parameters of water from the sensor(s) 108. The one or more operating parameters of the first separation apparatus 406 may

include, for example, current, flow rate, water recovery, or any other operating parameter or parameters.

Concentrate from the first separation apparatus 406 may be carried through a distribution system or conduit 430 as a waste stream which may be discharged to drain, directed to one or more ancillary points of use 412, for example, for use in the electrochemical production of chlorine, alkali (such as sodium hydroxide), chlorine gas, hypochlorite ion, hydrochloric acid, or sulfuric acid, or returned to the source 402, through, for example, distribution system or conduit 444. The first separation apparatus 406 may operate at very high water recoveries, for example, between about 20% and about 60% or 90% or more because monovalent ions may be present in the concentrate delivered through distribution system or conduit 430 at high concentrations without precipitation.

The partially desalinated water output as the first diluate stream from the first separation apparatus 406 may remain unacceptably high in monovalent salts and/or high in divalent salts, for example, calcium sulfates and/or magnesium sulfates for some uses. At least a portion of the first diluate stream may thus be further treated in a second separation apparatus 408. The second separation apparatus 408 may be a pressure-driven separation apparatus, for example, a membrane filtration system. The second separation apparatus 408 may include nanofiltration membranes or reverse osmosis membranes. The second separation apparatus 408 may be a low-pressure nanofiltration system, for example, a nanofiltration system operated with a transmembrane pressure of between about 1 and about 15 bar, between about 2 and about 4 bar, or less than about 3 bar. The second separation apparatus 408 may treat water provided through the distribution system or conduit 424 to produce a permeate and a retentate. The second separation apparatus 408 may remove a large fraction, for example, between about 10% and about 95% of remaining monovalent ions from the influent diluate from the first separation apparatus 406 and substantially all the divalent ionic constituents, and produce a permeate stream with a very low concentration, for example, less than about 250 mg/L or less than 150 ppm or less than 50 ppm of monovalent ionic contaminants. The second separation apparatus 408 may also remove additional undesirable components from the portion of the diluate

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from the first treatment apparatus, for example, an additional 10% to 95% of nitrates from the portion of the diluate from the first separation apparatus 406. Retentate from the second separation apparatus 408 may be directed through distribution system or conduit 436 to point of use 414 where it may be used alone or combined with the first
5 portion of the diluate from the first separation apparatus 406 for agricultural use.

Point of use 414 may include a mixer (not shown) configured to blend diluate produced in the electro dialysis apparatus 406 with any retentate produced in the nanofiltration apparatus 408 and produce irrigation water when mixed having a SAR value intermediate of a SAR value of the diluate and a SAR value of the retentate.

10 Permeate from the second separation apparatus 408 may be low in divalent ionic contaminants and/or particulates and/or microbial contaminants. Permeate from the second separation apparatus 408 may be suitable for potable use, and may have characteristics defined in recognizable standards, for example, those provided by the WHO. The filtrate from the second separation apparatus 408 may include, for
15 example, a TDS level of less than about 500 ppm or less than about 250 ppm, low amounts of divalent ions, low amount of metals and trace elements, low levels of nitrates, for example, less than about 10 ppm of nitrates, low levels of particulates and/or colloidal or microbiological impurities, and/or a low organic content. The characteristics of the filtrate from the second separation apparatus 408 may meet
20 regulatory requirements for potable water in a jurisdiction in which the system 400 is operated. Filtrate from the second separation apparatus 408 may be directed through a distribution system or conduit 426 to a point of use 418 for use as potable water.

In some embodiments a portion of the permeate from the second separation apparatus 408 may be directed through a distribution system or conduit 442 to be used
25 as concentrate makeup for the first separation apparatus 406. The amount or fraction of permeate from the second separation apparatus 408 utilized as concentrate makeup for the first separation apparatus 406 may be selected as desired. The amount or fraction of permeate from the second separation apparatus 408 used as concentrate makeup for the first separation apparatus 406 may be selected based on, for example,
30 the quality of the filtrate from the second separation apparatus 408 and the need for the concentrate makeup. A valve or other diverting system 440 may provide for the

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amount of filtrate from the second separation apparatus 408 utilized as concentrate makeup for the first separation apparatus 406 to be adjusted as desired. Valve or diverting system 440 may be operated by the controller 106 responsive to information from any of sensors 108 or other sensors located in various portions of the system.

5 The amount of the permeate from the second separation apparatus 408 used as concentrate makeup for the first separation apparatus 406 may be selected based on, for example, the quality of the permeate from the second separation apparatus 408 and/or the need for the concentrate makeup.

The first separation apparatus 406 may include one or more monovalent ion selective membranes to selectively promote transfer of one or more monovalent cationic or anionic species. The first separation apparatus 406 may include, for example, an electro dialysis system. In other embodiments, the water treated in the first separation apparatus 406 may be sufficiently low in monovalent ions that monovalent ion selective membranes would not be required in the first separation apparatus 406. The first separation apparatus 406 may thus utilize standard ion exchange membranes known in the art. These standard ion exchange membranes may be not monovalent ion selective.

The system illustrated in FIG. 4 has advantages over many alternate approaches to producing irrigation water and/or potable water from brackish water. As compared to conventional electro dialysis systems including conventional or monovalent selective membranes, the system illustrated in FIG. 4 may operate with a higher recovery, may require less salt removal in the electrically-driven separation apparatus than conventional electro dialysis systems, and may exhibit less loss of calcium than conventional electro dialysis systems. The electrically-driven separation apparatus of the system illustrated in FIG. 4 may thus be smaller and may consume less power than conventional electro dialysis systems for producing irrigation water and/or potable water from brackish water. The system illustrated in FIG. 4 provides for a controllable ratio of irrigation water vs. potable water produced from influent water to be treated. The system produces waste that is low in species such as calcium, magnesium, boron, selenium, and silica. The majority of calcium, magnesium, and a fraction of nitrate in water to be treated introduced into the system is included in the

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irrigation water produced by the system. The system may utilize a nanofiltration filter as the pressure-driven separation apparatus rather than a more expensive and high operating cost reverse osmosis system to produce potable water suitable for use as drinking water. The system illustrated in FIG. 4 can produce irrigation water with a lower SAR value than conventional electro dialysis systems alone.

In another embodiment, indicated generally at 500 in FIG. 5, a treatment system may include a source of water to be treated 502. The source of water to be treated 502 may be, for example, a groundwater well. The water to be treated in the source of water to be treated 502 may be brackish water or saline water having a concentration of dissolved salts of between about 500 mg/L and about 5,000 mg/L or about 10,000 mg/L as well undesired species such as nitrates (in concentrations of up to about 40 ppm), bicarbonates (in concentrations up to about 400 ppm), sulfates (in concentrations up to about 2,000 ppm), iron, boron, manganese, selenium, calcium, magnesium, arsenic, silica, and/or sulfur or fluoride compounds. The source of water to be treated 502 may be fluidly connected to a break tank 506 via a conduit 504.

A booster pump 510 may be positioned and arranged to direct water to be treated from the break tank 506 through conduits 508 and 512 to an air stripper 514. Air stripper 514 may include a column and inert media, for example, gravel, greensand, or beads or pieces of porous silica through which the water to be treated downwardly flows while air flows upward in a countercurrent direction.

Alternatively, the greensand may be included in a separate unit process for removal of iron and manganese, independent of the air stripper 514. The air stripper 514 may remove bicarbonate, manganese, arsenic, iron, and/or sulfide from the water to be treated. To facilitate removal of the bicarbonate, an acid, for example, hydrochloric acid or sulfuric acid from a source of pH adjustment agent 516 may be injected into conduit 512 (or optionally into break tank 506 or conduit 508) to reduce the pH of the water to be treated entering the air stripper 514 to an acidic pH as low as about 4.2 to facilitate conversion of bicarbonate to carbon dioxide gas that would be removed from the water to be treated in the air stripper 514. The amount of pH adjustment agent added to the water to be treated may be controlled based on pH and/or flow reading from one or more sensors 108A, 108B upstream or downstream of the pH adjustment

agent injection point. The iron, manganese, arsenic, and/or sulfide may be adsorbed onto the media within the air stripper 514 or may form oxidized solid precipitates that may be removed by filtration. A trickling stream of water passing through the air stripper 514 may flush removed contaminants into a drain or sump 518. In

5 implementations where it is desirable to concentrate chlorides in the electro dialysis concentrate, it may be preferable to utilize hydrochloric acid to downwardly adjust the pH. In other implementations, particularly with waters having high monovalent anion levels relative to monovalent cation levels, it may be preferable to utilize sulfuric acid to downwardly adjust the pH.

10 After passing through the air stripper 514, the water to be treated passes through conduit 520 to a pre-filter 522, which may include a screen filter with apertures with sizes of, for example, about 10 μm . Pre-filter 522 may also or alternatively include any one or more of a, microfilter, settler, sand filter, or screen filter, microsand filter, or coarse particle filter.

15 The pre-filtered water to be treated may be then directed through conduit 524 and may be split to be introduced into diluting compartments 526A and concentrating compartments 526B of electro dialysis apparatus 526. The pre-filtered water to be treated introduced into the concentrating compartments 526B of electro dialysis apparatus 526 may be supplement by recycled concentrate makeup water from
20 concentrate loop 528. Alternately, the influent water may be exclusively directed into the diluting compartments, and make up water to the electrode and/or concentrate compartments may be sourced from the nanofilter permeate through line 554.

The electro dialysis apparatus 526 may remove a large fraction of contaminants from the water to be treated to produce a first diluate stream. The contaminants
25 removed may include one or more ionic species. The electro dialysis apparatus 526 may remove, for example, about 80% of sodium chloride and potassium chloride and about 80% of the nitrate from the water to be treated supplied through distribution system or conduit 524. The electro dialysis apparatus 526 may produce a diluate which may be high in divalent ions as compared to monovalent ions and may be
30 acceptable for agricultural use without further treatment. To be suitable for agricultural use, the diluate from the electro dialysis apparatus 526 may have a SAR

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value of, for example, less than about 20, less than about 18, less than about 9, less than about 8, less than about 3, or less than about 2, and may have a TDS value of less than about 3,500 ppm and/or greater than about 1,500 ppm. A ratio of divalent selenate or selenite ion concentration to chloride concentration in diluate from the electro dialysis apparatus 526 may be less than that of the water to be treated (if using non selective anion membranes) and a ratio of calcium ion concentration to sodium ion concentration in the diluate from the electro dialysis apparatus 526 may be greater than that of the water to be treated (if using monovalent selective cation membranes).

A first portion of the diluate from the electro dialysis apparatus 526 may be directed through a distribution system or conduit 530 and a distribution system or conduit 532 to a point of use 534 for agricultural use, for example, an irrigation water system or a holding system for irrigation water. The first portion of the diluate from the electro dialysis apparatus 526 directed to the point of use 534 for agricultural use may have a lower concentration ratio of monovalent ions such as sodium and chlorine and a higher concentration of divalent ions such as calcium and magnesium than the original water to be treated. The first portion of the diluate from the electro dialysis apparatus 526 directed to the point of use 534 for agricultural use may have a substantially similar concentration of boron, selenium (or divalent selenate or selenite), and silica as the original water to be treated. In some embodiments the portion of the diluate from the electro dialysis apparatus 526 directed to the point of use 534 for agricultural use may be further treated to reduce levels of boron, for example, using an ion exchange system 555. Unlike RO, since silica levels are not increased in the concentrate, silica does not limit water recovery through the system.

A second portion of the diluate from the electro dialysis apparatus 526 may be directed through the distribution system or conduit 536 to a nanofiltration apparatus 538, optionally after further treatment to further reduce boron concentration. The relative amounts of the diluate from the electro dialysis apparatus 526 directed to point of use 534 and to the nanofiltration apparatus 538 may be selected in any amount in a range of from about 0% to about 100%, from about 10% to about 90%, from about 25% to about 75%, or from about 40% to about 60% as desired. A diverting valve or other form of diverting system 540 may be utilized to vary the amount of diluate from

the electro dialysis apparatus 526 directed to point of use 534 and to the nanofiltration apparatus 538 as desired. The diverting valve or system 540 may be operated by a controller 106. The controller 106 may receive an indication of one or more parameters of water from one or more of distribution systems or conduits 532 or 542 or from the point of use 534 from one or more sensors 108C, 108D, 108E. The one or more parameters of water may include, for example, salinity, conductivity, pH, SAR value, TDS value, or any other parameter or parameters of interest. The controller 106 may adjust the diverting valve or system 540 and thus the ratio of diluate from the electro dialysis apparatus 526 directed to point of use 534 and to the nanofiltration apparatus 538 based on the indication of the one or more parameters of water from the sensor(s) 108C, 108D, 108E. Controller 106 may also adjust one or more operating parameters of the electro dialysis apparatus 526 based on the indication of the one or more parameters of water from the sensor(s) 108C, 108D, 108E. The one or more operating parameters of the electro dialysis apparatus 526 may include, for example, current, flow rate, or any other operating parameter or parameters.

Concentrate from the electro dialysis apparatus 526 may be carried through a distribution system or conduit 544 as a waste stream which may be discharged to drain, directed to one or more ancillary points of use 546, for example, for use in the electrochemical production of chlorine, alkali, chlorine gas, hypochlorite ions, hydrochloric acid, or sulfuric acid. Acid produced in the ancillary points of use 546, which may be or include an electrochemical generation system, may be used to supply source of pH adjustment agent 516 through, for example, conduit 517. In other embodiments the concentrate from the electro dialysis apparatus 526 may meet water quality standards for use in produced oil recovery operations and ancillary point of use 546 may include an oil drilling operation. The concentrate from the electro dialysis apparatus 526 may be softened to reduce a concentration of divalent cations prior to use in produced oil recovery operations in the ancillary point of use 546, for example, by treatment in ion exchange system 551. System 551 may alternatively or additionally include a pH adjustment system to adjust a pH of the concentrate from the electro dialysis apparatus 526 to a pH suitable for use in produced oil recovery operations. The amount of concentrate directed to the one or

more ancillary points of use 546 may be between about 5% and about 10% of the amount of concentrate produced in the electro dialysis apparatus 526. The concentrate directed to the one or more ancillary points of use 546 may have an elevated concentration of sodium chloride as compared to the original water to be treated and a substantially similar concentration of boron, selenium (or divalent selenate or selenite) and silica as the original water to be treated. A ratio of divalent selenate or selenite ion concentration to chloride ion concentration in the concentrate from the electro dialysis apparatus 526 may be less than that of the water to be treated.

The electro dialysis apparatus 526 may operate at very high water recoveries, for example, between about 20% and about 60% or 90% or more because monovalent ions may be present in the concentrate delivered through distribution system or conduit 544 at high concentrations without precipitation.

The partially desalinated water output as the first diluate stream from the electro dialysis apparatus 526 may remain unacceptably high in monovalent salts and/or high in divalent salts, for example, calcium sulfates and/or magnesium sulfates for some uses. At least a portion of the first diluate stream may thus be further treated in nanofiltration apparatus 538. The nanofiltration apparatus 538 may be a low-pressure nanofiltration system, for example, a nanofiltration system operated with a transmembrane pressure of between about 1 and about 10 bar, between about 2 and about 4 bar, less than about 4 bar, or less than about 3 bar. The nanofiltration apparatus 538 may treat water provided through the distribution system or conduit 536 to produce a permeate and a retentate. A ratio of retentate outlet flow to permeate flow through the nanofiltration apparatus 538 may be controlled so as to limit a scaling potential of dissolved constituents of the nanofiltration retentate. The nanofiltration apparatus 538 may remove a large fraction, for example, between about 50% and about 95% or up to about 97% of remaining monovalent ions from the influent diluate from the electro dialysis apparatus 526 and produce a permeate stream with a very low concentration, for example, less than about 500 mg/L (500 ppm) or less than about 200 ppm of monovalent ionic contaminants. In other implementations, the nanofiltration apparatus 538 may remove substantially all divalent ions and about 10% - 25% of monovalent ions to produce a permeate stream

with substantially no divalent ions and only slightly reduced monovalents. The nanofiltration apparatus 538 may produce retentate having a higher concentration of divalent cations such as calcium and magnesium than the concentration of the divalent cations in the diluate from the electro dialysis apparatus 526 and a lower concentration of monovalent cations such as sodium than the concentration of the monovalent cations in the diluate. The nanofiltration apparatus 538 may also remove additional undesirable components from the portion of the diluate from the electro dialysis apparatus 526, for example, an additional 10% to 90% of nitrates from the portion of the diluate from the electro dialysis apparatus 526. Retentate from the nanofiltration apparatus 538 may be directed through distribution system or conduit 542 to point of use 534 where it may be combined with the first portion of the diluate from the electro dialysis apparatus 526 for agricultural use. In some embodiments the retentate from the nanofiltration apparatus 538 and/or the diluate from the electro dialysis apparatus directed to the point of use 534 for agricultural use may be further treated to reduce levels of boron, for example, using an ion exchange system 557 and/or 555. In other cases, the electro dialysis feed or the nanofilter feed may be operated at elevated pH by addition of sodium hydroxide, such that dissolved boron is ionized to borate and then removed either during the electro dialysis step through the anion membrane or during the nanofiltration step. It is to be noted that side advantages of the bicarbonate removal step through acidification and stripping results in a feed water to the electro dialysis and nanofilter apparatus that has low scaling potential at elevated pH and is unbuffered, and thus will require a relatively cost effectively low amount of caustic soda to increase the pH of the water to levels sufficient to remove boron.

The retentate from the nanofiltration apparatus 538 may have a SAR value lower than the SAR value of diluate from the electro dialysis apparatus 526 and lower than the SAR value of the permeate from the nanofiltration apparatus 538. Point of use 534 may include a mixer (not shown) configured to blend diluate produced in the electro dialysis apparatus 526 with any retentate produced in the nanofiltration apparatus 538 and produce irrigation water having a SAR value intermediate of a SAR value of the diluate and a SAR value of the retentate.

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In some embodiments a portion of the retentate from the nanofiltration apparatus 538 may be directed to an electrochemical generation system 556 for production of sulfuric acid via diverting valve or system 558 and conduit 560 under control of controller 106. The electrochemical generation system 556 may include a
5 water softening apparatus upstream of the inlet to electrochemical generation apparatus included therein for treating the retentate prior to electrochemical generation of the sulfuric acid. Sulfuric acid produced in the electrochemical generation system 556 may be used to supply source of pH adjustment agent 516.

Permeate from the nanofiltration apparatus 538 may be low in divalent ionic
10 contaminants and/or particulates and/or microbial contaminants. Permeate from the nanofiltration apparatus 538 may be suitable for potable use, and may have characteristics defined in recognizable standards, for example, those provided by the WHO. The permeate from the nanofiltration apparatus 538 may include, for example, a TDS level of less than about 500 ppm, low amounts of divalent ions, low amount of
15 metals and trace elements, low levels of nitrates, for example, less than about 10 ppm of nitrates, low levels of particulates and/or colloidal or microbiological impurities, and/or a low organic content. The characteristics of the permeate from the nanofiltration apparatus 538 may meet regulatory requirements for potable water in a jurisdiction in which the system 500 is operated. Permeate from the nanofiltration
20 apparatus 538 may be directed through a distribution system or conduit 548 to a point of use 550 for use as potable water. In some embodiments, the permeate from the nanofiltration apparatus 538 may be further treated, for example in post-treatment unit 559, to achieve at least one of upward pH adjustment, lowered pH adjustment, addition of chlorine gas or addition of hypochlorite ion.

25 In some embodiments a portion of the permeate from the nanofiltration apparatus 538 may be directed through a distribution system or conduit 554 to be used as concentrate makeup and/or an electrode stream for the electrodialysis apparatus 526. The portion of the permeate from the nanofiltration apparatus 538 used as concentrate makeup for the electrodialysis apparatus 526 may be directed through
30 conduit 554 into the concentrate recirculation loop 528 of the electrodialysis apparatus 526. The amount or fraction of the permeate from the nanofiltration apparatus 538

utilized as concentrate makeup for the electro dialysis apparatus 526 may be selected as desired. The amount or fraction of permeate from the nanofiltration apparatus 538 used as concentrate makeup for the electro dialysis apparatus 526 may be selected based on, for example, the quality of the permeate from the nanofiltration apparatus 538 and the need for the concentrate makeup. The quality of the permeate from the nanofiltration apparatus 538 may be measured by sensor 108F, which may, in various embodiments, measure any one or more of pH, salinity, flow rate, ionic species, SAR value, or any other desired parameter of the permeate from the nanofiltration apparatus 538. The amount or fraction of permeate from the nanofiltration apparatus 538 used as concentrate makeup for the electro dialysis apparatus 526 may also or alternatively be selected based on, for example, the quality of the diluate from the electro dialysis apparatus 526. The quality of the diluate from the electro dialysis apparatus 526 may be measured by sensor 108C, which may, in various embodiments, measure any one or more of pH, salinity, flow rate, ionic species, SAR value, or any other desired parameter of the diluate from the electro dialysis apparatus 526.

A valve or other diverting system 552 may provide for the amount of permeate from the nanofiltration apparatus 538 utilized as concentrate makeup for the electro dialysis apparatus 526 to be adjusted as desired. Valve or diverting system 552 may be operated by the controller 106 responsive to information from any of sensors 108C, 108D, 108E, 108F or other sensors located in various portions of the system.

The electro dialysis apparatus 526 may include one or more monovalent ion selective membranes to selectively promote transfer of one or more monovalent cationic or anionic species. In other embodiments, the water treated in the electro dialysis apparatus 526 may be sufficiently low in monovalent ions that monovalent ion selective membranes would not be required in the electro dialysis apparatus 526. The electro dialysis apparatus 526 may thus utilize standard ion exchange membranes known in the art. These standard ion exchange membranes may be not monovalent ion selective.

In each of the systems described above, total water recoveries of between about 66% and about 95% or higher may be possible with low processing costs and high energy efficiency, for example, an energy efficiency of about 2.0 kwh/m³ of

purified water or less. In some embodiments none of the water purified to a level suitable for agricultural or potable use would need to be used for wastewater makeup. Rather, makeup water for the concentrate or electrode streams for the electro dialysis systems could be made up from influent feed. The provision of this makeup water
5 from these downstream sources may improve the overall electrical efficiency of the disclosed system as compared to systems not including the provision of makeup water as described.

Potable water produced in the above systems may be low in salinity and substantially or completely free of particulates, colloids, dissolved organic materials,
10 viruses, or microbes. In implementations where selenium levels from the electro dialysis apparatus meet minimum standards for irrigation but are in excess of requirements for potability, this selenium in the form of divalent selenate or selenite will be adequately removed in the nanofiltration step. In implementations where selenium levels are in excess of requirements for irrigation, the electro dialysis
15 apparatus can incorporate non-selective anion membranes and be operated such that the divalent selenate or selenite is adequately reduced in the diluate. In implementation where selenium levels are acceptable in the feed water for irrigation, but concentration of selenium would be harmful to wildlife in the effluent from the system, the electro dialysis system can incorporate monovalent selective anion
20 membranes to prevent divalent selenate or selenite from entering the concentrating compartments. The potable water may have characteristics defined in recognizable standards, for example, those provided by the WHO. The fraction of purified water produced that is suitable for agricultural use may include high levels of divalent ions and minerals, but have very low concentrations of impurities such as sodium chloride
25 or potassium chloride. The fraction of purified water produced that is suitable for agricultural use may not require processing by softening membranes. Since only a small fraction of the total water influent to the treatment systems described above may be processed by water softening membranes to produce potable water, the size and capital and operational costs of the water softening membrane units may be low
30 relative to other systems where a greater quantity of water would need to be processed through water softening membrane units. Also, in some embodiments, only a small

portion of the influent water to be treated would be processed through electro dialysis units including monovalent ion selective membranes, providing further for reduced treatment unit sizes and reduced capital and operational costs.

In some embodiments transport of divalent ions through electro dialysis including monovalent ion selective membranes may be substantially avoided or
5 eliminated. This may result in a reduced buildup of scale in the electro dialysis units and may reduce pH shifts and high voltages caused by calcium or sulfate transport through a typical cation or anion selective ion exchange membrane.

Each of the embodiments of treatment systems described above may be used
10 partially within a total system such that combinations of the various embodiments can be used for customized agricultural and potable water production quantities or qualities, depending on the feed water makeup and demand for the water for agricultural or potable uses. The quantities of water suitable for potable or agricultural consumption may be changed based on demand by running more or less
15 water to be treated through each of the various embodiments, changing the relative water recoveries or adjusting based on salinity changes in the water to be treated.

As noted above, ancillary systems may be utilized in the systems and techniques of the invention in post-treatment operations. For example, one or more
20 disinfecting systems such as those that irradiate (such as by UV treatment), oxidize, or otherwise reduce microbiological activity in the water may be disposed to further treat the water. Further, one or more storage systems may be also used as discussed above.

Because some embodiments of the invention can selectively remove monovalent species, any resultant secondary or concentrate streams would be less susceptible to scaling and fouling. This feature advantageously allows some
25 separation embodiments of the invention to operate at higher water recovery rates, compared to non-selective techniques, because the volumetric rate of any secondary streams can be effectively reduced without or with less concern for undesirable precipitation. Thus, some embodiments of the invention directed to utilizing systems and techniques that selectively separate monovalent species can be operated at higher
30 recovery rates compared to non-selective ED and distillation based separation apparatus, and even much higher recovery rates compared RO and NF based

separation apparatus. Significantly, because RO and NF based separation systems selectively reduce the concentration of non-monovalent species, these processes cannot effectively provide treated water having low SAR values.

A further advantage of the selective separation systems and techniques of the invention pertains to the reduction or removal of non-ionized species that have little or no influence on crop growth. For example, silica is typically not preferentially removed in the ED-based systems of the invention or the nanofilter-based systems of the invention, thereby avoiding any scaling or fouling concerns in secondary streams that typically arise when treating silica-containing water in RO and distillation apparatus. In addition, because secondary streams of some embodiments of the invention typically have reduced scaling tendencies, the recovery rates in the separation systems and techniques of the invention may be greater than the recovery rates of RO and distillation based systems.

A further advantage of the nanofiltration system is that the water retentate from the nanofilter may be useful for irrigation water. Thus, in cases where irrigation needs are substantial, all or a substantial part of the diluate from the electro dialysis system may be sent to the nanofilter and the nanofilter operated at low water recovery. The result of operating at relatively low water recovery of the nanofilter is that the retentate flow is increased and the concentration of the retentate is reduced and thus less prone to scaling or fouling or operation at high osmotic pressure loss. Overall water usage is maintained because the entire effluent from the retentate of the nanofilter may be an excellent source of low SAR water for irrigation.

Controller 106 of the systems of the invention may be implemented using one or more computer systems. The computer system may be, for example, a general-purpose computer such as those based on an Intel PENTIUM®-type processor, a Motorola PowerPC® processor, a Sun UltraSPARC® processor, a Hewlett-Packard PA-RISC® processor, or any other type of processor or combinations thereof. The computer system may be implemented using specially-programmed, special-purpose hardware, for example, an application-specific integrated circuit (ASIC) or controllers intended for water treatment system.

The computer system can include one or more processors typically connected to one or more memory devices, which can comprise, for example, any one or more of a disk drive memory, a flash memory device, a RAM memory device, or other device for storing data. The memory component or subsystem is typically used for storing
5 programs and data during operation of the system 100 and/or the computer system. For example, the memory component may be used for storing historical data relating to the parameters over a period of time, as well as operating data. Software, including programming code that implements embodiments of the invention, can be stored on a computer readable and/or writeable nonvolatile recording medium, and then typically
10 copied into the memory subsystem wherein it can then be executed by one or more processors. Such programming code may be written in any of a plurality of programming languages, for example, Java, Visual Basic, C, C#, or C++, Fortran, Pascal, Eiffel, Basic, or any of a variety of combinations thereof.

Components of the computer system may be coupled by an interconnection
15 mechanism, which may include one or more busses that provide communication between components that are integrated within a same device and/or a network that provide communication or interaction between components that reside on separate discrete devices. The interconnection mechanism typically enables communications, including but not limited to data and instructions to be exchanged between
20 components of the system.

The computer system can also include one or more input devices, for example, a keyboard, mouse, trackball, microphone, touch screen, and one or more output devices, for example, a printing device, display screen, or speaker. In addition, computer system may contain one or more interfaces that can connect the computer
25 system to a communication network, in addition or as an alternative to the network that may be formed by one or more of the components of the system.

According to one or more embodiments of the invention, the one or more input devices may include sensors for measuring parameters. Alternatively, the sensors, the metering valves and/or pumps, or all of these components may be
30 connected to a communication network that is operatively coupled to the computer system. For example, one or more sensors 108 (or 108A-108F) may be configured as

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input devices that are directly connected to controller 106, metering valves, pumps, and/or components of the various embodiments disclosed herein may be configured as output devices that are connected to controller 106. Any one or more of such subcomponents or subsystems may be coupled to another computer system or
5 component so as to communicate with the computer system over a communication network. Such a configuration permits one sensor to be located at a significant distance from another sensor or allow any sensor to be located at a significant distance from any subsystem and/or the controller, while still providing data therebetween.

The controller can include one or more computer storage media such as
10 readable and/or writeable nonvolatile recording medium in which signals can be stored that define a program to be executed by the one or more processors. The medium may, for example, be a disk or flash memory. In typical operation, the processor can cause data, such as code that implements one or more embodiments of the invention, to be read from the storage medium into a memory that allows for faster
15 access to the information by the one or more processors than does medium. The memory is typically a volatile, random access memory such as a dynamic random access memory (DRAM) or static memory (SRAM) or other suitable devices that facilitates information transfer to and from the one or more processors.

Although the control system is described by way of example as one type of
20 computer system upon which various aspects of the invention may be practiced, it should be appreciated that the invention is not limited to being implemented in software, or on the computer system as exemplarily shown. Indeed, rather than implemented on, for example, a general purpose computer system, the controller, or components or subsections thereof, may alternatively be implemented as a dedicated
25 system or as a dedicated programmable logic controller (PLC) or in a distributed control system. Further, it should be appreciated that one or more features or aspects of the invention may be implemented in software, hardware or firmware, or any combination thereof. For example, one or more segments of an algorithm executable by controller 106 can be performed in separate computers, which in turn, can be
30 communication through one or more networks.

Prophetic example – Tranquility Water District, California

Tranquility Water District has a need for both potable and irrigation water with a scarce supply of fresh water capable of meeting its needs for both types of water. There is, however, an ample and sustainable supply of shallow inexpensive to
5 obtain ground water but that is saline and unusable for either irrigation or potable water without removal of impurities. Presently this shallow ground water is a nuisance because as the water table increases through irrigation from other alternative and expensive or unsustainable fresh water sources such as from deep wells, the water table can rise to reach the root zones of crops and reduce crop yields despite the use of
10 the expensive alternative pure water sources. An embodiment of the present invention may solve these problems by inexpensively treating a very high amount of this shallow ground water to sustainably provide both high quality irrigation and potable water while reducing the effects of increased water table, and providing a low quantity of water for disposal that is relatively non-toxic to wildlife and easy to
15 manage.

Referring to the embodiment of FIG. 4 the system design there described will be used to calculate expected concentration of contaminants at an overall 90% water recovery and where the electro dialysis system is controlled to an 70% TDS removal in the electro dialysis diluate, a 90% recovery in the concentrate, contains monovalent
20 selective anion and cation membranes using commercially available selectivities, and the pretreatment to the electro dialysis system consists of screen type prefiltration followed by acidification using hydrochloric acid to pH 4.5, and air stripping to remove incoming bicarbonate ion, and any iron, manganese, or sulfide and pH increase to pH 5.5. The concentrate makeup to the electro dialysis concentrate is taken
25 from the pretreated influent and there is internal recirculation around the concentrate compartments.

A water analysis from Tranquility's shallow saline source is summarized below:

TDS: 3941 ppm; temperature 68 F; feed pH 6.5; Calcium 240 ppm as ion,
30 magnesium 28 ppm as ion, sodium 1080 as ion, potassium 12 ppm as ion, sulfate

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1480 ppm as ion, bicarbonate (post stripper) 25 ppm as ion, chloride (post acidification) 1076 as ion. (SAR of 17.6)

Calculated diluate from the electro dialysis apparatus is as follows:

TDS: 1209 ppm; calcium 209 ppm as ion, magnesium 25 ppm as ion, sodium
5 140 ppm as ion, potassium 2 ppm as ion, sulfate 617 ppm as ion, bicarbonate 7 ppm
as ion, chloride 209 ppm as ion. (SAR of 1.7) – this water is also sent in part to the
nanofiltration apparatus.

Calculated concentrate effluent from the electro dialysis apparatus is as
follows:

10 TDS 27660 ppm; calcium 520 ppm as ion, magnesium 16 ppm as ion, sodium
9443 ppm as ion, potassium 100 ppm as ion, sulfate 8630 ppm as ion, bicarbonate 186
ppm as ion, chloride 8795 as ion. (SAR of 111).

As can be seen, this diluate water is ideal for use as irrigation water with SAR
of 1.7 and low alkalinity and pH to aid in reducing soil sodicity. It is possible that the
15 district would wish to remove only 50% or 60% of the TDS in such a case and operate
and control the system to a somewhat higher but still acceptable SAR value, to utilize
a still smaller electro dialysis system or save still more energy, or capture even more of
the divalents for use in agriculture. The concentrate is low volume, and with low
scaling potential (saturation index of 0.35) with only 10% of the influent water being
20 sent to drain or for further high concentration use with predominantly sodium as the
cation to be potentially used as a feed to prepare hydrochloric/sulfuric acid, caustic
soda, or chlorine. The concentrate will also be relatively low in selenium and boron
as compared to for example that what would be expected from reverse osmosis
retentate, and thus more safely disposable.

25 Using a “loose” nanofilter a part of the electro dialysis diluate is then used to
produce potable water. A loose nanofilter will typically reject 97.5% of divalents and
10% of the monovalents. Calculating from the diluate as feed and such rejections, the
permeate from the nanofilter will have a calculated TDS of 333, well within the
acceptable potability range. If even lower TDS values are desired for potable use,
30 then a somewhat “tighter” nanofilter can be used, to reject monovalents for example
up to 20% or 40% or higher. Since in the nanofilter, divalents are selectively rejected,

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the SAR of the nanofilter retentate will have an SAR even lower than the 1.7 SAR of the influent, and thus permit the retentate to be fully useful for irrigation water either by itself or in combination with the diluate of the electro dialysis apparatus. Recovery from the nanofilter apparatus can be determined strictly by the safety factor to reduce the precipitation potential in the retentate without regard to the amount of internal
5 water recovery. This is due to the realization that the retentate may be fully used for irrigation and thus none of the retentate is wasted.

Total DC power to the electro dialysis system is a very low 3.653 kWh/kgal diluate owing to the fact that only 70% of the TDS need be removed. Pumping
10 energy is also quite inexpensive because the wells are shallow, the electro dialysis system is generally a low pressure device, and the nanofilter is also operates at low pressure. Further, since 70% of the TDS is removed prior to entering the nanofilter apparatus, the osmotic pressure losses are low in the nanofilter.

15 Although various embodiments exemplarily shown have been described as using sensors, it should be appreciated that the invention is not so limited. The invention contemplates the modification of existing facilities to retrofit one or more systems, subsystems, or components and implement the techniques of the invention. Thus, for example, an existing facility, especially an agricultural or crop-growing
20 facility, can be modified to include one or more systems configured to provide irrigation water, potable water, or both, accordance with any one or more embodiments exemplarily discussed herein. Alternatively, existing systems and/or components or subsystems thereof can be modified to perform any one or more acts of the invention.

25 Moreover, it should also be appreciated that the invention is directed to each feature, system, subsystem, or technique described herein and any combination of two or more features, systems, subsystems, or techniques described herein and any combination of two or more features, systems, subsystems, and/or methods, if such features, systems, subsystems, and techniques are not mutually inconsistent, is
30 considered to be within the scope of the invention as embodied in the claims.

Use of ordinal terms such as “first,” “second,” “third,” and the like in the claims to modify a claim element does not by itself connote any priority, precedence, or order of one claim element over another or the temporal order in which acts of a method are performed, but are used merely as labels to distinguish one claim element
5 having a certain name from another element having a same name (but for use of the ordinal term) to distinguish the claim elements.

Those skilled in the art should appreciate that the parameters and configurations described herein are exemplary and that actual parameters and/or configurations will depend on the specific application in which the systems and
10 techniques of the invention are used. Those skilled in the art should also recognize or be able to ascertain, using no more than routine experimentation, equivalents to the specific embodiments of the invention. It is therefore to be understood that the embodiments described herein are presented by way of example only and that, within the scope of the appended claims and equivalents thereto; the invention may be
15 practiced otherwise than as specifically described.

As used herein, the term “plurality” refers to two or more items or components. The terms “comprising,” “including,” “carrying,” “having,” “containing,” and “involving,” whether in the written description or the claims and the like, are open-ended terms, i.e., to mean “including but not limited to.” Thus, the use
20 of such terms is meant to encompass the items listed thereafter, and equivalents thereof, as well as additional items. Only the transitional phrases “consisting of” and “consisting essentially of,” are closed or semi-closed transitional phrases, respectively, with respect to the claims. Further the use of the term “potable” with reference to water, especially treated water, does not limit the scope of the inventive
25 subject matter and can refer to water suitable for livestock use, including consumption.

What is claimed is:

CLAIMS

1. A water treatment system for producing potable water and water for irrigation
5 from saline water, the system comprising:
 - an electro dialysis apparatus including one or more monovalent selective
membranes and having an inlet fluidly connectable to a source of water to be treated
having a concentration of dissolved salts of between 500 mg/L and 10,000 mg/L, a
diluate outlet, and a concentrate outlet;
 - 10 a low-pressure nanofiltration apparatus positioned downstream of the
electrodialysis apparatus and having an inlet fluidly connectable to the diluate outlet
of the electro dialysis apparatus, a permeate outlet, and a retentate outlet;
 - an irrigation distribution system fluidly connectable to the retentate outlet of
the nanofiltration apparatus and with the diluate outlet of the electro dialysis apparatus;
 - 15 and
 - a diverting system configured to vary an amount of diluate from the
electrodialysis apparatus directed to the irrigation distribution system and an amount
of diluate from the electro dialysis apparatus directed to the nanofiltration apparatus.
- 20 2. The water treatment system of claim 1, wherein one or both of diluate of the
electrodialysis system and permeate of the nanofiltration system is post-treated to
remove excess boron.
3. The water treatment system of claim 1, further comprising a pre-filtration
25 system in fluid communication between the source of water to be treated and the inlet
of the electro dialysis apparatus.
4. The water treatment system of claim 3, wherein the pre-filtration system
includes at least one of a microfilter, a settler, a screen filter, a microsand filter, and a
30 coarse particle filter.

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5. The water treatment system of claim 3, wherein the pre-filtration system is preceded by a pH adjusting system and an air stripping system.

6. The water treatment system of claim 1, further comprising a conduit fluidly
5 connecting the permeate outlet of the nanofiltration apparatus to at least one of an inlet of a concentrating compartment or inlet of an electrode stream compartment of the electro dialysis apparatus.

7. The water treatment system of claim 1, wherein the concentrate outlet of the
10 electro dialysis apparatus is fluidly connectable to an inlet of an electrochemical generation system for production of one or more of chlorine gas, hypochlorite ion, sodium hydroxide, sulfuric acid, or hydrochloric acid.

8. The water treatment system of claim 7, wherein the electrochemical generation
15 system is fluidly connectable to a source of pH adjustment agent configured to introduce pH adjustment agent into the water to be treated.

9. The water treatment of one of claims 7 or 8, further comprising a water
softening apparatus upstream of the inlet to the electrochemical generation system.

10. The water treatment system of claim 1, further comprising a system for
20 treating permeate of the nanofiltration apparatus to achieve at least one of upward pH adjustment, lowered pH adjustment, addition of chlorine gas or addition of hypochlorite ion.

11. The water treatment system of claim 1, wherein the electro dialysis apparatus
25 further includes one or more monovalent cation selective membranes.

12. The water treatment system of claim 11, wherein the one or more monovalent
30 cation selective membranes have a selectivity coefficient for sodium over calcium of at least about 2.

13. The water treatment system of claim 1, 11, or 12, wherein the electro dialysis apparatus further includes one or more monovalent anion selective membranes.

5 14. The water treatment system of claim 13, wherein the one or more monovalent anion selective membranes have a selectivity coefficient for chloride over sulfate of at least about 2.

15. The water treatment system of claim 1, wherein the electro dialysis apparatus is
10 configured to produce diluate having a sodium adsorption ratio (SAR) value of less than about 10.

16. The water treatment system of claim 15, wherein the electro dialysis apparatus is configured to produce diluate having a SAR value of less than about 5.

15

17. The water treatment system of claim 15, wherein the nanofiltration apparatus is configured to produce retentate having a SAR value less than the SAR value of the diluate.

20 18. The water treatment system of claim 15, wherein the nanofiltration apparatus is configured to produce retentate and permeate, the retentate having a SAR value less than the SAR value of the permeate.

25 19. The water treatment system of claim 1, wherein the electro dialysis apparatus is configured to produce diluate having a higher ratio of divalent cations including magnesium and calcium to monovalent cations including sodium and a lower total concentration of ions than in the water to be treated.

30 20. The water treatment system of claim 19, wherein the nanofiltration apparatus is configured to produce retentate having a higher concentration of the divalent cations than the concentration of the divalent cations in the diluate and a lower

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concentration of the monovalent cations than the concentration of the monovalent cations in the diluate.

21. The water treatment system of claim 1, further comprising a controller
5 configured to adjust the amount of diluate from the electro dialysis apparatus directed to the irrigation distribution system and the amount of diluate from the electro dialysis apparatus directed to the nanofiltration apparatus with the diverting system in any amount from 0% to 100% of an amount of diluate produced in the electro dialysis apparatus.

10

22. The water treatment system of claim 21, wherein the controller is configured to adjust the amount of diluate directed to the irrigation distribution system and the amount of diluate directed to the nanofiltration apparatus depending on a relative demand for irrigation water and potable water.

15

23. The water treatment system of claim 1, further including a mixer configured to blend diluate produced in the electro dialysis apparatus with any retentate produced in the nanofiltration apparatus and produce irrigation water having a SAR value intermediate of a SAR value of the diluate and a SAR value of the retentate.

20

24. The water treatment system of claim 1, further comprising a sensor in communication with the controller, the sensor configured to provide the controller with an indication of one or more parameters of one of diluate from the electro dialysis apparatus and retentate from the nanofiltration apparatus, the controller configured to
25 adjust the diverting system responsive to the indication of the one or more parameters.

25. The water treatment system of claim 1, further comprising a sensor in communication with a controller, the sensor configured to provide the controller with an indication of one or more parameters of one of diluate from the electro dialysis
30 apparatus and retentate from the nanofiltration apparatus, the controller configured to

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adjust one or more operating parameters of the electro dialysis apparatus responsive to the indication of the one or more parameters.

26. A method of treating saline water to produce potable water and irrigation
5 water, the method comprising:
directing water to be treated having a concentration of dissolved salts of
between 500 mg/L and 10,000 mg/L to an inlet of an electro dialysis apparatus
including one or more monovalent cation selective membranes;
treating the water to be treated in the electro dialysis apparatus to produce a
10 diluate;
determining an amount of the diluate to direct to an irrigation distribution
system and an amount of the diluate to direct to an inlet of a low-pressure
nanofiltration apparatus;
treating any diluate directed to the inlet of the low-pressure nanofiltration
15 apparatus in the nanofiltration apparatus to produce a permeate and a retentate; and
combining any retentate with any of the diluate directed to the irrigation
distribution system at the irrigation distribution system.

27. The method of claim 26, further comprising directing at least a portion of any
20 permeate produced in the nanofiltration apparatus to a concentrating compartment of
the electro dialysis apparatus.

28. The method of claim 26, wherein producing the diluate in the electro dialysis
apparatus comprises producing treated water having a SAR value of less than about
25 10.

29. The method of claim 28, wherein producing the diluate in the electro dialysis
apparatus comprises producing treated water having a SAR value of less than about 5.

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30. The method of claim 26, wherein producing the retentate in the nanofiltration apparatus comprises producing treated water having a SAR value of less than the SAR value of the diluate.

5 31. The method of claim 26, further comprising directing at least a portion of the diluate to the irrigation distribution system.

32. The method of claim 31, wherein combining any retentate with any of the diluate directed to the irrigation distribution system comprises producing irrigation
10 water with a SAR value intermediate of a SAR value of the diluate and a SAR value of the retentate.

33. The method of claim 30, wherein combining any retentate with any of the diluate directed to the irrigation distribution system comprises producing irrigation
15 water with a SAR value of less than about 8 and a total dissolved solids level of greater than about 750 ppm.

34. The method of claim 26, further comprising adjusting an amount of diluate from the electro dialysis apparatus directed to the irrigation distribution system and the
20 amount of diluate from electro dialysis apparatus directed to the nanofiltration apparatus based on one of one or more parameters of the diluate from the electro dialysis apparatus and flow rate of retentate from the nanofiltration apparatus.

35. The method of claim 26, further comprising adjusting one or more operating
25 parameters of the electro dialysis apparatus based on one or more parameters of one of the diluate from the electro dialysis apparatus and any retentate from the nanofiltration apparatus.

36. The method of claim 26, further comprising pre-filtering the water to be
30 treated prior to directing the water to be treated to the inlet of the electro dialysis apparatus.

37. The method of claim 26, further comprising directing permeate suitable for use as potable water from the permeate outlet of the nanofiltration apparatus to a potable water point of use.

5

38. The method of claim 26, wherein a pH of the water to be treated is reduced prior to introduction into the electro dialysis apparatus such that a concentration of bicarbonate ions in the water to be treated is reduced.

10 39. The method of claim 38, wherein the pH is reduced using a controlled amount of one of either hydrochloric or sulfuric acid.

40. The method of claim 39, wherein an amount of hydrochloric or sulfuric acid added to the water to be treated is determined based on a relationship between the pH
15 of the water to be treated, and one of either of a concentrate water recovery of the electro dialysis apparatus or a SAR value of diluate of the electro dialysis apparatus.

41. The method of claim 38, wherein the reduced pH level water to be treated is treated in an air stripping apparatus to remove a portion of dissolved or gaseous
20 carbon dioxide prior to introduction of the water to be treated into the electro dialysis apparatus.

42. The method of claim 26, wherein the water to be treated is subjected to aeration followed by filtration such that a portion of one or more of elements or
25 compounds of iron, manganese, arsenic, or sulfur are reduced in concentration prior to introduction of the water to be treated into the electro dialysis apparatus.

43. The method of claim 26, wherein a ratio of retentate outlet flow to permeate flow through the nanofiltration apparatus is controlled so as to limit a scaling potential
30 of dissolved constituents of the nanofiltration retentate.

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44. The method of claim 26, wherein effluent from one of more of the nanofiltration permeate stream or the electro dialysis diluate outlet stream is further treated to reduce dissolved concentration of boron compounds.

5 45. The method of claim 26, wherein a system including the nanofiltration apparatus and the electro dialysis apparatus is controlled and operated such that permeate from the nanofiltration apparatus meets water quality standards for potable drinking water.

10 46. The method of claim 26, wherein concentrations of one or both of dissolved boron or silica compounds in the water to be treated remain essentially the same in concentrating and diluting streams of the electro dialysis apparatus, and one or more of retentate and permeate of the nanofiltration apparatus.

15 47. The method of claim 26, wherein a system including the nanofiltration apparatus and the electro dialysis apparatus is controlled and operated such that retentate from the nanofiltration apparatus, diluate from the electro dialysis apparatus, or a combination of both meet water quality requirements for agricultural water.

20 48. The method of claim 26, wherein a system including the nanofiltration apparatus and the electro dialysis apparatus is controlled and operated such that concentrate from the electro dialysis apparatus meets water quality standards for use in produced oil recovery operations.

25 49. The method of claim 48, wherein the concentrate is treated with one of ion exchange softening to reduce a concentration of divalent cations or pH adjustment prior to use in produced oil recovery operations.

30 50. The method of claim 26, wherein a ratio of divalent selenate or selenite ion concentration to chloride ion concentration in the concentrate from the electro dialysis apparatus is less than that of the water to be treated.

51. The method of claim 26, wherein a ratio of divalent selenate or selenite ion concentration to chloride concentration in diluate from the electro dialysis apparatus is less than that of the water to be treated and a ratio of calcium ion concentration to
5 sodium ion concentration in the diluate from the electro dialysis apparatus is greater than that of the water to be treated.

52. The method of claim 26, further comprising removing at least about 50% of nitrite present in the water to be treated in the electro dialysis apparatus.

10

53. The method of claim 52, further comprising removing at least about 50% of nitrate present in diluate of the electro dialysis apparatus directed to the nanofiltration apparatus in the nanofiltration apparatus.

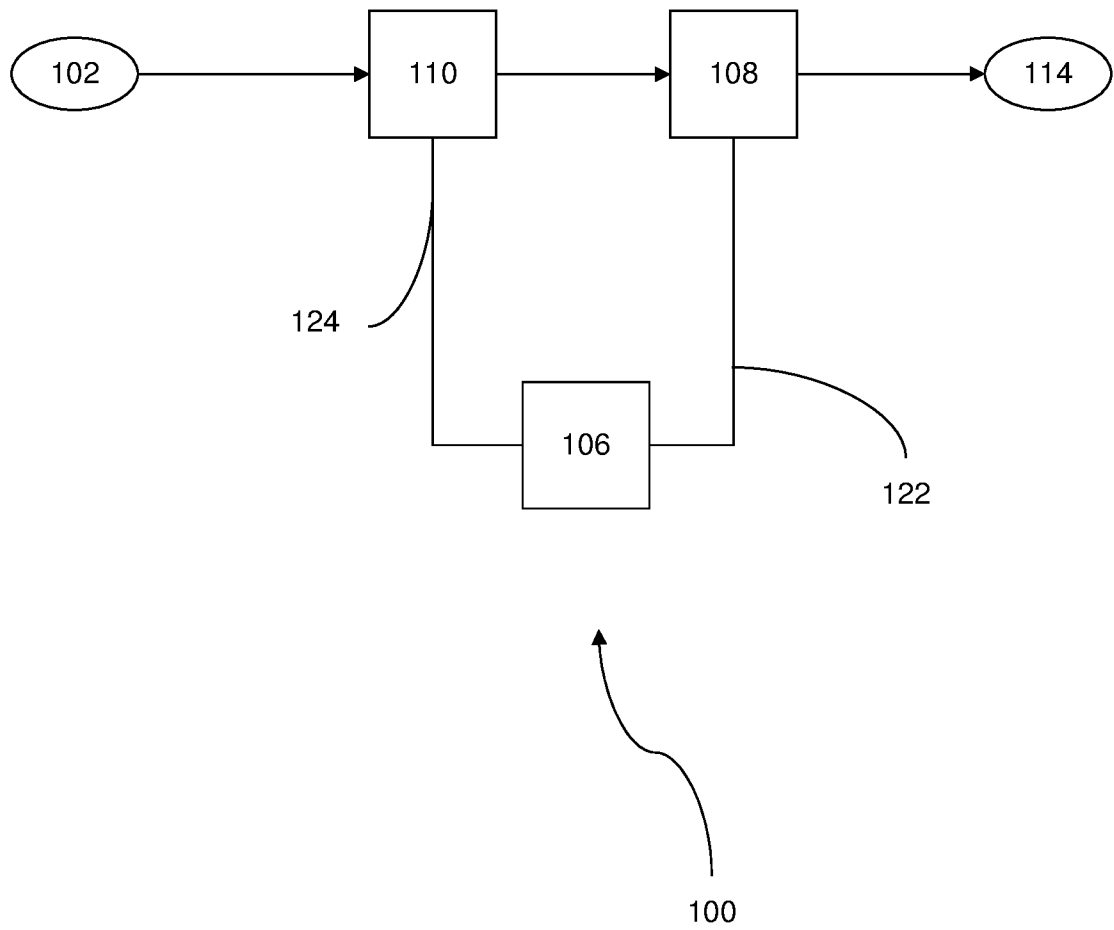


FIG. 1

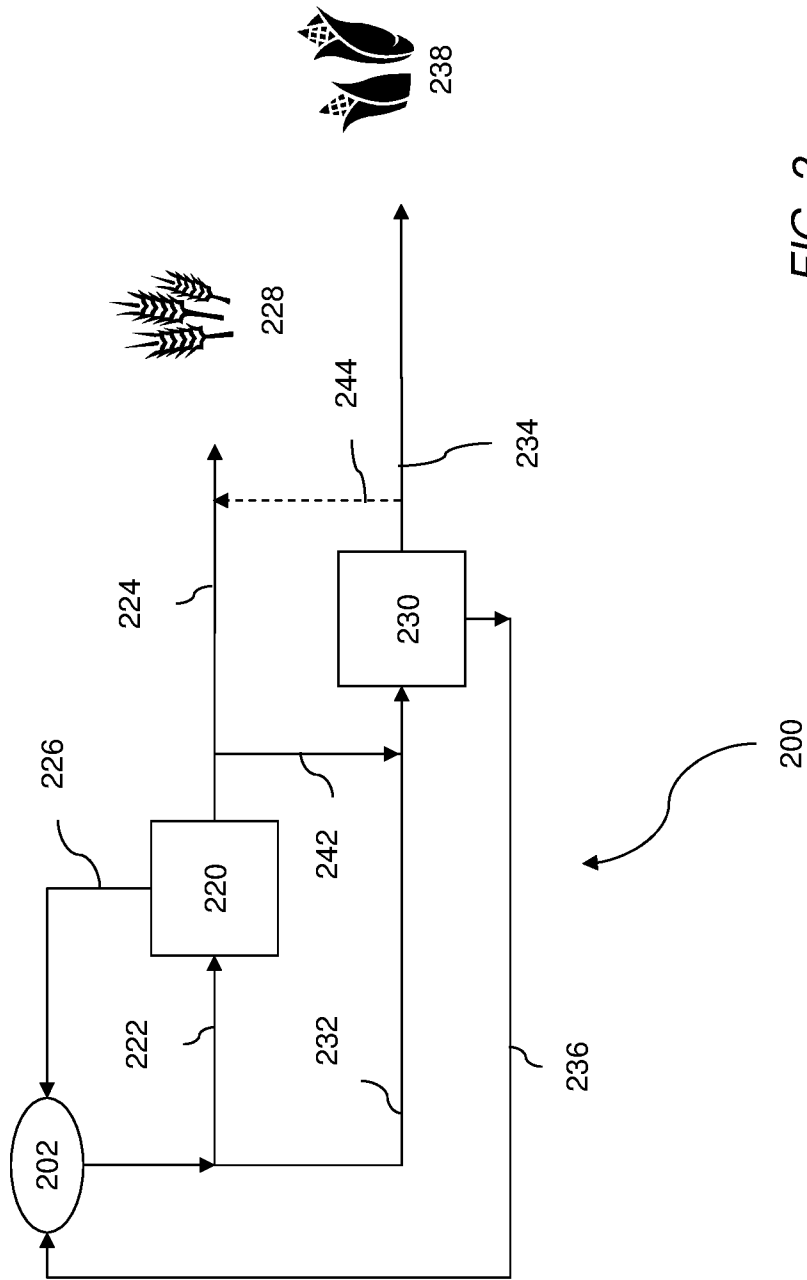


FIG. 2

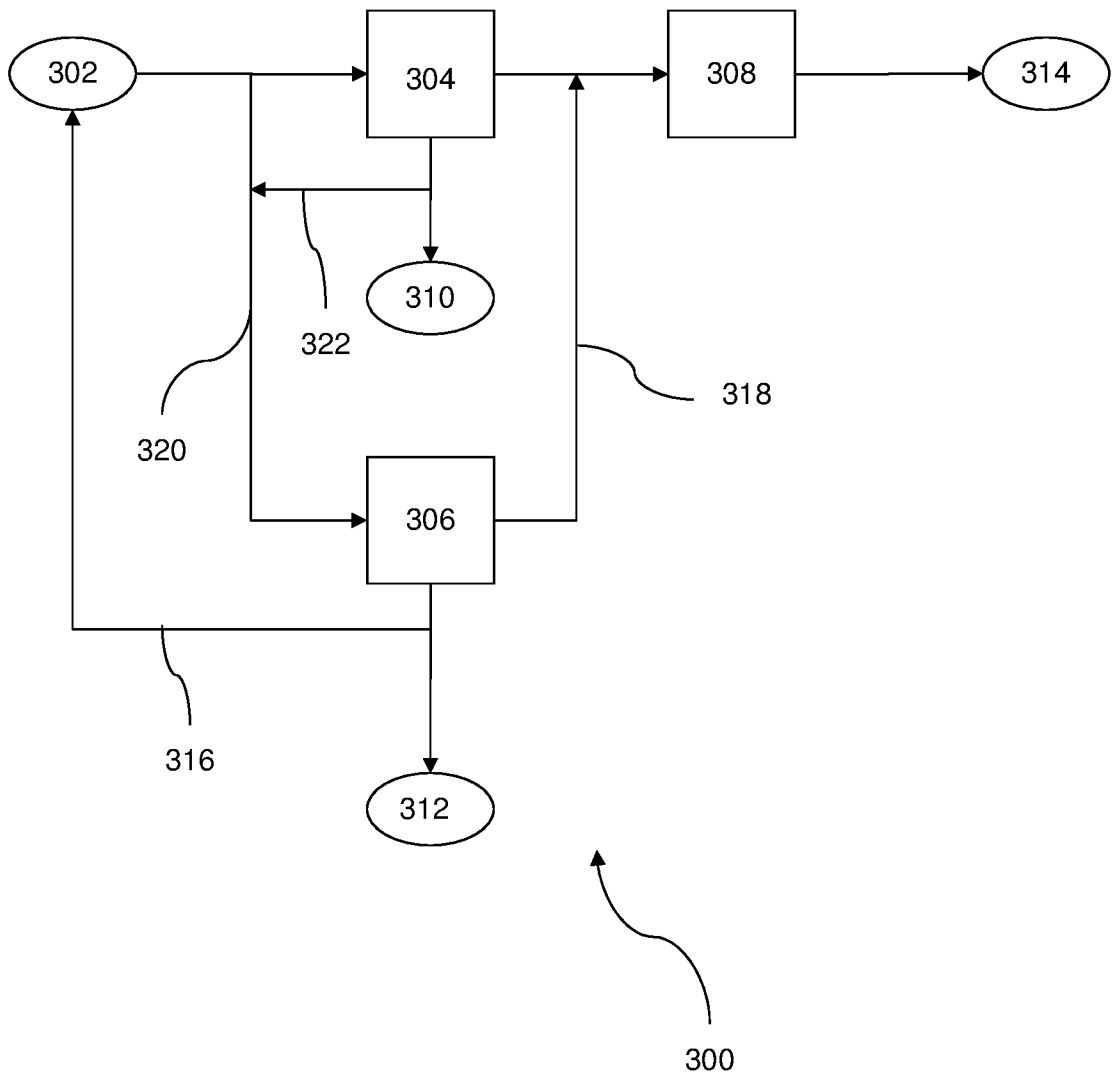


FIG. 3

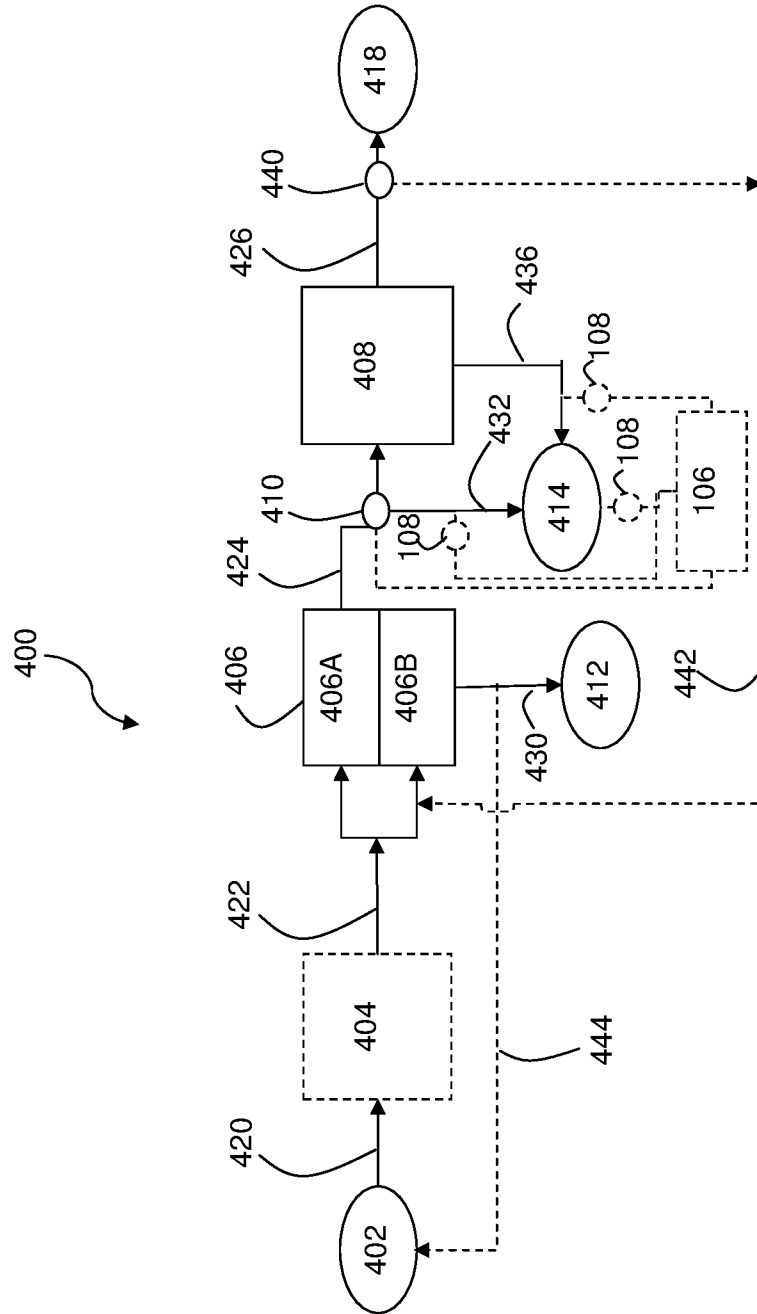


FIG. 4

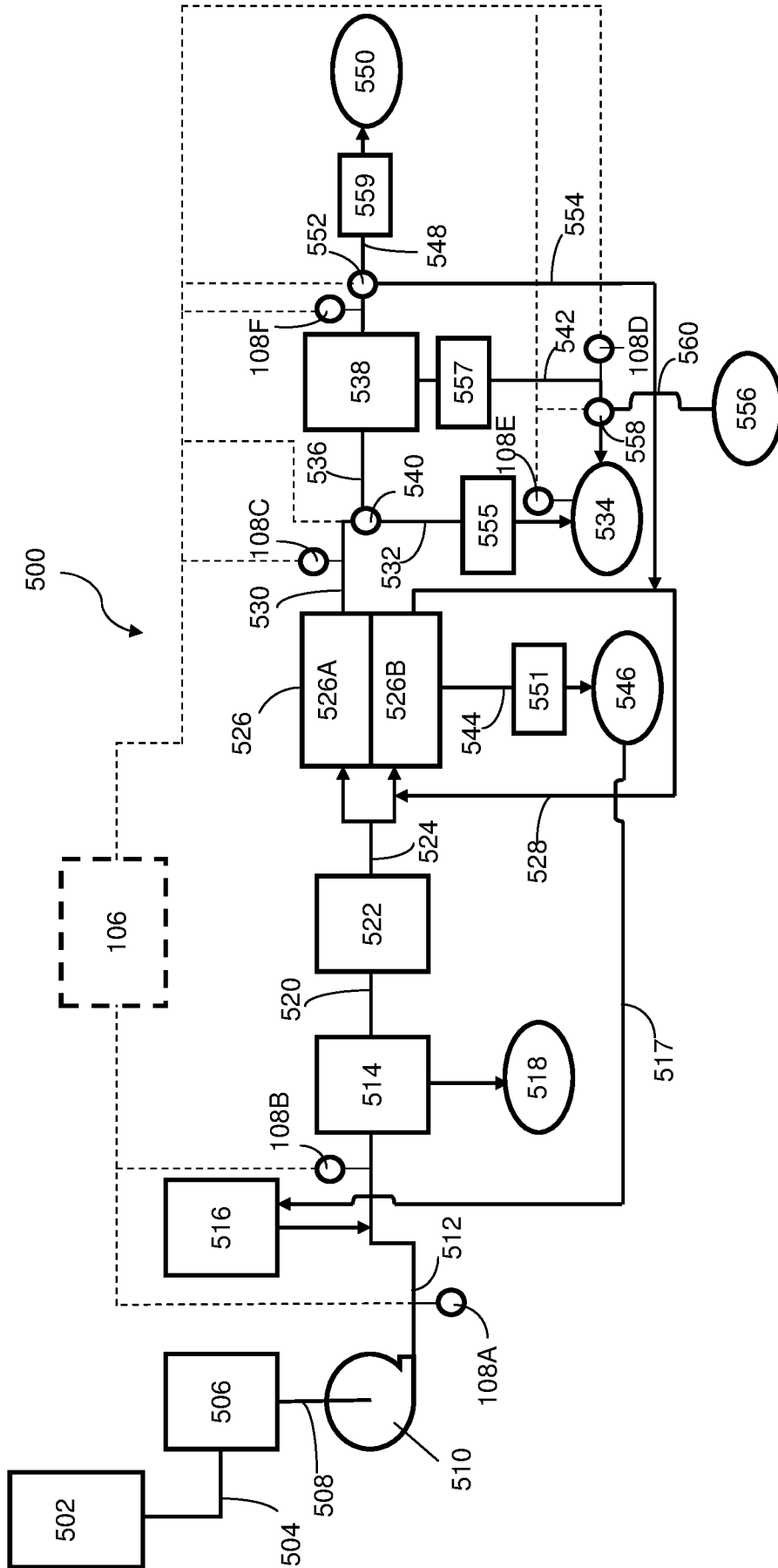


FIG. 5

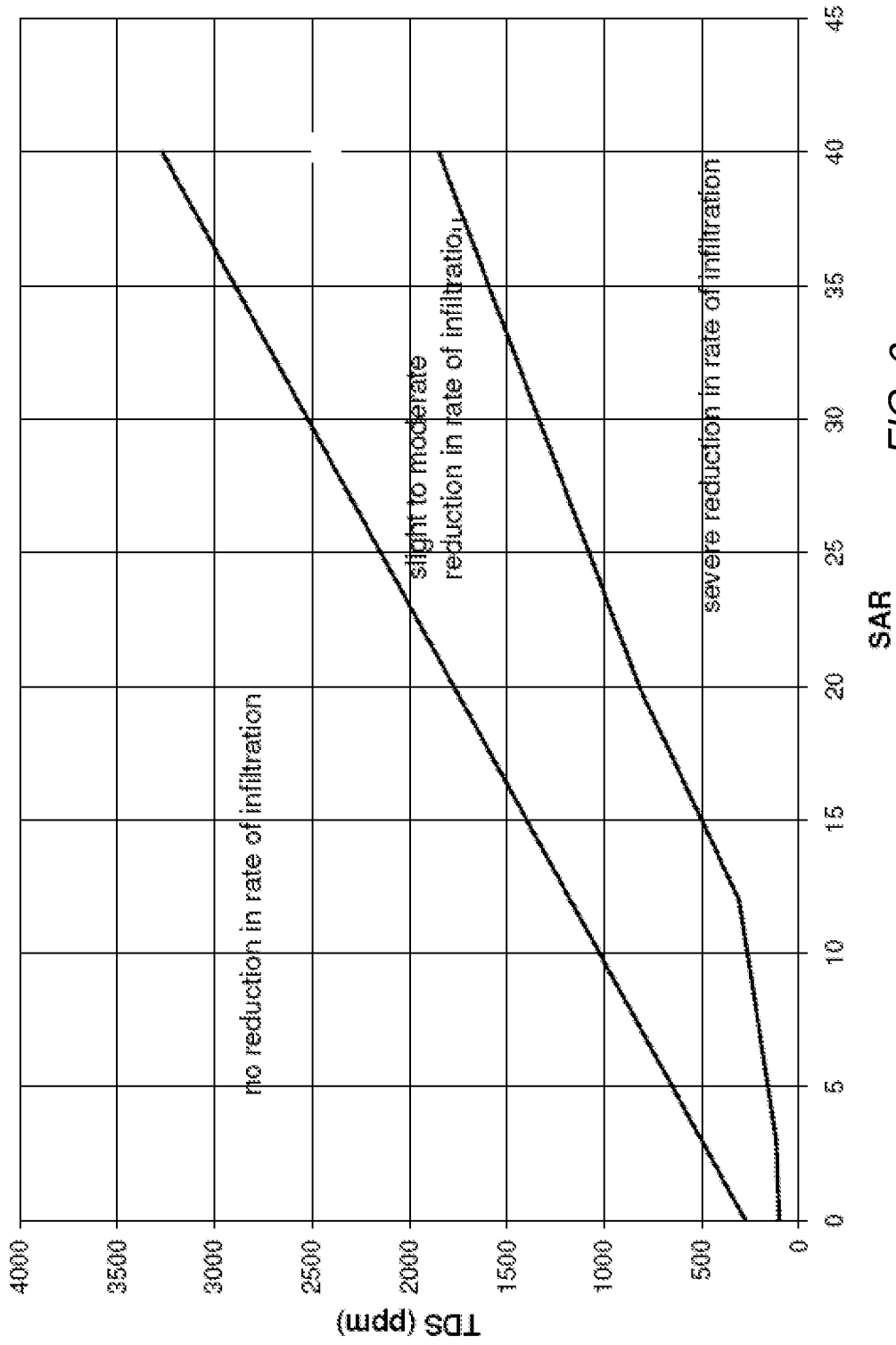


FIG. 6

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US 18/47034

A. CLASSIFICATION OF SUBJECT MATTER
 IPC(8) - C02F 1/20, C02F 1/26, C02F 9/00 (2018.01)
 CPC - C02F 1/20, C02F 1/265, C02F 2101/12

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

See Search History Document

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

See Search History Document

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

See Search History Document

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X -- Y -- A	US 2013/0015135 A1 (GANZI et al.) 17 January 2013 (17.01.2013) Abstract, para[0006]-[0007], [0009], [0011], [0013], [0016], [0018], [0025], [0034], [0041]-[0042], [0045]-[0046], [0049], [0053]-[0054], [0058], [0061], [0063]-[0064], [0066]-[0067], [0069], [0072], [0074]-[0075], [0077], [0084]-[0086], [0088], [0090]-[0092], [0094], [0098], [0109], [0118], [0129]-[0130]; Figures 4-5, 8; Tables 1-2;	1-4, 6-37, 44-45, 47, 51-53 ----- 5, 38-39, 41-42, 48-49 ----- 40, 43, 46, 50
Y	US 2012/0055875 A1 (LIEN) 08 March 2012 (08.03.2012) Abstract, para[0007], [0027]	5
Y -- A	US 2014/0045248 A1 (ENVIRO WATER MINERALS COMPANY, INC.) 13 February 2014 (13.02.2014) para[0022], [0030], [0035], [0048]; Figure 2	38-39, 41 ----- 40
Y	CN 101955282 A (JIANGSU PROVINCIAL ACADEMY ENVIRONMENTAL SCIENCE) 26 January 2011 (26.01.2011) Abstract, para[0001], [0008]	42
Y	US 2014/0216934 A1 (EVOQUA WATER TECHNOLOGIES LLC) 07 August 2014 (07.08.2014) Title, para[0001], [0005], [0007], [0023], [0028]	48-49
A	WO 2014/110429 A1 (MURKUTE et al.) 17 July 2014 (17.07.2014) Entire document, especially para[125]	1-53
A	US 2008/0067125 A1 (WILKINS et al.) 20 March 2008 (20.03.2008) Entire document	1-53

Further documents are listed in the continuation of Box C. See patent family annex.

* Special categories of cited documents:	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"A" document defining the general state of the art which is not considered to be of particular relevance	"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"E" earlier application or patent but published on or after the international filing date	"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"&" document member of the same patent family
"O" document referring to an oral disclosure, use, exhibition or other means	
"P" document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search
 18 October 2018

Date of mailing of the international search report

09 NOV 2018

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