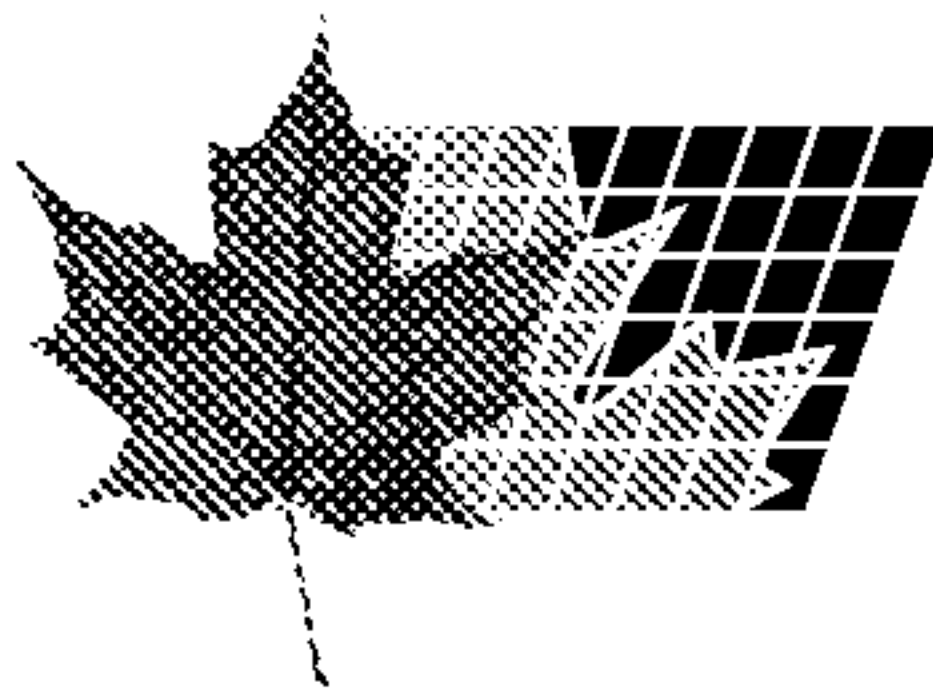




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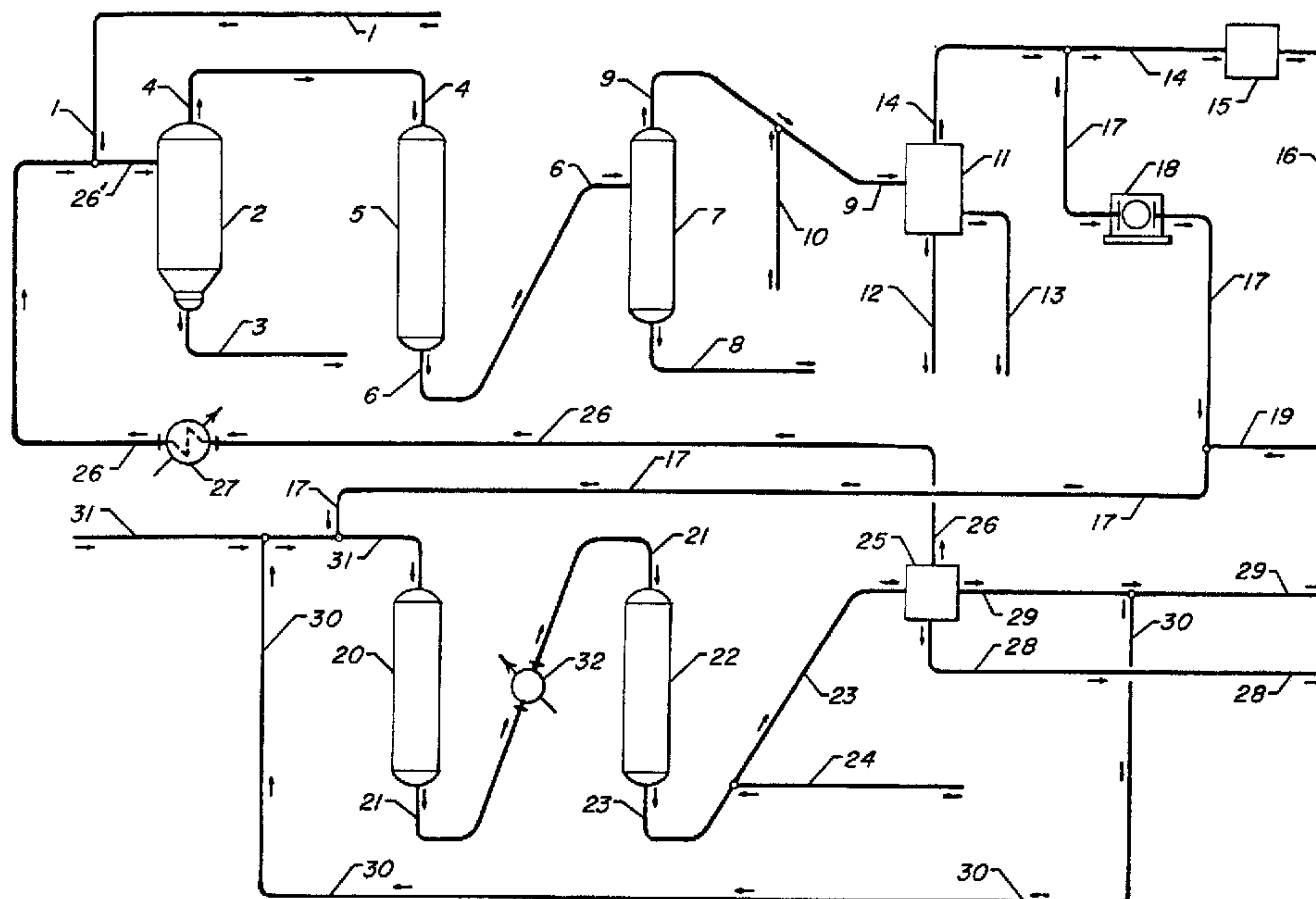
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(54) **PROCEDE DE TRAITEMENT SIMULTANE DE DEUX
CHARGES D'ALIMENTATION DANGEREUSES**

(54) **PROCESS FOR THE SIMULTANEOUS TREATMENT OF TWO
HAZARDOUS FEEDSTOCKS**



(57) Hydrogenated distillable hydrocarbonaceous product is produced from a first feed comprising hydrocarbonaceous compounds and having a non-distillable component, and a second feed comprising halogenated organic compounds by contacting the first feed with a hot hydrogen-rich gaseous stream to increase the temperature of this feed stream and to vaporize at least a portion of the distillable hydrocarbonaceous compounds thereby producing a distillable hydrocarbonaceous product which is immediately hydrogenated in an integrated first hydrogenation zone and thereafter contacting the second feed with hydrogen in a second hydrogenation zone at hydrogenation conditions to produce a hydrogenated hydrocarbonaceous product and at least one water-soluble inorganic halide compound.

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**"PROCESS FOR THE SIMULTANEOUS
TREATMENT OF TWO HAZARDOUS FEEDSTOCKS"**

FIELD OF THE INVENTION

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The field of art to which this invention pertains is the production of hydrogenated distillable hydrocarbonaceous compounds from a hydrocarbonaceous feed having a non-distillable component and a feed comprising
10 halogenated organic compounds. More specifically, the invention relates to a novel process for the simultaneous hydrogenation of a first feedstock comprising hydrocarbonaceous compounds and having a non-distillable component, and a second feedstock comprising halogenated
15 organic compounds.

BACKGROUND OF THE INVENTION

There is a steadily increasing demand for technology which is capable of the simultaneous
20 hydrogenation of a first feedstock comprising hydrocarbonaceous compounds and having a non-distillable component and a second feedstock comprising halogenated organic compounds. Previous techniques utilized to dispose of such feedstocks which are often undesirable
25 waste effluents such as used lubricating oils and spent solvents, for example, have frequently become environmentally unpopular or illegal and, in general, have always been expensive. With the increased environmental emphasis for the treatment and recycle of chlorinated
30 organic product and waste oils, there is an increased need for the conversion of these products in the event that they become unwanted or undesirable. For example, during the disposal or recycle of potentially environmentally harmful halogenated organic waste streams, an important
35 step in the total solution to the problem is the conditioning of the halogenated organic stream which

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facilitates the ultimate resolution to provide product streams which may be handled in an environmentally acceptable manner. In another example, large quantities of used motor oil are generated and discarded which oil would provide a large potential supply of feedstock for the present invention while providing an environmentally responsible disposal. Therefore, those skilled in the art have sought to find feasible techniques to convert such feedstocks to provide hydrocarbonaceous product streams which may be safely and usefully employed or recycled. Previous techniques which have been employed include incineration which, in addition to potential pollution considerations, fails to recover valuable hydrocarbonaceous materials.

Examples of prior art techniques include desulurization and hydrorefining as taught in the following patents.

In U.S. Patent No. 3,133,013, hydrorefining of hydrocarbons is disclosed for the purpose of removing diverse contaminants therefrom and/or reacting such hydrocarbons to improve the chemical and physical characteristics thereof. In addition, the process is directed toward the selective hydrogenation of unsaturated, coke-forming hydrocarbons through the use of particular conditions whereby the formation of coke, otherwise resulting from the hydrorefining of such hydrocarbon fractions and distillates, is effectively inhibited.

In U.S. Patent No. 3,992,285, desulfurization of a hydrocarbonaceous black oil containing sulfur and asphaltic material is performed by preheating the oil by indirect heat exchange to a temperature not in excess of 288°C (550°F), commingling the preheated oil with a steam-containing gas to raise the temperature of the oil to a

desulfurization temperature of 316-427°C (600 to 800°F) and contacting the thus heated oil at hydrocarbon conversion conditions with a desulfurization catalyst.

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BRIEF SUMMARY OF THE INVENTION

The invention provides an improved process for the production of hydrogenated, distillable hydrocarbonaceous product from a first feed comprising hydrocarbonaceous compounds and having a non-distillable component, and a second feed comprising halogenated organic compounds by contacting the first feed with a hot hydrogen-rich gaseous stream derived at least in part from processing of the second feed to increase the temperature of this feed stream and to vaporize at least a portion of the distillable hydrocarbonaceous compounds thereby producing a distillable hydrocarbonaceous product which is immediately hydrogenated in an integrated first hydrogenation zone. The second feed is then contacted with hydrogen derived from the first zone in a second hydrogenation zone at hydrogenation conditions to produce a hydrogenated hydrocarbonaceous product and at least one water-soluble inorganic halide compound. Important elements of the process are the integrated hydrogenation reaction zones which reduce capital and utility costs, and the recycle of the hydrogen-rich gas stream recovered from the second hydrogenation zone to the first feed processing steps. This recycle gas stream may contain small quantities of unconverted volatile organic halide compounds and the first hydrogenation zone serves to ensure complete destruction of these compounds. The consecutive passage of this recycle gas stream through both a thermal zone for heating the gas stream followed by a catalytic hydrogenation zone will convert greater than 99% of the organic halide compounds to hydrogen halide.

One embodiment of the invention may be characterized as a process for the simultaneous hydrogenation of a first feedstock comprising hydrocarbonaceous compounds and having a non-distillable component, and a second feedstock comprising halogenated organic compounds which process comprises: (a) contacting the first feedstock with a first hydrogen-rich gaseous stream having a temperature greater than the first feedstock in a flash zone at flash conditions selected to increase the temperature of the first feedstock and to vaporize at least a portion thereof to provide a hydrocarbonaceous vapor stream comprising hydrogen, and a heavy product comprising the non-distillable component; (b) contacting the hydrocarbonaceous vapor stream comprising hydrogen with a hydrogenation catalyst in a first hydrogenation reaction zone at hydrogenation conditions to increase the hydrogen content of the hydrocarbonaceous compounds contained in the hydrocarbonaceous vapor stream; (c) condensing at least a portion of the resulting effluent from the first hydrogenation reaction zone to produce a second hydrogen-rich gaseous stream and a first liquid hydrogenated stream comprising hydrogenated distillable hydrocarbonaceous compounds; (d) reacting the second feedstock and at least a portion of the second hydrogen-rich gaseous stream with a hydrogenation catalyst in a second hydrogenation reaction zone at hydrogenation conditions selected to produce hydrocarbonaceous compounds and at least one water-soluble inorganic halide compound; (e) contacting the resulting effluent from the second hydrogenation zone with a halide-lean aqueous scrubbing solution; (f) introducing the resulting admixture of the effluent and the aqueous scrubbing solution into a separation zone to provide a third hydrogen-rich gaseous stream, a second liquid hydrogenated stream comprising hydrocarbonaceous compounds and a halide-rich aqueous scrubbing solution

containing at least a portion of the water-soluble inorganic halide compound; (g) recycling and heating at least a portion of the third hydrogen-rich gaseous stream recovered in step (f) to step (a) as at least a portion of the first hydrogen-rich gaseous stream; and (h) recovering the first liquid hydrogenated stream from step (c) and the second liquid hydrogenated stream from step (f).

Other embodiments of the present invention encompass further details such as preferred feedstocks, hydrogenation catalysts, aqueous scrubbing solutions and operating conditions, all of which are hereinafter disclosed in the following discussion of each of these facets of the invention.

BRIEF DESCRIPTION OF THE DRAWING

The drawing is a simplified process flow diagram of a preferred embodiment of the present invention.

DETAILED DESCRIPTION OF THE INVENTION

The present invention provides an improved integrated process for the simultaneous hydrogenation of a first feedstock comprising hydrocarbonaceous compounds and having a non-distillable component, and a second feedstock comprising halogenated organic compounds.

A wide variety of hydrocarbonaceous streams having a non-distillable component are candidates for the first feedstock. Examples of such hydrocarbonaceous streams which are suitable for treatment by a process of the present invention are dielectric fluids, hydraulic fluids, heat transfer fluids, used lubricating oil, used cutting oils, used solvents, still bottoms from solvent recycle operations, coal tars, atmospheric residuum, oils contaminated with polychlorinated biphenyls (PCB), and other hydrocarbonaceous industrial waste. Many of these

hydrocarbonaceous streams may contain non-distillable components which include, for example, organometallic compounds, inorganic metallic compounds, finely divided particulate matter and non-distillable hydrocarbonaceous compounds. The present invention is particularly advantageous when the non-distillable components comprise sub-micron particulate matter and the conventional techniques of filtration or centrifugation tend to be highly ineffective.

10 The presence of a non-distillable component including finely divided particulate matter in a hydrocarbonaceous feed to a hydrogenation zone greatly increases the difficulty of hydrogenation. A non-distillable component tends (1) to foul the hot heat exchange surfaces which are used to heat the feed to hydrogenation conditions, (2) to form coke or in some other manner deactivate the hydrogenation catalyst thereby shortening its active life and (3) to otherwise hinder a smooth and facile hydrogenation operation. Particulate matter in a feed stream tends to deposit within the hydrogenation zone and to plug a fixed hydrogenation catalyst bed thereby abbreviating the time on stream.

 Once the first feedstock is separated into a distillable hydrocarbonaceous stream and a heavy non-distillable product, the resulting distillable hydrocarbonaceous stream is introduced into a hydrogenation zone. If the first feedstock contains metallic compounds such as those that contain metals such as zinc, copper, iron, barium, phosphorous, magnesium, aluminum, lead, mercury, cadmium, cobalt, arsenic, vanadium, chromium, and nickel, these compounds will be isolated in the relatively small volume of recovered non-distillable product which may then be treated for metals recovery or otherwise disposed of as desired. In the event that the feedstock contains distillable hydrocarbonaceous compounds which include sulfur, oxygen,

nitrogen, metal or halogen components, the resulting recovered distillable hydrocarbonaceous stream is hydrogenated to remove or convert such components as desired. In a preferred embodiment of the present invention, the hydrogenation of the resulting distillable hydrocarbonaceous stream is preferably conducted immediately without intermediate separation or condensation. The advantages of the integrated process of the present invention will be readily apparent to those skilled in the art and include the economy of greatly reduced utility costs.

In accordance with the present invention, a hydrocarbonaceous stream containing a non-distillable component is contacted with a hot hydrogen-rich gaseous stream having a temperature greater than the hydrocarbonaceous stream in a flash zone at flash conditions thereby increasing the temperature of the hydrocarbonaceous stream and vaporizing at least a portion thereof to provide a hydrocarbonaceous vapor stream comprising hydrogen and a heavy non-distillable product. The hot hydrogen-rich gaseous stream preferably comprises more than about 70 mol.% hydrogen and preferably more than about 90 mol.% hydrogen. In a preferred embodiment, the hot hydrogen-rich gaseous stream is comprised of a recycle hydrogen gas stream which contains trace quantities of halogenated organic compounds.

The hot hydrogen-rich gaseous stream is multi-functional and serves as (1) a heat source used to directly heat the hydrocarbonaceous feed stream to preclude the coke formation that could otherwise occur when using an indirect heating apparatus such as a heater or heat-exchanger, (2) a diluent to reduce the partial pressure of the hydrocarbonaceous compounds during vaporization in the flash zone, (3) a reactant to minimize the possible formation of hydrocarbonaceous polymers at elevated temperatures, (4) a stripping medium, and (5) at

least a portion of the hydrogen required in the hydrogenation reaction zone. In addition, when the hot hydrogen-rich gaseous stream is composed of a recycle hydrogen gas stream which contains halogenated organic compounds, the subsequent thermal and catalytic zones through which this stream passes is a valuable technique to ensure essentially complete conversion of halogenated organic compounds in the present process. In accordance with the present invention, the first feedstock is preferably maintained at a temperature less than 482°F (250°C) before being introduced into the flash zone in order to prevent or minimize the thermal degradation. Depending upon the characteristics and composition of the first feedstock, the hot hydrogen-rich gaseous stream is introduced into the flash zone at a temperature greater than the hydrocarbonaceous feed stream and preferably at a temperature from 200°F (93°C) to 1200°F (649°C).

The flash zone is preferably maintained at flash conditions which include a temperature from 150°F (65°C) to 860°F (460°C), a pressure from atmospheric to 2000 psig (103 to 13893 kPa), a hydrogen circulation rate of 1000 SCFB (168 normal m³/m³) to 60,000 SCFB (10,110 normal m³/m³) based on the hydrocarbonaceous feed stream to the flash zone and an average residence time of the hydrogen-containing, hydrocarbonaceous vapor stream in the flash zone from about 0.1 seconds to about 50 seconds. A more preferred average residence time of the hydrogen-containing hydrocarbonaceous vapor stream in the flash zone is from about 1 second to about 10 seconds.

The resulting heavy non-distillable portion of the first feedstock is removed from the bottom of the flash zone as required to yield a heavy non-distillable product. The heavy non-distillable product may contain a relatively small amount of distillable components, but since essentially all of the non-distillable components contained in the first feedstock are recovered in this

product stream, the term "heavy non-distillable product" is nevertheless used for the convenient description of this product stream. The heavy non-distillable product preferably contains a distillable component of less than 10 weight percent and more preferably less than 5 wt.%. Under certain circumstances with a feed stream not having an appreciable amount of liquid non-distillable components, it is contemplated that an additional liquid may be utilized to flush the heavy non-distillables from the flash zone. An example of this situation is when the hydrocarbonaceous feed stream comprises a very high percentage of distillable hydrocarbonaceous compounds and relatively small quantities of finely divided particulate matter "solid" and essentially no liquid non-distillable component for use as a carrier for the solids. Such a flush liquid may, for example, be a high boiling range vacuum gas oil having a boiling range from 700°F (371°C) to 1000°F (538°C) or a vacuum tower bottom stream boiling at a temperature greater than 1000°F (538°C). In the event when the non-distillable fraction is flushed with vacuum resid (bitumen), the properties of the resid are enhanced for use as an asphalt cement and thus provides a useful outlet for the bottoms. In addition, toxic metals are stabilized and made non-leachable. The selection of a flush liquid depends upon the composition of the hydrocarbonaceous feed stream and the prevailing flash conditions in the flash separator, and the volume of the flush liquid is preferably limited to that required for removal of the heavy non-distillable component.

The resulting hydrogen-containing hydrocarbonaceous vapor stream is removed from the flash zone and is introduced into a first catalytic hydrogenation zone containing hydrogenation catalyst and maintained at hydrogenation conditions. The catalytic hydrogenation zone may contain a fixed, ebullated or fluidized catalyst bed. This reaction zone is preferably maintained under an imposed pressure from atmospheric to 2000 psig (103 to 13893 kPa) and more preferably under a pressure from 100 psig to 1800 psig (739 to 12514 kPa). Suitably, such reaction is conducted with a maximum catalyst bed temperature in the range of 122°F (50°C) to 850°F (454°C) selected to perform the desired hydrogenation conversion to reduce or eliminate the undesirable characteristics or components of the hydrocarbonaceous vapor stream. In accordance with the present invention, it is contemplated that the desired hydrogenation conversion includes, for example, dehalogenation, desulfurization, denitrification, olefin saturation, oxygenate conversion and hydrocracking. Further preferred operating conditions include liquid hourly space velocities (LHSV) in the range from 0.05 hr.⁻¹ to 20 hr.⁻¹ and hydrogen circulation rates from 200 standard cubic feet per barrel (SCFB) (33.71 normal m³/m³) to 70,000 SCFB (11,796 normal m³/m³), preferably from 300 SCFB (50.6 normal m³/m³) to about 20,000 SCFB (3371 normal m³/m³).

In the event that the temperature of the hydrogen-containing hydrocarbonaceous vapor stream which is removed from the flash zone is not deemed to be exactly the temperature selected to operate the catalytic hydrogenation zone, the temperature of the vapor stream may be adjusted either upward or downward in order to achieve the desired temperature in the catalytic

hydrogenation zone. Such a temperature adjustment may be accomplished, for example, by the addition of either cold or hot hydrogen.

The preferred catalytic composite disposed
5 within the first hydrogenation zone can be characterized as containing a metallic component having hydrogenation activity, which component is combined with a suitable refractory inorganic oxide carrier material of either synthetic or natural origin. Preferred carrier materials
10 are alumina, silica and mixtures thereof. Suitable metallic components having hydrogenation activity are those selected from the group comprising the metals of Groups VI-B and VIII of the Periodic Table, as set forth in the Periodic Table of the Elements, E.H. Sargent and
15 Company, 1964. Thus, the catalytic composites may comprise one or more metallic components from the group of molybdenum, tungsten, chromium, iron, cobalt, nickel, platinum, palladium, iridium, osmium, rhodium, ruthenium, and mixtures thereof. The concentration of the
20 catalytically active metallic component, or components, is primarily dependent upon a particular metal as well as the physical and/or chemical characteristics of the particular hydrocarbon feedstock. For example, the metallic components of Group VI-B are generally present in the
25 catalytic composite in an amount within the range of from 1 to 20 wt.%, the iron-group metals in an amount within the range of 0.2 to 10 wt.%, whereas the noble metals of Group VIII are preferably present in an amount within the range of from 0.1 to 5 wt.%, calculated on an elemental
30 basis. In addition, any catalyst employed commercially for hydrogenating middle distillate hydrocarbonaceous compounds to remove nitrogen and sulfur may function effectively in the hydrogenation zone of the present invention. It is further contemplated that hydrogenation
35 catalytic composites may comprise one or more of the following components: cesium, francium, lithium,

potassium, rubidium, sodium, copper, gold, silver, cadmium, mercury and zinc.

The effluent from the first hydrogenation zone is preferably partially condensed in a hot separator and then contacted with an aqueous scrubbing solution and the admixture is admitted to a separation zone in order to separate a spent aqueous stream, a hydrogenated hydrocarbonaceous liquid phase and a hydrogen-rich gaseous phase. The contact of the hydrocarbonaceous effluent from the hydrogenation zone with the aqueous scrubbing solution may be performed in any convenient manner and is preferably conducted by co-current, in-line mixing which may be promoted by inherent turbulence, mixing orifices or any other suitable mixing means. The aqueous scrubbing solution is preferably introduced in an amount from 1 to 100 volume percent based on the hydrocarbonaceous effluent from the hydrogenation zone. The aqueous scrubbing solution is selected depending on the characteristics of the hydrocarbonaceous vapor stream introduced into the hydrogenation zone. For example, if the hydrocarbonaceous vapor stream to the hydrogenation zone comprises halogenated compounds, the aqueous scrubbing solution preferably contains a basic compound such as calcium hydroxide, potassium hydroxide, potassium carbonate, sodium carbonate or sodium hydroxide in order to neutralize the acid such as hydrogen chloride, hydrogen bromide and hydrogen fluoride, for example, which is formed during the hydrogenation of the halogen compounds. In the event that the hydrocarbonaceous vapor stream contains only sulfur and nitrogen compounds, water may be a suitable aqueous scrubbing solution to dissolve the resulting hydrogen sulfide and ammonia.

The resulting hydrogenated hydrocarbonaceous liquid phase is preferably recovered from the hydrogen-rich gaseous phase in a separation zone which is maintained at essentially the same pressure as the first

hydrogenation reaction zone and as a consequence contains dissolved hydrogen and low molecular weight normally gaseous hydrocarbons if present. In accordance with the present invention, it is preferred that the hydrogenated hydrocarbonaceous liquid phase comprising the hereinabove mentioned gases be stabilized in a convenient manner, such as, for example, by stripping or flashing to remove the normally gaseous components to provide a stable hydrogenated distillable hydrocarbonaceous product.

10 A wide variety of halogenated organic compounds, both unsaturated and saturated, are candidates for the second feedstock. Examples of organic streams comprising halogenated organic compounds are dielectric fluids, hydraulic fluids, heat transfer fluids, used lubricating oil, used cutting oils, used solvents, halogenated hydrocarbonaceous by-products, oils contaminated with polychlorinated biphenyls (PCB), halogenated wastes, petrochemical by-products and other halogenated hydrocarbonaceous industrial waste. The halogenated organic feed streams may also contain organic compounds which include sulfur, oxygen, nitrogen or metal components which may be hydrogenated to remove or convert such components as desired. The halogenated organic compounds may also contain hydrogen and are therefore then referred to as hydrocarbonaceous compounds.

Preferred second feedstocks comprise fractionation column bottoms in the production of allyl chloride, fractionation column bottoms in the production of ethylene dichloride, fractionation column bottoms in the production of trichloroethylene and perchloroethylene, used dielectric fluid containing polychlorinated biphenyls (PCB) and chlorinated benzene, used chlorinated solvents, and mixtures thereof.

Other preferred second feedstocks comprise fractionation bottoms from the purification column in epichlorohydrin production, carbon tetrachloride, 1,1,1-

trichloroethane, chlorinated alcohols, chlorinated ethers, chlorofluorocarbons, ethylene dibromide and admixtures thereof. The second feedstock preferably contain a halogen selected from the group consisting of chlorine, 5 fluorine and bromine.

The second feedstock is introduced in admixture with a hydrogen-rich gaseous stream into a second catalytic hydrogenation zone containing hydrogenation catalyst and maintained at hydrogenation conditions. This 10 second catalytic hydrogenation zone may contain a fixed, ebullated or fluidized catalyst bed. The operating conditions selected for this catalytic hydrogenation zone are selected primarily to dehalogenate the halogenated organic compounds which are introduced thereto. This 15 catalytic hydrogenation zone is preferably maintained under an imposed pressure from atmospheric to 2000 psig (103 to 13893 kPa) and more preferably under a pressure from 100 psig to 1800 psig (793 to 12514 kPa). Suitably, such reaction is conducted with a maximum catalyst bed 20 temperature in the range of 122 to 850°F (50 to 454°C) selected to perform the desired hydrogenation and dehalogenation conversion to reduce or eliminate the concentration of halogenated organic compounds contained in the second feedstock and to perform the desired 25 hydrogenation conversion includes, for example, dehalogenation, desulfurization, denitrification, olefin saturation, oxygenate conversion and hydrocracking. Further preferred operating conditions include LHSV in the range from 0.05 hr.⁻¹ to 20 hr.⁻¹ and hydrogen circulation 30 rates from 200 standard cubic feet per barrel (SCFB) (33.71 normal m³/m³) to 100,000 SCFB (16851 normal m³/m³), preferably from 200 SCFB (33.71 normal m³/m³) to 50,000 SCFB (8427 normal m³/m³). When the second feedstock demonstrates thermal instability characteristics, it is 35 preferred that the conversion temperatures be increased in stages to prevent decomposition of the feedstock on heat-

exchange surfaces and catalyst by means of using two or more catalyst zones with interstage heating, for example.

In a preferred embodiment of the present invention, at least a portion of the hydrogen-rich gaseous stream which is introduced into the second hydrogenation zone is provided via a recycle stream which is recovered from the first hydrogenation zone.

In the event that the temperature of the second feedstock is not deemed to be exactly the temperature selected to operate the second catalytic hydrogenation zone, the temperature may be adjusted either upward or downward by either indirect heat exchange or by the addition of either cool or hot hydrogen.

The hydrogen-rich gaseous stream which is ultimately recovered from the effluent of the second hydrogenation zone in one embodiment of the present invention is recycled to the hot flash zone as described hereinabove.

Either of the hydrogenation zones utilized in the present invention may contain one or more catalyst beds or stages. The preferred catalytic composites disposed within the second hydrogenation zone can be selected from the preferred catalytic composites which are preferably used in the first hydrogenation zone.

The hydrocarbonaceous effluent from the second hydrogenation zone is preferably contacted with an aqueous scrubbing solution and the admixture is admitted to a separation zone in order to separate a halide-rich aqueous stream, a hydrogenated hydrocarbonaceous liquid phase and a hydrogen-rich gaseous phase which contains trace quantities of halogenated organic compounds. The contact of the effluent from the second hydrogenation zone with the aqueous scrubbing solution may be performed in any convenient manner and is preferably conducted by co-current, in-line mixing which may be promoted by inherent turbulence, mixing orifices or any other suitable

mixing means. The aqueous scrubbing solution is preferably introduced in an amount from about 1 to about 100 vol.% of the total feedstock charged to the hydrogenation zone based on the quantity of hydrogen halide compounds present in the effluent from the hydrogenation zone. The aqueous scrubbing solution is selected depending on the characteristics of the organic feed stream introduced into the second hydrogenation zone. In accordance with the present invention, at least some halogenated organic compounds are introduced as feedstock and therefore the aqueous scrubbing solution in one embodiment preferably contains a basic compound such as calcium hydroxide, potassium hydroxide or sodium hydroxide in order to neutralize the acid such as hydrogen chloride, hydrogen bromide and hydrogen fluoride, for example, which is formed during the hydrogenation of the halogenated organic compounds. In another preferred embodiment, the halide component is recovered by dissolution in water or a lean aqueous solution of the halide compound. This embodiment permits the subsequent recovery and use of a desirable and valuable halide compound. The final selection of the aqueous scrubbing solution is dependent upon the particular halide compounds which are present and the desired end product. The resulting hydrogenated hydrocarbonaceous liquid phase is recovered and at least a portion of the hydrogen-rich gaseous phase is heated and recycled to the flash zone and subsequently to the first hydrogenation zone.

The separation zone that follows the second hydrogenation zone is preferably maintained at essentially the same pressure as the second hydrogenation zone and as a consequence the liquid phase contains dissolved hydrogen and low molecular weight normally gaseous hydrocarbons. It is preferred that the liquid phase be stabilized in a convenient manner, such as, for example, by stripping or flashing to remove the normally gaseous components to

provide a stable hydrogenated distillable hydrocarbonaceous product. In some cases, a significant portion of the hydrogenated hydrocarbonaceous product may comprise methane, ethane, propane, butane, hexane and admixtures thereof and an adsorbent/stripper arrangement may conveniently be used to recover methane and ethane.

In the drawing, the process of the present invention is illustrated by means of a simplified flow diagram in which some details have been deleted as being non-essential to an understanding of the techniques involved.

With reference now to the drawing, a first feedstock is introduced into the process via conduit 1 and is contacted with a hot gaseous hydrogen-rich recycle stream which is provided via conduit 26 and hereinafter described. The mixture of the first feedstock and the hydrogen-rich recycle stream are introduced via conduit 26' and intimately contacted in flash separator 2. A hydrocarbonaceous vapor stream comprising hydrogen is removed from hot hydrogen flash separator 2 via conduit 4 and introduced into first hydrogenation reaction zone 5 without intermediate separation thereof. A heavy non-distillable stream is removed from the bottom of flash separator 2 via conduit 3 and recovered. A hydrogenated hydrocarbonaceous vapor stream is removed from the first hydrogenation reaction zone 5 via conduit 6 and is introduced into hot separator 7. A liquid hydrocarbonaceous stream containing high molecular weight hydrocarbons is removed from hot separator 7 via conduit 8. A gaseous stream containing hydrogen and hydrocarbons having lower molecular weights is removed from hot separator 7 via conduit 9 and are contacted with an aqueous scrubbing solution which is introduced via conduit 10. The resulting admixture of the gaseous effluent from hot separator 7 and the aqueous scrubbing solution is passed via conduit 9 into vapor-liquid separator 11. A

hydrogen-rich gaseous stream is removed from vapor-liquid separator 11 via conduit 14 and at least a portion of this stream is introduced via conduit 14 into guard bed 15. A fuel gas stream is removed from guard bed 15 via conduit 5 16 and recovered. At least a portion of the gaseous stream flowing in conduit 14 is diverted via conduit 17 and introduced into compressor 18 and the resulting compressed gas is transported from compressor 18 via conduit 17. Since hydrogen is lost in the process by 10 means of a portion of the hydrogen being dissolved in the exiting liquid hydrocarbon streams and the hydrogen being consumed during the hydrogenation reactions, it is necessary to supplement the hydrogen-rich gaseous stream with make-up hydrogen from some suitable external source, 15 and make-up hydrogen is introduced via conduit 19. A hydrocarbon stream containing lower molecular weight compounds is removed from vapor-liquid separator 11 via conduit 13 and recovered.

The second feedstock is introduced into the 20 process via conduit 31 and is contacted with the hydrogen-rich gaseous recycle stream which is provided via conduit 17, and the resulting mixture introduced into the first stage of second hydrogenation zone, zone 20, via conduit 31. A hydrocarbon recycle stream is provided via conduit 25 30 and is hereinafter described is also introduced into zone 20 via conduit 30 and conduit 31. A resulting hydrogenated stream is removed from zone 20 via conduit 21, further heated in heat exchanger 32 and introduced into the second stage of the second hydrogenation reaction 30 zone, zone 22. The resulting hydrogenated hydrocarbonaceous stream is removed from zone 22 via conduit 23 and is contacted with an aqueous halide-lean scrubbing solution which is introduced via conduit 24. The resulting admixture of the hydrogenated 35 hydrocarbonaceous effluent and the aqueous scrubbing solution is passed via conduit 23 and introduced into

vapor-liquid separator 25. A hydrogen-rich gaseous stream which may contain small quantities of organic halide compounds is removed from vapor-liquid separator 25 via conduit 26 and passed through heat exchanger 27 to raise the temperature of the flowing stream. The resulting heated flowing stream is continued to be transported via conduit 26 and is subsequently introduced into hot flash separator 2 as described hereinabove. A halide-rich aqueous scrubbing solution is removed from vapor-liquid separator 25 via conduit 28 and recovered. A liquid hydrogenated hydrocarbonaceous stream comprising hydrogen in solution is removed from vapor-liquid separator 25 via conduit 29 and at least a portion of this stream is removed from the process and recovered. Another portion of the liquid hydrogenated hydrocarbonaceous stream which is removed from vapor-liquid separator 25 via conduit 29 is recycled via conduit 30 and conduit 31 to zone 20 as described hereinabove. In the event that the liquid distillable hydrogenated hydrocarbonaceous product stream removed via conduit 29 contains propane, for example, and is therefore not accurately described as normally liquid, the vapor-liquid separator 25 may be necessarily operated at a pressure in the range from 300 psig to 1000 psig (2172 to 6998 kPa).

The process of the present invention is further demonstrated by the following more detailed illustrative embodiment.

ILLUSTRATIVE EMBODIMENT

The first feedstock is a waste lube oil having the characteristics presented in Table 1 and contaminated with 20 ppm by weight of polychlorinated biphenyl (PCB) is charged at a rate of 100 mass units per hour to a hot

hydrogen flash separation zone. The hot hydrogen is introduced into the hot hydrogen flash separation zone at a rate of 31 mass units per hour.

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TABLE 1
WASTE LUBE OIL FEEDSTOCK PROPERTIES

10	Specific Gravity @ 60°F (15°C)	0.8827	
	Vacuum Distillation Boiling Range, (ASTM) D-1160	°F	(°C)
	IBP	338	(170)
	10%	516	(269)
15	20%	628	(331)
	30%	690	(367)
	40%	730	(388)
	50%	750	(399)
	60%	800	(421)
20	70%	831	(444)
	80%	882	(474)
	% Over	80	
	% Bottoms	20	
25	Sulfur, weight percent		0.5
	Polychlorinated Biphenyl Concentration, wppm		20
	Lead, wppm		863
	Zinc, wppm		416
	Cadmium, wppm		1
30	Copper, wppm		21
	Chromium, wppm		5

The waste lube oil is preheated to a temperature of <482°F (<250°C) before introduction into the hot hydrogen flash separation zone which temperature precluded any significant detectable thermal degradation. The waste lube oil is intimately contacted in the hot flash

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separation zone with a hot hydrogen-rich gaseous stream having a temperature upon introduction into the hot hydrogen flash separation zone of $>748^{\circ}\text{F}$ ($>398^{\circ}\text{C}$). In addition, the hot hydrogen flash separation zone is

5 operated at conditions which included a temperature of 788°F (420°C), a pressure of 810 psig (5688 kPa), a hydrogen circulation rate of 18,000 SCFB (3034 normal m^3/m^3) and an average residence time of the vapor stream of 5 seconds.

10 A hydrocarbonaceous vapor stream comprising hydrogen is recovered from hot hydrogen flash separation zone, and is directly introduced without separation into a first hydrogenation zone containing a hydrogenation catalyst comprising alumina, nickel and molybdenum.

15 Properties of C_7^+ fraction entering the reaction zone are presented in Table 2. The hydrogenation reaction is conducted with a catalyst peak temperature of 662°F (350°C), a pressure of 800 psig (5619kPa), a LHSV of 0.5 hr.^{-1} based on hydrocarbon feed and a hydrogen to oil

20 ratio of 20,000 SCFB (3370 normal m^3/m^3). The hydrogenated effluent from the first hydrogenation zone including small quantities of hydrogen chloride is passed into a hot flash zone to produce a liquid hydrocarbonaceous stream and a gaseous stream containing

25 hydrogen, hydrogen chloride, hydrogen sulfide and lower molecular weight hydrocarbons. The resulting gaseous stream is thereafter contacted with an aqueous scrubbing solution containing sodium hydroxide, cooled to about 100°F (38°C), and sent to a vapor-liquid separator wherein

30 a gaseous hydrogen-rich stream is separated from the normally liquid hydrocarbonaceous products and spent aqueous scrubbing solution containing sodium, sulfide and chloride ions. The resulting gaseous hydrogen-rich stream is bifurcated to provide a first stream which is passed

35 through an adsorption zone to remove any trace quantities of organic halide compounds and to provide a fuel gas

stream, and a second stream which is compressed and admixed with a fresh supply of hydrogen in an amount sufficient to maintain the specified second hydrogenation zone conditions.

TABLE 2
 PROPERTIES OF C₇⁺ FRACTION OF REACTION ZONE FEED

	Specific Gravity @ 60°F (15°C)	0.866	
5			
	Vacuum Distillation Boiling Range, (ASTM D-1160)	°F	(°C)
	IBP	225	(107)
	10%	433	(223)
10	20%	538	(280)
	30%	633	(334)
	40%	702	(372)
	50%	741	(394)
	60%	770	(410)
15	70%	801	(427)
	80%	837	(447)
	90%	896	(479)
	95%	943	(506)
	EP	982	(527)
20	% Over	97	
	% Bottoms	3	
	Sulfur, weight percent		0.31
	Polychlorinated Biphenyl Concentration, wppm		22
25			
	Lead, wppm		3.7
	Zinc, wppm		1.5
	Cadmium, wppm		<0.04
	Copper, wppm		0.1
30	Chromium, wppm		0.6

A non-distillable liquid stream is recovered from the bottom of the flash separation zone in an amount of 12 mass units per hour and having the characteristics presented in Table 3.

5

TABLE 3
ANALYSIS OF NON-DISTILLABLE STREAM

	Specific Gravity @60°F (15°C)	0.95
10	Polychlorinated Biphenyl Concentration, wppm	< 2

A halogenated organic second feedstock having the characteristics presented in Table 4 in an amount of 100 mass units per hour is admixed with the second hydrogen stream and the resulting admixture is charged to a second hydrogenation zone containing a palladium on alumina catalyst and operating at hydrogenation conditions which include a maximum temperature of 572°F (300°C), a pressure of 850 psig (5964 kPa) and a hydrogen to feed ratio of about 60,000 SCFB (10,110 normal m³/m³). A recycle hydrocarbon stream containing hydrocarbons recovered from the effluent from the second hydrogenation zone in an amount of 100 mass units per hour is also introduced into the second hydrogenation zone.

25

TABLE 4
SATURATED, HALOGENATED HYDROCARBONACEOUS
FEEDSTOCK PROPERTIES

5	Specific Gravity @60°F (15°C)	1.250
	Distillation, °C	
	IBP	95
	50%	<u>110</u>
		259
10	EP	97
	% Over	3
	% Residue	
	<u>Composition, Weight Percent</u>	
15	Chlorinated Propenes	44.0
	Chlorinated Propanes	34.2
	Chlorinated Alcohols	3.9
	Chlorinated Ethers	10.0
	Chlorinated Hexadiene	0.5
		1.1
20	Chlorinated Hexane	0.2
	Chlorinated Benzene	6.1
	Other	

The resulting effluent from the second
25 hydrogenation zone was neutralized with an aqueous
solution containing potassium hydroxide and was found to
contain 38 mass units of hydrocarbonaceous products having
the characteristics presented in Table 5.

TABLE 5
HYDROCARBONACEOUS PRODUCT STREAM PROPERTIES

<u>Composition, Weight Percent</u>		
5	Ethane	0.3
	Propane	96.6
	Chlorinated Propane	Trace
	Butane	Trace
	Pentane	0.0
10	Hexane and Nonane	<u>3.1</u>
		100.0

The foregoing description, drawing and illustrative embodiment clearly demonstrate the advantages encompassed by the process of the present invention and the benefits to be afforded with the use thereof.

THE EMBODIMENTS OF THE INVENTION IN WHICH AN EXCLUSIVE PROPERTY OR PRIVILEGE IS CLAIMED ARE DEFINED AS FOLLOWS:

1. A process for the simultaneous hydrogenation of a first feedstock comprising hydrocarbonaceous compounds and having a non-distillable component, and a second feedstock comprising halogenated organic compounds which process comprises:

(a) contacting said first feedstock with a first hydrogen-rich gaseous stream having a temperature greater than said first feedstock in a flash zone at flash conditions thereby increasing the temperature of said first feedstock and vaporizing at least a portion thereof to provide a hydrocarbonaceous vapor stream comprising hydrogen, and a heavy product comprising said non-distillable component;

(b) contacting said hydrocarbonaceous vapor stream comprising hydrogen with a hydrogenation catalyst in a first hydrogenation reaction zone at hydrogenation conditions effective to increase the hydrogen content thereof;

(c) condensing at least a portion of the resulting effluent from said first hydrogenation reaction zone to produce a second hydrogen-rich gaseous stream and a first liquid hydrogenated stream comprising hydrogenated distillable hydrocarbonaceous compounds;

(d) reacting said second feedstock and at least a portion of said second hydrogen-rich gaseous stream with a hydrogenation catalyst in a second hydrogenation reaction zone at hydrogenation conditions selected to produce an effluent

containing hydrocarbonaceous compounds and at least one water-soluble inorganic halide compound;

(e) contacting the resulting effluent from said second hydrogenation zone with a halide-lean aqueous scrubbing solution;

5 (f) introducing a resulting admixture of said effluent and said halide-lean aqueous scrubbing solution into a separation zone to provide a third hydrogen-rich gaseous stream, a second liquid hydrogenated stream comprising hydrocarbonaceous compounds and a halide-rich aqueous scrubbing solution containing at least
10 a portion of said water-soluble inorganic halide compound;

(g) recycling and heating at least a portion of said third hydrogen-rich gaseous stream recovered in step (f) to step (a) as at least a portion of said first hydrogen-rich gaseous stream; and

15 (h) recovering said first liquid hydrogenated stream from step (c) and said second liquid hydrogenated stream from step (f).

2. The process of Claim 1 wherein said first feedstock comprises dielectric fluids, hydraulic fluids, heat transfer
20 fluids, used lubricating oil, used cutting oils, used solvents, still bottoms from solvent recycle operations, coal tars, atmospheric residuum, PCB-contaminated oils, halogenated wastes or other hydrocarbonaceous industrial waste and wherein said non-distillable component comprises organometallic compounds,
25 inorganic metallic compounds, finely divided particulate matter or non-distillable hydrocarbonaceous compounds.

3. The process of Claim 1 or 2 wherein said first feedstock is introduced into said flash zone at a temperature less than 250°C and wherein the temperature of said first hydrogen-rich stream is from 93°C to 649°C.

5 4. The process of Claim 1, 2 or 3 wherein at least a portion of the resulting effluent from said first hydrogenation zone is contacted with an aqueous scrubbing solution.

10 5. The process of Claim 1 wherein said second feedstock comprises a component selected from the group consisting of fractionation column bottoms in the production of allyl chloride, fractionation column bottoms in the production of ethylene dichloride, fractionation column bottoms in the production of trichloroethylene and perchloroethylene, used dielectric fluid containing polychlorinated biphenyls (PCB) and
15 chlorinated benzene, used chlorinated solvents, fractionation bottoms from the purification column in epichlorohydrin production, carbon tetrachloride, 1,1,1 trichloroethane, chlorinated alcohols, chlorinated ethers, chlorofluorocarbons, ethylene dibromide and mixtures thereof.

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