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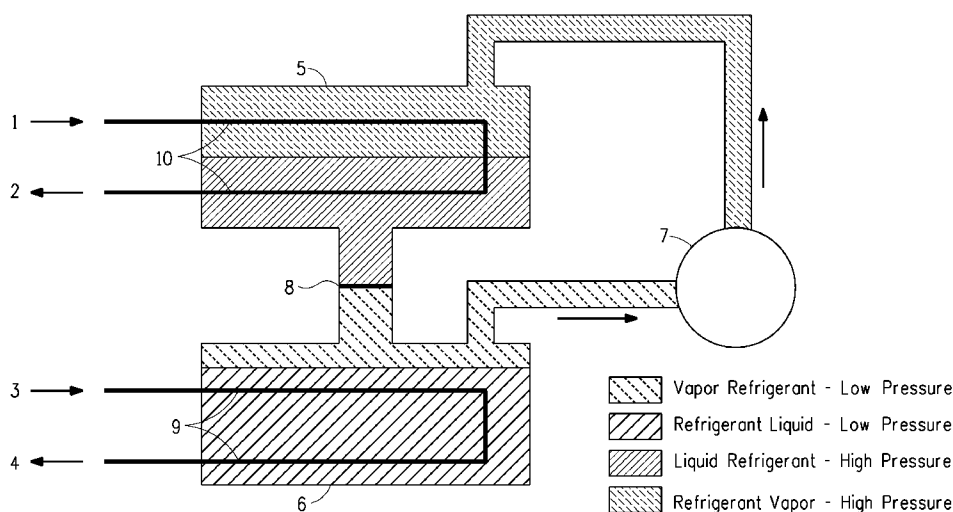


FIG. 2

(57) Abstract: Disclosed are compositions comprising iodotrifluoromethane and other compounds that are useful as heat transfer fluids, including refrigerants, in a stationary air conditioning or refrigeration system, a flooded evaporator chiller or a direct expansion chiller. Also disclosed are methods for producing cooling and methods for replacing R22, R407C, R410A, R404A or R507A in such equipment.

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TITLE OF INVENTION**COMPOSITIONS COMPRISING IODOTRIFLUOROMETHANE
AND USES THEREOF**

5

BACKGROUND OF THE INVENTION

1. Field of the Invention.

The present disclosure relates to the field of low GWP refrigerant compositions comprising iodotrifluoromethane. These compositions are useful as low GWP replacements in equipment designed for R22, R407C,
10 R410A, R404A or R507A, including stationary air conditioning or refrigeration systems, flooded evaporator chillers or direct expansion chillers.

2. Description of Related Art.

Working fluids for various applications are being sought that have
15 little if any environmental impact. The hydrofluorocarbon working fluids adopted as replacements for chlorofluorocarbons, have no ozone depletion potential, but have been found to contribute to global warming.

Therefore, replacements are sought for the hydrofluorocarbons currently in use as refrigerants, heat transfer fluids, cleaning solvents,
20 aerosol propellants, foam blowing agents and fire extinguishing or suppression agents.

In order to serve as drop-in replacements in existing equipment, replacements must be close to or match the properties of the original working fluid for which the equipment was designed. It would be desirable
25 to identify compositions that provide a balance of properties that will allow replacement of existing refrigerants and also to serve as refrigerants in new equipment designed for similar applications.

SUMMARY OF THE INVENTION

30

The present invention provides for particular iodotrifluoromethane compositions, and in particular, refrigerants for replacing R22, R407C, R410A, R404A or R507A, which particular iodotrifluoromethane compositions have a low global warming potential (GWP) and similar

energy efficiency and refrigeration capacity to the refrigerant being replaced. In addition, the present invention provides for refrigerants having low or a specified amount of glide for heat transfer systems with heat exchangers (i.e., evaporators or condensers) that are optimized to
5 take advantage of glide.

In particular, the compositions disclosed herein may be useful for replacing R22, R404A, R407C, R410A, or R507A as a working fluid in a stationary air conditioning or a stationary refrigeration system, or for replacing R22, R407C or R410A in a flooded evaporator chiller or a direct
10 expansion (DX) chiller. Compositions as disclosed herein may be useful in new or existing equipment.

According to the present invention, there is provided a composition comprising iodotrifluoromethane, difluoromethane and an additional compound selected from the group consisting of pentane, butane,
15 isobutane, propylene, cyclopropylene and propane.

In a particular embodiment, the additional compound is propane. In this embodiment, the composition may further comprise pentafluoroethane.

Further in accordance with the present invention, there is provided
20 a composition consisting essentially of either iodotrifluoromethane and 1,1,1,2-tetrafluoroethane; or iodotrifluoromethane, 1,1,1,2-tetrafluoroethane and difluoromethane; or iodotrifluoromethane and pentafluoroethane; or iodotrifluoromethane, difluoromethane, pentafluoroethane and 1,1,1,2-tetrafluoroethane.

25 Further in accordance with the present invention, there is provided a composition consisting essentially of about 0.01 to about 67.25 weight difluoromethane and about 32.75 to about 99.99 % iodotrifluoromethane.

Also in accordance with the present invention, there is provided a method for replacing R22, R407C, R-410A, R404A or R507A in a flooded
30 evaporator chiller, a direct expansion chiller or a stationary air conditioning system or a stationary refrigeration system. The method comprises providing any of the above compositions to a stationary air conditioning system or a stationary refrigeration system, wherein the composition is a

refrigerant. According to this method, R22, R404A, R407C, R410A or R507A are replaced by any of the above compositions in a stationary air conditioning or a stationary refrigeration system, and R22, R407C or R410A are replaced by any of the above compositions in a flooded
5 evaporator chiller or a direct expansion chiller.

Further in accordance with the present invention, there is provided a method for producing cooling in either a stationary air conditioning system or a stationary refrigeration system, a flooded evaporator or a direct expansion chiller. The method comprises evaporating any of the
10 compositions above in an evaporator, wherein the composition is a refrigerant, to form a vapor refrigerant, condensing the refrigerant vapor to form a refrigerant liquid, and returning the refrigerant liquid to the evaporator.

In one embodiment, the method for producing cooling comprises
15 producing cooling in a stationary refrigeration or stationary air conditioning system

According to another embodiment, the method of producing cooling comprises producing cooling in a flooded evaporator chiller. In this method, a first cooling medium is circulated through an inlet in the
20 evaporator, through a coil in the evaporator and to an outlet of the evaporator, thereby lowering the temperature of the first cooling medium as it passes from the inlet to the outlet of the evaporator. The first cooling medium is then passed to a body to be cooled, thereby producing cooling. This embodiment of the method of the present invention may be practiced
25 in a flooded evaporator chiller.

According to yet another embodiment, the method of producing cooling comprises producing cooling in a direct expansion chiller. In this method, the step of evaporating the refrigerant composition comprises circulating the refrigerant composition through an inlet in the
30 evaporator, through a coil in the evaporator, and through an outlet in the evaporator, thereby lowering the temperature of a first liquid cooling medium contained in the evaporator. The first liquid cooling medium is then passed out of the evaporator to a body to be cooled, thereby

producing cooling. This embodiment of the method of the present invention may be practiced in a direct expansion chiller.

BRIEF DESCRIPTION OF THE DRAWINGS

5 FIG.1 is a schematic diagram of a stationary air conditioning or a stationary refrigeration system which utilizes the refrigerant compositions of the present invention.

 FIG. 2 is a schematic diagram of a flooded evaporator chiller which utilizes the refrigerant compositions of the present invention.

10 FIG. 3 is a schematic diagram of a direct expansion chiller which utilizes the refrigerant compositions of the present invention.

DETAILED DESCRIPTION OF THE INVENTION

 Before addressing details of embodiments described below, some
15 terms are defined or clarified.

 Global warming potential (GWP) is an index for estimating relative global warming contribution due to atmospheric emission of a kilogram of a particular greenhouse gas compared to emission of a kilogram of carbon dioxide. GWP can be calculated for different time horizons showing the
20 effect of atmospheric lifetime for a given gas. The GWP for the 100 year time horizon is commonly the value referenced. For mixtures, a weighted average can be calculated based on the individual GWPs for each component.

 Refrigeration capacity (sometimes referred to as cooling capacity)
25 is a term to define the change in enthalpy of a refrigerant in an evaporator per pound of refrigerant circulated, i.e., the heat removed by the refrigerant in the evaporator per a given time. The refrigeration capacity is a measure of the ability of a refrigerant or heat transfer composition to produce cooling. Therefore, the higher the capacity the greater the
30 cooling that may be produced.

 Coefficient of performance (COP) is the amount of heat removed divided by the required energy input to operate the cycle. The higher the COP, the higher the energy efficiency. COP is directly related to the

energy efficiency ratio (EER) that is the efficiency rating for refrigeration or air conditioning equipment at a specific set of internal and external temperatures.

5 Glide (also sometimes referred to as temperature glide) is defined as the absolute value of the difference between the starting and ending temperatures of a phase change process by a refrigerant within a component of a refrigerating or air conditioning system exclusive of any subcooling or superheating. This term usually describes condensation or evaporation of a zeotrope. Specifically, refrigerant glide in a condenser is
10 the difference between its dew point and bubble point temperatures at the condensing pressure, while in an evaporator, it is the difference between the inlet temperature and the saturated vapor temperature at the evaporating pressure. Pure compound refrigerants have zero glide as do azeotrope compositions at specific temperatures and pressures. Near-
15 azeotrope (sometimes referred to as azeotrope-like) compositions that behave similarly to azeotropes, will have low glide. Compositions that are non-azeotropes (or zeotropes) may have significantly higher glide. Average glide is meant to mean the average of glide in the evaporator and glide in the condenser.

20 The term "subcooling" refers to the reduction of the temperature of a vapor below that vapor's saturation point for a given pressure. The saturation point is the temperature at which the vapor usually would condense to a liquid, but subcooling produces a lower temperature vapor at the given pressure. By cooling a vapor below the saturation point, the
25 net refrigeration capacity can be increased. Subcooling thereby improves refrigeration capacity and energy efficiency of a system, such as vapor compression heat transfer systems (meaning refrigeration or air conditioning systems). Superheat is a term defines how far above its saturation vapor temperature a vapor composition is heated.

30 As used herein, a non-azeotropic composition comprises one that is not azeotropic and also not near-azeotropic, meaning that it behaves as a simple mixture of components and thus will fractionate during evaporation or boiling off. During leakage from a heat transfer system this fractionation

will cause the lower boiling (higher vapor pressure) component to leak out of the apparatus first. Thus, the vapor pressure of the heat transfer composition remaining inside the heat transfer system will be reduced.

This drop in pressure can be measured and used as an early indication of a leak.

As used herein, an azeotropic composition comprises a constant-boiling mixture of two or more substances that behave as a single substance. One way to characterize an azeotropic composition is that the vapor produced by partial evaporation or distillation of the liquid has the same composition as the liquid from which it is evaporated or distilled, i.e., the mixture distills/refluxes without compositional change. Constant-boiling compositions are characterized as azeotropic because they exhibit either a maximum or minimum boiling point, as compared with that of the non-azeotropic mixture of the same compounds. An azeotropic composition will not fractionate within a heat transfer system during operation, which may reduce efficiency of the system. Additionally, an azeotropic composition will not fractionate upon leakage from a heat transfer system.

As used herein, a near-azeotropic composition (also commonly referred to as an "azeotrope-like composition") comprises a substantially constant boiling liquid admixture of two or more substances that behaves essentially as a single substance. One way to characterize a near-azeotropic composition is that the vapor produced by partial evaporation or distillation of the liquid has substantially the same composition as the liquid from which it was evaporated or distilled, that is, the admixture distills/refluxes without substantial composition change. Another way to characterize a near-azeotropic composition is that the bubble point vapor pressure and the dew point vapor pressure of the composition at a particular temperature are substantially the same. Herein, a composition is near-azeotropic if, after 50 weight percent of the composition is removed, such as by evaporation or boiling off, the difference in vapor pressure between the original composition and the composition remaining

after 50 weight percent of the original composition has been removed is less than about 10 percent.

As used herein, a heat transfer system may be any refrigeration system, refrigerator, air conditioning system, air conditioner, heat pump, flooded evaporator chiller, direct expansion chiller and the like utilizing a heat transfer composition.

As used herein, a heat transfer composition comprises a composition used to carry heat from a heat source to a heat sink.

As used herein, a refrigerant comprises a compound or mixture of compounds that function as a heat transfer composition in a cycle wherein the composition undergoes a phase change from a liquid to a gas and back.

As used herein, the terms "comprises," "comprising," "includes," "including," "has," "having" or any other variation thereof, are intended to cover a non-exclusive inclusion. For example, a process, method, article, or apparatus that comprises a list of elements is not necessarily limited to only those elements but may include other elements not expressly listed or inherent to such process, method, article, or apparatus. Further, unless expressly stated to the contrary, "or" refers to an inclusive or and not to an exclusive or. For example, a condition A or B is satisfied by any one of the following: A is true (or present) and B is false (or not present), A is false (or not present) and B is true (or present), and both A and B are true (or present).

Also, use of "a" or "an" are employed to describe elements and components described herein. This is done merely for convenience and to give a general sense of the scope of the invention. This description should be read to include one or at least one and the singular also includes the plural unless it is obvious that it is meant otherwise. Group numbers corresponding to columns within the Periodic Table of the elements use the "New Notation" convention as seen in the *CRC Handbook of Chemistry and Physics*, 81st Edition (2000-2001).

Unless otherwise defined, all technical and scientific terms used herein have the same meaning as commonly understood by one of

ordinary skill in the art to which this invention belongs. Although methods and materials similar or equivalent to those described herein can be used in the practice or testing of embodiments of the present invention, suitable methods and materials are described below. All publications, patent applications, patents, and other references mentioned herein are incorporated by reference in their entirety, unless a particular passage is cited. In case of conflict, the present specification, including definitions, will control. In addition, the materials, methods, and examples are illustrative only and not intended to be limiting.

10

Compositions

The present disclosure relates generally to compositions comprising iodotrifluoromethane (CF₃I). Iodotrifluoromethane is commercially available from various sources or may be prepared by methods known in the art.

15

According to one embodiment, the composition of the present invention comprises iodotrifluoromethane (CF₃I), difluoromethane (R-32) and at least one hydrocarbon. The compositions of this embodiment will be hereinafter referred to as the compositions of Group A. The additional hydrocarbons are shown in Table 1.

20

TABLE 1

Code	Structure	Name	Other designation
	CH ₃ CH ₂ CH ₂ CH ₂ CH ₃	n-pentane	
	CH ₃ CH ₂ CH ₂ CH ₃	n-butane	
i-butane	CH ₃ CH(CH ₃)CH ₃	Isobutane	
	CH ₃ CH(CH ₃)CH ₂ CH ₃	2-methylbutane	
HC-1270	CH ₃ CH=CH ₂	Propylene	R1270
HC-C270	cyclo-CH ₂ CH ₂ CH ₂ -	Cyclopropane	RC270
HC-290	CH ₃ CH ₂ CH ₃	Propane	R290

The hydrocarbons of Table 1 may be prepared by methods known in the art or are commercially available.

According to one particular embodiment, the additional hydrocarbon may comprise in particular propane. In this embodiment, the composition
5 of the present invention may additionally comprise pentafluoroethane.

Alternatively, the compositions of the present invention may consist essentially of iodotrifluoromethane (CF_3I), and at least one hydrofluorocarbon. These hydrofluorocarbons are shown in Table 2. The compositions according to this embodiment may include one or more of
10 the hydrofluorocarbons listed in Table 2.

TABLE 2

Code	Structure	Name	Other designation
HFC-134a	$\text{CF}_3\text{CH}_2\text{F}$	1,1,1,2-tetrafluoroethane	R134a
HFC-125	CF_3CHF_2	pentafluoroethane	R125
HFC-32	CH_2F_2	Difluoromethane	R32

In one embodiment, the composition of the present invention consists essentially of iodotrifluoromethane and 1,1,1,2-tetrafluoroethane.

15 In another embodiment, the composition of the present invention consists essentially of iodotrifluoromethane, 1,1,1,2-tetrafluoroethane and difluoromethane.

In another embodiment, the composition of the present invention consists essentially of iodotrifluoromethane, 1,1,1,2-tetrafluoroethane, difluoromethane and pentafluoroethane.
20

In another embodiment, the composition of the present invention consists essentially of iodotrifluoromethane and pentafluoroethane.

Any of the compositions listed in the column titled "Composition" in Table 3 are hereinafter referred to as the compositions of Group B.

25 Specific weight percent ranges for the compositions of Group B of the present invention are given in Table 3. It is within the scope of the present

invention to include those ranges which are included within any of the ranges given below.

TABLE 3

Composition	Range (wt%)	Alternate Range A (wt%)	Alternate Range B (wt%)
CF ₃ I/32	32.75-99.9/0.01-67.25	40-60/60-40	60-99.9/40-0.1
CF ₃ I/R32/R125/R290	30-40/0.1-60/1-10/1-10	30-40/0.1-55/1-10/1-6	30-40/0.1-35/1-10/1-3
CF ₃ I/R134a	30-60/70-40	30-50/70-50	
CF ₃ I/R134a/R32	5-35/5-40/30-70	5-35/5-40/30-60	5-35/5-40/30-40
CF ₃ I/R32/R125/R134a	25-50/0.1-60/0-50/5-50	25-50/40-60/5-10/5-50	25-50/0.1-40/10-50/5-50

5 The compositions of the present invention may be prepared by any convenient method to combine the desired amounts of the individual components as set forth in Table 3. A preferred method is to weigh the desired component amounts and thereafter combine the components in an appropriate vessel. Agitation may be used, if desired.

10 Further in accordance with the present invention, there is provided a composition consisting essentially of about 0.01 to about 67.25 weight percent difluoromethane and about 32.75 to about 99.99 weight percent iodotrifluoromethane. The composition of this embodiment is hereinafter referred to as the composition of Group C.

15 The present invention provides compositions that have zero or low ozone depletion potential and low global warming potential (GWP). The compositions as disclosed herein will have global warming potentials that are less than many hydrofluorocarbon refrigerants currently in use. One aspect of the present invention is to provide a composition with a global
20 warming potential of less than 1000, and less than 500, and in some

cases less than 350, or even less than 150, or less than 100, or less than 50.

In addition, non-flammability and low GWP are both desirable properties for compositions when used as refrigerants. R32 and propane (R290) are both known to be flammable compounds. In one embodiment, those compositions provided in Alternate range A of Table 2 that contain these flammable compounds are expected to be non-flammable or less flammable. In another embodiment, those compositions provided in Alternate range B of Table 2 are expected to be non-flammable. Overall, compositions as described in Table 2 have lower GWP than the existing refrigerant compounds or blends they are meant to replace (see Examples 1 and 2 herein).

The compositions of the present invention as disclosed herein may be used in combination with a desiccant in a refrigeration, air-conditioning, or heat pump system to aid in removal of moisture. Desiccants may be composed of activated alumina, silica gel, or zeolite-based molecular sieves. Representative molecular sieves include MOLSIV XH-7, XH-6, XH-9 and XH-11 (UOP LLC, Des Plaines, IL). For refrigerants with small molecular size such as HFC-32, XH-11 desiccant is preferred.

The compositions of the present invention may further comprise at least one lubricant selected from the group consisting of polyalkylene glycols, polyol esters, polyvinylethers, mineral oils, alkylbenzenes, synthetic paraffins, synthetic naphthenes, and poly(alpha)olefins.

Lubricants for use with the compositions of the present invention comprise those suitable for use with refrigeration or air-conditioning apparatus. Among these lubricants are those conventionally used in vapor compression refrigeration apparatus utilizing chlorofluorocarbon refrigerants. Such lubricants and their properties are discussed in the 1990 ASHRAE Handbook, Refrigeration Systems and Applications, chapter 8, titled "Lubricants in Refrigeration Systems", pages 8.1 through 8.21, herein incorporated by reference. Lubricants suitable for use with compositions of the present invention may comprise those commonly known as "mineral oils" in the field of compression refrigeration lubrication.

Mineral oils comprise paraffins (i.e., straight-chain and branched-carbon-chain, saturated hydrocarbons), naphthenes (i.e. cyclic paraffins) and aromatics (i.e. unsaturated, cyclic hydrocarbons containing one or more rings characterized by alternating double bonds). Lubricants suitable for use with the compositions of the present invention further comprise those commonly known as "synthetic oils" in the field of compression refrigeration lubrication. Synthetic oils comprise alkylaryls (i.e. linear and branched alkyl alkylbenzenes), synthetic paraffins and naphthenes, and poly(alphaolefins). Representative conventional lubricants suitable for use with compositions of the present invention are the commercially available BVM 100 N (paraffinic mineral oil sold by BVA Oils), naphthenic mineral oil commercially available from Crompton Co. under the trademarks Suniso[®] 3GS and Suniso[®] 5GS, naphthenic mineral oil commercially available from Pennzoil under the trademark Sontex[®] 372LT, naphthenic mineral oil commercially available from Calumet Lubricants under the trademark Calumet[®] RO-30, linear alkylbenzenes commercially available from Shrieve Chemicals under the trademarks Zerol[®] 75, Zerol[®] 150 and Zerol[®] 500, and HAB 22 (branched alkylbenzene sold by Nippon Oil).

Lubricants suitable for use with the compositions of the present invention may further comprise those which have been designed for use with hydrofluorocarbon refrigerants and are miscible with refrigerants of the present invention under compression refrigeration and air-conditioning apparatus' operating conditions. Such lubricants and their properties are discussed in "Synthetic Lubricants and High-Performance Fluids", R. L. Shubkin, editor, Marcel Dekker, 1993. Such lubricants include, but are not limited to, polyol esters (POEs) such as Castrol[®] 100 (Castrol, United Kingdom), polyalkylene glycols (PAGs) such as RL-488A from Dow (Dow Chemical, Midland, Michigan), polyvinyl ethers (PVEs), and polycarbonates (PCs).

Lubricants used with the compositions of the present invention are selected by considering a given compressor's requirements and the environment to which the lubricant will be exposed.

The compositions of the present invention described herein containing hydrocarbons may provide improved miscibility with conventional refrigeration lubricants, such as mineral oil. Thus, use of these hydrocarbon-containing compositions for retrofit of existing
5 equipment would not require the costly and time consuming lubricant change out process.

The compositions of the present invention may further comprise an additive selected from the group consisting of compatibilizers, UV dyes, solubilizing agents, tracers, stabilizers, perfluoropolyethers (PFPE), and
10 functionalized perfluoropolyethers.

The compositions of the present invention may further comprise about 0.01 weight percent to about 5 weight percent of a stabilizer, free radical scavenger or antioxidant. Such other additives include but are not limited to, nitromethane, hindered phenols, hydroxylamines, thiols,
15 phosphites, or lactones. Single additives or combinations may be used.

Optionally, certain refrigeration or air-conditioning system additives may be added, as desired, to compositions of the present invention in order to enhance performance and system stability. These additives are known in the field of refrigeration and air-conditioning, and include, but are
20 not limited to, anti wear agents, extreme pressure lubricants, corrosion and oxidation inhibitors, metal surface deactivators, free radical scavengers, and foam control agents. In general, these additives may be present in the inventive compositions in small amounts relative to the overall composition. Typically concentrations of from less than about 0.1
25 weight percent to as much as about 3 weight percent of each additive are used. These additives are selected on the basis of the individual system requirements. These additives include members of the triaryl phosphate family of EP (extreme pressure) lubricity additives, such as butylated triphenyl phosphates (BTPP), or other alkylated triaryl phosphate esters,
30 e.g. Syn-0-Ad 8478 from Akzo Chemicals, tricresyl phosphates and related compounds. Additionally, the metal dialkyl dithiophosphates (e.g., zinc dialkyl dithiophosphate (or ZDDP), Lubrizol 1375 and other members of this family of chemicals may be used in compositions of the present

invention. Other antiwear additives include natural product oils and asymmetrical polyhydroxyl lubrication additives, such as Synergol TMS (International Lubricants). Similarly, stabilizers such as antioxidants, free radical scavengers, and water scavengers may be employed.

- 5 Compounds in this category can include, but are not limited to, butylated hydroxytoluene (BHT), epoxides, and mixtures thereof. Corrosion inhibitors include dodeceyl succinic acid (DDSA), amine phosphate (AP), oleoyl sarcosine, imidazone derivatives and substituted sulfphonates. Metal surface deactivators include areoxalyl bis (benzylidene) hydrazide
- 10 (CAS reg. no. 6629-10-3), N,N'-bis(3,5-di-tert-butyl-4-hydroxyhydrocinnamoylhydrazine (CAS reg no. 32687-78-8) , 2,2,' -oxamidobis-ethyl-(3,5-di-tert-butyl-4-hydroxyhydrocinnamate (CAS reg no. 70331-94-1), N,N'-(disalicyclidene)-1,2-diaminopropane (CAS reg no. 94-91-7) and ethylenediaminetetra-acetic acid (CAS reg no. 60-00-4) and its
- 15 salts, and mixtures thereof.

- Additional additives include stabilizers comprising at least one compound selected from the group consisting of hindered phenols, thiophosphates, butylated triphenylphosphorothionates, organophosphates, or phosphites, aryl alkyl ethers, terpenes, terpenoids,
- 20 epoxides, fluorinated epoxides, oxetanes, ascorbic acid, thiols, lactones, thioethers, amines, nitromethane, alkylsilanes, benzophenone derivatives, aryl sulfides, divinyl terephthalic acid, diphenyl terephthalic acid, ionic liquids, and mixtures thereof. Representative stabilizer compounds include but are not limited to tocopherol; hydroquinone; t-butyl
 - 25 hydroquinone; monothiophosphates; and dithiophosphates, commercially available from Ciba Specialty Chemicals, Basel, Switzerland, hereinafter "Ciba", under the trademark Irgalube[®] 63; dialkylthiophosphate esters, commercially available from Ciba under the trademarks Irgalube[®] 353 and Irgalube[®] 350, respectively; butylated triphenylphosphorothionates,
 - 30 commercially available from Ciba under the trademark Irgalube[®] 232; amine phosphates, commercially available from Ciba under the trademark Irgalube[®] 349 (Ciba); hindered phosphites, commercially available from Ciba as Irgafos[®] 168; a phosphate such as (Tris-(di-tert-butylphenyl),

commercially available from Ciba under the trademark Irgafos[®] OPH; (Di-n-octyl phosphite); and iso-decyl diphenyl phosphite, commercially available from Ciba under the trademark Irgafos[®] DDPP; anisole; 1,4-dimethoxybenzene; 1,4-diethoxybenzene; 1,3,5-trimethoxybenzene; d-
5 limonene; retinal; pinene; menthol; Vitamin A; terpinene; dipentene; lycopene; beta carotene; bornane; 1,2-propylene oxide; 1,2-butylene oxide; n-butyl glycidyl ether; trifluoromethyloxirane; 1,1-bis(trifluoromethyl)oxirane; 3-ethyl-3-hydroxymethyl-oxetane, such as OXT-101 (Toagosei Co., Ltd); 3-ethyl-3-((phenoxy)methyl)-oxetane, such
10 as OXT-211 (Toagosei Co., Ltd); 3-ethyl-3-((2-ethyl-hexyloxy)methyl)-oxetane, such as OXT-212 (Toagosei Co., Ltd); ascorbic acid; methanethiol (methyl mercaptan); ethanethiol (ethyl mercaptan); Coenzyme A; dimercaptosuccinic acid (DMSA); grapefruit mercaptan ((R)-2-(4-methylcyclohex-3-enyl)propane-2-thiol)); cysteine ((R)-2-amino-3-
15 sulfanyl-propanoic acid); lipoamide (1,2-dithiolane-3-pentanamide); 5,7-bis(1,1-dimethylethyl)-3-[2,3(or 3,4)-dimethylphenyl]-2(3H)-benzofuranone, commercially available from Ciba under the trademark Irganox[®] HP-136; benzyl phenyl sulfide; diphenyl sulfide; diisopropylamine; dioctadecyl 3,3'-thiodipropionate, commercially available from Ciba under the trademark
20 Irganox[®] PS 802 (Ciba); didodecyl 3,3'-thiopropionate, commercially available from Ciba under the trademark Irganox[®] PS 800; di-(2,2,6,6-tetramethyl-4-piperidyl)sebacate, commercially available from Ciba under the trademark Tinuvin[®] 770; poly-(N-hydroxyethyl-2,2,6,6-tetramethyl-4-hydroxy-piperidyl succinate, commercially available from Ciba under the
25 trademark Tinuvin[®] 622LD (Ciba); methyl bis tallow amine; bis tallow amine; phenol-alpha-naphthylamine; bis(dimethylamino)methylsilane (DMAMS); tris(trimethylsilyl)silane (TTMSS); vinyltriethoxysilane; vinyltrimethoxysilane; 2,5-difluorobenzophenone; 2',5'-dihydroxyacetophenone; 2-aminobenzophenone; 2-chlorobenzophenone;
30 benzyl phenyl sulfide; diphenyl sulfide; dibenzyl sulfide; ionic liquids; and others as disclosed in International Patent Application No. PCT/US07/07477, filed March 26, 2007.

Ionic liquid stabilizers comprise at least one ionic liquid. Ionic liquids are organic salts that are liquid at room temperature (approximately 25 ° C). In another embodiment, ionic liquid stabilizers comprise salts containing cations selected from the group consisting of pyridinium, pyridazinium, pyrimidinium, pyrazinium, imidazolium, pyrazolium, thiazolium, oxazolium and triazolium; and anions selected from the group consisting of $[\text{BF}_4]^-$, $[\text{PF}_6]^-$, $[\text{SbF}_6]^-$, $[\text{CF}_3\text{SO}_3]^-$, $[\text{HCF}_2\text{CF}_2\text{SO}_3]^-$, $[\text{CF}_3\text{HFCF}_2\text{SO}_3]^-$, $[\text{HCClFCF}_2\text{SO}_3]^-$, $[(\text{CF}_3\text{SO}_2)_2\text{N}]^-$, $[(\text{CF}_3\text{CF}_2\text{SO}_2)_2\text{N}]^-$, $[(\text{CF}_3\text{SO}_2)_3\text{C}]^-$, $[\text{CF}_3\text{CO}_2]^-$, and F^- . Representative ionic liquid stabilizers include emim BF_4 (1-ethyl-3-methylimidazolium tetrafluoroborate); bmim BF_4 (1-butyl-3-methylimidazolium tetraborate); emim PF_6 (1-ethyl-3-methylimidazolium hexafluorophosphate); and bmim PF_6 (1-butyl-3-methylimidazolium hexafluorophosphate), all of which are available from Fluka (Sigma-Aldrich).

The compositions of the present invention may also include as additives perfluoropolyethers which are miscible with hydrofluorocarbon refrigerants or heat transfer fluids. A common characteristic of perfluoropolyethers is the presence of perfluoroalkyl ether moieties. Perfluoropolyether is synonymous to perfluoropolyalkylether. Other synonymous terms frequently used include "PFPE", "PFAE", "PFPE oil", "PFPE fluid", and "PFPPE". For example, Krytox[®], available from DuPont, is a perfluoropolyether having the formula of $\text{CF}_3-(\text{CF}_2)_2-\text{O}-[\text{CF}(\text{CF}_3)-\text{CF}_2-\text{O}]_j-\text{R}'_f$. In the formula, j is 2 - 100, inclusive and R'_f is CF_2CF_3 , a C3 to C6 perfluoroalkyl group, or combinations thereof.

Other PFPE's, including the PFPE fluids sold under the trademarks Fomblin[®] and Galden[®], available from Ausimont, Milan, Italy and produced by perfluoroolefin photooxidation, can also be used. Fomblin[®]-Y can have the formula of $\text{CF}_3\text{O}(\text{CF}_2\text{CF}(\text{CF}_3)-\text{O})_{m'}(\text{CF}_2-\text{O})_{n'}-\text{R}_{1f}$. Also suitable is $\text{CF}_3\text{O}[\text{CF}_2\text{CF}(\text{CF}_3)\text{O}]_{m'}(\text{CF}_2\text{CF}_2\text{O})_{o'}(\text{CF}_2\text{O})_{n'}-\text{R}_{1f}$. In the formulae R_{1f} is CF_3 , C_2F_5 , C_3F_7 , or combinations of two or more thereof; $(m' + n')$ is 8 - 45, inclusive; and m/n is 20 - 1000, inclusive; o' is 1; $(m'+n'+o')$ is 8 - 45, inclusive; m'/n' is 20 - 1000, inclusive.

Fomblin[®]-Z can have the formula of $\text{CF}_3\text{O}(\text{CF}_2\text{CF}_2\text{-O})_p(\text{CF}_2\text{-O})_q\text{CF}_3$ where $(p' + q')$ is 40 - 180 and p'/q' is 0.5 - 2, inclusive.

Another family of PFPE fluids, available from Daikin Industries, Japan, under the trademark Demnum[™], can also be used as additives.

- 5 Such PFPE fluids can be produced by sequential oligomerization and fluorination of 2,2,3,3-tetrafluorooxetane, yielding the formula of $\text{F}-[(\text{CF}_2)_3\text{-O}]_t\text{-R}_{2f}$ where R_{2f} is CF_3 , C_2F_5 , or combinations thereof and t' is 2 - 200, inclusive.

- The two end groups of the perfluoropolyether, independently, can be functionalized or unfunctionalized. In an unfunctionalized perfluoropolyether, the end group can be branched or straight chain perfluoroalkyl radical end groups. Examples of such perfluoropolyethers can have the formula of $\text{C}_{r'}\text{F}_{(2r'+1)}\text{-A-C}_{r'}\text{F}_{(2r'+1)}$ in which each r' is independently 3 to 6; A can be $\text{O}-(\text{CF}(\text{CF}_3)\text{CF}_2\text{-O})_{w'}$, $\text{O}-(\text{CF}_2\text{-O})_{x'}(\text{CF}_2\text{CF}_2\text{-O})_{y'}$, $\text{O}-(\text{C}_2\text{F}_4\text{-O})_{w'}$, $\text{O}-(\text{C}_2\text{F}_4\text{-O})_{x'}(\text{C}_3\text{F}_6\text{-O})_{y'}$, $\text{O}-(\text{CF}(\text{CF}_3)\text{CF}_2\text{-O})_{x'}(\text{CF}_2\text{-O})_{y'}$, $\text{O}-(\text{CF}_2\text{CF}_2\text{CF}_2\text{-O})_{w'}$, $\text{O}-(\text{CF}(\text{CF}_3)\text{CF}_2\text{-O})_{x'}(\text{CF}_2\text{CF}_2\text{-O})_{y'}\text{-(CF}_2\text{-O)}_{z'}$, or combinations of two or more thereof; preferably A is $\text{O}-(\text{CF}(\text{CF}_3)\text{CF}_2\text{-O})_{w'}$, $\text{O}-(\text{C}_2\text{F}_4\text{-O})_{w'}$, $\text{O}-(\text{C}_2\text{F}_4\text{-O})_{x'}(\text{C}_3\text{F}_6\text{-O})_{y'}$, $\text{O}-(\text{CF}_2\text{CF}_2\text{CF}_2\text{-O})_{w'}$, or combinations of two or more thereof; w' is 4 to 100; x' and y' are each independently 1 to 100. Specific examples include, but are not limited to, $\text{F}(\text{CF}(\text{CF}_3)\text{-CF}_2\text{-O})_9\text{-CF}_2\text{CF}_3$, $\text{F}(\text{CF}(\text{CF}_3)\text{-CF}_2\text{-O})_9\text{-CF}(\text{CF}_3)_2$, and combinations thereof. In such PFPE's, up to 30% of the halogen atoms can be halogens other than fluorine, such as, for example, chlorine atoms.

- The two end groups of the perfluoropolyether, independently, can also be functionalized. A typical functionalized end group can be selected from the group consisting of esters, hydroxyls, amines, amides, cyanos, carboxylic acids and sulfonic acids.

- Representative ester end groups include $-\text{COOCH}_3$, $-\text{COOCH}_2\text{CH}_3$, $-\text{CF}_2\text{COOCH}_3$, $-\text{CF}_2\text{COOCH}_2\text{CH}_3$, $-\text{CF}_2\text{CF}_2\text{COOCH}_3$, $-\text{CF}_2\text{CF}_2\text{COOCH}_2\text{CH}_3$, $-\text{CF}_2\text{CH}_2\text{COOCH}_3$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{COOCH}_3$, $-\text{CF}_2\text{CH}_2\text{CH}_2\text{COOCH}_3$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{CH}_2\text{COOCH}_3$.

Representative hydroxyl end groups include $-\text{CF}_2\text{OH}$, $-\text{CF}_2\text{CF}_2\text{OH}$, $-\text{CF}_2\text{CH}_2\text{OH}$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{OH}$, $-\text{CF}_2\text{CH}_2\text{CH}_2\text{OH}$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{CH}_2\text{OH}$.

Representative amine end groups include $-\text{CF}_2\text{NR}^1\text{R}^2$, $-\text{CF}_2\text{CF}_2\text{NR}^1\text{R}^2$, $-\text{CF}_2\text{CH}_2\text{NR}^1\text{R}^2$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{NR}^1\text{R}^2$, $-\text{CF}_2\text{CH}_2\text{CH}_2\text{NR}^1\text{R}^2$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{CH}_2\text{NR}^1\text{R}^2$, wherein R^1 and R^2 are independently H, CH_3 , or CH_2CH_3 .

- 5 Representative amide end groups include $-\text{CF}_2\text{C}(\text{O})\text{NR}^1\text{R}^2$, $-\text{CF}_2\text{CF}_2\text{C}(\text{O})\text{NR}^1\text{R}^2$, $-\text{CF}_2\text{CH}_2\text{C}(\text{O})\text{NR}^1\text{R}^2$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{C}(\text{O})\text{NR}^1\text{R}^2$, $-\text{CF}_2\text{CH}_2\text{CH}_2\text{C}(\text{O})\text{NR}^1\text{R}^2$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{CH}_2\text{C}(\text{O})\text{NR}^1\text{R}^2$, wherein R^1 and R^2 are independently H, CH_3 , or CH_2CH_3 .

- 10 Representative cyano end groups include $-\text{CF}_2\text{CN}$, $-\text{CF}_2\text{CF}_2\text{CN}$, $-\text{CF}_2\text{CH}_2\text{CN}$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{CN}$, $-\text{CF}_2\text{CH}_2\text{CH}_2\text{CN}$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{CH}_2\text{CN}$.

Representative carboxylic acid end groups include $-\text{CF}_2\text{COOH}$, $-\text{CF}_2\text{CF}_2\text{COOH}$, $-\text{CF}_2\text{CH}_2\text{COOH}$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{COOH}$, $-\text{CF}_2\text{CH}_2\text{CH}_2\text{COOH}$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{CH}_2\text{COOH}$.

- 15 Representative sulfonic acid end groups include $-\text{S}(\text{O})(\text{O})\text{OR}^3$, $-\text{S}(\text{O})(\text{O})\text{R}^4$, $-\text{CF}_2\text{O S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{CF}_2\text{O S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{CH}_2\text{O S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{O S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{CH}_2\text{CH}_2\text{O S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{CH}_2\text{O S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{CF}_2\text{S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{CH}_2\text{S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{CH}_2\text{CH}_2\text{S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{CH}_2\text{S}(\text{O})(\text{O})\text{OR}^3$, $-\text{CF}_2\text{O S}(\text{O})(\text{O})\text{R}^4$, $-\text{CF}_2\text{CF}_2\text{O S}(\text{O})(\text{O})\text{R}^4$, $-\text{CF}_2\text{CH}_2\text{O S}(\text{O})(\text{O})\text{R}^4$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{O S}(\text{O})(\text{O})\text{R}^4$, $-\text{CF}_2\text{CH}_2\text{CH}_2\text{O S}(\text{O})(\text{O})\text{R}^4$, $-\text{CF}_2\text{CF}_2\text{CH}_2\text{CH}_2\text{O S}(\text{O})(\text{O})\text{R}^4$, wherein R^3 is H, CH_3 , CH_2CH_3 , CH_2CF_3 , CF_3 , or CF_2CF_3 , R^4 is CH_3 , CH_2CH_3 , CH_2CF_3 , CF_3 , or CF_2CF_3 .
- 20

- In one embodiment, the compositions of the present invention (meaning those of Groups A, B and C) may be used as blowing agents for use in preparing foams. Thus, according to the present invention, there is provided a foam prepared from such blowing agents, and preferably polyurethane and polyisocyanate foams, and a method of preparing such foams. In such foam embodiments, one or more of the compositions of the present invention is included as a blowing agent and is added to a foamable composition, and the foamable composition is reacted under conditions effective to form a foam. Such conditions may include the use of one or more additional components capable of reacting and foaming
- 25
- 30

under the proper conditions to form a foam or cellular structure. Any of the methods well known in the art, such as those described in "Polyurethanes Chemistry and Technology," Volumes I and II, Saunders and Frisch, 1962, John Wiley and Sons, New York, N.Y., may be used or adapted for use in accordance with the foam embodiments of the present invention.

In another embodiment, the compositions of the present invention may be used as propellants in sprayable compositions. In this embodiment, the present invention relates to a sprayable composition comprising the compositions of the present invention. In this embodiment, the sprayable composition may further comprise an active ingredient to be sprayed together with inert ingredients, solvents and other materials. The sprayable composition may be an aerosol. Suitable active ingredients to be sprayed include, without limitations, cosmetic materials, such as deodorants, perfumes, hair sprays, cleaners, and polishing agents as well as medicinal materials such as anti-asthma and anti-halitosis medications.

The present disclosure also provides a process for producing aerosol products comprising the step of adding a composition of the present invention to an active ingredient in an aerosol container, wherein said composition functions as a propellant.

In one embodiment, a sterilant is an azeotrope or azeotrope-like composition comprising ethylene oxide and a composition of Group A, Group B or Group C. In another embodiment, a sterilant mixture is a non-azeotrope (or zeotrope) composition comprising ethylene oxide and a composition of Group A, Group B or Group C.

In another embodiment, the compositions of the present invention may be used as sterilants. The sterilant may be used to sterilize a great many articles, including but not limited to medical equipment and materials, such diagnostic endoscopes, plastic goods such as syringes, gloves, test tubes, incubators and pacemakers; rubber goods such as tubing, catheters and sheeting; instruments such as needles, scalpels and oxygen tests; and other items such as dilators, pumps, motors and intraocular lenses. In another embodiment, the sterilant of this invention may be used as a fumigant for items outside the medical field, including

but not limited to certain food stuffs, such as species, and other items such as furs, bedding, paper goods, and transportation equipment such as the cargo area of airplanes, trains, and ships. The sterilant may be effective against all forms of life, particularly unwanted insects, bacteria, virus,
5 molds, fungi, and other microorganisms.

In one embodiment, a sterilant is an azeotrope or azeotrope-like composition comprising ethylene oxide and a composition of the present invention. In another embodiment, a sterilant is a non-azeotrope (or zeotrope) composition comprising ethylene oxide and a composition of the
10 present invention.

The present invention also provides a method for sterilizing an article which comprises contacting the article with a sterilant comprising ethylene oxide and a composition of the present invention. In one embodiment, the method of sterilizing an article may be accomplished in
15 any manner known in the art, including contacting the article to be sterilized with the sterilant under conditions and for a period of time as to be effective in achieving the desired degree of sterility. In another embodiment, the method is effected by placing the articles to be sterilized in a vessel, evacuating the air from the vessel, humidifying the vessel, and
20 contacting the articles to the sterilant for an effective period of time. In one embodiment the humidifying creates a relative humidity within the vessel of from about 30 to about 80 percent. An effective period of time for sterilizing will depend upon a number of factors including temperature, pressure, relative humidity, the specific sterilant mixture employed and the
25 material being sterilized. Alternatively, some porous articles may require shorter contact times than do articles sealed in polyethylene bags. Further, in another embodiment, certain bacteria are especially resistant and may thus require longer contact times for sterilization.

The compositions of the present invention (meaning the
30 compositions of Groups A, B and C) are particularly useful as refrigerants. The use of such refrigerants in cooling systems and in methods for producing cooling will be described below.

Cooling Systems

Fig. 1 is a schematic illustration of a stationary refrigeration or a stationary air conditioning system which may employ the compositions of the present invention. In this system, liquid refrigerant flows through an expansion valve 212, from which the refrigerant exits as part liquid and part vapor, into an evaporator 214, which has an inlet and an outlet. The liquid refrigerant is vaporized in the evaporator, exiting as a vapor, and enters a suction line 240. The refrigerant vapor is then drawn into a coupled compressor 250, which increases the pressure and temperature of the refrigerant vapor. The compressor compresses this vapor so that it may be condensed at a higher pressure and temperature than the pressure and temperature of the refrigerant vapor when it comes out of the evaporator. The refrigerant vapor flows out of the compressor into a coupled hot gas line 260, and then flows into a condenser 270, whereby the refrigerant vapor is condensed and returned to the liquid phase. A liquid refrigerant line 280 returns the liquid refrigerant to the expansion valve, and the cycle is repeated.

The compositions of the present invention may also be useful in other air conditioning/refrigeration systems, such as small coolers which have less than 5 to 10 kW cooling capacity, or in closed loop heat transfer systems, which re-use refrigerant in multiple steps to produce a cooling effect in one step and a heating effect in a different step. Such systems are typically used in mobile air conditioning systems. As used herein, a mobile air conditioning system refers to any refrigeration or air-conditioning apparatus incorporated into a transportation unit for the road, rail, sea or air.

In another embodiment, the compositions of the present invention may be used as refrigerants in a chiller. A chiller is a type of air conditioning/refrigeration apparatus. Two types of water chillers are available, vapor-compression chillers and absorption chillers. The present disclosure is directed to a vapor compression chiller. Such vapor compression chiller may be either a flooded evaporator chiller, which is shown in Fig. 2, or a direct expansion chiller, which is shown in Fig. 3.

Both a flooded evaporator chiller or a direct expansion chiller may be air-cooled or water-cooled. In the embodiment where chillers are water cooled, such chillers are generally associated with cooling towers for heat rejection from the system. In the embodiment where chillers are air-cooled, the chillers are equipped with refrigerant-to-air finned-tube condenser coils and fans to reject heat from the system. Air-cooled chiller systems are generally less costly than equivalent-capacity water-cooled chiller systems including cooling tower and water pump. However, water-cooled systems can be more efficient under many operating conditions due to lower condensing temperatures.

Chillers, including both flooded evaporator and direct expansion chillers, may be coupled with an air handling and distribution system to provide comfort air conditioning (cooling and dehumidifying the air) to large commercial buildings, including hotels, office buildings, shopping centers, hospitals, universities and the like. In another embodiment, chillers, most likely air-cooled direct expansion chillers, have found additional utility in naval submarines and surface vessels.

To illustrate how chillers operate, reference is made to the Figures. A water-cooled, flooded evaporator chiller is shown illustrated in Fig. 2. In this chiller a warm first liquid cooling medium liquid, (usually water, but may have additives, such as glycol) enters the chiller from a cooling system, such as a building cooling system, shown entering at arrow 3, through a coil 9 in an evaporator 6, which has an inlet and an outlet. The warm first liquid cooling medium is delivered to the evaporator, where it is chilled by liquid refrigerant, which is shown in the lower portion of the evaporator. The liquid refrigerant evaporates at a lower temperature than the warm first liquid cooling medium which flows through coil 9. The chilled first liquid cooling medium re-circulates back to the building cooling system, as shown by arrow 4, via a return portion of coil 9. The liquid refrigerant, shown in the lower portion of evaporator 6 in Fig. 2, vaporizes and is drawn into a compressor 7, which increases the pressure and temperature of the refrigerant vapor. The compressor compresses this vapor so that it may be condensed in a condenser 5 at a higher pressure

and temperature than the pressure and temperature of the refrigerant vapor when it comes out of the evaporator. A second cooling medium, which is a liquid in the case of a water-cooled chiller, enters the condenser via a coil 10 in condenser 5 from a cooling tower at arrow 1 in Fig. 2. The second cooling medium is warmed in the process and returned via a return loop of coil 10 and arrow 2 to a cooling tower. This second cooling medium cools the vapor in the condenser and turns the vapor to liquid refrigerant, so that there is liquid refrigerant in the lower portion of the condenser as shown in Fig. 2. The condensed liquid refrigerant in the condenser flows back to the evaporator through an expansion device or an orifice 8, and the cycle is repeated. Orifice 8 reduces the pressure of the liquid refrigerant, and converts the liquid refrigerant partially to vapor, that is to say that the liquid refrigerant partially changes to vapor (flashes) as pressure drops between the condenser and the evaporator. Flashing cools the refrigerant both the liquid and vapor to the saturated temperature at evaporator pressure, so that both liquid refrigerant and refrigerant vapor are present in the evaporator.

It should be noted that for a single component refrigerant composition, the composition of the vapor refrigerant in the evaporator is the same as the composition of the liquid refrigerant in the evaporator. In this case, evaporation will occur at a constant temperature. However, if a refrigerant blend is used, as in the case of the compositions of the present invention, the liquid refrigerant and the refrigerant vapor in the evaporator (or in the condenser) may have different compositions. Such compositions depend on the properties of the components such as boiling points, structures and ability to form azeotropes, etc.

Chillers with capacities above 700 kW generally employ flooded evaporators, where the refrigerant is contained in the evaporator and the condenser (i.e., on the shell side). Flooded evaporators require higher charges of refrigerant, but permit closer approach temperatures and higher efficiencies. Chillers with capacities below 700 kW commonly employ evaporators with refrigerant flowing inside the tubes and chilled cooling medium in the evaporator and the condenser, i.e., on the shell

side. Such chillers are called direct-expansion (DX) chillers. A water-cooled direct expansion chiller is illustrated in Fig. 3. In the chiller as illustrated in Fig. 3, a first liquid cooling medium, which is a warm liquid, such as warm water, enters an evaporator 6' at inlet 14. Mostly liquid
5 refrigerant enters a coil 9' in the evaporator at arrow 3' and evaporates. As a result, the first liquid cooling medium is cooled in the evaporator, and a cooled first liquid cooling medium exits the evaporator at outlet 16, and is sent to a body to be cooled, such as a building. In this embodiment of Fig. 3, this first liquid cooling medium is the working fluid that produces
10 cooling. The refrigerant vapor exits the evaporator at arrow 4' and is drawn into a compressor 7', where it is compressed and exits as high temperature, high pressure vapor. This refrigerant vapor enters the condenser through a condenser coil at 1'. The refrigerant vapor is cooled by a second liquid cooling medium in the condenser and becomes a liquid.
15 The second liquid cooling medium enters the condenser through a condenser cooling medium inlet 20, and it extracts heat from the condensed refrigerant vapor, which heats the cooling medium. The second liquid cooling medium exits through the condenser cooling medium outlet 18. The condensed refrigerant liquid exits the condenser and flows
20 through an expansion valve 12, which reduces the pressure of the liquid refrigerant. A small amount of refrigerant vapor, produced as a result of the expansion, enters the evaporator with liquid refrigerant, and the cycle is repeated.

Vapor-compression chillers are identified by the type of compressor
25 they employ. In one embodiment, the compositions of the present invention are useful in chillers, which utilize centrifugal compressors, as will be described below. In another embodiment the compositions of the present invention are useful in chillers which utilize positive displacement compressors, i.e., either reciprocating, screw, or scroll compressors.

30 A centrifugal compressor uses rotating elements to accelerate the refrigerant radially, and typically includes an impeller and diffuser housed in a casing. Centrifugal compressors usually take fluid in at an impeller eye, or central inlet of a circulating impeller, and accelerate it radially

outward. Some static pressure rise occurs in the impeller, but most of the pressure rise occurs in the diffuser section of the casing, where velocity is converted to static pressure. Each impeller-diffuser set is a stage of the compressor. Centrifugal compressors are built with from one to twelve or
5 more stages, depending on the final pressure desired and the volume of refrigerant to be handled.

The pressure ratio, or compression ratio, of a compressor is the ratio of absolute discharge pressure to the absolute inlet pressure. Pressure delivered by a centrifugal compressor is practically constant over
10 a relatively wide range of capacities. The pressure a centrifugal compressor can develop depends on the tip speed of the impeller. Tip speed is the speed of the impeller measured at its tip and is related to the diameter of the impeller and its revolutions per minute. The capacity of the centrifugal compressor is determined by the size of the passages
15 through the impeller. This makes the size of the compressor more dependent on the pressure required than the capacity.

Positive displacement compressors draw vapor into a chamber, and the chamber decreases in volume to compress the vapor. After being compressed, the vapor is forced from the chamber by further decreasing
20 the volume of the chamber to zero or nearly zero.

Reciprocating compressors use pistons driven by a crankshaft. They can be either stationary or portable, can be single or multi-staged, and can be driven by electric motors or internal combustion engines. Small reciprocating compressors from 5 to 30 hp are seen in automotive
25 applications and are typically for intermittent duty. Larger reciprocating compressors up to 100 hp are found in large industrial applications. Discharge pressures can range from low pressure to very high pressure (>5000 psi or 35 MPa).

Screw compressors use two meshed rotating positive-displacement
30 helical screws to force the gas into a smaller space. Screw compressors are usually for continuous operation in commercial and industrial application and may be either stationary or portable. Their application can

be from 5 hp (3.7 kW) to over 500 hp (375 kW) and from low pressure to very high pressure (>1200 psi or 8.3 MPa).

Scroll compressors are similar to screw compressors and include two interleaved spiral-shaped scrolls to compress the gas. The output is
5 more pulsed than that of a rotary screw compressor.

For chillers which use scroll compressors or reciprocating compressors, capacities below 150 kW, brazed-plate heat exchangers are commonly used for evaporators instead of the shell-and-tube heat exchangers employed in larger chillers. Brazed-plate heat exchangers
10 reduce system volume and refrigerant charge.

Methods

According to another aspect of the present invention, the compositions of the present invention (meaning the compositions of
15 Groups A, B and C) are useful in methods to produce cooling. In these methods, the compositions of the present invention are refrigerants.

In all of the embodiments as illustrated with respect to Figs. 1 – 3, there is provided a method for producing cooling, comprising evaporating the composition of any of Groups A, B or C in an evaporator, wherein the
20 composition is a refrigerant, to form a vapor refrigerant, condensing the vapor refrigerant to form a refrigerant liquid, and returning the refrigerant liquid to the evaporator. In all of the embodiments, the refrigerant vapor may be compressed in order to increase its temperature and pressure before it is condensed.

25 In one embodiment, the method for producing cooling comprises producing cooling in a stationary refrigeration or stationary air conditioning system as described with respect to Fig. 1 above. This method comprises the steps of evaporating a refrigerant composition of the present invention in an evaporator to form a refrigerant vapor, thereby producing cooling.
30 The refrigerant vapor exits the evaporator and is sent to a compressor, which increases the pressure and temperature of the refrigerant vapor. The refrigerant composition is thereafter condensed to a liquid refrigerant and returned to the evaporator.

In another embodiment, the method for producing cooling comprises producing cooling in a flooded evaporator chiller as described above with respect to Fig. 2. In this method, a refrigerant composition of the present invention is evaporated in an evaporator to form a refrigerant vapor. A first cooling medium, which is a warm liquid, is brought to the evaporator from a cooling system, and is circulated through an inlet of the evaporator, through a coil in the evaporator, and to an outlet of the evaporator, thereby lowering the temperature of the cooling medium as it passes from the inlet to the outlet of the evaporator. The warm first cooling medium is cooled in the evaporator and is passed to a body to be cooled, such as a building, thereby producing cooling. The refrigerant vapor exits the evaporator and is sent to a compressor, which increases the pressure and temperature of the refrigerant vapor. The refrigerant composition is then condensed in a condenser. A second cooling medium, which is a chilled liquid, or air, is transported into the condenser from a cooling tower. The second cooling medium cools the refrigerant vapor in the condenser, and the condenser condenses the refrigerant vapor to a liquid refrigerant. The liquid refrigerant is sent back to the evaporator, and the cycle is then repeated.

In another embodiment, the method for producing cooling comprises producing cooling in a direct expansion chiller as described above with respect to Fig. 3. In this method, a refrigerant composition of the present invention is evaporated in an evaporator. The step of evaporating the refrigerant composition comprises circulating the refrigerant composition through an inlet in the evaporator, through a coil in the evaporator, and through an outlet in the evaporator, thereby lowering the temperature of a first liquid cooling medium contained in the evaporator. The first liquid cooling medium is then passed out of the evaporator to a body to be cooled, thereby producing cooling. The refrigerant vapor is drawn into the compressor, which increases the pressure and the temperature of the refrigerant vapor before it is condensed in a condenser. A second liquid cooling medium is circulated through an inlet in the condenser, through a coil in the condenser, and

through an outlet in the condenser. The temperature of the second liquid cooling medium is increased as it passes from the inlet to the outlet of the condenser. The second liquid cooling medium is then sent from the outlet of the condenser to a cooling tower. The refrigerant vapor in the
5 condenser is condensed to refrigerant liquid, and is returned to the evaporator through an expansion valve. The cycle is then repeated.

As noted above, the compositions of the present invention have global warming potentials that are less than many hydrofluorocarbon refrigerants currently in use. The compositions of the present invention
10 have low ozone depletion potential and reduced low global warming potential (GWP) as compared to such currently used hydrocarbon refrigerants. A high GWP refrigerant would be any compound capable of functioning as a refrigerant or heat transfer fluid having a GWP at the 100 year time horizon of about 1000 or greater. One aspect of the present
15 invention is to provide a refrigerant with a global warming potential of less than 1000, less than 500, and in some cases less than 350, or even less than 150, or less than 100, or less than 50, which could be used to replace higher global warming potential refrigerants.

Based upon GWP calculations published by the Intergovernmental
20 Panel on Climate Change (IPCC), Third Assessment Report, the compositions of the present invention are particularly useful for replacing R22, R404A, R407C, R410A or R507A as a working fluid in a stationary heat transfer system, including a stationary air conditioning system or a stationary refrigeration system, or for replacing R22, R407C or R410A in a
25 flooded evaporator chiller or a direct expansion chiller.

Therefore, in accordance with the present invention, there is provided a method for replacing R22, R404A, R407C, R410A or R507A in a stationary air conditioning or refrigeration system. The method comprises providing a refrigerant composition of the present invention to a
30 stationary air conditioning or refrigeration system, in place of R22, R404A, R407C, R410A or R507A.

In addition, in accordance with the present invention, there is provided a method for replacing R-22, R407C or R410A in a flooded

evaporator chiller or a direct expansion chiller. The method comprises providing a refrigerant composition of the present invention to a flooded evaporator chiller or a direct expansion chiller in place of R22, R407C or R410A.

- 5 In one embodiment of this method, the compositions of the present invention are useful in flooded evaporator chillers that have centrifugal compressors that may have been originally designed and manufactured to operate with R22, R407C or R410A. In another embodiment, the compositions of the present invention are useful in direct expansion
- 10 chillers that use reciprocating, screw or scroll compressors that may have been originally designed and manufactured to operate with R22, R407C or R410A.

Alternatively, the compositions of the present invention as disclosed herein may be useful in new equipment, such as a new stationary air

15 conditioning or refrigeration system, a new flooded evaporator chiller or a new direct expansion chiller. In such new equipment, either a centrifugal compressor or a positive displacement compressor, including reciprocating, screw or scroll compressors, and the heat exchangers used therewith, could be designed for use with the compositions of the present

20 invention.

EXAMPLES

The sample compositions have the concentrations of components as shown in TABLE 4.

25

TABLE 4

Sample #	Concentration (weight fraction)				
	R32	R125	R134a	R290	CF ₃ I
1	0.500				0.500
2	0.600				0.400
3	0.400				0.600
4	0.700				0.300

5			0.500		0.500
6			0.600		0.400
7			0.400		0.600
8			0.700		0.300
9	0.700		0.050		0.250
10	0.700		0.250		0.050
11	0.700		0.150		0.150
12	0.600		0.050		0.350
13	0.600		0.350		0.050
14	0.600		0.200		0.200
15	0.500		0.100		0.400
16	0.500		0.400		0.100
17	0.500		0.250		0.250
18	0.330		0.330		0.340
19	0.700	0.050			0.250
20	0.600	0.050			0.350
21	0.500	0.050			0.450
22	0.550	0.050		0.050	0.350
23	0.600	0.025		0.025	0.350
24	0.600	0.050	0.050		0.300
25	0.600	0.050	0.100		0.250
26	0.500	0.050	0.050		0.400
27	0.500	0.050	0.100		0.350
28	0.400	0.050	0.100		0.450
29	0.400	0.050	0.050		0.500
30	0.400	0.100	0.050		0.450

EXAMPLE 1**Refrigeration Performance Data**

- 5 Table 5 shows cooling performance, as energy efficiency (COP), cooling capacity (Cool Cap), compressor discharge pressure (Dis Press),

compressor suction pressure (Suct Press), and compressor discharge temperature (Dis T) for compositions described herein as compared to R507A and R22 (HCFC-22) for low and medium temperature refrigeration systems. Additionally, average temperature glide (Avg glide, which is the average glide for evaporator and compressor) is included. Finally, a GWP value was calculated from the GWP values for the individual components for the 100 year time horizon values.

The data for the present compositions is provided as compared to R507A (a blend of 50 weight% of R125 and 50 weight% R143a (1,1,1-trifluoroethane)), R404A (a blend of 44 weight% of R125, 52 weight% of R143a, and 4.0 weight% of R134a), and R22 (chlorodifluoromethane, CHF₂Cl).

The performance data are based on the following conditions:

Evaporator temperature -17.8 °C
 Condenser temperature 46.1 °C
 Subcool temperature 5.5 °C
 Return gas temperature 15.6 °C
 Compressor efficiency 70%

20

TABLE 5

<u>Sample #</u>	<u>Calc GWP</u>	<u>Avg Glide (° C)</u>	<u>COP</u>	<u>Cool Cap (kJ/m3)</u>	<u>Dis Press (kPa)</u>	<u>Suct Press (kPa)</u>	<u>Dis T (° C)</u>
R507A	3900	5.00	1.35	1801	2151	342	100.0
R404A	3780	0.30	1.36	1765	2104	330	101.2
R22	2000	0.00	1.46	1697	1774	267	144.0
1	276	3.00	1.36	2341	2543	396	159.2
2	330	1.27	1.36	2506	2672	424	162.4
3	221	5.95	1.38	2136	2364	358	155.6
4	385	0.47	1.36	2594	2760	437	166.4
5	651	0.09	1.50	1129	1258	174	108.9
6	780	0.18	1.50	1114	1264	170	108.6

7	521	0.34	1.51	1127	1234	174	109.4
8	910	0.34	1.50	1094	1257	164	108.6
9	450	1.02	1.36	2517	2703	420	165.5
10	710	2.89	1.38	2255	2490	360	161.1
11	580	2.04	1.37	2377	2594	388	163.3
12	395	1.58	1.37	2447	2623	411	160.8
13	785	3.79	1.39	2088	2330	329	154.0
14	590	2.79	1.38	2253	2473	366	157.4
15	405	2.79	1.38	2283	2466	380	154.5
16	795	4.30	1.40	1964	2206	308	148.2
17	600	3.53	1.39	2116	2336	342	151.0
18	611	4.50	1.42	1851	2059	296	139.1
19	555	0.32	1.36	2595	2778	439	164.0
20	500	0.09	1.36	2524	2700	430	159.4
21	445	2.29	1.36	2382	2583	407	155.5
22	474	2.27	1.34	2565	2794	460	149.9
23	416	1.60	1.35	2579	2771	450	156.7
24	565	1.34	1.37	2454	2646	414	158.2
25	630	1.81	1.37	2384	2593	397	157.1
26	510	2.25	1.37	2350	2541	397	153.3
27	575	2.45	1.38	2297	2495	385	151.9
28	520	3.65	1.39	2159	2357	361	146.7
29	456	4.01	1.37	2178	2389	367	148.9
30	625	3.24	1.37	2216	2431	377	145.8

- Many compositions have similar energy efficiency (COP) as compared to R22, R404A or R507A while maintaining lower discharge pressures and temperatures. Refrigeration capacity for several of the compositions listed in Table 5 is also similar to R22, R404A and R507A, indicating these compositions could be replacement refrigerants for R22, R404A or R507A in stationary air conditioning or stationary refrigeration systems.

EXAMPLE 2**Air Conditioning Performance Data**

- 5 Table 6 shows cooling performance, as energy efficiency (COP), cooling capacity (Cool Cap), compressor discharge pressure (Dis Press), compressor suction pressure (Suction Press), and compressor discharge temperature (Dis T) for compositions described herein as compared to R410A and R22 (HCFC-22) for stationary air conditioning systems.
- 10 Additionally, average temperature glide (Avg glide, which is the average glide for evaporator and compressor) is included. Finally, a GWP value was calculated from the GWP values for the individual components for the 100 year time horizon values.

- The data for the present compositions is provided as compared to
- 15 R410A (a blend of 50 weight% of R125 and 50 weight% of R32), R407C (a blend of 25 weight% of R125, 23 weight% of R32, and 52 weight% of R134a), and R22 (chlorodifluoromethane, CHF₂Cl).

The performance data are based on the following conditions:

- Evaporator temperature 4 °C
- 20 Condenser temperature 43 °C
- Subcool temperature 6 °C
- Return gas temperature 18 °C
- Compressor efficiency is 70%

25

TABLE 6

<u>Sample #</u>	<u>Calc GWP</u>	<u>Avg Glide (° C)</u>	<u>COP</u>	<u>Cool Cap (kJ/m3)</u>	<u>Dis Press (kPa)</u>	<u>Suct Press (kPa)</u>	<u>Dis T (° C)</u>
R410A	1700	0.10	2.72	5488	2571	900	90.9
R407C	1653	4.84	2.86	3855	1754	564	80.6
R22	2000	0.00	2.92	3808	1648	565	88.1
1	276	3.40	2.74	5205	2366	826	99.7
2	330	1.53	2.74	5512	2486	875	101.6

3	221	6.43	2.73	4802	2198	756	97.8
4	385	0.58	2.74	5710	2566	902	103.9
5	651	0.05	2.97	2644	1166	382	71.8
6	780	0.09	2.97	2644	1170	378	71.7
7	521	0.41	2.98	2609	1144	378	72.2
8	910	0.26	2.97	2623	1162	370	71.6
9	450	1.09	2.75	5592	2513	875	103.2
10	710	2.98	2.78	5159	2310	775	100.0
11	580	2.10	2.76	5366	2410	822	102.0
12	395	1.76	2.76	5419	2439	852	100.8
13	785	3.96	2.80	4827	2161	716	97.2
14	590	2.90	2.78	5112	2297	780	98.9
15	405	3.06	2.77	5091	2293	795	97.4
16	795	4.51	2.82	4568	2045	674	93.9
17	600	3.69	2.80	4827	2168	733	95.4
18	611	4.78	2.84	4264	1911	639	89.0
19	555	0.39	2.74	5726	2583	907	102.5
20	500	1.11	2.74	5551	2511	885	101.0
21	445	2.60	2.74	5277	2403	843	97.9
22	474	2.47	2.89	5601	2603	936	95.2
23	416	1.78	2.72	5644	2581	919	98.8
24	565	1.48	2.75	5445	2461	859	99.4
25	630	1.91	2.76	5336	2410	833	98.8
26	510	2.52	2.76	5213	2363	825	96.8
27	575	2.66	2.77	5129	2320	804	96.0
28	520	4.03	2.78	4834	2191	758	93.2
29	456	4.45	2.76	4875	2222	770	94.3
30	625	3.63	2.76	4947	2261	787	92.7

Many compositions have similar energy efficiency (COP) as compared to R22, R407C or R410A while maintaining lower discharge pressures and temperatures. Refrigeration capacity for several of the

compositions listed in Table 6 is also similar to R22, R407C or R410A, indicating these compositions could be replacement refrigerants for R22, R410A or R407C in air-conditioning and chiller systems. Additionally, several of the compositions have low average glide thus allowing use in

5 flooded evaporator type chillers.

CLAIMS

What is claimed is:

1. A composition comprising iodotrifluoromethane, difluoromethane,
and at least one hydrocarbon selected from the group consisting of
5 pentane, butane, isobutane, propylene, cyclopropylene and
propane.
2. The composition of claim 1, wherein the hydrocarbon comprises
propane, and wherein the composition further comprises
10 pentafluoroethane.
3. A composition consisting essentially of:
iodotrifluoromethane and 1,1,1,2-tetrafluoroethane;
iodotrifluoromethane, 1,1,1,2-tetrafluoroethane and
15 difluoromethane;
iodotrifluoromethane and pentafluoroethane; and
iodotrifluoromethane, difluoromethane; pentafluoroethane and
1,1,1,2-tetrafluoroethane.
- 20 4. A composition consisting essentially of about 0.01 to about 67.25
weight difluoromethane and about 32.75 to about 99.99 %
iodotrifluoromethane.
5. The composition of claim 2, comprising:
25 35% iodotrifluoromethane, 55% difluoromethane, 5%
pentafluoroethane and 5% propane; or
35% iodotrifluoromethane, 60% difluoromethane, 2.5%
pentafluoroethane and 2.5% propane.
- 30 6. The composition of claim 3, consisting essentially of:
30 - 50 weight percent iodotrifluoromethane and 50 - 70 weight
percent 1,1,1,2-tetrafluoroethane;

- 5 - 35 weight percent iodotrifluoromethane, 5 - 40 weight percent 1,1,1,2-tetrafluoroethane and 33 - 70 weight percent difluoromethane; or
25 - 50 weight percent iodotrifluoromethane, 40 - 60 weight percent difluoromethane, 0 - 50 weight percent pentafluoroethane and 5 - 50 weight percent 1,1,1,2-tetrafluoroethane.
- 5
7. The composition of claims 1, 3 or 4, further comprising a lubricant selected from the group consisting of polyalkylene glycols, polyol esters, polyvinylethers, mineral oils, alkylbenzenes, synthetic paraffins, synthetic naphthenes, and poly(alpha)olefins.
- 10
8. The composition of claims 1, 3 or 4, further comprising at least one additive selected from the group consisting of compatibilizers, UV dyes, solubilizing agents, tracers, stabilizers, perfluoropolyethers, and functionalized perfluoropolyethers.
- 15
9. A method for producing cooling, comprising evaporating the composition of claims 1, 3 or 4 in an evaporator, wherein the composition is a refrigerant, to form a vapor refrigerant, condensing the vapor refrigerant to form a liquid refrigerant, and returning the liquid refrigerant to the evaporator.
- 20
10. The method of claim 9, further comprising the step of compressing the refrigerant vapor before it is condensed.
- 25
11. The method of claim 9, wherein the cooling is produced in a stationary air conditioning or a stationary refrigeration system.
- 30
12. The method of claim 9, further comprising circulating a first cooling medium through an inlet in the evaporator, through a coil in the evaporator and to an outlet of the evaporator, thereby lowering the temperature of the first cooling medium as it passes from the inlet

to the outlet of the evaporator, and passing the first cooling medium to a body to be cooled, thereby producing cooling.

5 13. The method of claim 9, wherein the cooling is produced in a flooded evaporator chiller.

10 14. The method of claim 9, wherein the step of evaporating the refrigerant composition comprises circulating the refrigerant composition through an inlet in the evaporator, through a coil in the evaporator, and through an outlet in the evaporator, thereby lowering the temperature of a first liquid cooling medium contained in the evaporator, and passing the first liquid cooling medium out of the evaporator to a body to be cooled, thereby producing cooling.

15 15. The method of claim 14, wherein the cooling is produced in a direct expansion chiller.

20 16. A method for replacing R22, R407C, R-410A, R404A or R507A in a flooded evaporator chiller, a direct expansion chiller or a stationary air conditioning system or a stationary refrigeration system, said method comprising providing a composition of claims 1, 3 or 4 to said flooded evaporator chiller, direct expansion chiller or stationary air conditioning or refrigeration system, wherein said composition comprises a refrigerant.

25 17. The method of claim 16, wherein the refrigerant composition is provided to a stationary air conditioning system or stationary refrigeration system in place of R22, R404A, R407C, R410A or R507A.

30 18. The method of claim 16, wherein the refrigerant composition is provide to a flooded evaporator chiller or a direct expansion chiller in place of R22, R407C or R410A.

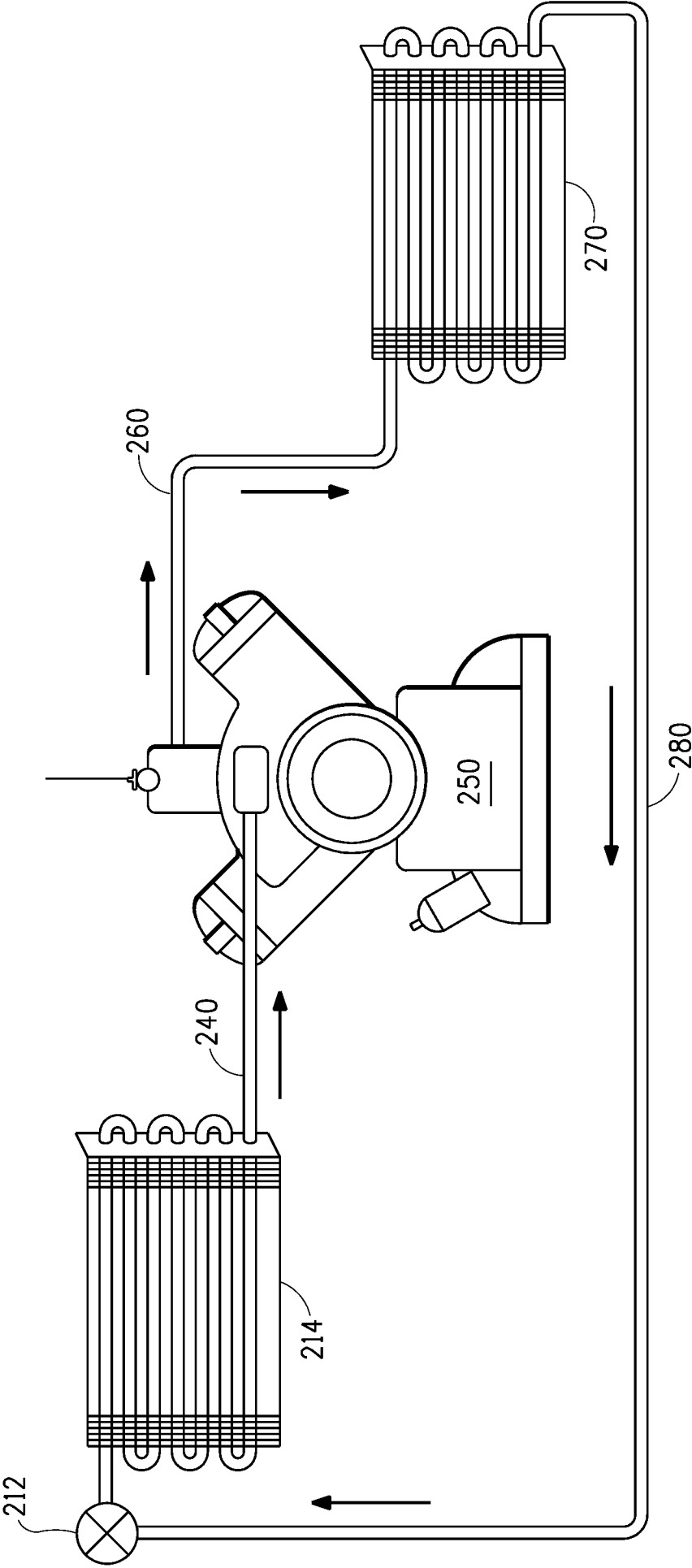


FIG. 1

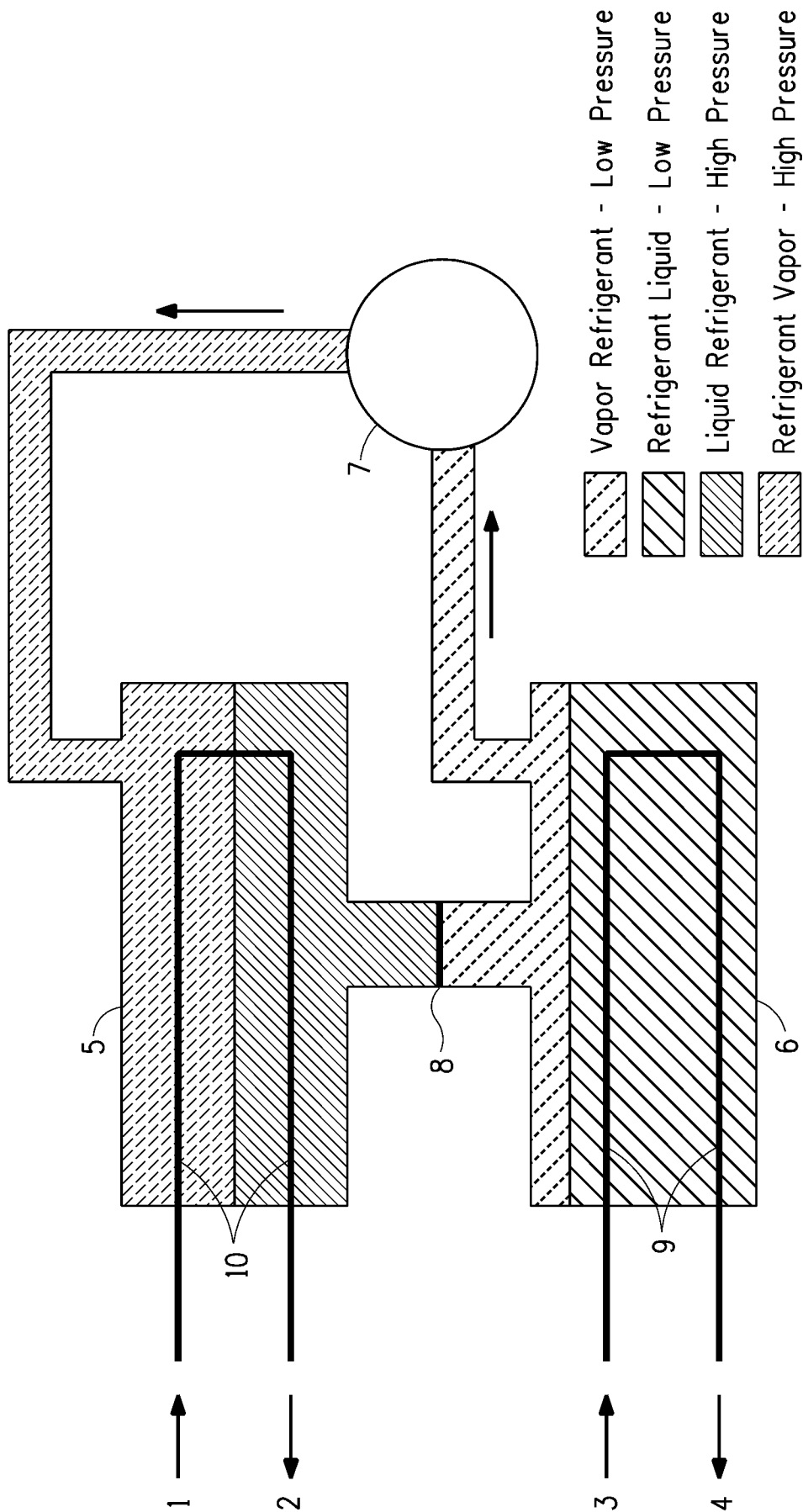
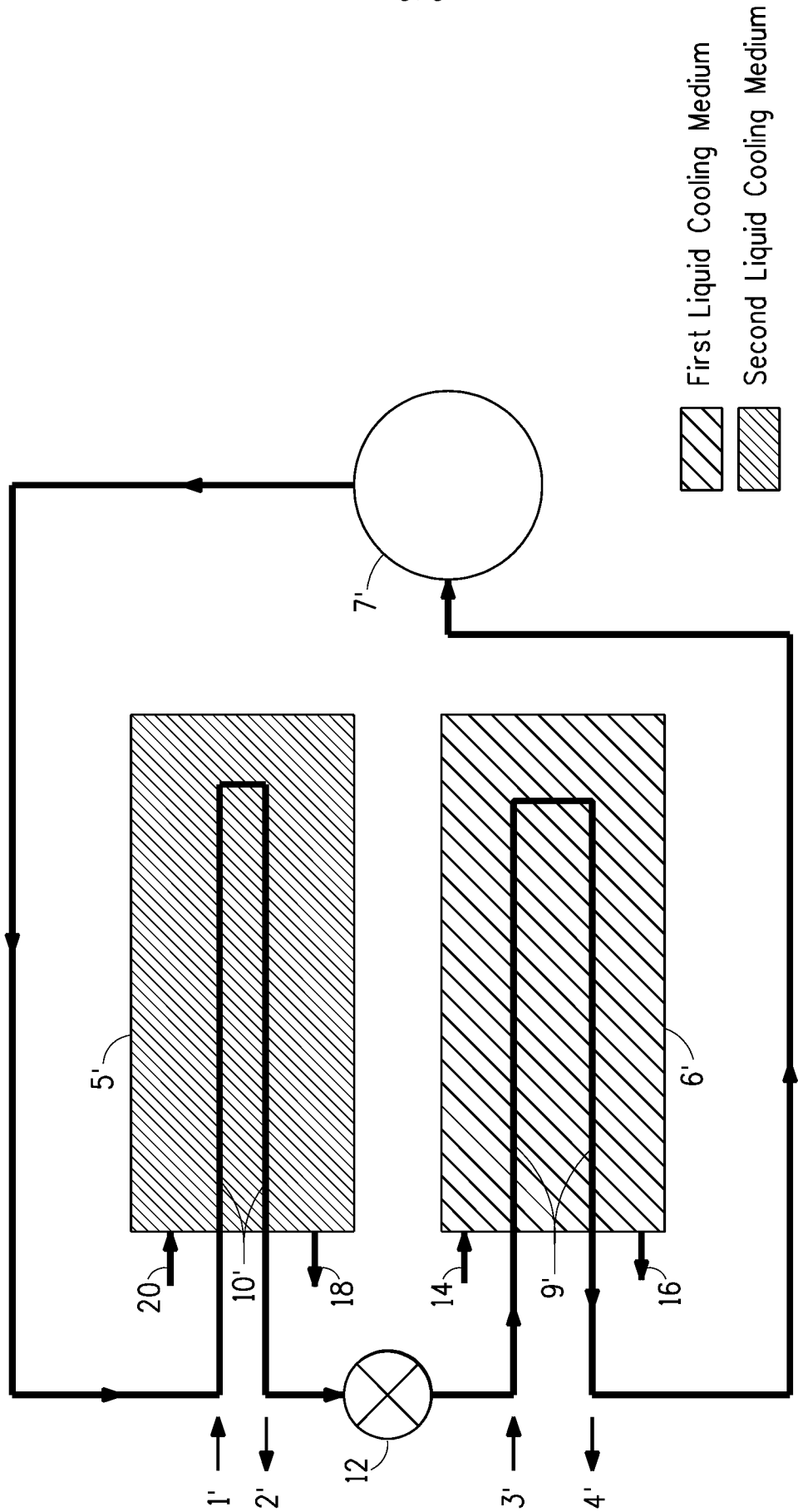


FIG. 2



INTERNATIONAL SEARCH REPORT

International application No
PCT/US2008/081955

A. CLASSIFICATION OF SUBJECT MATTER
INV. C09K5/04

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
C09K A61L C08J

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 2006/033071 A1 (WILSON DAVID P [US] ET AL) 16 February 2006 (2006-02-16) paragraphs [0004] - [0010], [0018], [0019], [0050] - [6845]; claims 1,2,9-12,18,22,30-34; table 1	1-18
X	WO 2006/069362 A (HONEYWELL INT INC [US]; SINGH RAJIV R [US]; NAIR HARIDASAN K [US]; THO) 29 June 2006 (2006-06-29) page 1, lines 5-28 page 3, lines 10-27 page 4, line 25 - page 5, line 11 page 6, line 15 - page 7, line 29 page 18, line 25 - page 21, line 10 page 29, line 4 - page 32, line 2 page 38, line 24 - page 39, line 22; claims 50-52; tables 1,3,4 ----- -/-	1-15

☒ Further documents are listed in the continuation of Box C.

☒ See patent family annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

- *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *&* document member of the same patent family

Date of the actual completion of the international search

24 March 2009

Date of mailing of the international search report

30/03/2009

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INTERNATIONAL SEARCH REPORT

International application No

PCT/US2008/081955

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 5 716 549 A (NIMITZ JONATHAN SHELLEY [US] ET AL) 10 February 1998 (1998-02-10) column 6, line 66 - column 7, line 19; claim 31; tables 2,5 column 8, line 31 - column 9, line 59 column 13, lines 22-44 -----	3,4,6,7, 9-18
A	EP 1 193 305 A (HONEYWELL INT INC [US]) 3 April 2002 (2002-04-03) paragraphs [0001], [0009] - [0014], [0018], [0022] - [0027]; claims 1-4,8; examples 1,4-7; tables I-IV -----	1-18
A	EP 1 028 152 A (DU PONT [US]) 16 August 2000 (2000-08-16) paragraphs [0001] - [0005], [0009], [0010], [0030] - [0032], [0037], [0038], [0048], [0053], [0058] - [0065], [0076]; claims 1,2,6-8; example 4 -----	1-15
A	WO 00/56834 A (DU PONT [US]) 28 September 2000 (2000-09-28) page 1, line 1 - page 2, line 25 page 3, line 24 - page 4, line 21 page 5, line 31 - page 7, line 14; example 1; tables 1,2 -----	1-15

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US2008/081955

Box No. II Observations where certain claims were found unsearchable (Continuation of item 2 of first sheet)

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:
because they relate to subject matter not required to be searched by this Authority, namely:
2. ☐ Claims Nos.:
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
3. ☐ Claims Nos.:
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

Box No. III Observations where unity of invention is lacking (Continuation of item 3 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

see additional sheet

1. ☐ As all required additional search fees were timely paid by the applicant, this international search report covers allsearchable claims.
2. ☒ As all searchable claims could be searched without effort justifying an additional fees, this Authority did not invite payment of additional fees.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search reportcovers only those claims for which fees were paid, specifically claims Nos.:
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest and, where applicable, the payment of a protest fee.
- ☐ The additional search fees were accompanied by the applicant's protest but the applicable protest fee was not paid within the time limit specified in the invitation.
- ☐ No protest accompanied the payment of additional search fees.

FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

This International Searching Authority found multiple (groups of) inventions in this international application, as follows:

1. claims: 1, 2 and 5, 7-16 (all in part)

directed to compositions comprising CF3I, HFC32 and at least one hydrocarbon and methods for producing cooling or replacing a refrigerant using said compositions.

2. claims: 3, 6-16 (all in part)

directed to compositions comprising CF3I, HFC32 and at least a second HFC and methods for producing cooling or replacing a refrigerant using said compositions.

3. claims: 4 and 7-16 (all in part)

directed to compositions consisting essentially of CF3I and HFC32 and methods for producing cooling or replacing a refrigerant using said compositions.

4. claims: 3, 6-16 (all in part)

directed to compositions comprising CF3I and HFC-134a and methods for producing cooling or replacing a refrigerant using said composition.

5. claims: 3, 6-16 (all in part)

directed to compositions comprising CF3I and HFC-125 and methods for producing cooling or replacing a refrigerant using said composition.

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

PCT/US2008/081955

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