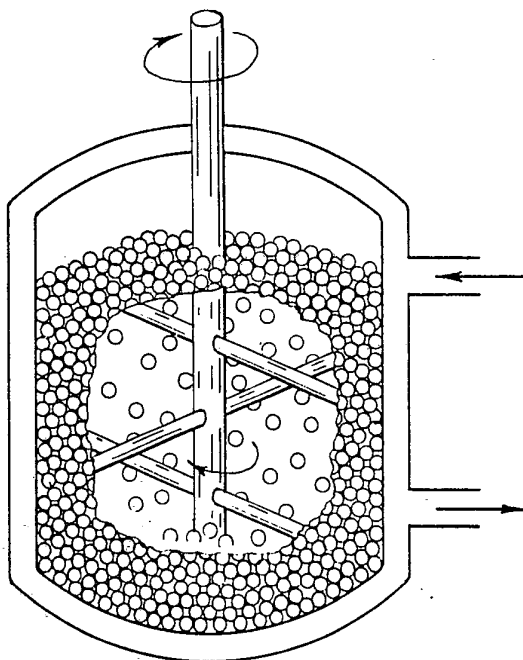


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(21) International Application Number: PCT/US80/01061 (22) International Filing Date: 18 August 1980 (18.08.80) (31) Priority Application Number: 081,723 (32) Priority Date: 4 October 1979 (04.10.79) (33) Priority Country: US (71) Applicant: OWENS-CORNING FIBERGLASS CORPORATION [US/US]; Law Department, Fiberglass Tower, Toledo, OH 43659 (US). (72) Inventor: ROEHRIG, Frederick, Karl; 4800 Hayden Boulevard, Columbus, OH 43220 (US). (74) Agent: PACELLA, Patrick, P.; Law Department, Fiberglass Tower, Toledo, OH 43659 (US).		(81) Designated States: GB, JP, SE. Published <i>With international search report</i>
(54) Title: PROCESS FOR PRODUCING DISPERSION STRENGTHENED PRECIOUS METAL ALLOYS <div data-bbox="568 1220 1093 1877" data-label="Image">  </div> (57) Abstract <p>Process for producing dispersion-strengthened precious metal alloys having superior creep resistance. According to this invention precious metal powders and dispersoids are mechanically alloyed together.</p>		

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D E S C R I P T I O N

PROCESS FOR PRODUCING DISPERSION
10 STRENGTHENED PRECIOUS METAL ALLOYS

TECHNICAL FIELD

This invention relates to a process for producing dispersion strengthened precious metal alloys. The present
15 invention can provide alloys containing platinum, palladium, rhodium and gold which are useful in the production of glass fibers.

BACKGROUND ART

One of the most exacting applications of platinum
20 is in the production of glass fibers. Molten glass often at temperatures ranging from 1200 to 1600°C passes through a series of orifices in a bushing. Advances in glass fiber production are demanding both larger bushings and higher operating temperatures.

25 Structural components such as these at elevated temperatures under constant loads experience continuous dimensional changes or creep during their lives. This creep behavior depends upon the interaction between the external conditions (load, temperature) and the
30 microstructure of the component. In recent times, increased resistance to creep of material systems has been accomplished by using a dispersion of very small, hard particles (called dispersoids) to strengthen the microstructure of the component. These systems have become
35 to be known as dispersion-strengthened metals and alloys and the dispersoids used are usually oxides.



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1 A recent development in dispersion-strengthening
is called mechanical alloying. Generally, the process uses
a high energy ball mill to achieve the intimate mechanical
mixing typical of the process. An attritor mill or
5 vibratory mill also can be used. While mechanical alloying
has been applied to some of the transition metals, no
actual work has been reported on precious metals such as
platinum.

DISCLOSURE OF THE INVENTION

10 The present invention provides a process for
producing dispersion-strengthened precious metal alloys
having creep resistance superior to known
dispersion-strengthened platinum alloys.

 According to the process of this invention,
15 precious metal powder and dispersoids are mechanically
alloyed together. The mechanical alloying uses a high
energy ball mill to achieve the intimate mechanical mixing
of this process. The oxide particles are forged into the
precious metal matrix powder particle to form a composite
20 powder particle.

BRIEF DESCRIPTION OF DRAWINGS

 FIG. 1 illustrates the internal arrangement in an
attritor mill showing the impeller, grinding media and
external cooling jacket. Impact events occur in the
25 dynamic interstices of the media created by the impeller
during stirring.

BEST MODE OF CARRYING OUT INVENTION

 There are several high-energy ball mills
commercially available either using a stirrer to induce the
30 deformation events or vibratory motion. FIG. 1 shows an
overall view of the attritor mill. The stainless steel
bearings or grinding media and the powder charge go into
the cylindrical container of the mill. The high-energy
impacts are effected by the rotating impeller. FIG. 1 also
35 illustrates the internal arrangement in the attritor mill,
impact events occur in the dynamic interstices of the media
created by the impeller during stirring.



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1 Dispersion strengthened precious metals are known
in the art and are commercially available. One such
material is that available from Johnson, Matthey & Co.
Limited, under their designation ZGS. The above indicated
5 ZGS material consists essentially of platinum in which the
dispersoid is zirconia; the latter is present in an amount
of about 0.5% by volume.

The dispersion strengthened precious metals of
this invention generally comprise a precious metal, or
10 precious metal alloy, preferably platinum, as the
dispersing medium, or matrix, and a dispersoid of a metal
oxide, metal carbide, metal silicide, metal nitride, metal
sulfide or a metal boride which dispersoid is present in
effective dispersion strengthening amounts. Usually such
15 amounts will be between about 0.1 percent to about 5.0
percent by volume. Preferably the dispersoid will be an
oxide. Exemplary of metal compounds which may be employed
as the dispersoid are compounds of metals of Group IIA,
IIIA, IIIB (including non-hazardous metals of the Actinide
20 and Lanthanide classes), IVB, VB, VIB and VIIB. More
specifically exemplary of suitable metals are the
following: Be, Mg, Ca, Ba, Y, La, Ti, Zr, Hf, Mo, W, Ce,
Nd, Gd, and Th as well as Al.

Several mechanical alloying experiments were
25 performed using the attritor mill to generate a composite
powder for consolidation. Wash heats intended to coat a
thin layer of platinum on the internal workings surfaces of
the attritor mill were carried out. This "conditioning"
treatment was intended to prevent iron contamination of
30 subsequent milling experiments, but several washes were
required before the iron contamination was reduced to what
was believed to be an acceptable level.

The samples then are consolidated by vacuum hot
pressing (VHP) at elevated temperatures and pressures. In
35 the alternative, the samples can be consolidated by first
cold pressing at elevated pressures followed by sintering
at elevated temperatures. VHP generally is carried out at



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1 a temperature ranging from 1300 to 1700°C under a pressure
ranging from 500 to 10,000 psi for a time ranging from 10
to 30 minutes. Preferably, the temperature ranges from
1400 to 1500°C under a pressure of 3,000 to 6,000 psi for a
5 time of 15 to 25 minutes. Generally, the cold pressing is
carried out at a pressure ranging from 2,000 to 10,000 psi
for up to 5 minutes followed by sintering at a temperature
ranging from 1200 to 1700°C for 2 to 6 hours.

EXAMPLE I

10 Approximately one kgm of -325 mesh (-44 micron)
platinum sponge from Englehard was blended with an amount
of yttria (Y_2O_3) to give nominally 0.65 volume percent
(0.15 weight percent) oxide loading in the final compact.
The yttria was 200-600 angstrom in size. The platinum
15 matrix starting powder for the experiment consisted of very
fine, near spherical particles or chained aggregates. Most
of the particles below 2 microns appeared to be single
crystals. The starting powder had a fairly high specific
surface area.

20 The powder mixture was charged into the
container of the attritor mill while it was running. The
grinding media had been previously loaded to give a volume
ratio of media to powder of about 20:1. The grinding media
used was a hardened 400 series stainless steel bearing
25 nominally 3/8 inch (0.953 cm) diameter. The impeller
rotational speed was selected at 130 rpm.

Samples of powder were removed at various times
to obtain information on the changes in particle morphology
and specific surface area with milling time. The first
30 sample was taken after one hour of milling and indicated
that flake generation was in progress.

After milling for three hours, another powder
sample was taken for metallographic characterization.
While more flakes were generated, the extent of plastic
35 deformation seemed to have increased. Flake cold welding
appeared to have taken place as well. The composite flake
appeared to have three or four component flakes cold welded



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1 together. No edge cracking appeared in the composite flake
suggesting that work hardening saturation had not been
reached at this point.

After milling for 23 hours, the composite flakes
5 appeared to thicken. This clearly demonstrates the cold
welding aspect of the milling action. Along with cold
welding, the flake diameter appeared to increase.

The experiment was continued for 71 hours then
terminated, and the powder was removed for further
10 processing.

There appeared to be a fairly high initial
surface area generation rate. The iron contamination in
the milled powder was greatly reduced compared to the
previous experiments and reflects the coating action that
15 appeared to minimize wear debris generation during milling.
The maximum iron contamination level in the powder was
approximately 300 wppm. The milled powder was consolidated
by vacuum hot pressing and thermomechanically processing
into sheet for creep testing, the details are to follow.

20 EXAMPLE II

Example I produced a powder of relatively low
iron contamination. Since this experiment resulted in
small powder lots (nominally 20 gms) taken at various times
during the milling experiment, each sample was individually
25 consolidated by vacuum hot pressing (VHP) at 1,450°C under
5,000 psi (34.5 MN/m²) for twenty minutes. The resultant
compacts were nominally 1 inch (2.54 cm) in diameter.

Relative density of specimens are listed.

	<u>Specimen</u>	<u>Milling Time (hr.)</u>	<u>Relative Density (%)</u>
30	A	0	95.2
	B	1	98.2
	C	2.5	99.6
	D	6	99.6

The thermomechanical processing (TMP) scheduled
35 Used on the compact consisted of several roll/anneal
cycles. The basic operation involved rolling a sheet
specimen and cropping pieces after various rolling passes



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1 for microstructural characterization. The procedure used
was to roll the compact for a 10 percent reduction in area
then anneal the rolled specimen for five minutes at
nominally 1,040°C before further rolling.

5 Specimen D was the most responsive to the TMP
cycles. After the 10th rolling pass, the grain structure
was fairly elongated. The lack of oxide clusters during
optical metallographic examination suggested that the
milling action had worked the yttria into the platinum
10 matrix. A metallographic analysis of the same region
showed the development of a moderate grain aspect ratio
(grain length to thickness ratio in the viewing plane). As
the number of roll/anneal cycles increased, the grain
aspect ratio (GAR) increased. At this stage a moderate GAR
15 also had been developed in a transverse direction. The
significance of this observation is that the grains took on
the shape of a pancake structure thin in a direction
perpendicular to the sheet yet extended in the other two
directions. Since a GAR seems to extend in two directions
20 in the rolled sheet and the state of stress in a bushing
tip plate is biaxial, this transverse GAR development may
be very beneficial for good creep resistance in bushing
applications.

After the 16th rolling pass, the elongation of
25 the grains had increased significantly. A higher
magnification view of the same region revealed the degree
of grain elongation and fineness of the grain size. The
transverse GAR had also significantly increased. These
elongated grain morphologies are desirable microstructures
30 for good creep resistance.

INDUSTRIAL APPLICABILITY

EXAMPLE III

Creep Testing

All the creep testing was done in air using
35 Constant load machines, the elongation was measured by an
LVDT connected to a multi-point recorder and a precision
digital voltmeter. Specimen temperature was monitored with



13063A

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1 a calibrated Pt/Pt-Rh thermocouple attached so that the
bead was adjacent to the gage section of the creep
specimen. The creep specimen was a flat plate type with a
gage length of approximately 2.25 inch (5.72 cm). The
5 tensile stress was applied parallel to the rolling
direction (longitudinal direction). The general procedure
was to hang the specimen in the furnace to reach thermal
equilibrium then start the rig timer upon application of
the load. Periodic temperature and extension measurements
10 were made either until the specimen failed or the test was
terminated (specimen removal or furnace burn-out).

Creep results were obtained from specimens that
were processed according to Example II except that these
specimens were milled 10 hours and received the above
15 thermomechanical processing treatment of 10% reduction in
area per pass with an intermediate anneal at nominally
1040°C for 5 minutes. The extent of deformation was
nominally an 85% reduction in area. The first specimen had
a varied creep history that started by applying a tensile
20 stress of 1,000 psi (6.89 Mn/m^2) at 2,400°F ($1,316^\circ\text{C}$). The
resultant secondary creep rate was too low to adequately
measure; therefore, the temperature was increased to
2,600°F ($1,427^\circ\text{C}$) and a secondary creep rate of 4.5×10^{-6}
 hr^{-1} was observed. After approximately 118 hours the
25 stress was increased to 1,400 psi (9.65 Mn/m^2) and a new
secondary creep rate of nominally $3 \times 10^{-5} \text{ hr}^{-1}$ was recorded.
These creep rates are two orders of magnitude less than
that for the previously indicated ZGS under the same
testing conditions. The ZGS material will have a stress
30 rupture life of at least 48 hours when tested at 1400°C and
1000 psi in the rolling direction of the sheet.

The general microstructure of the crept specimen
indicated that the grains were highly elongated in the
rolling direction (creep stress direction also) and the
35 grain boundaries were ragged. There appeared to be evidence
of subgrains in the structure as well. The microstructure
observed in this specimen was typical of that of a good



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1 creep resistant material as evidenced by the exceptionally good creep properties.

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CLAIMS

1. A process for producing dispersion
10 strengthened precious metal alloys comprising the step of
mechanically alloying precious metal powder and at least
one dispersoid together wherein the dispersoid is present
in effective dispersion strengthening amounts.
2. A process for producing dispersion
15 strengthened precious metal alloys comprising the steps of:
(1) mechanically alloying precious metal
powder and at least one dispersoid together wherein the
dispersoid is present in effective dispersion strengthening
amounts; and
20 (2) consolidating the resulting powder.
3. A process according to claim 2 wherein the
consolidating is carried out by vacuum hot pressing at
elevated temperature and pressures.
4. A process according to claim 2 wherein the
25 consolidating is carried out by first cold pressing at
elevated pressures and then sintering at elevated
temperatures.
5. A process according to claims 1 or 2 wherein
the precious metal powder is platinum or a platinum alloy.
- 30 6. A process according to claims 1 or 2 wherein
the disperoids include a metal oxide.
7. A process according to claims 1 or 2 wherein
the precious metal powder is platinum and the disperoids
include yttria (Y_2O_3).
- 35 8. A process according to claims 1 or 2 wherein
high energy ball milling is used to achieve the mechanical
alloying.

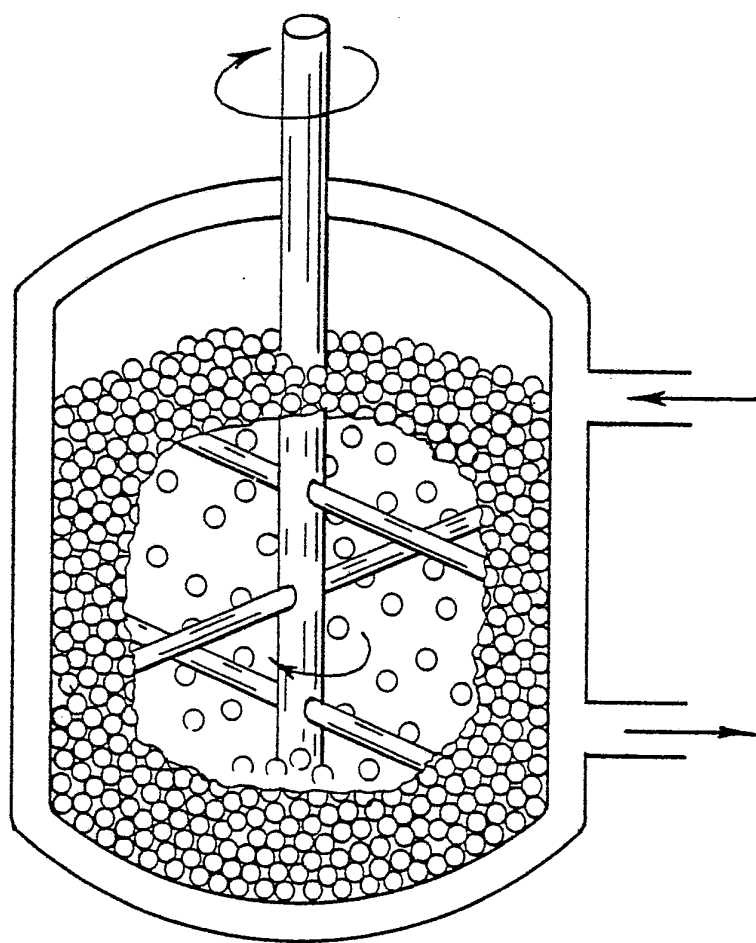


-10-

- 1 9. A process for producing dispersion
strengthened precious metal alloys comprising the steps of:
 (1) mechanically alloying platinum powder and
yttria (Y_2O_3) together wherein the yttria is present in
5 effective dispersion strengthening amounts; and
 (2) consolidating the resulting powder by
vacuum hot pressing at elevated temperatures and pressures.
- 10 10. A process according to claim 9 wherein the
amount of yttria ranges between 0.1 and 5.0 percent by
volume.
11. A process according to claim 9 wherein the
amount of yttria is 0.65 percent by volume (0.15 percent by
weight).
- 15 12. A process according to claim 9 wherein the
vacuum hot pressing is carried out at a temperature ranging
from 1300 to 1700°C under a pressure ranging from 500 to
10,000 psi for a time ranging from 10 to 30 minutes.
- 20 13. A process according to claim 9 wherein the
vacuum hot pressing is carried out at a temperature ranging
from 1400 to 1500°C under a pressure ranging from 3,000 to
6,000 psi for a time ranging from 15 to 25 minutes.
- 25 14. A process according to claim 9 wherein the
vacuum hot pressing is carried out at a temperature of
1,450°C under a pressure of 5,000 psi for a time of twenty
minutes.
15. A process according to claim 9 wherein high
energy ball milling is used to achieve the mechanical
alloying.
- 30 16. A process for producing dispersion
strengthened precious metal alloys comprising the steps of:
 (1) mechanically alloying platinum powder
and yttria (Y_2O_3) together wherein the yttria is present in
effective dispersion strengthening amounts; and
 (2) consolidating the resulting powder by
35 First cold pressing at a pressure ranging from 2,000 to
10,000 psi for up to 5 minutes and then sintering at a
temperature ranging from 1200 to 1700°C for 2 to 6 hours.



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INTERNATIONAL SEARCH REPORT

International Application No PCT/US 80/01061

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) ³ According to International Patent Classification (IPC) or to both National Classification and IPC B 22 F 9/04; B 22 F 3/16; C 22 C 1/05; C 22 C 5/04																							
II. FIELDS SEARCHED <div style="text-align: center; margin-top: 10px;">Minimum Documentation Searched ⁴</div> <table style="width: 100%; border-collapse: collapse;"> <tr> <th style="width: 20%; border: 1px solid black;">Classification System</th> <th style="border: 1px solid black;">Classification Symbols</th> </tr> <tr> <td style="border: 1px solid black; text-align: center; vertical-align: top;">U.S.</td> <td style="border: 1px solid black;">75/0.5R, 172E, 206, 211, 232, 235, 247</td> </tr> </table> <div style="text-align: center; margin-top: 10px;">Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched ⁵</div>			Classification System	Classification Symbols	U.S.	75/0.5R, 172E, 206, 211, 232, 235, 247																	
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<div style="font-size: small;"> <p>[*] Special categories of cited documents: ¹⁵</p> <div style="display: flex; justify-content: space-between;"> <div style="width: 45%;"> <p>"A" document defining the general state of the art</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document cited for special reason other than those referred to in the other categories</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> </div> <div style="width: 45%;"> <p>"P" document published prior to the international filing date but on or after the priority date claimed</p> <p>"T" later document published on or after the international filing date or priority date and not in conflict with the application, but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance</p> </div> </div> </div>																							
IV. CERTIFICATION <table style="width: 100%; border-collapse: collapse; margin-top: 5px;"> <tr> <td style="width: 50%; border: 1px solid black; padding: 5px;"> Date of the Actual Completion of the International Search ² <div style="text-align: center; font-size: large;">29 December 1980</div> </td> <td style="width: 50%; border: 1px solid black; padding: 5px;"> Date of Mailing of this International Search Report ² <div style="text-align: center; font-size: large;">16 JAN 1981</div> </td> </tr> <tr> <td style="border: 1px solid black; padding: 5px;"> International Searching Authority ¹ <div style="text-align: center;">ISA/US</div> </td> <td style="border: 1px solid black; padding: 5px;"> Signature of Authorized Officer ²⁰ <div style="text-align: center;"> Richard E. Schafer </div> </td> </tr> </table>			Date of the Actual Completion of the International Search ² <div style="text-align: center; font-size: large;">29 December 1980</div>	Date of Mailing of this International Search Report ² <div style="text-align: center; font-size: large;">16 JAN 1981</div>	International Searching Authority ¹ <div style="text-align: center;">ISA/US</div>	Signature of Authorized Officer ²⁰ <div style="text-align: center;"> Richard E. Schafer </div>																	
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