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<b>(21) International Application Number:</b> PCT/US89/01395 <b>(22) International Filing Date:</b> 6 April 1989 (06.04.89) <b>(30) Priority data:</b> 178,153                      6 April 1988 (06.04.88)                      US <b>(71) Applicant:</b> UNITED STATES OF AMERICA, represented by THE SECRETARY, UNITED STATES DEPARTMENT OF COMMERCE [US/US]; Washington, DC 20231 (US). <b>(72) Inventors:</b> MARQUEZ, Victor, E. ; 20020 Doolittle Street, Gaithersburg, MA 20879 (US). GODDARD, Amanda, J. ; 36 Central Street, Huntington, NY 11743 (US). <b>(74) Agent:</b> OLIFF, James, A.; Oliff & Berridge, 277 S. Washington Street, Alexandria, VA 22314 (US).		<b>(81) Designated States:</b> AU, JP.  <b>Published</b> <i>With international search report.</i>
<b>(54) Title:</b> PHOSPHORAMIDITE REAGENT FOR CHEMICAL SYNTHESIS OF MODIFIED DNA  <b>(57) Abstract</b>  5,6-dihydro-5-azacytidine phosphoramidite is useful in the synthesis of oligonucleotides and DNA containing dihydro-5-aza- and 5-azacytosine bases. The modified oligonucleotides which contain 5-azacytosine residues at specific sites can be used to determine the mechanism of selective gene activation and the relationship existing between the presence of the triazine base and inhibition of DNA methylation.		

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PHOSPHORAMIDITE REAGENT FOR CHEMICAL SYNTHESIS  
OF MODIFIED DNA

FIELD OF THE INVENTION

The present invention relates to the synthesis  
5 of DNA containing modified cytosine bases, namely 5-6-  
dihydro-5-azacytosine and 5-azacytosine, and more  
specifically to a reagent for use in the chemical syn-  
thesis of modified DNA which allows the incorporation of  
said bases at specific sites of a sequence.

10 BACKGROUND OF THE INVENTION

The ability to synthesize polynucleotide frag-  
ments having a desired nucleotide sequence is a useful  
tool in both research and applied molecular biology.  
Short synthetic polynucleotides, or oligonucleotides, are  
15 useful as adaptors or linkers in joining longer DNA seg-  
ments, and as hybridization probes and DNA synthesis  
primers. Longer polynucleotides can be constructed from  
shorter segments having overlapping cohesive ends and  
used as structural genes, regulatory regions such as  
20 promoters, terminators, operators, and the like. It is  
thus of great interest to provide convenient automatic  
techniques for producing synthetic DNA fragments with  
high yields in a relatively short time.

As the understanding of the function, struc-  
25 ture, and chemical makeup of nucleotide sequences, such  
as DNA, has evolved, so too has the awareness of the  
practicalities and feasibilities of genetic engineer-  
ing. These engineering efforts, however, require a com-  
plete understanding of the chemical and biological reac-  
30 tions in cells. One of these reactions is the  
postreplicative modification of newly synthesized DNA by  
the selective methylation of certain cytosine residues  
which is performed enzymatically by a specific DNA  
methylase. An understanding of the factors governing the  
35 formation of specific methylation patterns in eucaryotic  
DNA is very important if we are to understand the  
mechanisms of gene expression.

Basic to such genetic engineering efforts is

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the synthesis of desired nucleotide chains from single mononucleotides. In this regard, electromechanical apparatus has been developed for synthesizing desired oligonucleotide sequences via the sequential linking of  
5 desired bases to a starting nucleotide.

At present, a variety of approaches for polynucleotide synthesis are available. These approaches can be characterized based on several criteria. First, the synthesis is usually carried out either on a solid-phase  
10 substrate or in a solution. Solid-phase synthesis relies on sequential addition of mononucleotides to a growing chain attached at one end to the substrate. The solid phase permits easy separation of the reactants, but the method requires excess quantities of reactants and  
15 usually provides only small quantities (less than 1 mg) of the desired sequence. Solution phase synthesis, while it requires lesser amounts of the expensive reagents and can provide larger quantities of the product sequence, requires isolation and purification of the intermediate  
20 product after every addition. Virtually all automated polynucleotide systems rely on solid phase synthesis.

There are presently two synthesis chemistries in widespread used for automated polynucleotide synthesis. The triester method, as described by Catlin and  
25 Cramer J. Org. Chem. 38: 245-250 (1973) and Itakura et al., Can. J. Chem. 51: 3649-3651 (1973) which relies on the addition of suitable blocked phosphate-triester intermediates which are generally inexpensive and stable. The phosphite-triester method, as described by  
30 Letsinger and Lunsford in J. Am. Chem. Soc. 98:3655 (1975) is somewhat more complex, but generally provides higher yields than the phosphate triester method. The utility of the phosphite-triester method was greatly improved by the use of N,N-dialkylamino phosphites  
35 (amidites) which are more stable than the phosphor-chlorodite intermediates initially employed. While the phosphite-triester method is often favored because of the

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greater yield at each nucleotide addition, the phosphate-triester method is also suitable for automated polynucleotide synthesis.

Among the reactor systems that can be used in synthesizing polynucleotides are solid-phase reactor systems which use either a tight bed column, a loose bed column, or a batch reactor. The tight bed column is tightly packed with the solid-phase support and the reactants are introduced either in a single pass or by a recirculating stream.

Loose bed columns have been introduced to alleviate these problems partially. By slowly passing the reactant through the column, higher mass transfer rates are achieved and utilization of the expensive reactants is improved. Also, channelling is reduced, since the solid phase packing will shift to equalize the flow profile therethrough.

In a batch reactor, the support matrix is held in an enclosed vessel. Reactants are introduced and the vessel contents agitated, typically by bubbling an inert gas through the liquid in the reactor. While such a system can provide very efficient utilization of the reactants by increasing the retention time in the reactor, relatively large volumes of the reactant and solvent are necessary to fill the reactor.

Urdea et al., in U.S. Patent No. 4,517,338, disclose a method and system for sequential modification of a linear polymeric molecule attached to a dispersed solid phase support by adding individual nucleotides in a predetermined order to a nucleotide chain. The dispersed solid phase is retained within a reactor zone which is provided with access ports for the introduction and removal of reagents. Reagents are selectively delivered to the reactor zone through at least one of the access ports by a reagent manifold.

Another apparatus for programmably synthesizing selected nucleotide sequences is described in Zelinka et

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al., U.S. Patent 4, 598,049.

The well known instability of the triazine ring of 5-azacytosine deoxyribonucleoside makes it unsuitable for use as a building block in the aforementioned automated DNA syntheses. Despite this drawback, interest in the synthesis of single and double stranded DNA fragments containing 5-azacytosine residues constitutes an important goal in the understanding of the mechanism of action of this drug. DNA incorporation of 5-azacytosine in living cells has been associated with inhibition of DNA methylase activity and consequent gene activation.

Because of the well established relationship that exists between the DNA incorporation of 5-azacytosine residues and gene activation, it would be useful to develop a methodology for the synthesis of oligonucleotide fragments which contain this unnatural base. These modified oligonucleotides, which would contain 5,6-dihydro-5-azacytosine and 5-azacytosine residues at specific sites, could serve as tools for elucidating the mechanism of selective gene activation and the relationship that exists between the presence of these triazine bases and inhibition of DNA methylation. A direct incorporation of the phosphoramidite of 2'-deoxy-5-azacytidine in DNA synthesis would result in failure, since the 2'-deoxy-5-aza-cytidine is extremely sensitive to acid or alkaline conditions.

#### SUMMARY OF THE INVENTION

It is an object of the present invention to overcome deficiencies in the prior art, such as noted above.

It is a further object of the invention to provide a method for the synthesis of modified oligonucleotides containing either 5,6-dihydro-5-azacytosine or 5-azacytosine bases at specific sites of the sequence.

It is another object of the present invention to provide a reagent for the automated synthesis of DNA to accomplish the incorporation of 5,6-dihydro-5-azacyto-

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sine and 5-azacytosine bases.

It is also an object of the present invention to provide for a method of converting 5,6-dihydro-5-azacytosine to 5-azacytosine in an oligonucleotide structure.

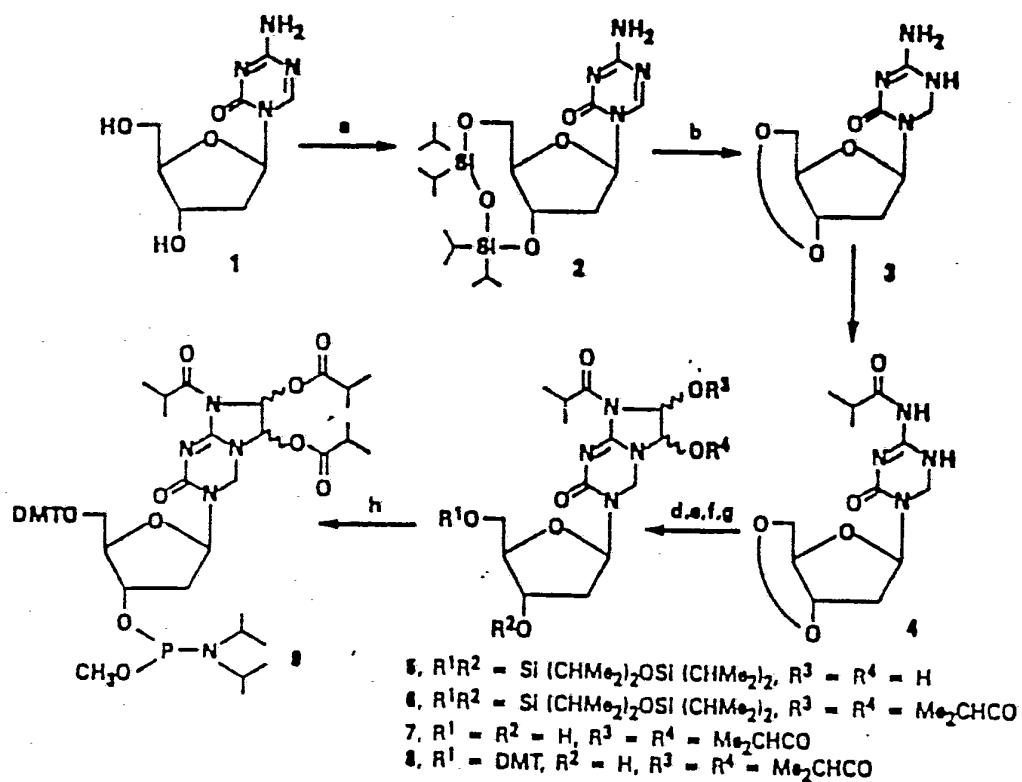
It is yet a further object of the present invention to provide compounds for use in studying the mechanism of selective gene activation.

The use of a phosphoramidite of 2'-deoxy-10 dihydro-5-aza-cytidine in DNA synthesis results in the successful formation of the desired internucleotide linkage and permits the synthesis of modified DNA fragments, since it is totally compatible with all of the chemical steps used in DNA synthesis. At the conclusion 15 of the synthesis, a very specific and easily performed oxidation generates the desired 5-aza-cytosine moiety. Since the hydrolytic instability of the triazine ring in 5-azacytosine nucleosides is very well documented, the use of a conventional phosphoramidite derivative of 5-20 azacytosinedeoxyribose, compound 1, is impractical, as this would have resulted in the base-catalyzed cleavage of the triazine ring during the last deprotection step of the synthesis. The process of the present invention overcomes this problem by using a stable phosphoramidite precursor of 5-azacytosinedeoxyribose that permits regen-25 eration of the desired 5-azacytosine base after the conclusion of the synthesis of the oligonucleotide.

The protected 5,6-dihydro-5-azacytidine phosphoramidite, compound 9, has a very stable triazine ring, 30 analogous to its parent nucleoside.

The reactions according to the present invention are shown in the following reaction schemes:

## Scheme 1



a) 2.2 eq. 1,3-dichloro-1,1,3,3-tetraisopropylidisiloxane, pyridine, rt, 1 h, 97%. b) 8 eq.  $\text{NaBH}_4$ , THF, rt, 1 h, 78%. c) i. 8 eq. isobutyryl chloride, pyridine/chloroform,  $0^\circ$ , ii.  $\text{MeOH}$ , rt, 16 h, 84%. d) i. 40 eq. glyoxal, pyridine, rt, thrice reduced to dryness, ii. chloroform/water extraction, e) 2 eq. isobutyryl chloride, pyridine, rt, 2 h, 81%. f) 1.2 eq. tetrabutylammonium fluoride, THF, rt, 1/2 h, 60%. g) 1.2 eq. 4,4'-dimethoxytrityl chloride, pyridine, rt, 2 h, 50%. h) 2.2 eq. chloro-N,N-diisopropylamino methoxyphosphine, 4.2 eq. tetrazole, chloroform, rt, 15 min, 71%.



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Scheme 1 outlines the synthesis of 5-azacytidine phosphoramidite starting with 5-aza-cytosinedeoxyribose. Protection with 1,3-dichloro-1,1,3,3-tetraisopropyldisiloxane, followed by borohydride reduction of compound 2, gave the desired dihydro analog, compound 3, after purification by silica gel flash chromatography with 5% methanol in ethyl acetate. The <sup>1</sup>H-NMR spectrum of compound 3 shows the newly generated C-6 methylene protons as an AB quartet centered at 4.40 ppm. The exocyclic amino group of compound 3 was then protected as the isobutyrylamide, compound 4, and purified by silica gel flash chromatography with 50% ethyl acetate in hexane. Complete protection of the triazine ring was accomplished with the introduction of the bis(isobutyryloxy)ethylene group, performed in the same manner as for 2'-deoxyguanosine. Thus, the intermediate diol, compound 5, isolated from the reaction of compound 4 with glyoxal, was reacted with isobutyryl chloride to give compound 6, which was purified by silica gel flash chromatography with 15% ethyl acetate in hexane. Removal of the tetraisopropyldisiloxane group in compound 6 with tetrabutylammonium fluoride gave compound 7, following a simple extraction in methylene chloride/water. Protection of the 5'-hydroxyl group was accomplished by the standard procedure using 4,4'-dimethoxytrityl chloride to yield compound 8 as a crystalline solid, mp 89-91°C (hexane). Finally, phosphitylation of compound 8 with chloro-N,N'-diisopropylaminomethoxyphosphite gave the desired phosphoramidite, compound 9, as a white solid, mp 67-69°C after purification by silica gel flash chromatography with 25% ethyl acetate in hexane.

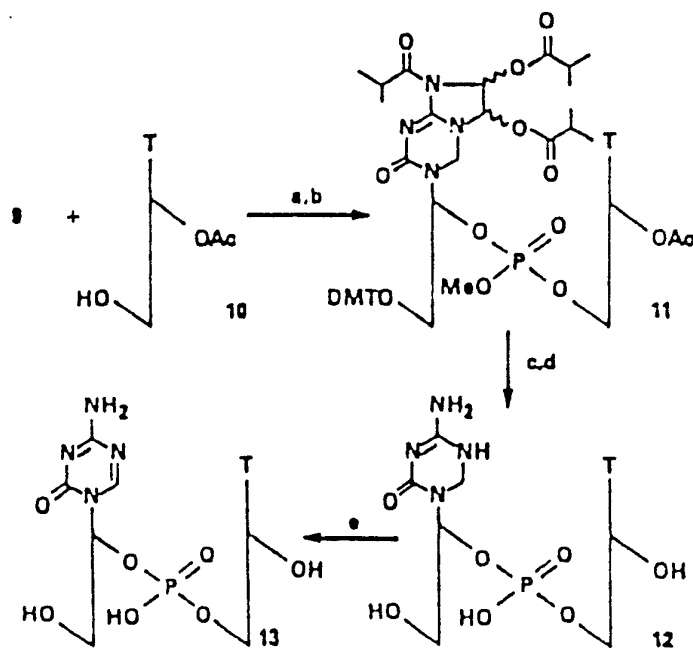
The reactivity of the new phosphoramidite, compound 9, was initially tested under the standard conditions used for DNA synthesis in a typical tetrazole-catalyzed condensation reaction with 3'-O-acetylthymidine, as shown in Scheme 2. After 15 minutes, the reaction was complete, and was immediately oxidized in

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situ to give a quantitative yield of the fully protected dimer, compound 11. Removal of the dimethoxytrityl group with trichloroacetic acid and further treatment of the residue with concentrated ammonium hydroxide yielded the

5 fully blocked dimer, compound 12.

## Scheme 2



a) 1.5 eq phosphoramidite 9, 5 eq tetrazole, MeCN, rt, 15 min. b) Iodine, 3%; water, 2%; 2,6-lutidine, 2% THF, 93%; rt, 10 min. c) trichloroacetic acid, 2% in dichloromethane, rt, 10 min. d) conc.  $\text{NH}_4\text{OH}$ ,  $50^\circ$ , 15 h. e) i) 250 eq bis(trimethylsilyl)-trifluoroacetamide, MeCN, reflux, 1 h, ii. 50% methanol in water, rt, 1 h.

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In order to test the new reagent, two decamers in which the cytosine base at positions 3 and 6 was replaced by the 5,6-dihydro-5-azacytosine moiety, were synthesized in an Applied Biosystems model 380A automated DNA synthesizer. Based on the trityl assay data, the stepwise yield was 98.5% and 98.4%, respectively, compared to 99.09% for the unmodified decamer.

The final conversion of the dihydrotriazine base to the aromatic triazine was successfully accomplished by the use of bis(trimethylsilyl)trifluoro-acetamide (BSTFA) and trimethylsilyl chloride as silylating reagents, and trimethylsilyl peroxide as an oxidizing reagent. For the transformation the dimeric compound 12 was used as a model and the resulting dimer 13 was prepared in quantitative yield.

#### BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 shows the fragmentation pattern for dimer 12 obtained by negative ion FAB mass spectroscopy, plotted as relative intensity vs. m/z.

Figure 2 shows autoradiography of synthetic oligonucleotides obtained after 5'-end labelling and polyacrylamide gel electrophoresis.

Lane 1: (CA)<sub>3</sub>, hexamer marker

Lane 2, (AT)<sub>4</sub>, octamer marker

Lane 3, TACGTCGCAG, parent decamer

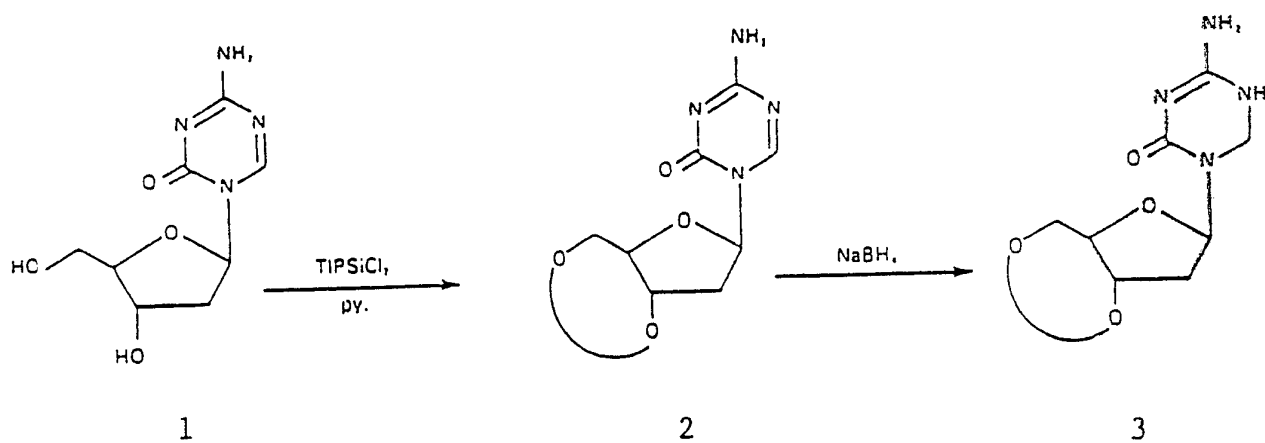
Lane 4, TAXGTCGCAG, 3-modified decamer

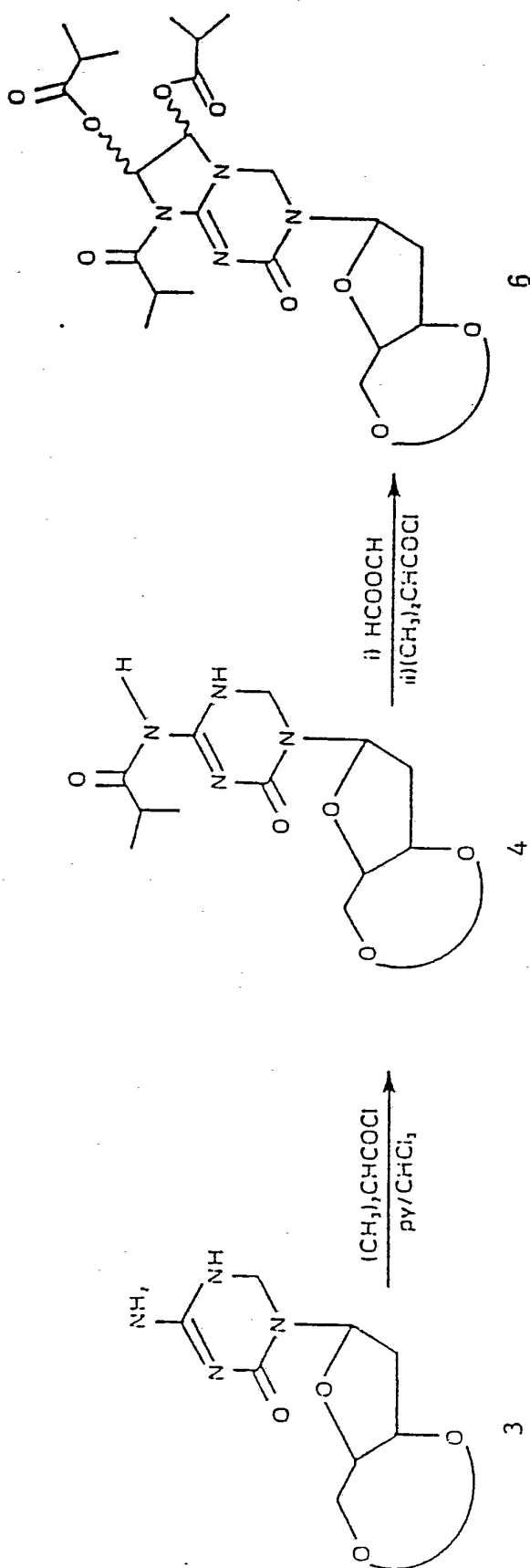
Lane 5, TACGTXGCAG, 6-modified decamer

X = 5,6-dihydro-5-azacytidine.

#### DETAILED DESCRIPTION OF THE INVENTION

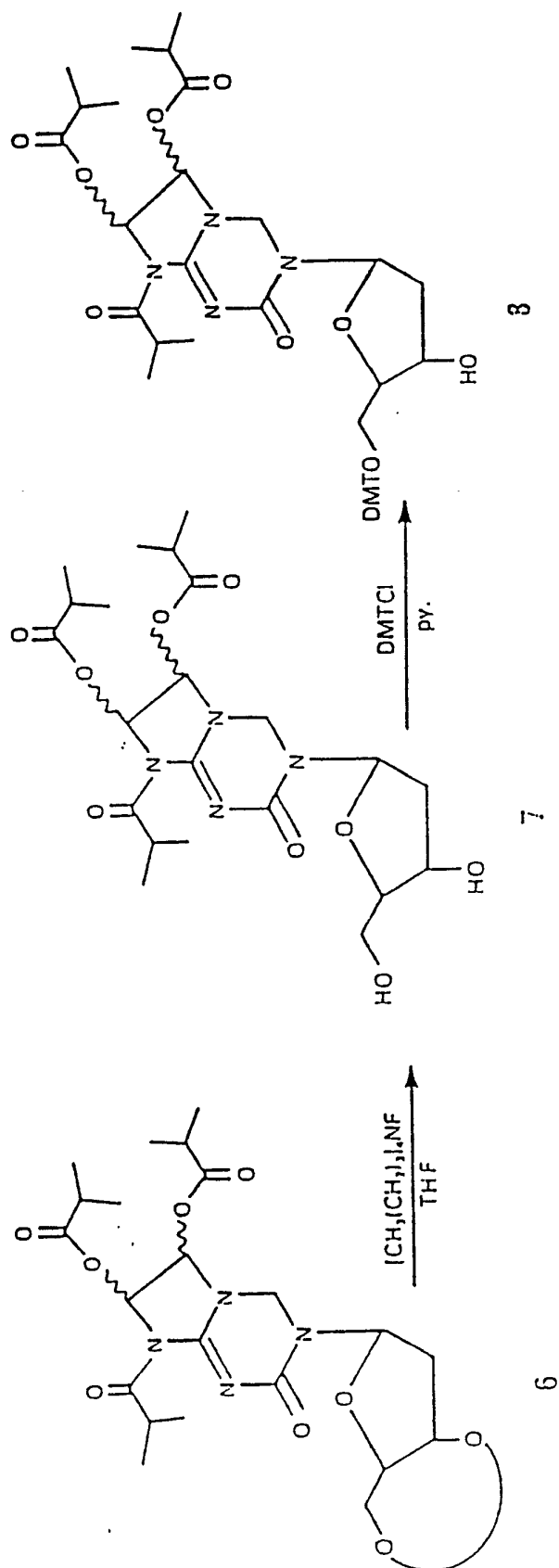
The detailed reaction schemes are shown as follows:

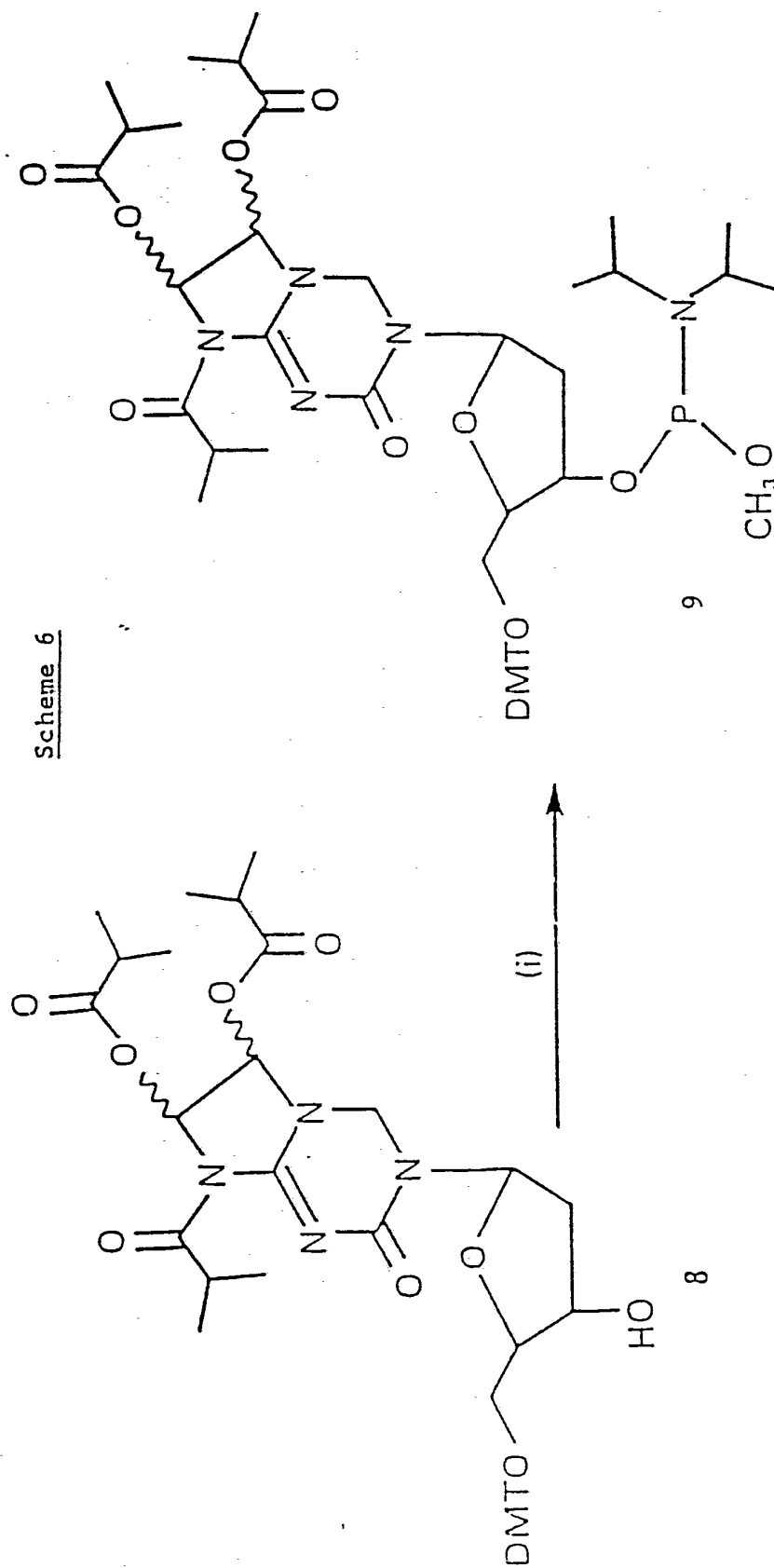
Scheme 3

Scheme 4

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Scheme 5



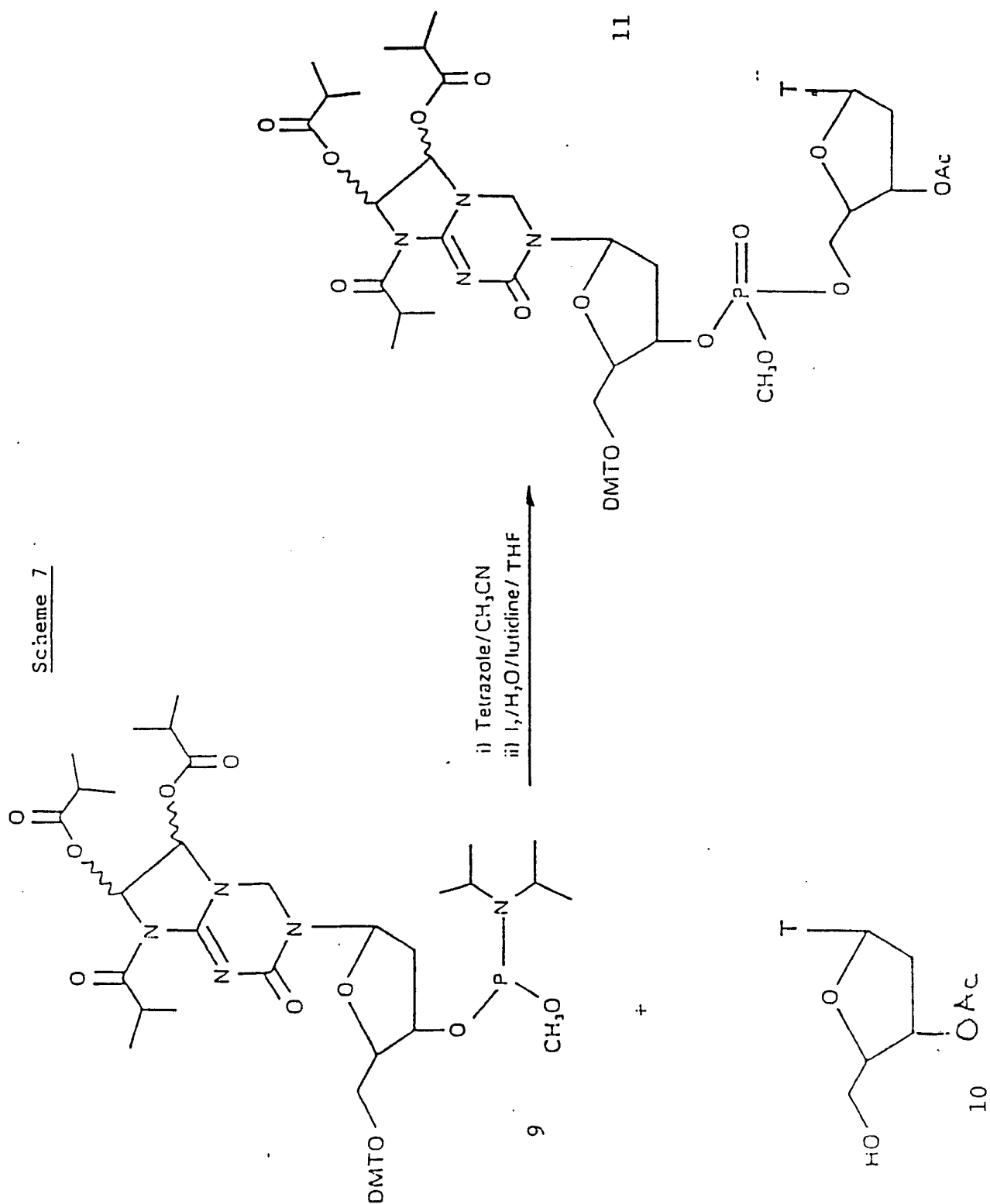


i) CH<sub>3</sub>OP(Cl)N[CH(CH<sub>3</sub>)<sub>2</sub>]<sub>2</sub>; [(CH<sub>3</sub>)<sub>2</sub>CH]<sub>2</sub> NCH<sub>2</sub>CH<sub>3</sub>; CH<sub>2</sub>Cl<sub>2</sub>

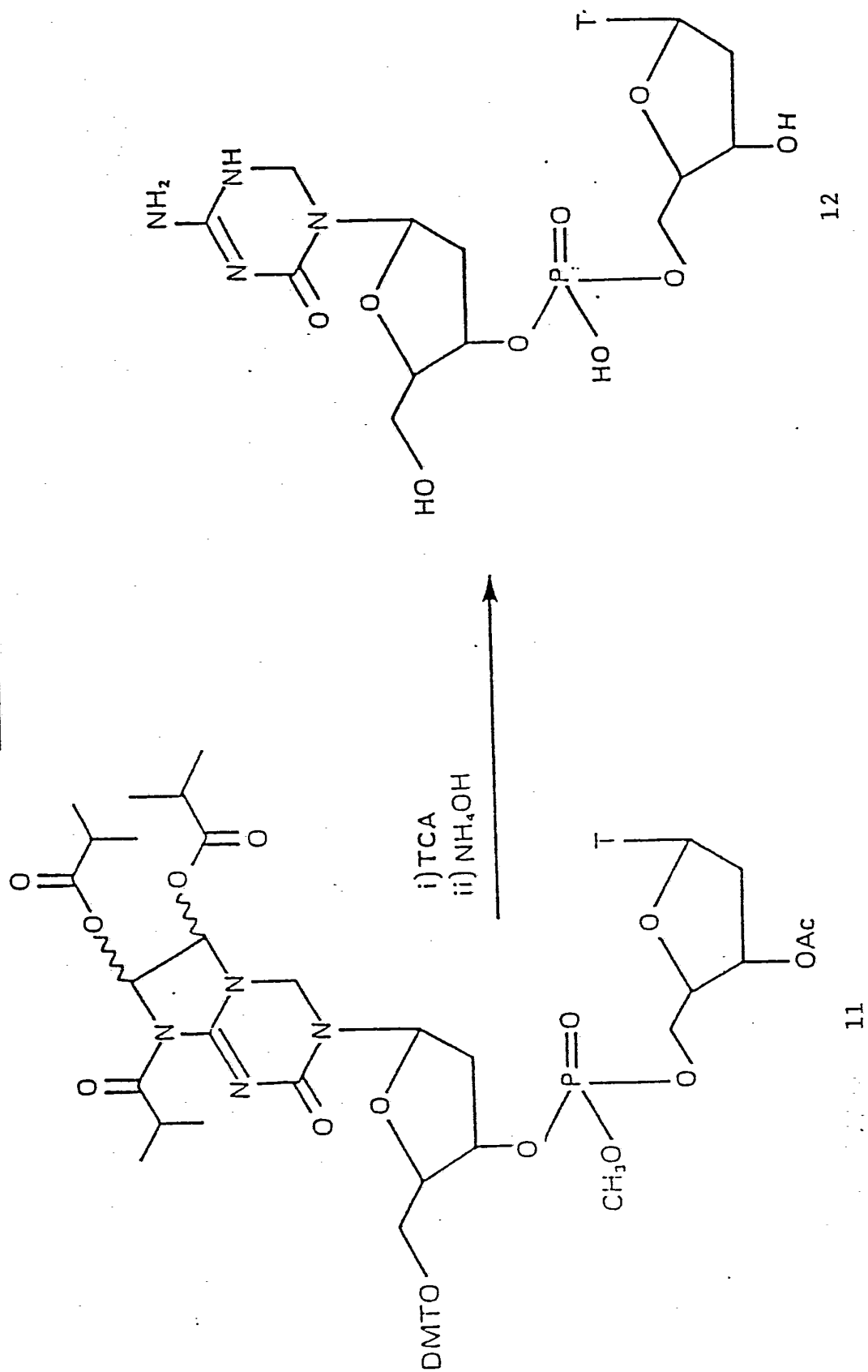
<sup>31</sup> P NMR	149.07
	148.97
	148.56
	148.48



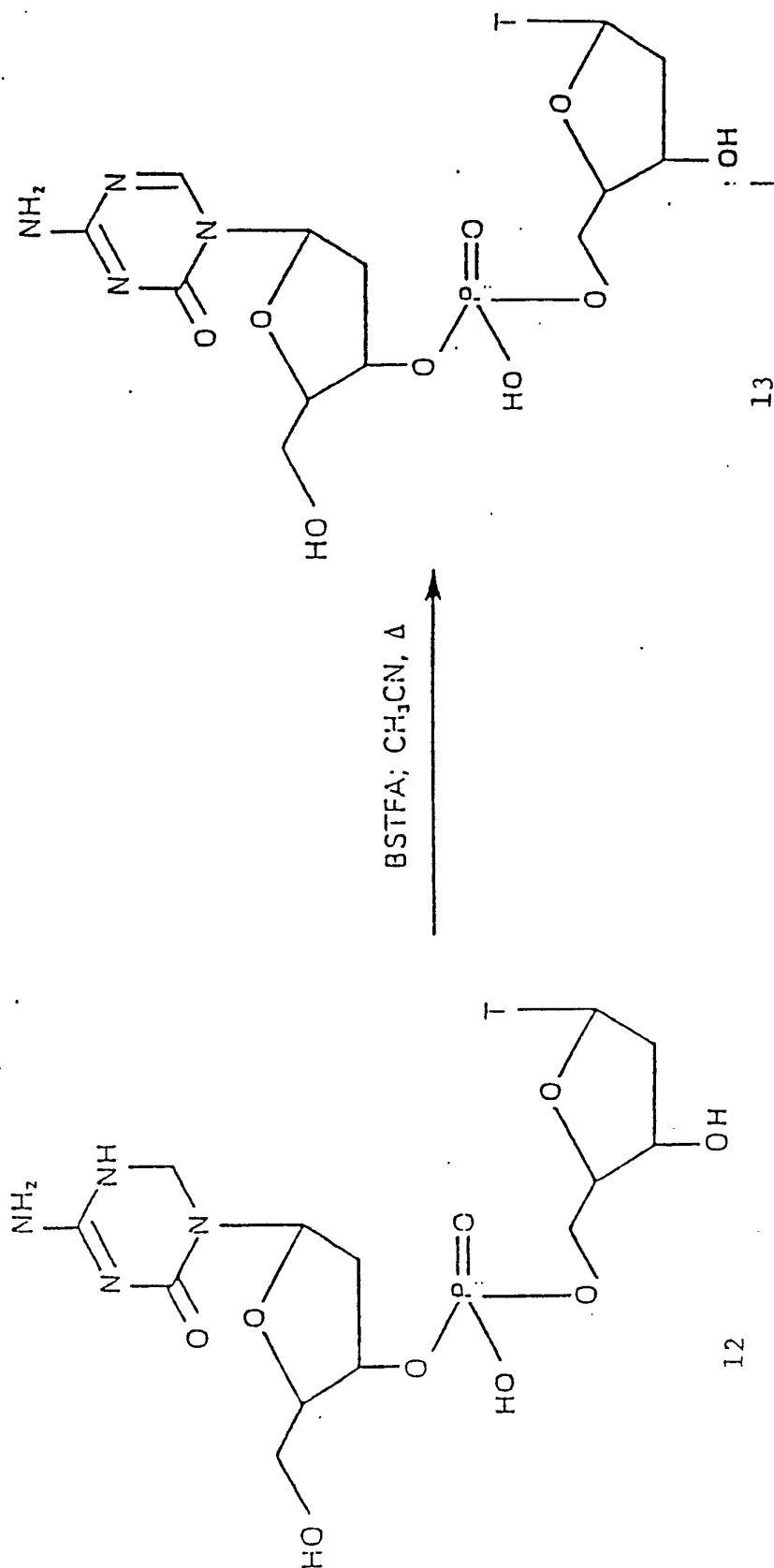
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Scheme 8

- 17 -

Scheme 9

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The synthesis of phosphoramidite according to the present invention started with 2'-deoxy-5-azacytidine, compound 1. As shown in Scheme 3, the 3' and 5' hydroxy groups were simultaneously protected with 1,3-dichloro-1,1,3,3-tetraisopropyldisiloxane using pyridine as solvent and base, according to the procedure of Markiewick et al. in Bull. Pol. Acad. Sci., 32: 433, 1984. This reaction proceeded in 97% yield, giving the desired compound 3 as a foam. In this and the following schemes, the tetraisopropyldisiloxane group is depicted as semicircle joining the 3' and 5' oxygen atoms of the nucleoside.

In the subsequent step, shown in Scheme 3, the double bond was reduced either catalytically with hydrogen over palladium on carbon, or more efficiently with sodium borohydride in tetrahydrofuran. After one hour of reaction, followed by treatment with methanol and water, workup and chromatography over silica gel with 5% methanol in ethyl acetate, the desired product, compound 4, was obtained in 78% yield as a foam.

Referring to Scheme 4, the exocyclic amino function was protected at this point in 84% yield by treatment of compound 3 with isobutyryl chloride in pyridine. After a conventional workup and silica gel chromatography with 50% hexane in ethyl acetate, compound 4 was obtained as a foam.

As shown in Scheme 4, complete protection of the aglycon moiety was achieved by introducing the isobutyryloxyethylene group. Reaction of compound 4 with glyoxal, followed by treatment of the cyclized intermediate with isobutyryl chloride in anhydrous pyridine, gave compound 6 after purification by silica gel column chromatography with 15% ethyl acetate in hexane. Compound 6 was isolated as a foam in 61% yield.

Referring to Scheme 5, compound 7 was prepared by removing the sugar tetraisopropyldisiloxane protective group with tetrabutylammonium fluoride at room tempera-

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ture in THF. This compound was purified by simple extraction in methylene chloride after the reaction mixture was reduced to dryness and partitioned between water and methylene chloride to give compound 7 as a foam in  
5 59% yield.

Selective protection of the 5'-hydroxy group, as required for DNA synthesis, was accomplished by the standard procedure using 4,4'-dimethoxytrityl chloride in dry pyridine to yield compound 8 in 50% yield as crystal-  
10 line solid, mp 89-91°C.

Scheme 6 shows the phosphitylation of compound 8 in the presence of diisopropylamine in methylene chloride with chloro(diisopropylamino)methoxy phosphine to give 71% yield of compound 9 as a glassy substance after  
15 purification by silica gel column chromatography with 25% ethyl acetate in hexane.

The phosphoramidite of the present invention, compound 9, is used in a typical condensation reaction to synthesize DNA. The phosphoramidite was mixed with 3'-O-acetyl thymidine, compound 10, as shown in Scheme 7, in  
20 acetonitrile in the presence of tetrazole as the condensing catalyst, according to the procedure of Pfleiderer and Schwarz (Tetrahedron Letters, 25: 5513, 1984). Thin layer chromatography showed complete reaction after fifteen minutes, and the dimeric product was immediately  
25 oxidized in situ with a mixture of iodine, lutidine, THF, and water to give a quantitative yield of the fully protected dimer phosphate, compound 11. Treatment of a solution of this dimer in dichloromethane with trichloroacetic acid removed the dimethoxytrityl group, and treatment of the residue with concentrated ammonium hydroxide at 55°C for fifteen hours yielded the fully deblocked target dimer, as shown in Scheme 8. An analytical sample of the deblocked dimer was obtained after reversed phase  
30 chromatography on J.T. Baker C-18 silical gel, 40 micrometers, 5% methanol in water, and as shown in Figure 1, the FAB/MS was consistent with the expected structure.

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Finally, the dihydro-5-azacytidine containing dimer, compound 12, was suspended in dry acetonitrile and treated with an excess of bis(trimethylsilyl)trifluoroacetamide, trimethylsilyl chloride, and trimethylsilylperoxide under reflux overnight, as shown in Scheme 9. Oxidation to the 5-azacytidine stage took place quantitatively as assessed by the dominance of the M-H peak in the mass spectrometer at m/z 531. The workup was very simple and involved evaporation of the volatile solvent and reagent and treatment of the residue with water to deblock the remaining oxygen and nitrogen to silicon linkages. Lyophilization of the aqueous solution yielded the desired dimer, compound 13.

In order to test the utility of the new reagent, two decamers, shown in Figure 2, lanes 4 and 5, in which the cytosine base at positions 3 and 6 was replaced by the 5,6-dihydro-5-azacytosine moiety, were synthesized in an Applied Biosystems model 380A automated DNA synthesizer. Based on the trityl assay data, the stepwise yield was 98.5% and 98.4%, respectively, compared to 99.09% for the unmodified decamer (Figure 2, lane 3).

While the invention is described above in relation to certain specific embodiments, it will be understood that many variations are possible, and that alternative materials and reagents can be used without departing from the invention. In some cases such variations and substitutions may require some experimentation, but such will only involve routine testing.

The foregoing description of the specific embodiments will so fully reveal the general nature of the invention that others can, by applying current knowledge, readily modify and/or adapt for various applications such specific embodiments without departing from the generic concept, and therefore such adaptations and modifications are intended to be comprehended within the meaning and range of equivalents of the disclosed

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embodiments. It is to be understood that the phraseology or terminology herein is for the purpose of description and not of limitation.

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## WHAT IS CLAIMED IS:

1. A 5,6-dihydro-5-azacytidine phosphoramidite suitable for use in the synthesis of modified DNA.
2. A method for the preparation of a 5,6-dihydro-5-azacytidine phosphoramidite comprising:
  - protecting a 5-aza-cytosinedeoxyribose with a siloxane;
  - reducing the protected 5-aza-cytosine-deoxyribose;
  - protecting the exocyclic amino group and the triazine ring of the reduced, protected 5-aza-cytosinedeoxyribose;
  - removing the siloxane group and protecting the hydroxyl group of the protected 5-aza-cytosine-deoxyribose;
  - phosphitylating the resultant compound to produce a 5-azacytosine phosphoramidite.
3. The method of claim 2 wherein the siloxane is 1,3-dichloro-1,1,3,3-tetraisopropyldisiloxane.
4. The method of claim 2 wherein the exocyclic amino group is protected as an isobutyrylamide.
5. The method of claim 2 wherein the triazine ring is protected by introducing a bis(isobutyryloxy)ethylene group.
6. A condensation reaction for preparation of a dimer oligonucleotide comprising reacting 3'-O-acetyl thymidine with 5,6-dihydro-5-azacytosine phosphoramidite.
7. A method for synthesizing DNA fragments comprising using a 5,6-dihydro-5-azacytidine phosphoramidite as a source of 5,6-dihydro-5 azacytosine and 5-azacytosine residues.
8. A method of oxidation for conversion of 5,6-dihydro-5-azacytosine residues to 5-azacytosine residues in an oligonucleotide.
9. The method of claim 8 wherein said 5,6-dihydro-5-azacytosine residues are produced by silylation using at least one silylating reagent selected from the



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group consisting of bis(trimethylsilyl) trifluoroacetamide and trimethylsilyl chloride, and which are oxidized by reaction with trimethylsilyl peroxide to produce said 5-azacytosine residues in an oligonucleotide.

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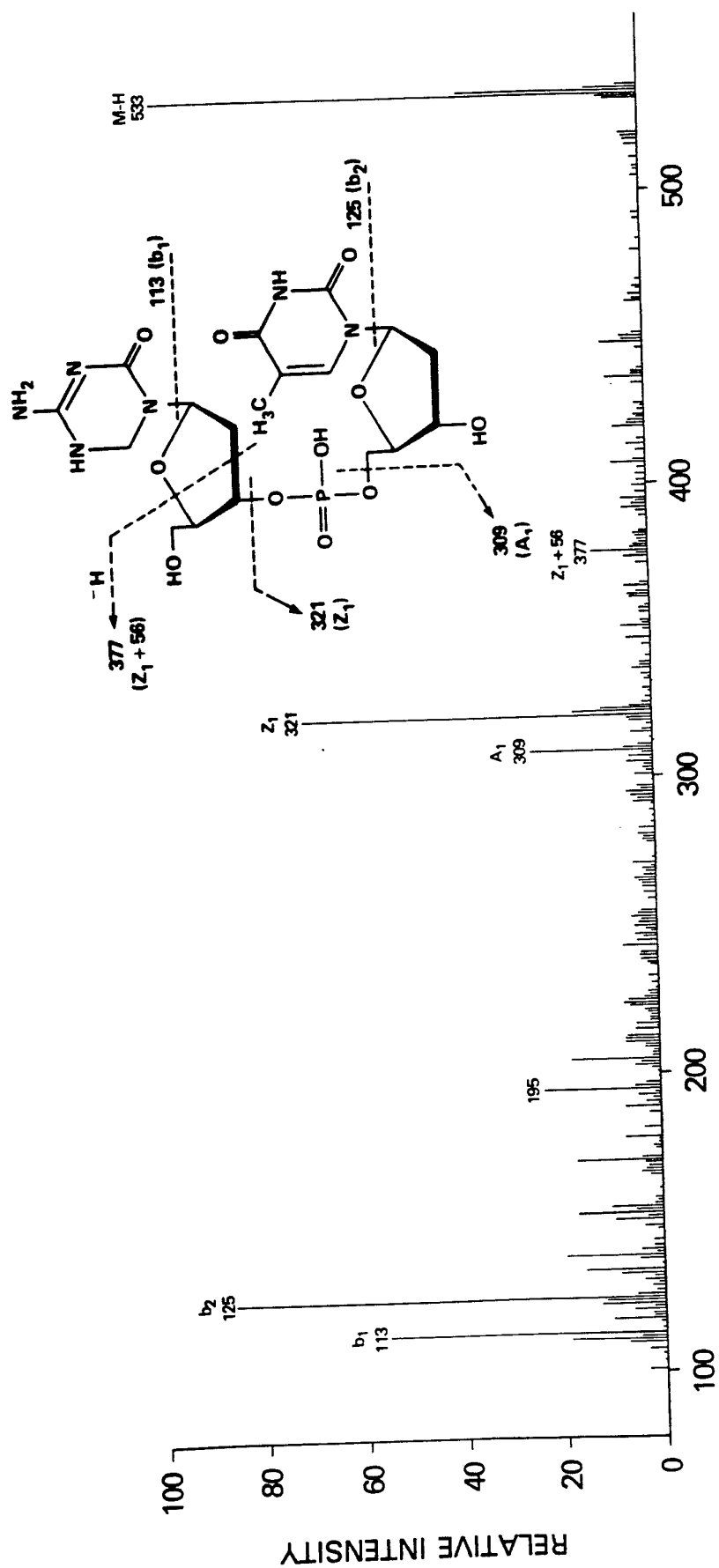


FIG. 1

1 2 3 4 5

FIG. 2

# INTERNATIONAL SEARCH REPORT

International Application No. **PCT/US89/01395**

<b>I. CLASSIFICATION OF SUBJECT MATTER</b> (if several classification symbols apply, indicate all) <sup>6</sup>		
According to International Patent Classification (IPC) or to both National Classification and IPC <b>IPC(4): C07H 19/20</b> <b>U.S. CL.: 536/23, 27. 29</b>		
<b>II. FIELDS SEARCHED</b>		
Minimum Documentation Searched <sup>7</sup>		
Classification System	Classification Symbols	
U.S.	536/23, 27, 29	
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched <sup>8</sup>		
<b>Data Bases Searched:</b> APS, CAS online		
<b>III. DOCUMENTS CONSIDERED TO BE RELEVANT <sup>9</sup></b>		
Category *	Citation of Document, <sup>11</sup> with indication, where appropriate, of the relevant passages <sup>12</sup>	Relevant to Claim No. <sup>13</sup>
Y	US, A, 4,058,602 (BEISLER) 15 November 1977, See entire document.	1-5
Y	US, A, 4,458,066 (CARUTHERS) 03 July 1984, See entire document.	1-7,9
X	A. Piskala, "Synthesis of 1-Glycosyl Derivatives of 5-Azauracil and 5-Azacytosine," published 1964, by Collection of Czechoslovak Chemical Communications. (Prague), see volume 29, pages 2060-2076.	8
<div style="display: flex; justify-content: space-between;"> <div style="width: 45%;"> <p>* Special categories of cited documents: <sup>10</sup></p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> </div> <div style="width: 45%;"> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&amp;" document member of the same patent family</p> </div> </div>		
<b>IV. CERTIFICATION</b>		
Date of the Actual Completion of the International Search  <div style="text-align: center; font-size: 1.2em;">06 June 1989</div>	Date of Mailing of this International Search Report  <div style="text-align: center; font-size: 1.5em;">10 JUL 1989</div>	
International Searching Authority  <div style="text-align: center; font-size: 1.2em;">ISA/US</div>	Signature of Authorized Officer <div style="text-align: center; font-weight: bold;">JAMES O. WILSON</div>	